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Thermostable and nonflammable silica-polyetherimidepolyurethane nanofibrous separators for high power lithium ion batteries

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Fig. S1 Stress-strain curves of the Celgard membrane.



Fig. S2 FT-IR spectra of SiO₂-PEI-PU nanofibrous membranes.

The FT-IR spectra of the SiO₂-PEI-PU nanofibrous membranes are shown in Fig. S2. The absorption spectra exhibit characteristic imide group absorptions at 1780 and 1720 cm⁻¹ (asymmetrical and symmetrical stretching of imide carbonyl bond), 1360 and 743 cm⁻¹ (C-N stretching and bending), and 1237 cm⁻¹ (aromatic ether C-O-C).¹ The typical absorption features for carbamate group were found at 3335 cm⁻¹ (hydrogen-bonded N-H stretching), 1720 cm⁻¹ (stretching vibrations of C=O), 1534 cm⁻¹ (CO-N-H bending), 1110 cm⁻¹ (C-O-C), 2945 and 2860 cm⁻¹ (asymmetrical and symmetric stretching of CH₂), respectively.² Meanwhile, the increased band intensity

at 1110 cm⁻¹ of SiO₂-PEI-PU membranes containing 8 wt% SiO₂ NPs may be ascribed to Si-O-Si stretching, which implies that the introduction of SiO₂ NPs.



Fig. S3 EDX spectra of SiO₂-PEI-PU nanofibrous membranes.

Further confirmation of the involvement of SiO₂ NPs has done by the energydispersive X-ray spectroscopy (EDX). Fig. S3 shows the typical EDX pattern for PEI-PU and SiO₂-PEI-PU nanofibrous membranes. Note that Pt signals come from the sputter-coating gold film. The EDX pattern for the SiO₂-PEI-PU membranes containing 0 wt% SiO₂ NPs do not show the characteristic signal of Si, whereas for the SiO₂-PEI-PU nanofibrous membranes containing 8 wt% SiO₂ NPs a clear signal of the presence of Si has been observed. All of these indicate that the successful introduction of SiO₂ NPs in the SiO₂-PEI-PU membranes.

References

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- 2. J. Wang, Y. Li, H. Tian, J. Sheng, J. Yu and B. Ding, RSC Adv., 2014, 4, 61068.