## **Electronic Supplementary Information**

## Synthesis of tapered tetragonal nanorods of anatase TiO<sub>2</sub> with enhanced photocatalytic activity via a sol-hydrothermal process mediated by H<sub>2</sub>O<sub>2</sub> and NH<sub>3</sub>

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Table S1 Summary of experiments carried out to investigate the effects of the volume of H<sub>2</sub>O<sub>2</sub> and NH<sub>3</sub>. All these experiments were performed with 20 mg of Ti powder under 200°C for 24 h.

$H_2O_2/mL$	NH <sub>3</sub> /mL	H <sub>2</sub> O/mL	Entry
2	1	12	a
	3	10	b
	5	8	c
	7	6	d
	8	5	e
3	1	11	f
	3	9	g
	5	7	h
	7	5	i
	9	3	j
5	1	9	k
	3	7	1
	5	5	m
	7	3	n
	9	1	0
7	1	7	р
	3	5	q
	5	3	r
	7	1	S
9	1	5	t
	3	3	u
	5	1	v



Fig. S1 SEM images of the synthesized nanocrystals showing shape evolution with contents of  $H_2O_2$  and  $NH_3$ . The detailed synthetic conditions corresponding to each image are displayed in Table S1. All the scale bars are 100 nm.



**Fig. S2** SEM images of nanocrystals synthesized with different volume ratios of  $H_2O_2$  to  $NH_3$  (x/y) for 24 h at varied reaction temperatures. For x/y = 2/8: a, 180 °C; d, 160 °C. For x/y = 5/5: b, 180 °C; e, 160 °C. For x/y = 9/1: c, 180 °C; f, 160 °C. All the scale bars are 60 nm.



**Fig. S3** SEM images of nanocrystals synthesized with different volume ratios of  $H_2O_2$  to  $NH_3$  (x/y) at 200°C for different reaction durations. For x/y = 2/8: a, 3 h; d, 12 h; g, 24 h; j, 48 h. For x/y = 5/5: b, 3 h; e, 12 h; h, 24 h; k, 48 h. For x/y = 9/1: c, 3 h; f, 12 h; i, 24 h; l, 48 h. All the scale bars are 60 nm.



**Fig. S4** XRD patterns of samples synthesized with 9 mL of  $H_2O_2$  and 1 mL of  $NH_3$  at varied reaction temperatures for 24 h (a) and at 200 °C for different reaction durations (b).



Fig. S5  $N_2$  adsorption-desorption isotherm curves of the samples of  $T_{2/8}$ ,  $T_{5/5}$  and  $T_{9/1}$ .



**Fig. S6** SEM (a) and TEM (b) images of anatase  $TiO_2$  nanorods  $(T_{9/1})$  loaded with Pt nanoparticles under UV light irradiation. The red circles indicate the locations of Pt nanoparticles.



**Fig. S7** Ti 2*p*, O1*s* and N 1*s* X-ray photoelectron spectra (XPS) corresponding to samples of  $T_{2/8}$  (a, b and c),  $T_{5/5}$  (d, e and f) and  $T_{9/1}$  (g, h and i).



**Fig. S8** Action spectrum of  $H_2$  evolution for sample of  $T_{5/5}$ , showing quantum efficiency variation with wavelength of incident light and the corresponding UV-visible absorption spectrum.