

**Pd nanoparticles decorating flower-like Co₃O₄ nanowire
clusters to form an efficient, carbon/binder-free cathode for the
Li–O₂ battery**

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Supporting information:

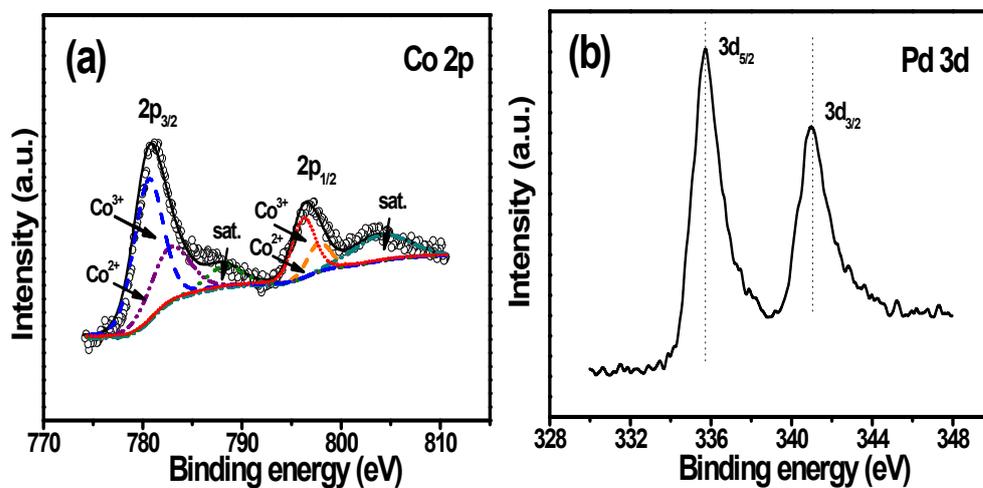


Fig. S1 (a) Co 2p XPS and (b) Pd 3d XPS of Pd/Co₃O₄.

In Figure S1a, the peaks at 780.4 eV and 795.8 eV are the Co 2p spectra of Co³⁺, and the peaks at 781.8 eV and 797.8 eV are the Co 2p spectra of Co²⁺. In Figure S1b, the peaks at 335.6 eV and 341.1 eV are the Pd 3d spectra of metallic Pd⁰.

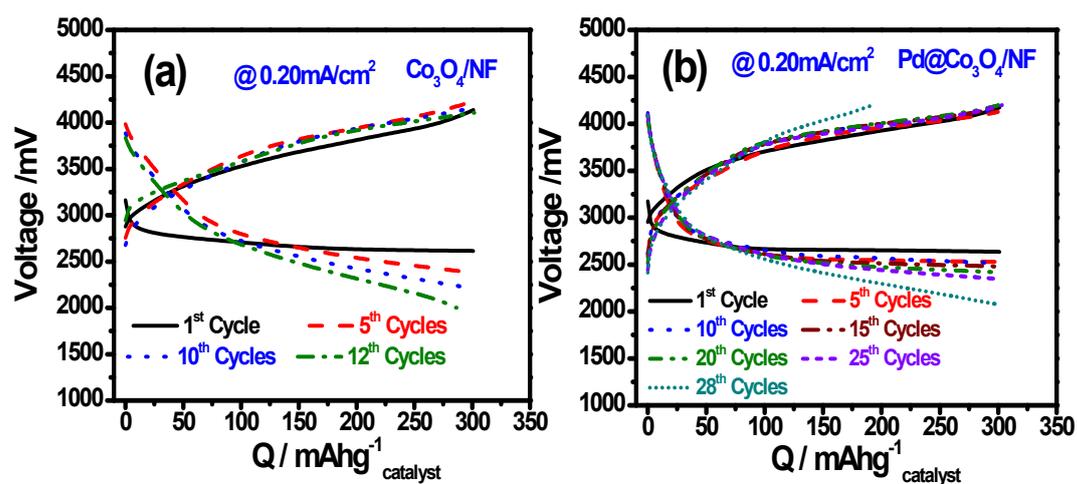


Fig. S2 Discharge-charge profiles at various cycles of Li-O₂ batteries with (a) Co₃O₄/NF and (b) Pd/Co₃O₄/NF cathodes at a limited capacity of 300 mAhg⁻¹ under current density of 0.20 mAcm⁻².

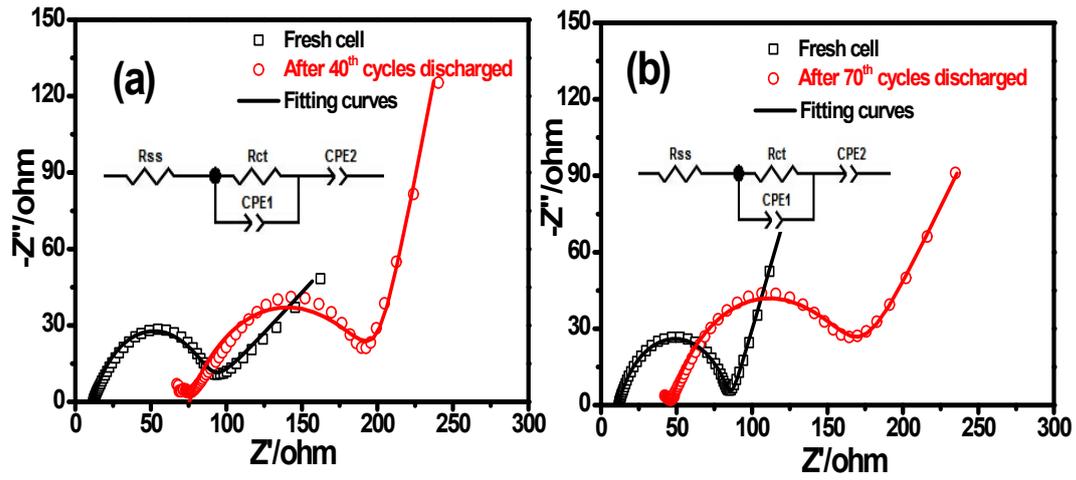


Fig. S3 Electrochemical impedance spectra (EIS) of Li-O₂ batteries based Co₃O₄ and Pd/Co₃O₄ catalysts after discharged with different cycles (The inset shows the equivalent circuit^{3, 4} that the data (circles) were fitted with (solid lines)).

R_{ss} is electronic resistance of the Li-O₂ cell consists of the electrodes, contacts, and electrolyte resistance. R_{ct} is the charge transfer resistance and CPE1, CPE2 are the constant phase elements. The CPE2 is related to diffusion of active species to the surfaces of the electrodes.

Table 1 Values of the fitting parameters evaluated from the equivalent circuit with different cathode catalysts after discharged at various cycles.

	Li/O ₂ batteries (Co ₃ O ₄)				Li/O ₂ batteries (Pd/Co ₃ O ₄)			
	Fresh cell	Error%	After 40 th cycles discharged	Error%	Fresh cell	Error%	After 70 th cycles discharged	Error%
R _{ss} /Ω	12.34	1.6335	76.3	0.725	11.76	0.52185	46.02	0.58281
R _{ct} /Ω	71.67	1.1697	124.9	1.7382	73.99	0.70773	119.5	1.3686
CPE1-T/F	2.34E-05	6.2539	2.05E-04	7.7034	2.18E-05	4.4269	1.43E-04	4.7314
CPE1-P	0.7928	1.0199	0.66842	1.9275	0.7771	0.63965	0.73634	1.1106
CPE2-T/F	0.013489	1.1566	0.01122	1.8377	0.0158	2.1261	0.01142	1.5853
CPE2-P	0.36323	1.854	0.81005	1.9107	0.70997	1.9711	0.57375	1.9495

References

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