Broad temperature range of cubic blue phase existed in

simple binary mixture systems containing rodlike Schiff

base mesogens with tolane moiety

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Electronic supplementary information (ESI)

Fig. S1 The DSC measurement of compounds **(S)-OH-TI** (blue line) and **OH-TI** (red line) on heating and cooling recycle.



Fig. S2 Microphotographs for compound OH-TI

(a) N texture at 149.3°C; (b) N - SmC texture at 110.4°C; (c) SmC texture at 100.0°C;

(d) CrG texture at 79.8°C; (e) CrH texture at 68.0°C.



Fig S3. Transition from mosaic texture of the soft crystal G to the soft crystal H of compound **OH-TI**₇ observed in the other area. (a) CrG texture at 78.6°C; (b) CrG-CrH texture at 75.8°C; (c) CrH texture at 70.0°C.



Fig. S4 The variable-temperature XRD measurements of compound OH-TI₇.



135.2°C; (e) SmC* texture in other area at 135.2°C; (f) CrG* texture at 126.6°C. This texture is of a mosaic kind, but retains schlieren characteristic from the smectic C phase ; (g) CrH* texture at 99.8°C.



Fig. S6 The DSC measurement of compounds **(S)-H-TI** (blue line) and **H-TI** (red line) on heating and cooling recycle.



Fig. S7 Microphotographs for compound H-TI

- (a) N texture at 149.8°C; (b) N-SmC texture at 139.8°C; (c) SmC texture at 136.1°C; (c) CrG texture at 121.9°C; (d) CrH texture at 101.8°C.



Fig. S8 The comparison of BP temperature range for the different blending mixture systems in heating (red line) and cooling processes (blue line).



Fig. S9 Typical reflectance profiles of the higher-temperature phase (gray line) and the lower-temperature phase (black line) for the blending system mixture (a) **OH-TI**₇ + 40% **S811** and (c) (*S*)-**OH-TI**₇ + 10% **IOS(6OBA)**₂. Temperature dependence of the Bragg reflection wavelength for the blending mixture consisting of **OH-TI**₇ + 40% **S811** (b) and (*S*)-**OH-TI**₇ + 10% **IOS(6OBA)**₂ (d) during cooling process with a rate of 0.5 °C min⁻¹.



Fig. S10 Microphotographs for compound (S)-OH-TI blended with 40.0 wt% R811.



Fig. S11 Contact test of Schiff base (*S*)-OH-TI₇ and the mixture of racemic Schiff base OH-TI₇ doped with 4 wt% **S811** (a) and **R811** (b).

Measurements of helical pitch and helical twisting power of two chiral Schiff base mesogens and binary mixture system composed of 60 wt% racemic Schiff base and 40 wt% S811by Cano's Wedge method.

The helical pitch (p) was evaluated by measuring the distance (a) between Cano lines as follows: $p = 2a \tan\theta$, where θ is the angle of the wedge of the cell.

In this experiment, the cell's tan θ is 0.0196, the concentration of the chiral dopant c is 4%.



Fig. S12 Cano's Wedge method to measure the helical pitch of two chiral Schiff base mesogens and binary mixture system composed of 60 wt% racemic Schiff base and 40 wt% **S811**.

The DSC diagram, POM texture and XRD variable-temperature XRD of Schiff base mesogen OH-TI₇₇.

The DSC (Fig. S13) and POM pictures (Fig. S14) have indicated that Schiff base compound **OH-TI**₇₇, on cooling from its isotropic phase, displayed nematic Schlieren texture (N phase, Fig. S14a), straited texture below N-SmC transition (Fig. S14b), Schlieren texture with narrow dark four brushes (SmC phase, Fig. S14c), Schlieren texture with broader dark four-brushes (SmF phase, Fig. S14d), mosaic terrace-like relief with larger domains (the soft crystal CrG phase, Fig. S14e), mosaic texture with small platelet area that are cross-hatched by grainings (the soft crystal CrH phase, Fig. S14f) and two crystal phase (Fig. S14g and Fig. S14h).



Fig. S13 The DSC measurement of compounds $OH-TI_{77}$ on heating (red line) and cooling recycle (blue line).





(g)

(h)



Fig. S14 Microphotographs of compound **OH-TI**₇₇ observed under the POM: (a) N texture at 210.0°C; (b) N - SmC transition texture at 181.6°C; (c) SmC texture at 130.0°C; (d) SmF texture at 120.0°C; (e) CrG texture at 112.0°C; (f) CrH texture at 100.0°C; (g) Cr₁ texture at 87.8°C; (h) Cr₂ texture at 74.2°C.



Fig. S15 The variable-temperature XRD measurements of compound OH-TI₇₇.





Fig. S16. HOMO and LUMO of Schiff base compound (a), Salicylaldimine compound (b) and compound OH-TI₇₇ (c). The simulation exchange functional and basis set are CAM-B3LYP and 6-311G(d, p), respectively. The isosurface is drawn at value of 0.02.

Table S1. A comparison table with different basis sets and exchange functionals on the calculated dipole moment in **Schiff base** compound, **Salicylaldimine** compound and compound **OH-TI**₇₇.

Exchange	Designet	Dipole Moment (Debye)				
functional	Dasis set	Schiff base	Salicylaldimine	OH-TI ₇₇		
CAM-B3LYP	6-311G(d,p)	1.8072 2.2053		1.9450		
	def2-tzvp	1.9057	2.1953	1.9520		
	6-31G(d)	1.8032	2.1847	1.9789		
	6-31+G(d)	1.9425	2.3394	2.1277		
ωB97X	6-311G(d,p)	1.6789	2.3066	1.9574		
	6-31G(d)	1.6739	2.0637	2.0152		
	6-31+G(d)	2.1211	2.1983	2.0575		

Table S2. DFT calculated HOMO, LUMO, energy gap, dipole moment components, μ_x , μ_y , μ_z and modulus (μ) for **Schiff base** compound, **Salicylaldimine** compound and compound **OH-TI**₇₇.

		Energy (eV)			Dipole moment (µ in Debye)			
Exchange functional and Basis set	Compound	НОМО	LUMO	ΔE (eV)	$\mu_{_{X}}$	μ_{y}	$\mu_{_z}$	$\mu_{_{ m total}}$
CAM-B3LYP 6-311G(d,p)	Schiff base	-6.697	-0.764	5.933	-1.058	-0.850	1.194	1.807
	Salicylaldimine	-6.736	-0.842	5.894	0.543	1.772	1.195	2.205
	OH-TI ₇₇	-6.766	-0.860	5.906	0.296	1.775	0.738	1.945
ωB97X 6-311G(d,p)	Schiff base	-7.641	0.116	7.757	-0.937	1.391	0.084	1.679
	Salicylaldimine	-7.689	0.034	7.723	0.478	1.671	1.518	2.307
	OH-TI ₇₇	-7.701	0.014	7.715	0.286	1.817	0.669	1.957

Table S3. DFT calculated principal polarizability components (α_{XX} , α_{YY} , α_{ZZ}), isotropic component $\alpha^{iso} = (\alpha_{XX} + \alpha_{YY} + \alpha_{ZZ})/3$, polarizability anisotropy $\Delta \alpha = [\alpha_{XX} - (\alpha_{YY} + \alpha_{ZZ})/2]$ and asymmetry parameter $\eta_{\alpha} = [(\alpha_{YY} - \alpha_{ZZ})/(\alpha_{XX} - \alpha^{iso})]$, relative to the molecular polarizability tensor α .

Exchange functional and Basis set	Compound	$\alpha_{_{\rm XX}}$	$lpha_{_{ m YY}}$	α ZZ	$lpha^{ m iso}$	$\Delta \alpha$	$\eta_{_{lpha}}$
CAM-B3LYP 6-311G(d,p)	Schiff base	874.25	374.57	256.32	501.71	558.81	0.31742
	Salicylaldimine	892.29	380.81	256.42	509.84	573.68	0.32525
	OH-TI ₇₇	897.57	353.14	246.43	499.05	597.79	0.26777
ωB97X 6-311G(d,p)	Schiff base	818.70	373.20	261.67	484.53	501.27	0.33375
	Salicylaldimine	837.45	376.35	264.11	492.64	517.22	0.32551
	OH-TI ₇₇	846.19	351.98	249.84	482.67	545.28	0.28098

References:

1. D. Demus, S. Diele, M. Klapperstück, V. Link a and H. Zaschke, *Mol. Cryst. Liq. Cryst.*, 1971, **15**, 161-174.

2. G. W. Gray and J. W. Goodby, *Smectic Liquid Crystals: Texture and structure*, Leonard Hill, Philadelphia, 1984.

3. I. Dierking, *Textures of Liquid Crystals*, Wiley-VCH, Weinheim, 2003.