

Supporting Information

Quantification of Residual Liquid on Repellent Cotton Fabrics after Liquid Roll Off

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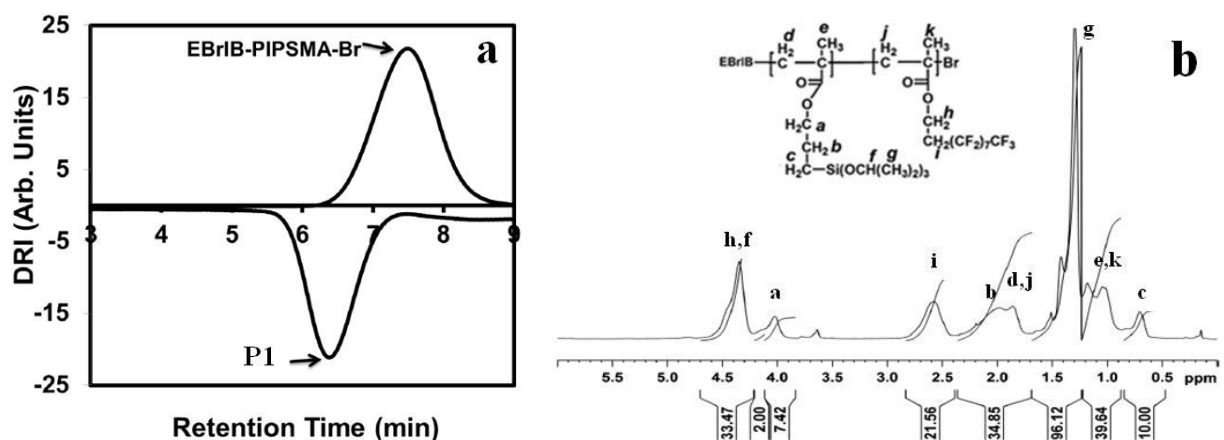


Figure S1. PIPSMA₁₅-*b*-PFOEMA₃₁ (P1) (a) SEC trace and (b) ¹H NMR.¹

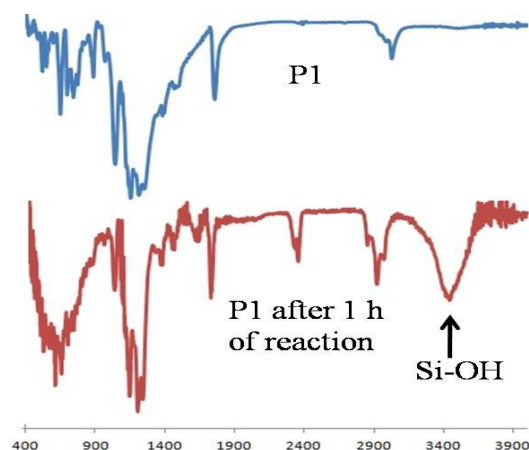


Figure S2. FTIR spectra of P1 before and after 1 h of sol-gel reaction. The peak at 3500 cm⁻¹ present in the spectra of the reacted polymer corresponds to the silanol groups indicative of P1 undergoing a hydrolysis reaction.²

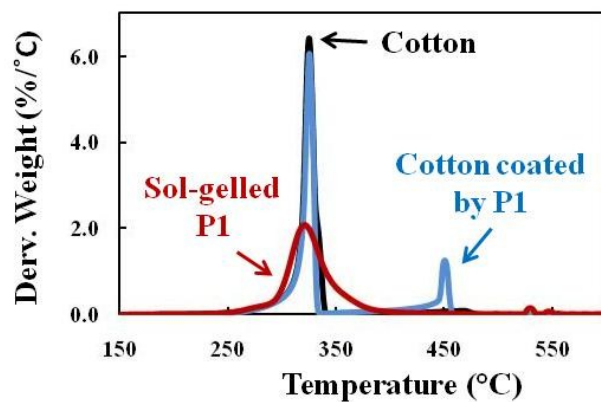


Figure S3. DTGA traces for cotton fabric, sol-gelled P1, and coated cotton at a P1 concentration of 5.00 mg/mL.

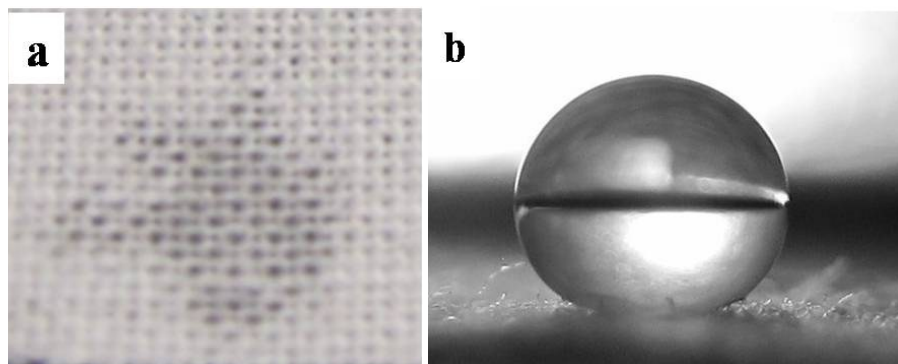


Figure S4. Photographs of 5.0 μL water droplets on (a) uncoated cotton (top view), and (b) coated cotton at a P1 concentration of 5.00 mg/mL (side view).

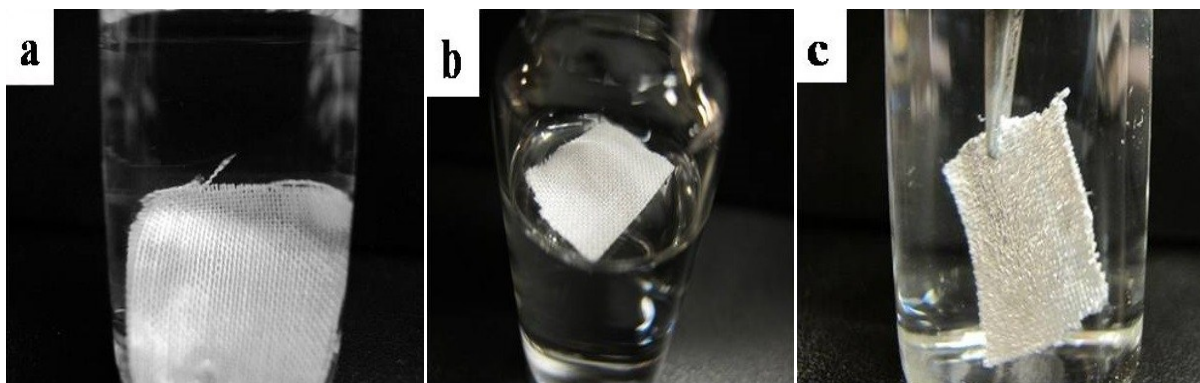


Figure S5. Photographs of (a) uncoated cotton, and (b) coated cotton at a P1 concentration of 5.00 mg/mL in a vial with water, and (c) same coated cotton piece forcibly submerged into the water. The coated cotton shows a plastron layer.

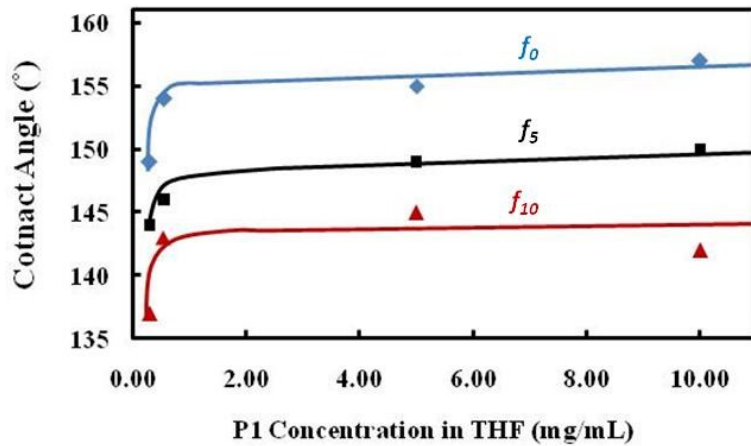


Figure S6. Changes in static contact angles for 5.0 μL droplets of f_0 , f_5 , and f_{10} loaded with QD at a 4.0 mg/mL concentration on coated cotton fabrics at different P1 concentrations.

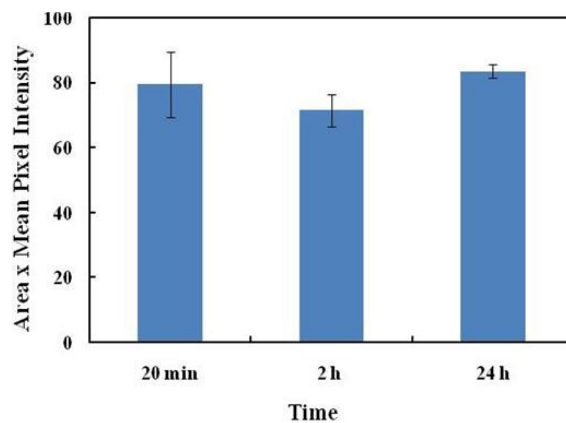


Figure S7. QD fluorescence intensity as a function of time for coated cotton fabrics at a P1 concentration of 5.00 mg/mL.

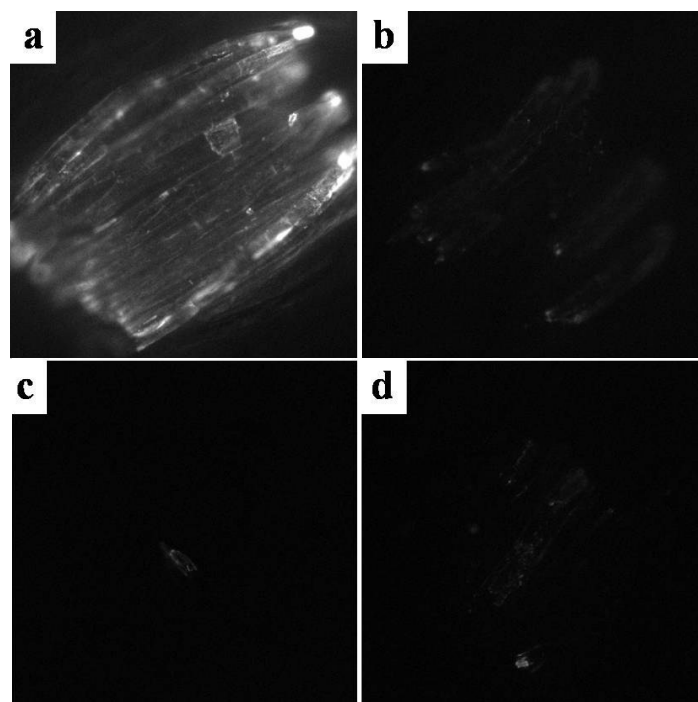


Figure S8. Fluorescence images of fabrics slanted at 60° after roll off of QD-containing f_5 . The fabrics were coated at a P1 concentration of (a) 0.30, (b) 0.55, (c) 5.00, and (d) 10.00 mg/mL. Images were cropped to $0.27 \times 0.27 \text{ mm}^2$.

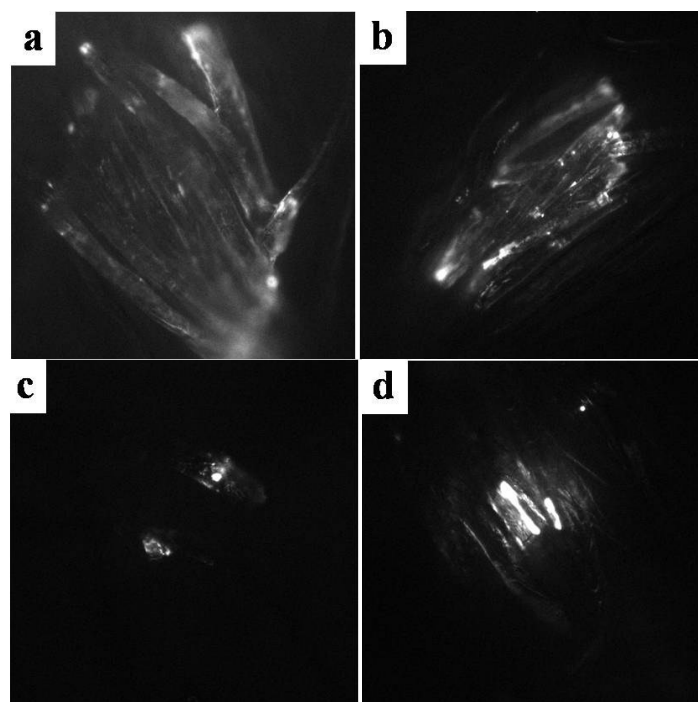


Figure S9. Fluorescence images of fabrics slanted at 60° after roll off of QD-containing f_{10} . The fabrics were coated at a P1 concentration of (a) 0.30, (b) 0.55, (c) 5.00, and (d) 10.00 mg/mL. Images were cropped to $0.27 \times 0.27 \text{ mm}^2$.

Table S1. Amount of residual liquid left behind on fabrics slanted at 60 ° after roll off using 10.0 μL droplets containing 4.0×10^{-2} mg QD.

P1 Concentration in THF (mg/mL)	f_0 (mg)	f_5 (mg)	f_{10} (mg)
0.30	$(8 \pm 5) \times 10^{-4}$	$(4 \pm 1) \times 10^{-3}$	$(2.4 \pm 0.8) \times 10^{-2}$
0.55	$(2 \pm 1) \times 10^{-5}$	$(1.2 \pm 0.2) \times 10^{-4}$	$(4 \pm 1) \times 10^{-3}$
5.0	$(2.0 \pm 0.4) \times 10^{-6}$	$(9 \pm 5) \times 10^{-6}$	$(3.1 \pm 8) \times 10^{-4}$
10.0	$(1.9 \pm 0.4) \times 10^{-6}$	$(2.1 \pm 0.3) \times 10^{-4}$	$(1.4 \pm 4) \times 10^{-3}$

Table S2. Percent of residual liquid left behind on fabrics slanted at 60 ° after roll off of using 10.0 μL droplets containing 4.0×10^{-2} mg QD.

P1 Concentration in THF (mg/mL)	f_0 (%)	f_5 (%)	f_{10} (%)
0.30	2 ± 1	10 ± 4	60 ± 19
0.55	0.06 ± 0.03	0.31 ± 0.06	10 ± 3
5.0	0.005 ± 0.001	0.02 ± 0.01	0.8 ± 0.2
10.0	0.005 ± 0.001	0.53 ± 0.08	3 ± 1

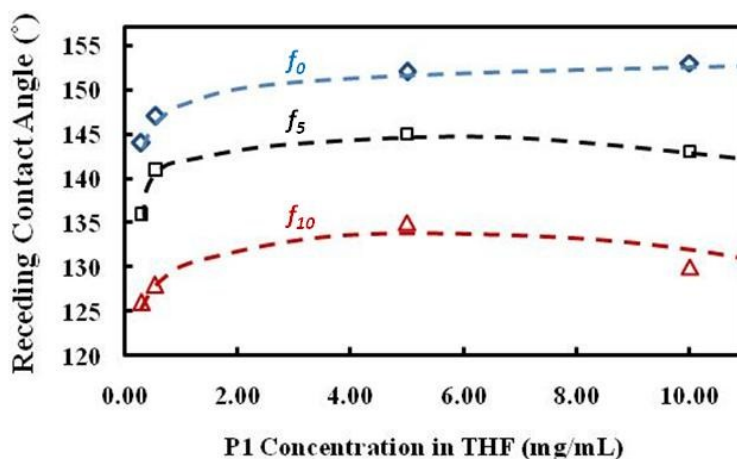


Figure S10. Changes in receding contact angles for 5.0 μL droplets of f_0 , f_5 , and f_{10} on coated cotton fabrics at different P1 concentrations.

References

1. C. M. Grozea, M. Rabnawaz, G. Liu and G. Zhang, *Polymer*, 2015, **64**, 153-162.
2. B. H. Stuart, *Infrared spectroscopy: fundamentals and applications*, John Wiley & Sons, Chichester, UK, 2004.