Electronic supplementary information of

Electrochemical amination of graphene using nanosized PAMAM dendrimers

for sensing applications

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Electrochemical surface area of GCE and graphene modified GCE

Randles-Sevcik equation was used to calculate electrochemical surface area of GCE and

graphene modified GCE

 $i_p = 2.69 \times 10^5 \ AC \ D^{1/2} \vartheta^{1/2} n^{3/2} \dots (S1)$

Where i_p =oxidation peak current of K₄[Fe(CN)₆] in ampere obtained from cyclic voltammogram of 5 mM K₄[Fe(CN)₆] in 0.5 M KCl at the scan rate of 50 mVs⁻¹, A=electrochemical surface area of electrode in cm², C= Concentration of analyte (5 mM K₄[Fe(CN)₆] in 0.5 M KCl) in mol cm⁻³, D- Diffusion coefficient of K₄[Fe(CN)₆] known as 6.5 χ 10⁻⁶ cm² s⁻¹, \Im scan rate in V s⁻¹ and n- number of electron transferred in the redox reaction (usually 1).



Figure S1. The wide scan of X-ray photoelectron spectra of graphene, aminated graphene and HRP immobilized aminated graphene surface.

Table s1.	The XPS	data analysis	of C1s,	Ols and Nls	of electrode surfaces
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Electrode Surface	C1s, BE (eV)	O1s, BE (eV)	N1s, BE (eV)
GCE-CtRGO	284.9, (C-C) 285.9, (C-O) 289.4, (C(O)OH)	531.3, (R-(C=O)-R) 532.7, (C-OH, C-O-C) 533.6, (O=C-OH, O=C-OR)	No peak
GCE-CtRGO-PAMAM	284.9, (C-C)	531.7, (R-(C=O)-R)	399.07, (N-C, sp ³)
	285.9, (C-O/C-N)	532.6, (C-OH, C-O-C)	400.3, (N-C, sp ²)
	289.4, (C(O)OH)	533.7, (O=C-OH, O=C-OR)	401.2, (protonated amine)
GCE-CtRGO-PAMAM-HRP	284.9, (C-C)	531.48, (R-(C=O)-R)	399.7, (N-C, sp ³)
	286.2, (C-O/C-N)	532.47, (C-OH, C-O-C)	400.4, (N-C, sp ²)
	287.9, (C(O)OH)	533.6, (O=C-OH, O=C-OR)	401.7, (protonated amine)



Figure S2. The current response of GCE-CtRGO-PAMAM-GA-HRP for 100 μ M H₂O₂ from chronoamperograms at different time intervals (A) and the corresponding chronoamperograms (B).

Sample	Added (µM)	Found (µM)	Recovery %
Serum	-	-	-
H ₂ O ₂ spiked Serum Sample	50	46	91
	50	49.5	99
	50	45.2	90.4
Standard H ₂ O ₂ sample	50	52.3	104.6
	50	50.2	100.4

Table S2. Determination of H₂O₂ in Human Serum

Mean recovery 96.28 %