

Supplementary Information

For

Magnetic nanomaterials with near-infrared pH-activatable fluorescence *via* iron-catalyzed AGET ATRP for tumor acidic microenvironment imaging†

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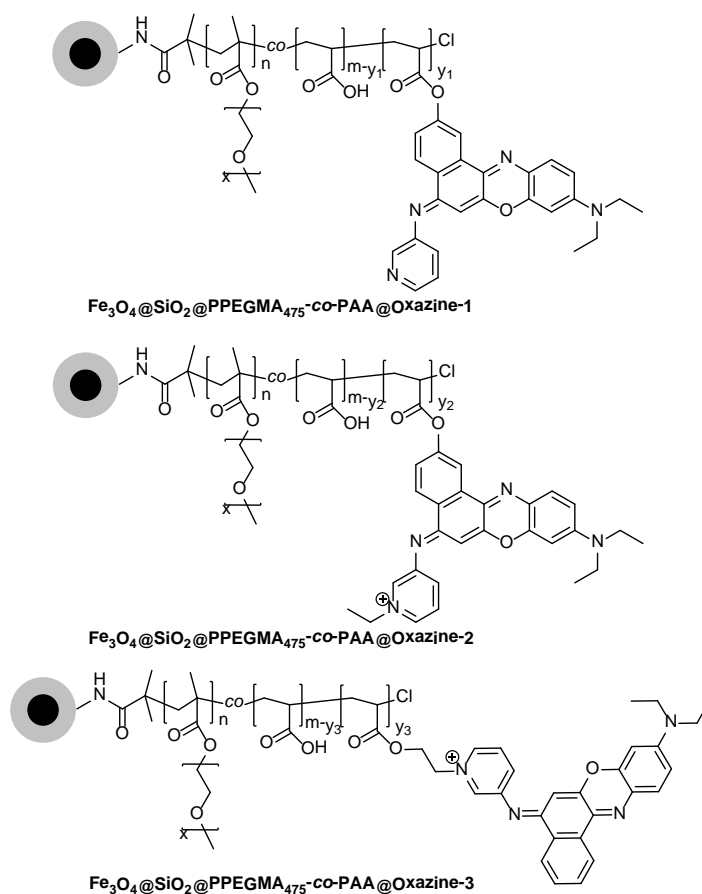


Fig. S1 Structures of benzo[*a*]phenoxazine-dotted Fe₃O₄@SiO₂@PPEGMA₄₇₅-co-PAA@Oxazine MNPs.

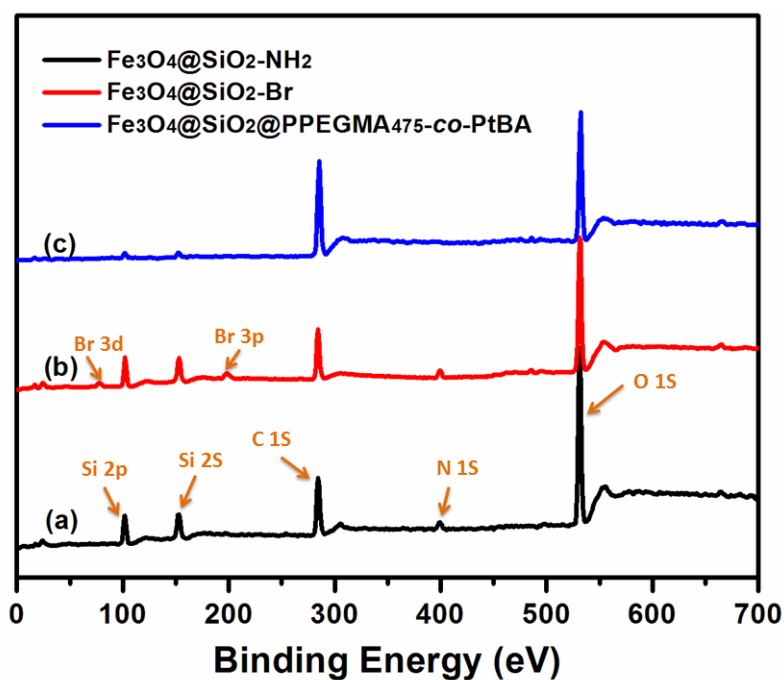


Fig. S2 XPS spectra of MNPs of (a) Fe₃O₄@SiO₂-NH₂, (b) Fe₃O₄@SiO₂-Br and (c) Fe₃O₄@SiO₂@PPEGMA₄₇₅-co-PtBA.

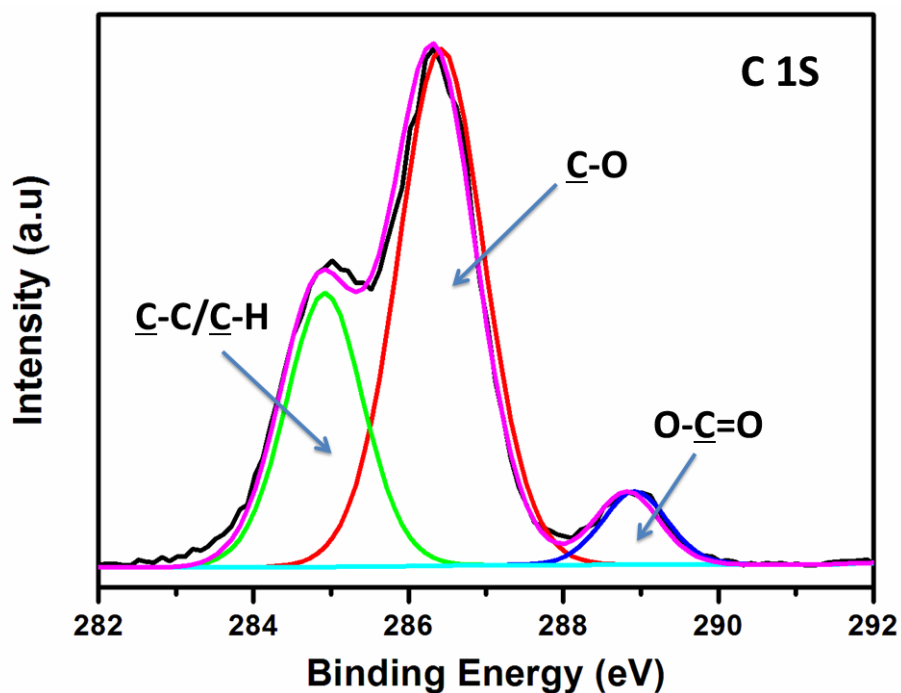


Fig. S3 XPS C 1s core-level spectra of $\text{Fe}_3\text{O}_4@\text{SiO}_2@\text{PPEGMA}_{475}\text{-CO-PtBA}$.

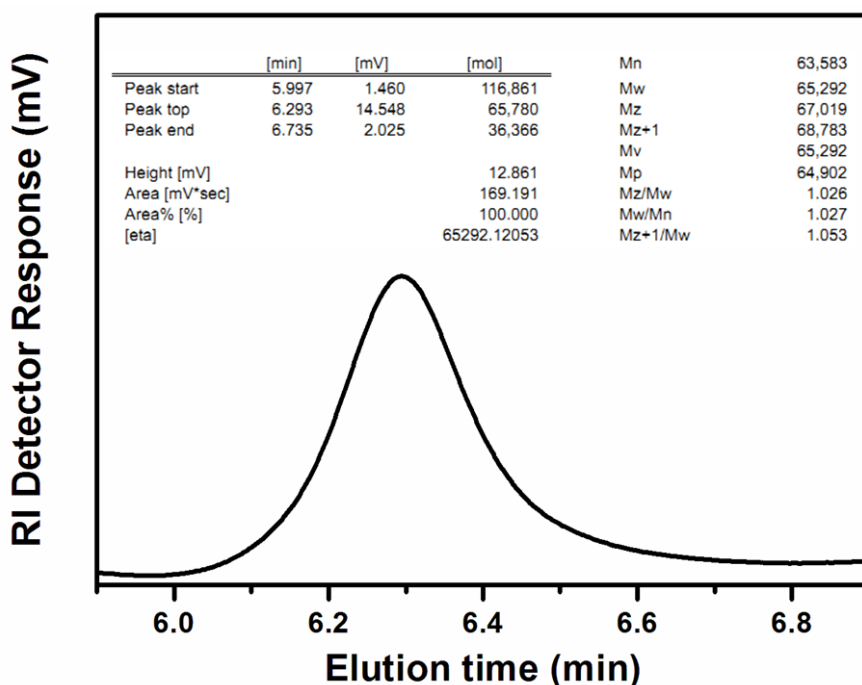


Fig. S4 GPC curve and data of $\text{PPEGMA}_{475}\text{-CO-PtBA}$ grafted on the surface of $\text{Fe}_3\text{O}_4@\text{SiO}_2@\text{PPEGMA}_{475}\text{-CO-PtBA}$. The sample was obtained by etching with hydrofluoric acid from corresponding MNPs.

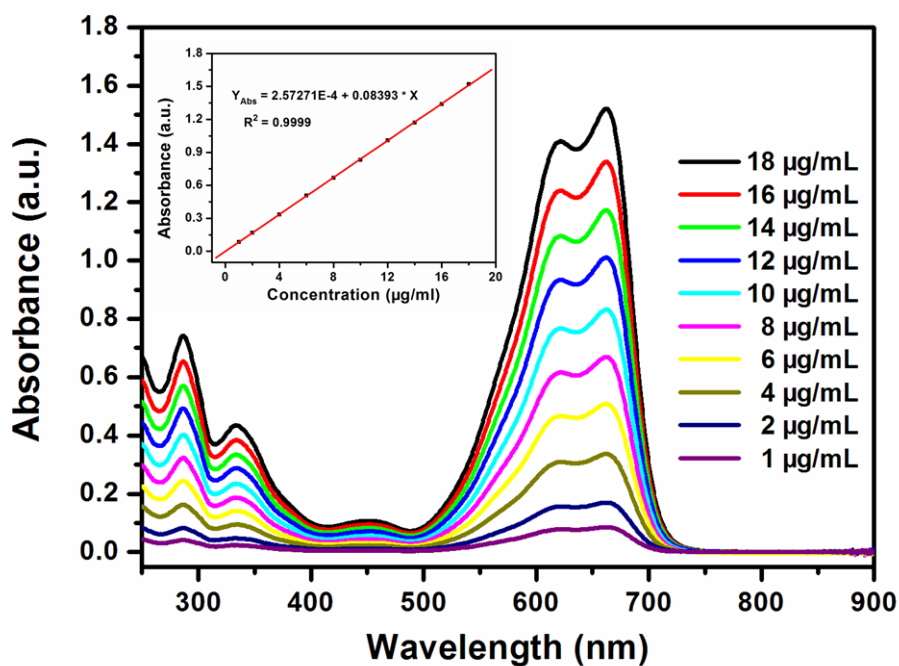


Fig. S5 Absorption spectra of **3a** (1, 2, 4, 6, 8, 10, 12, 14, 16 and 18 μg/mL) in Na₂HPO₄-citric acid buffer solution (pH 3.0); inset shows the linear fit of absorbance at 665 nm.

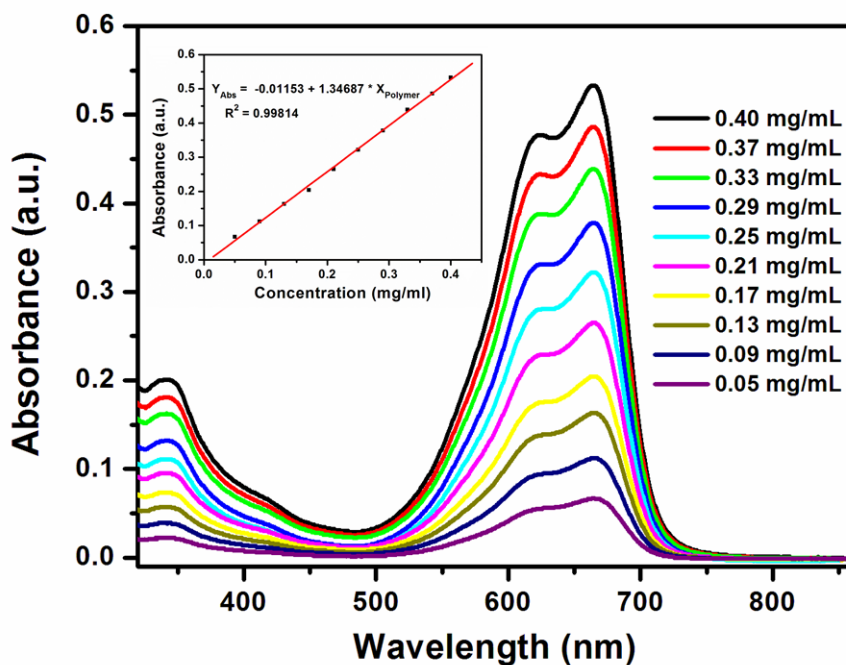


Fig. S6 Absorption spectra of PPEGMA₄₇₅-co-PAA@Oxazine-2 (0.05, 0.09, 0.13, 0.17, 0.21, 0.25, 0.29, 0.33, 0.37 and 0.40 mg/mL) in Na₂HPO₄-citric acid buffer solution (pH 3.0); inset shows the linear fit of absorbance at 665 nm.

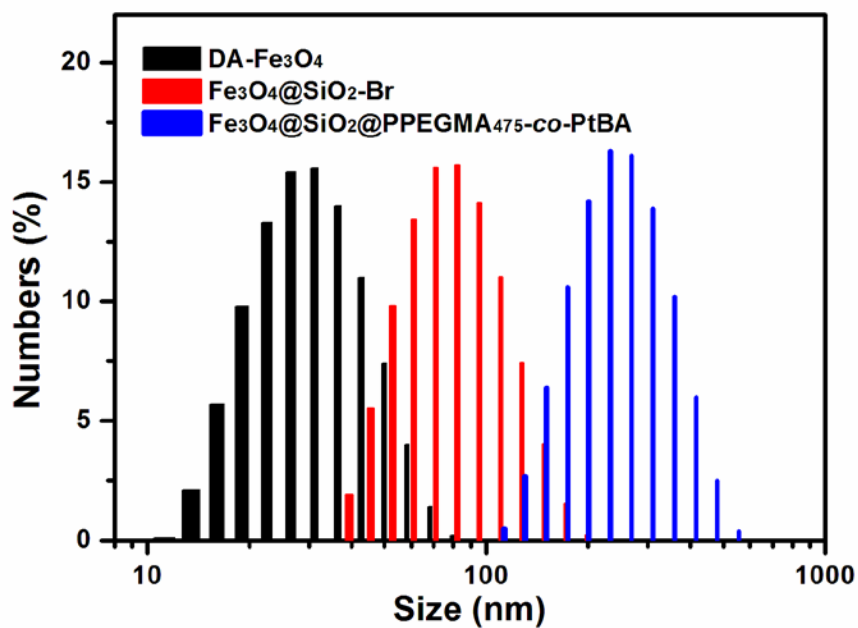


Fig. S7 Particle sizes of the as-prepared MNPs at different modification stages in water by DLS.

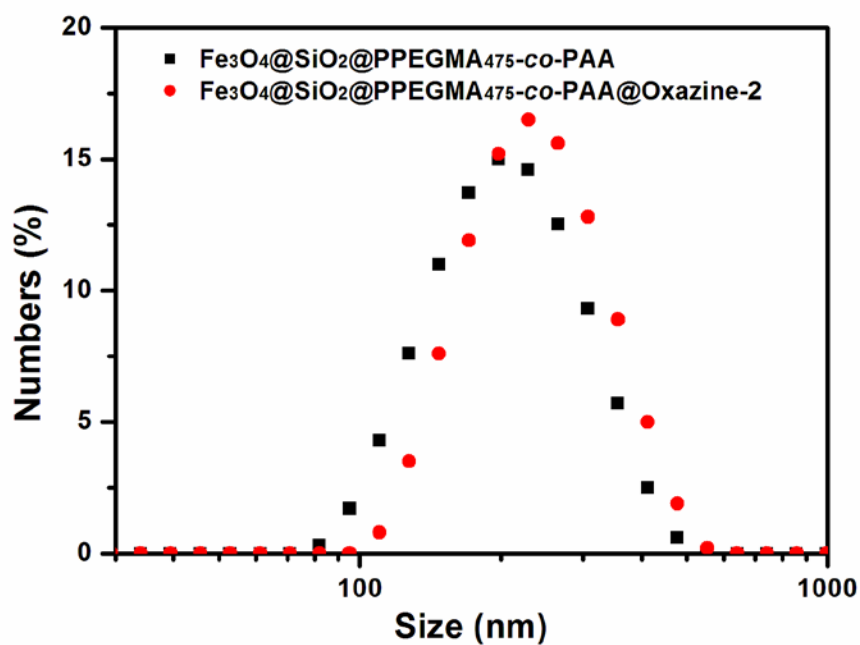


Fig. S8 Particle sizes of the as-prepared MNPs at different modification stages in water by DLS.

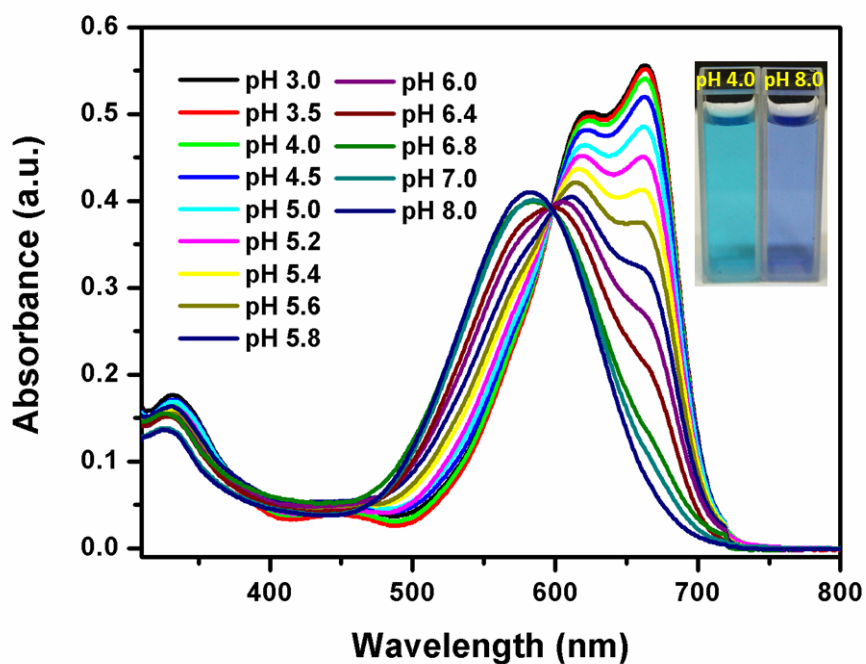


Fig. S9 Absorption properties of compound **3a** (10 μ M) toward different pH values in Na_2HPO_4 -citric acid buffer solution with 10% DMSO as a co-solvent.

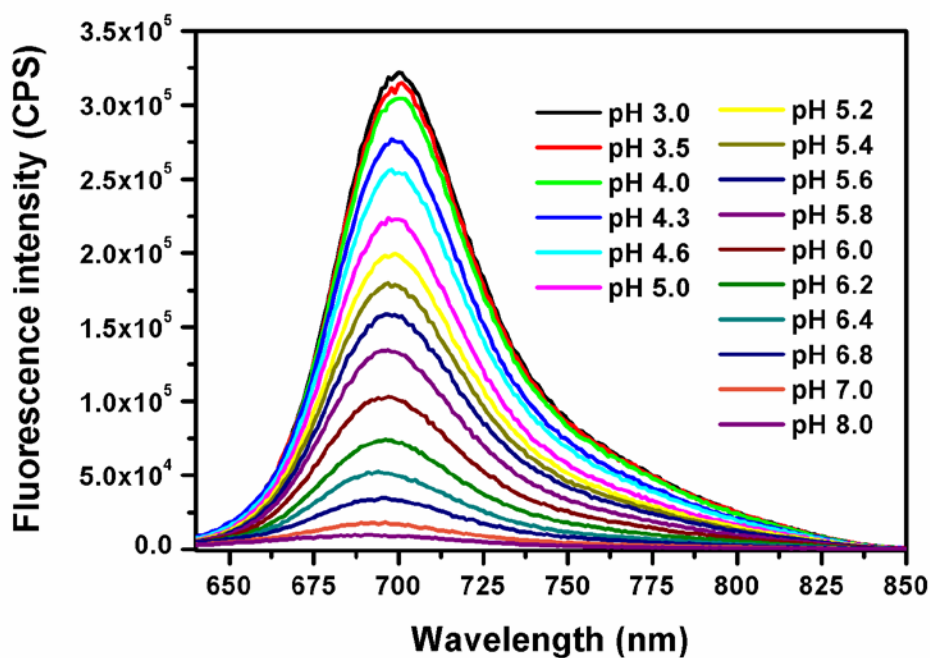


Fig. S10 Absorption properties of compound **3a** (10 μ M) toward different pH values in Na_2HPO_4 -citric acid buffer solution with 10% DMSO as a co-solvent ($\lambda_{\text{ex}} = 600$ nm).

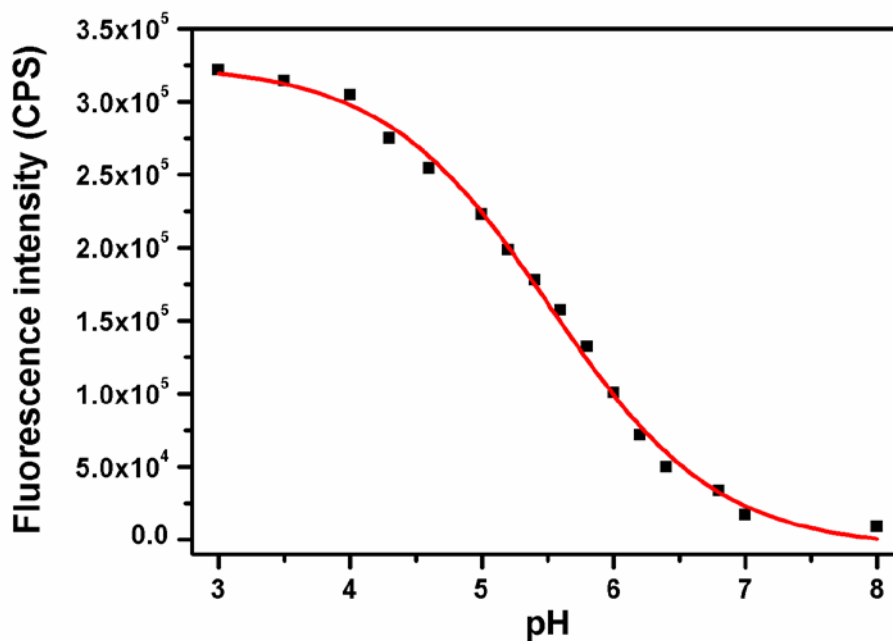


Fig. S11 Fluorescence intensity changes of compound **3a** (10 μ M) at 700 nm toward different pH values in Na_2HPO_4 -citric acid buffer solution with 10% DMSO as a co-solvent.

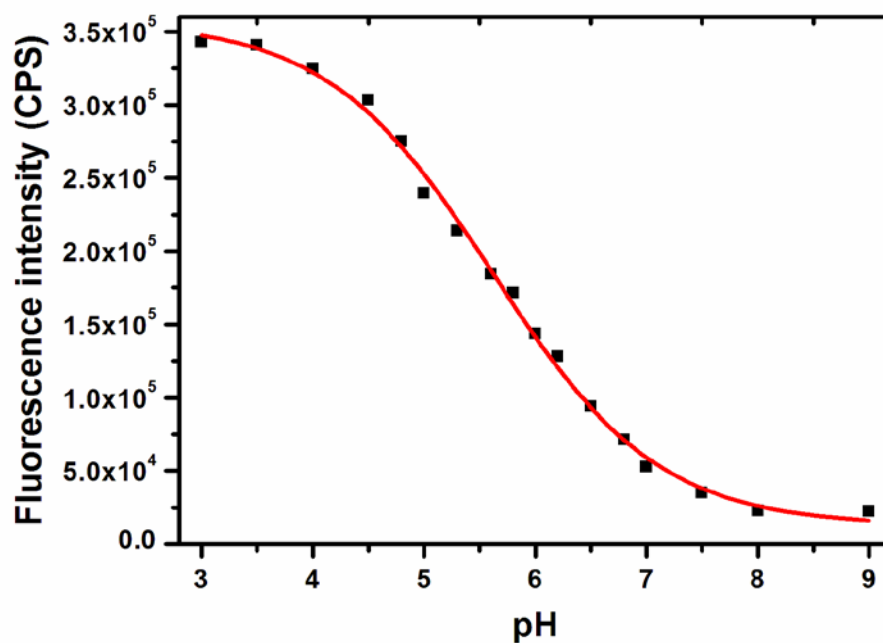


Fig. S12 Fluorescence intensity changes at 688 nm of $\text{Fe}_3\text{O}_4@\text{SiO}_2@\text{PPEGMA}_{475}\text{-co-PAA@Oxazine-2}$ toward different pH values in Na_2HPO_4 -citric acid buffer solution.

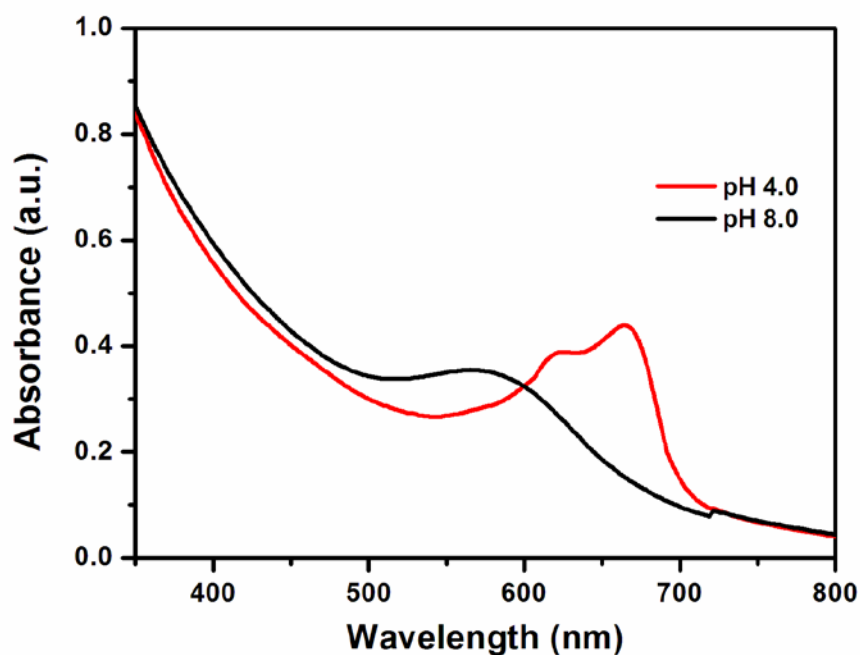


Fig. S13 Absorption properties of $\text{Fe}_3\text{O}_4@SiO_2@PPEGMA_{475-co-PAA@Oxazine-2}$ toward different pH values in Na_2HPO_4 -citric acid buffer solution. (Iron concentrations is 0.025 mg/mL, $\lambda_{ex} = 600$ nm).

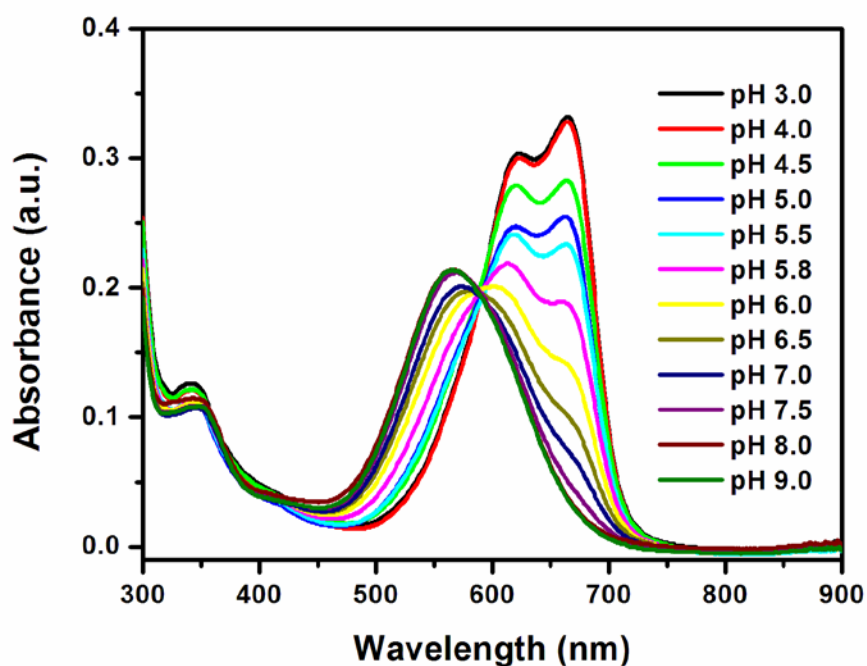


Fig. S14 Absorption spectra of grafted polymer $PPEGMA_{475-co-PAA@Oxazine-2}$ (0.25 mg/mL) toward different pH values in Na_2HPO_4 -citric acid buffer solution. The polymer was collected from the hybrid $\text{Fe}_3\text{O}_4@SiO_2@PPEGMA_{475-co-PAA@Oxazine-2}$ MNPs after treated with HF to remove the $\text{Fe}_3\text{O}_4@SiO_2$ core.

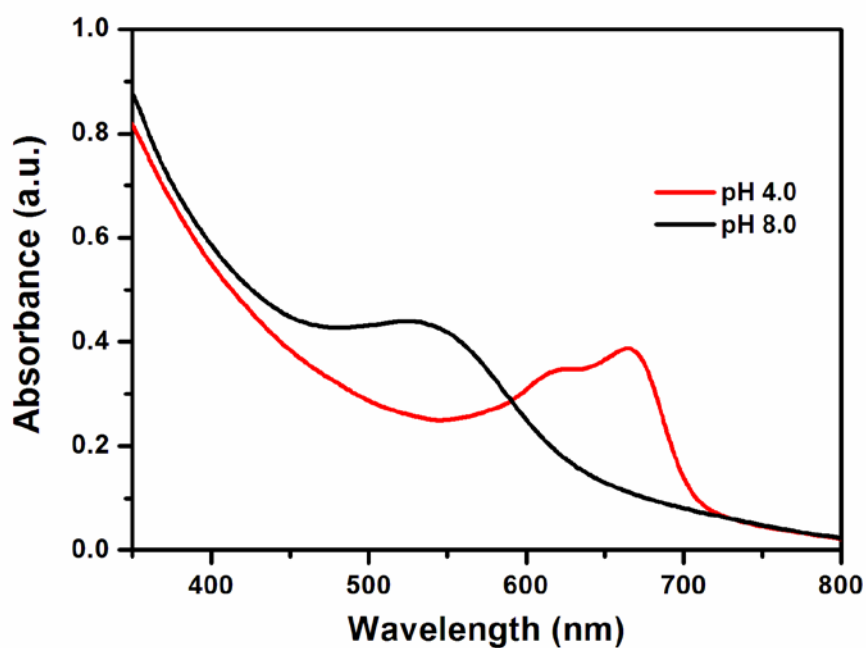


Fig. S15 Absorption properties of Fe₃O₄@SiO₂@PPEGMA_{475-co}-PAA@Oxazine-1 toward different pH values in Na₂HPO₄-citric acid buffer solution. (Iron concentrations is 0.025 mg/mL, λ_{ex} = 600 nm).

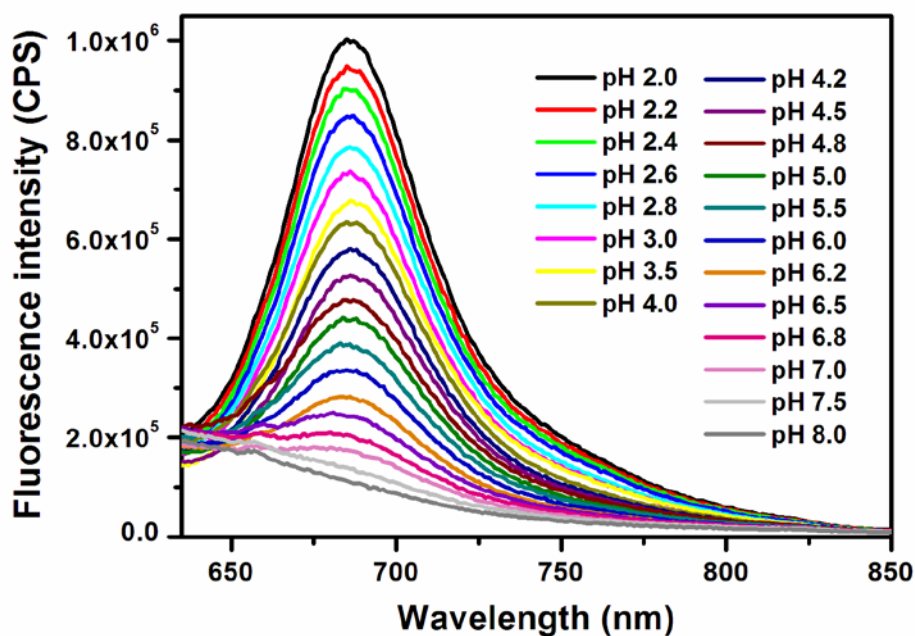


Fig. S16 Emission properties of Fe₃O₄@SiO₂@PPEGMA_{475-co}-PAA@Oxazine-1 toward different pH values in Na₂HPO₄-citric acid buffer solution. (Iron concentrations is 0.025 mg/mL, λ_{ex} = 600 nm).

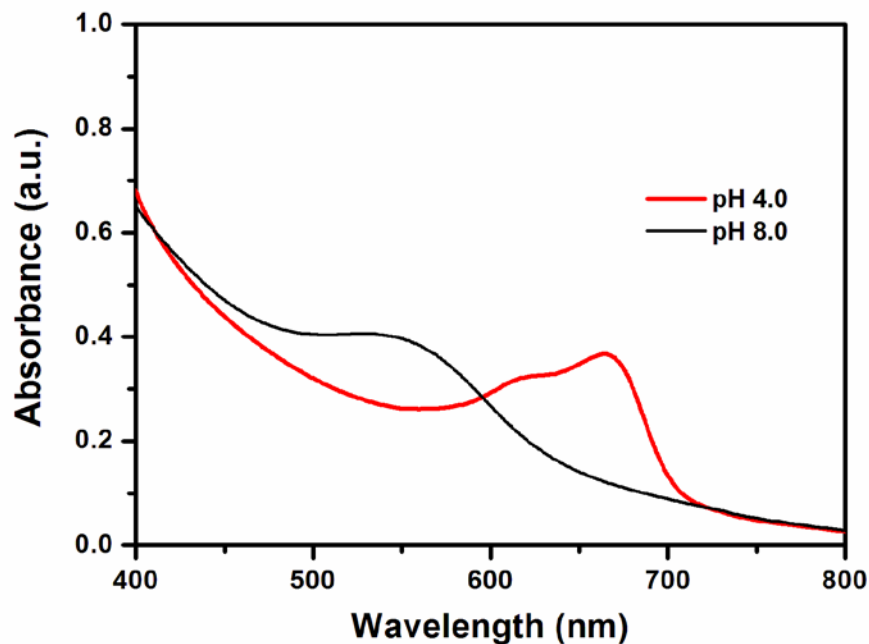


Fig. S17 Absorption properties of $\text{Fe}_3\text{O}_4@SiO_2@PPEGMA_{475-co-PAA}@Oxazine-3$ toward different pH values in Na_2HPO_4 -citric acid buffer solution. (Iron concentrations is 0.025 mg/mL, $\lambda_{ex} = 600$ nm).

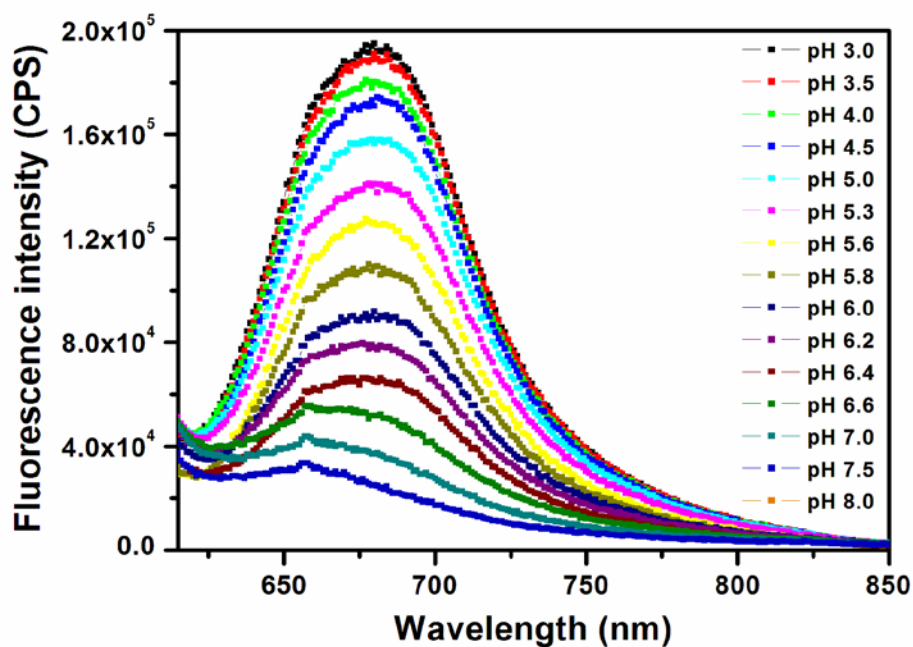


Fig. S18 Emission properties of $\text{Fe}_3\text{O}_4@SiO_2@PPEGMA_{475-co-PAA}@Oxazine-3$ toward different pH values in Na_2HPO_4 -citric acid buffer solution. (Iron concentrations is 0.012 mg/mL, $\lambda_{ex} = 600$ nm).

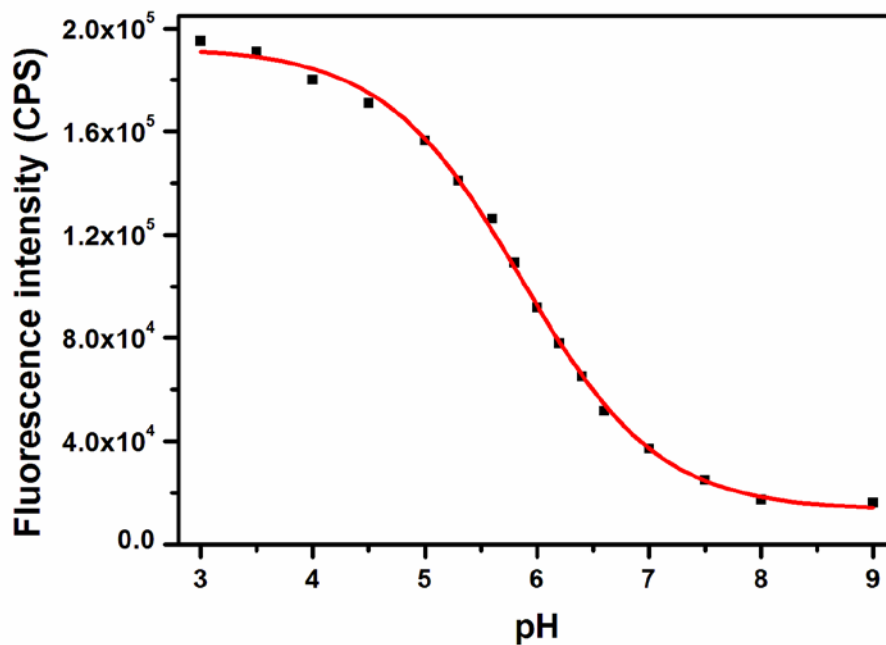


Fig. S19 Fluorescence intensity changes at 680 nm of $\text{Fe}_3\text{O}_4@\text{SiO}_2@\text{PPEGMA}_{475}\text{-co-PAA}@\text{Oxazine-3}$ toward different pH values in Na_2HPO_4 -citric acid buffer solution.

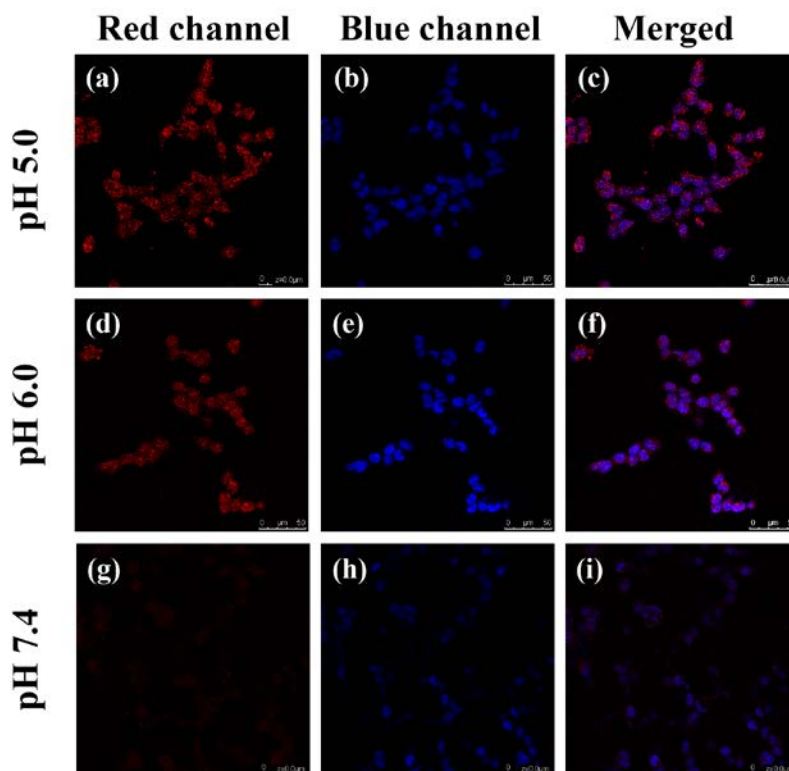


Fig. S20 Confocal fluorescent images of fixed 293T cells incubated with $\text{Fe}_3\text{O}_4@\text{SiO}_2@\text{PPEGMA}_{475}\text{-co-PAA}@\text{Oxazine-2}$ and Hoechst 33342 at pH 5.0 (a-c), pH 6.0 (d-f), and pH 7.4 (g-i), respectively.

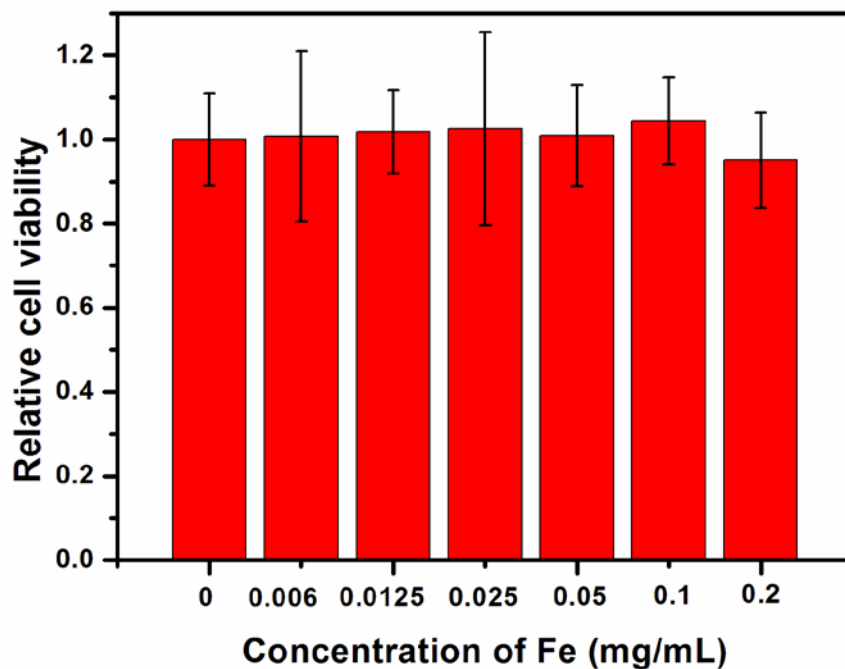


Fig. S21 Relative cell viability data of 4T1 cells incubated with a series of iron concentrations of $\text{Fe}_3\text{O}_4@\text{SiO}_2@\text{PPEGMA}_{475}\text{-CO-PAA}@ \text{Oxazine-2}$ measured by the MTT cell viability assay.

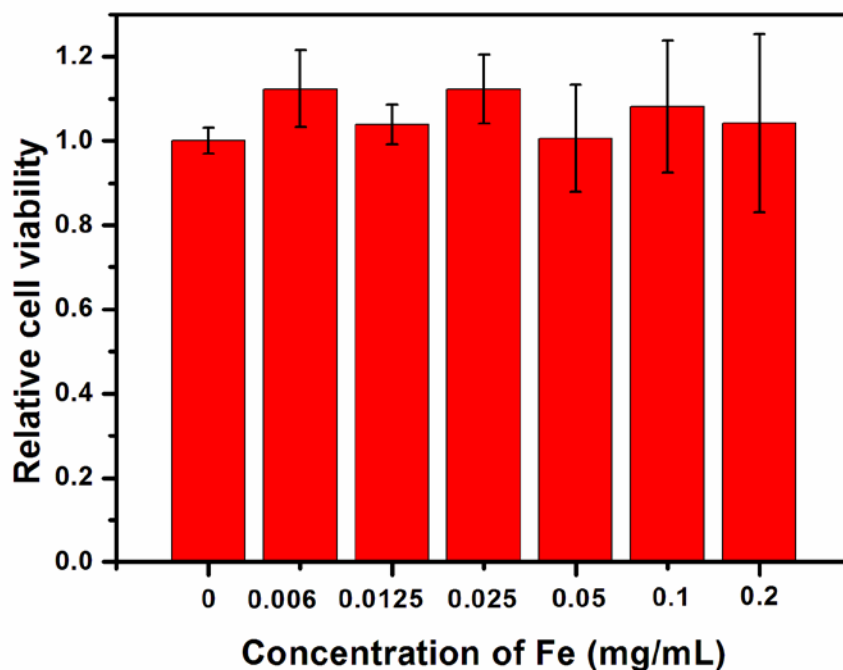


Fig. S22 Relative cell viability data of 293T cells incubated with a series of iron concentrations of $\text{Fe}_3\text{O}_4@\text{SiO}_2@\text{PPEGMA}_{475}\text{-CO-PAA}@ \text{Oxazine-2}$ measured by the MTT cell viability assay.

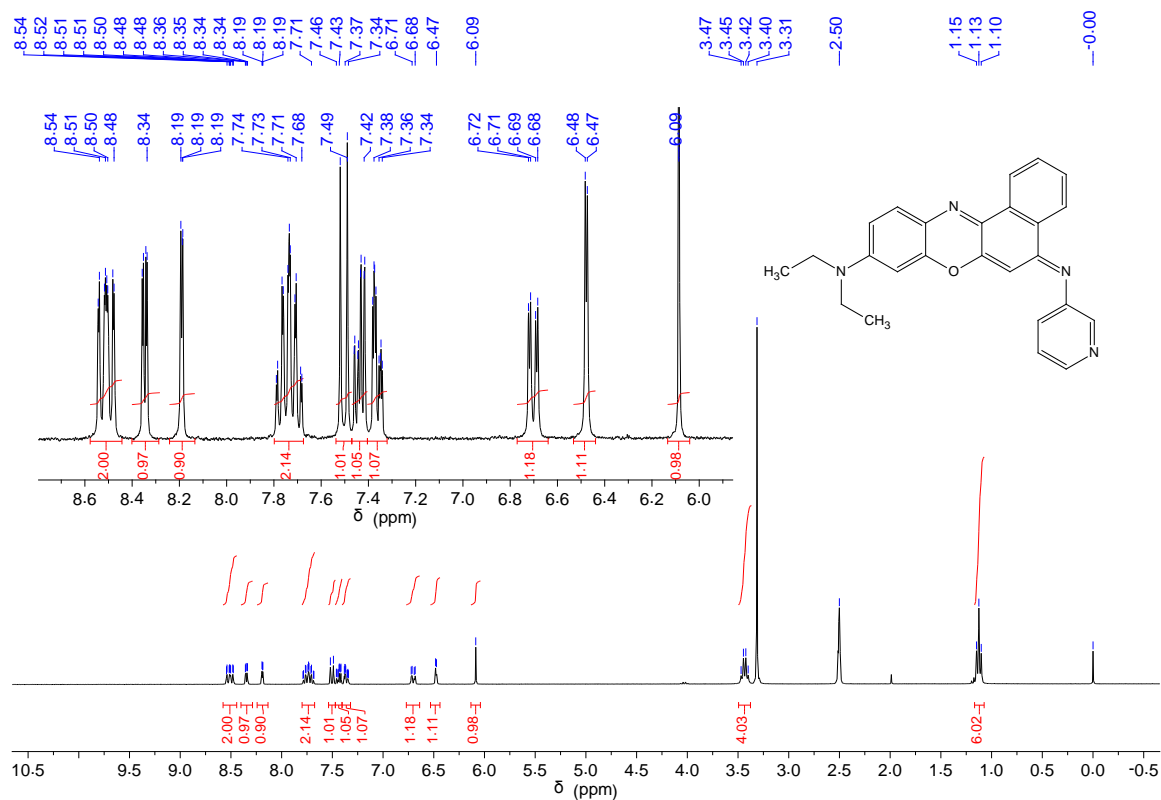


Fig. S23 ¹H NMR (300 MHz) of **2a** in DMSO-*d*₆.

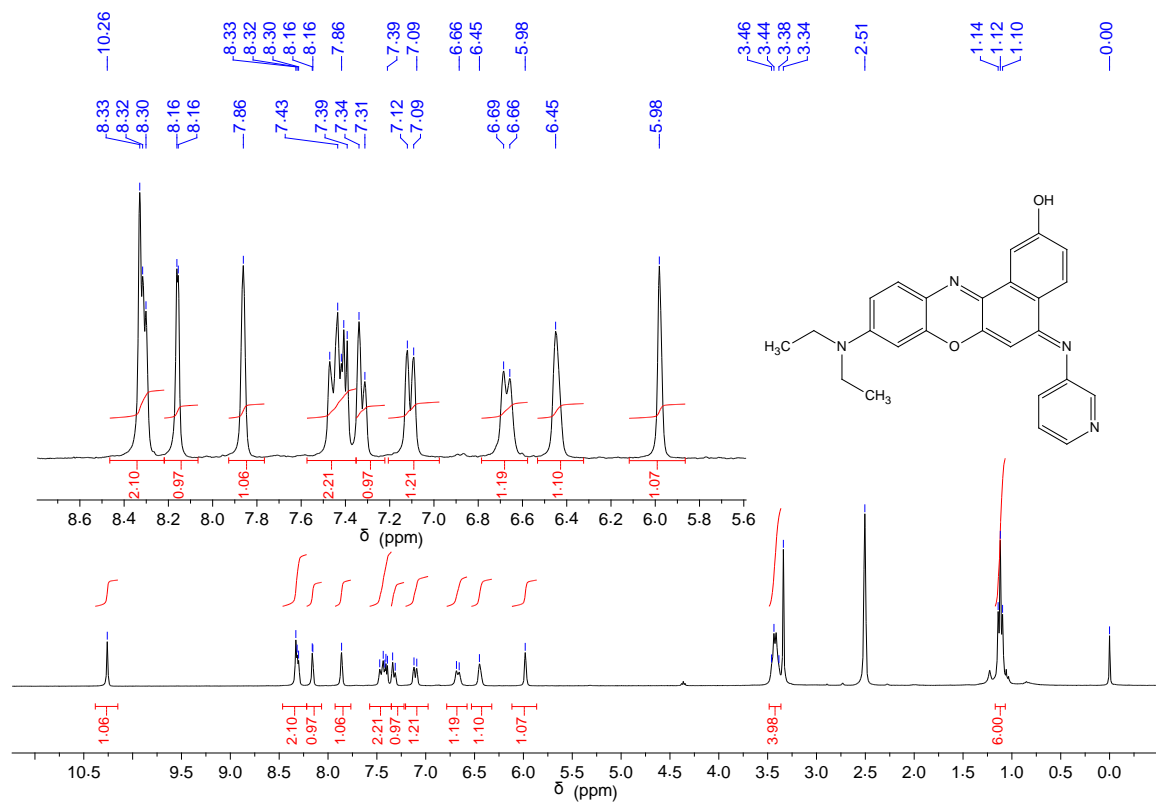


Fig. S24 ¹H NMR (300 MHz) of **2b** in DMSO-*d*₆.

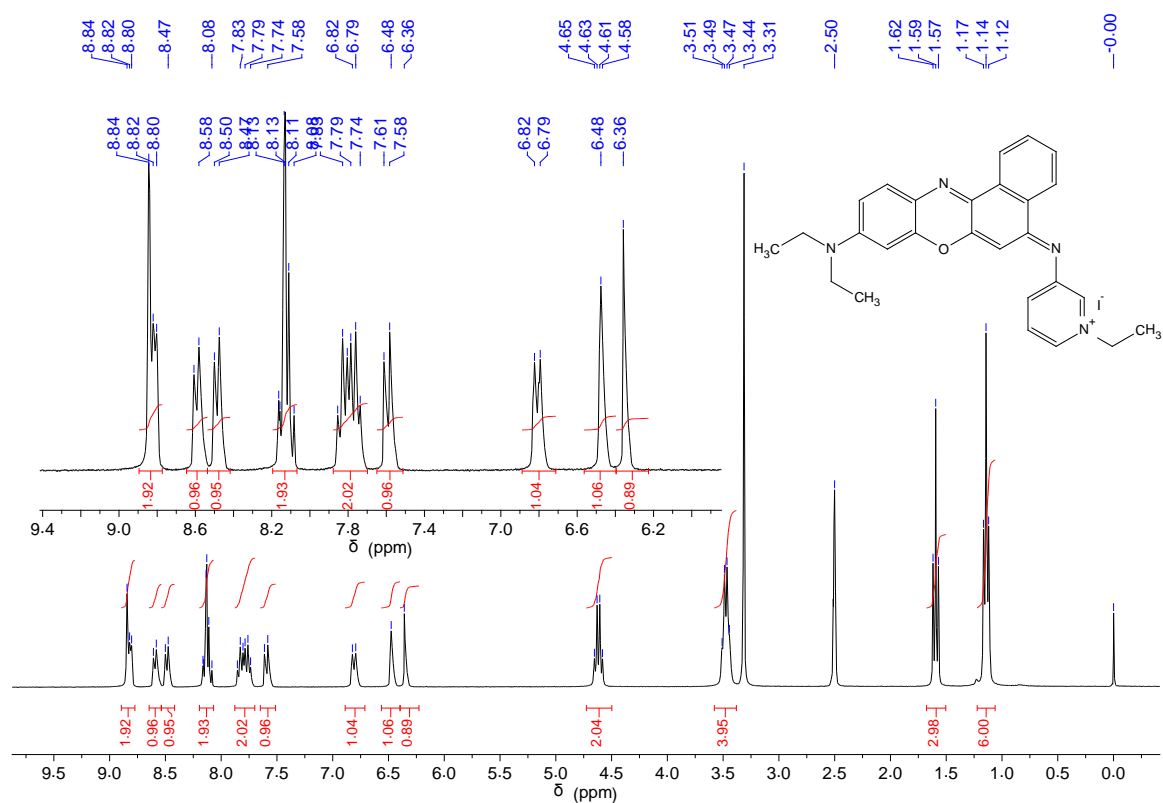


Fig. S25 ¹H NMR spectra (300 MHz) of **3a** in DMSO-*d*₆.

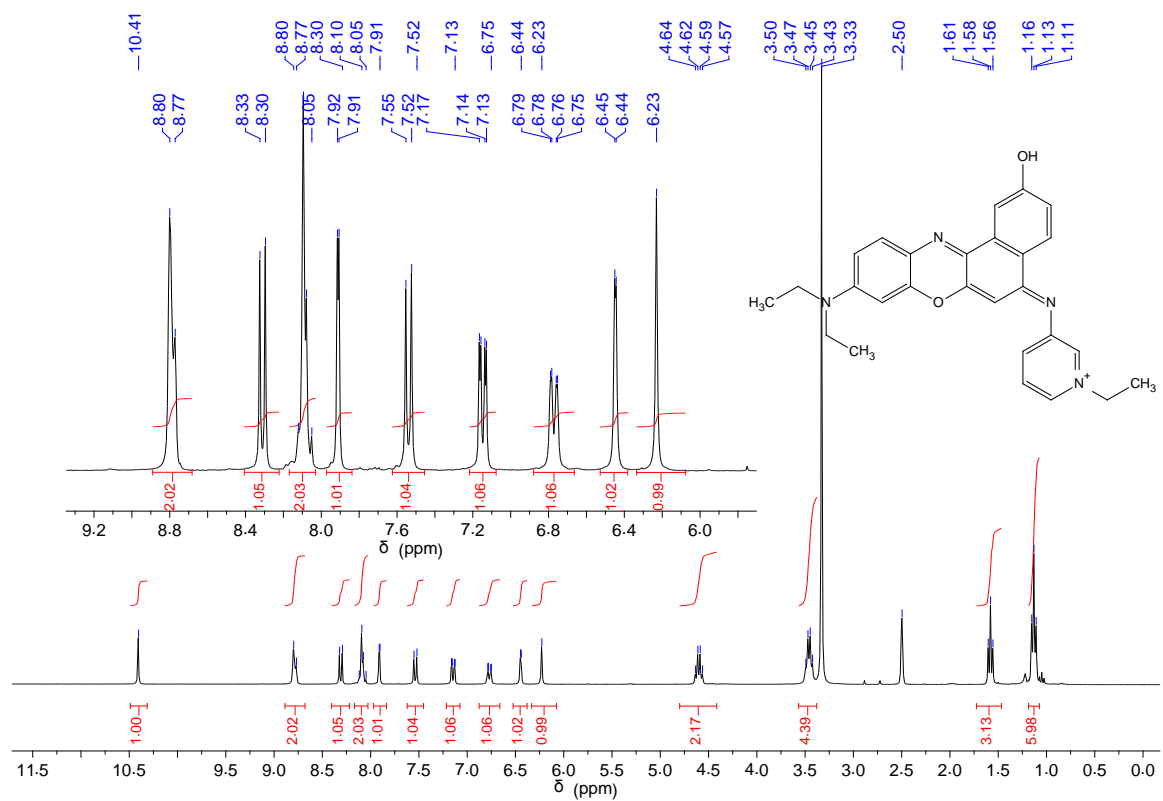


Fig. S26 ¹H NMR spectra (300 MHz) of **3b** in DMSO-*d*₆.

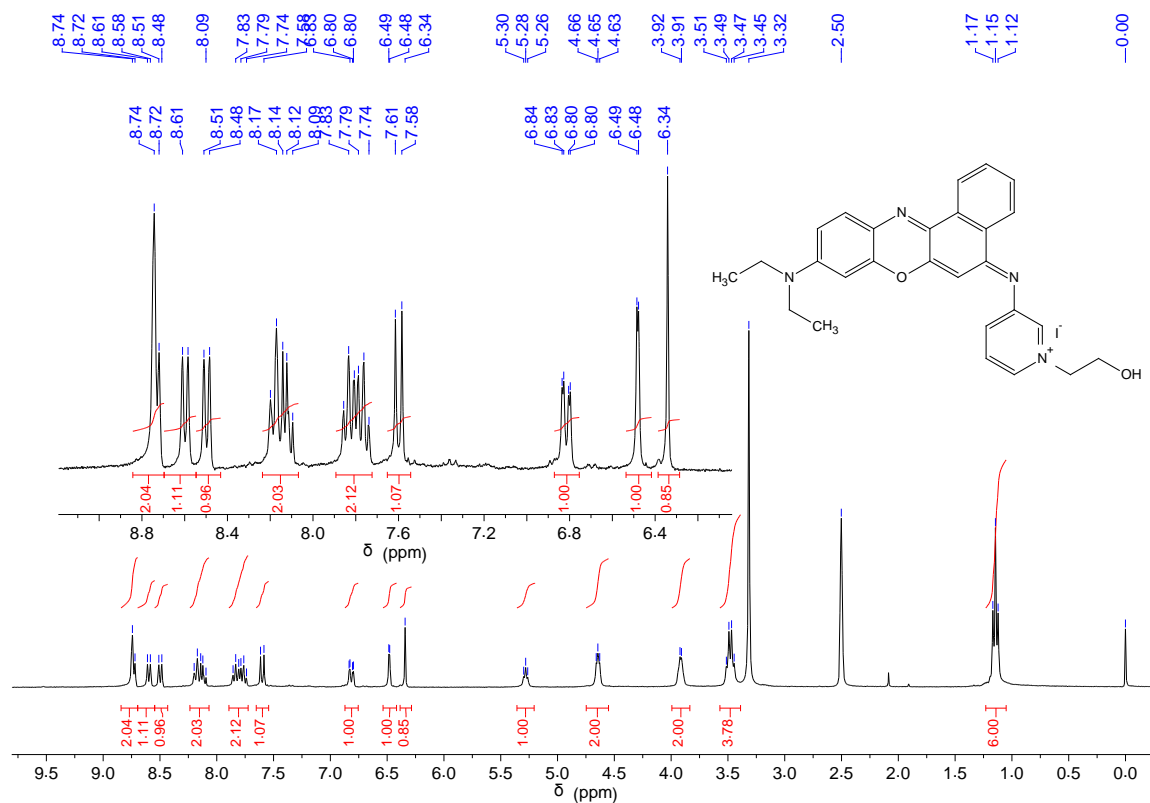


Fig. S27 ¹H NMR (300 MHz) of **3c** in DMSO-*d*₆.

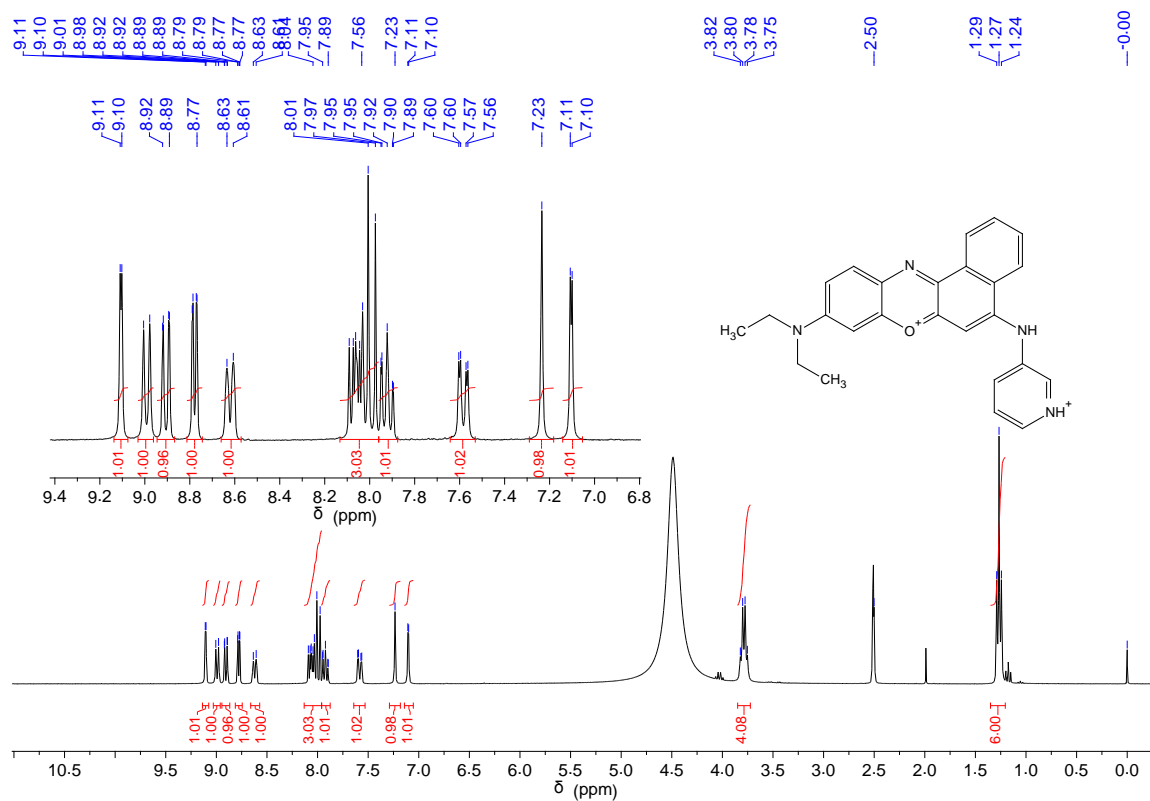


Fig. S28 ¹H NMR spectra (300 MHz) of **2a+H⁺** in DMSO-*d*₆.

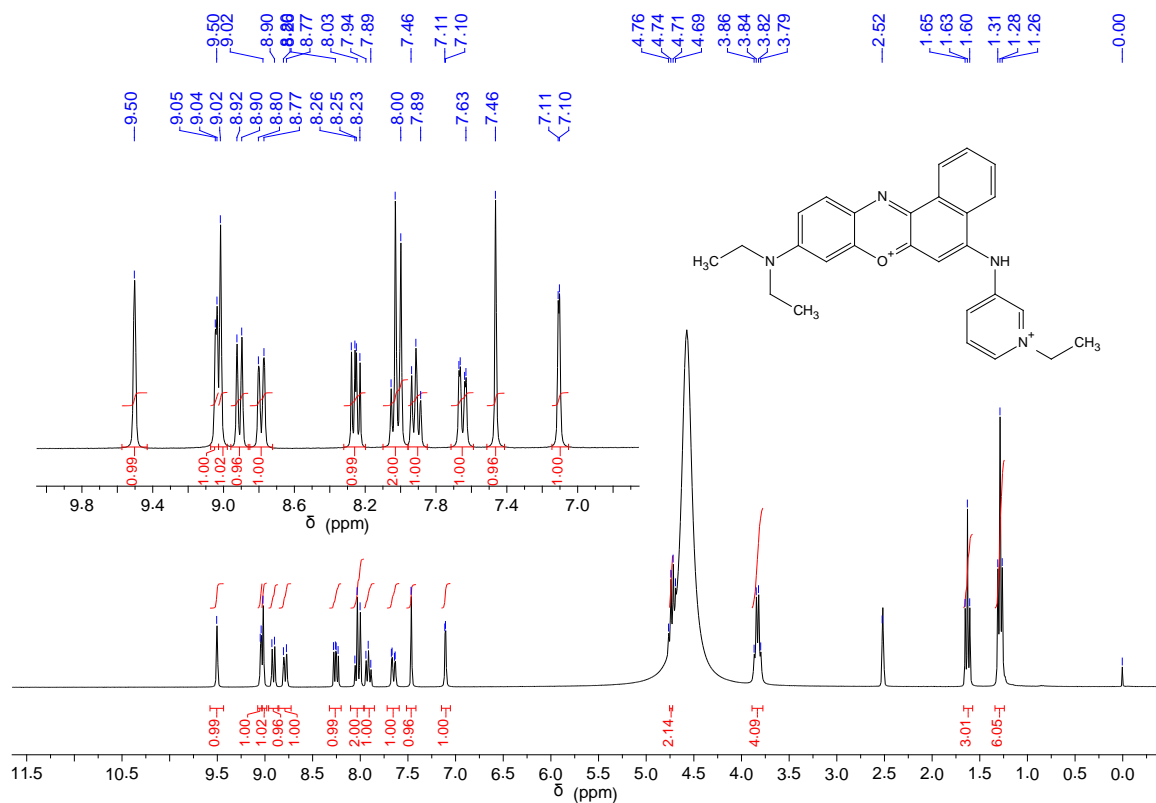


Fig. S29 ¹H NMR spectra (300 MHz) of **3a+H⁺** in DMSO-*d*₆.

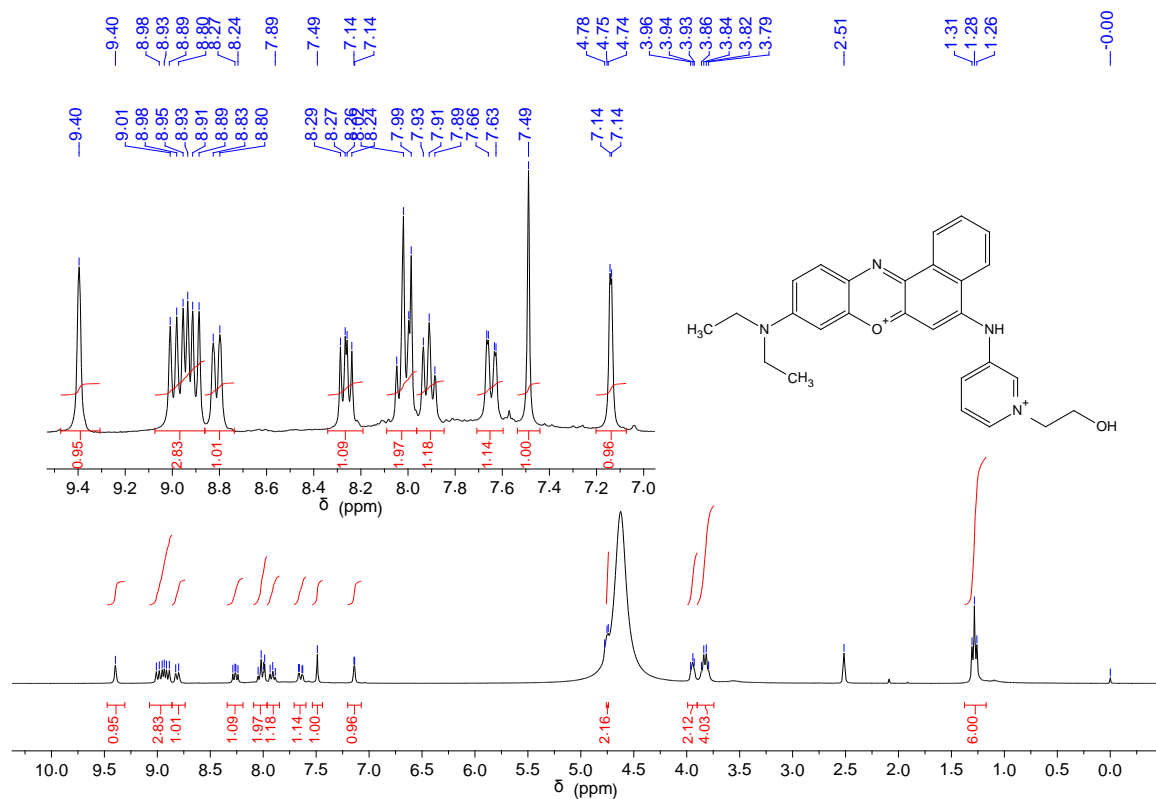


Fig. S30 ¹H NMR spectra (300 MHz) of **3c+H⁺** in DMSO-*d*₆.

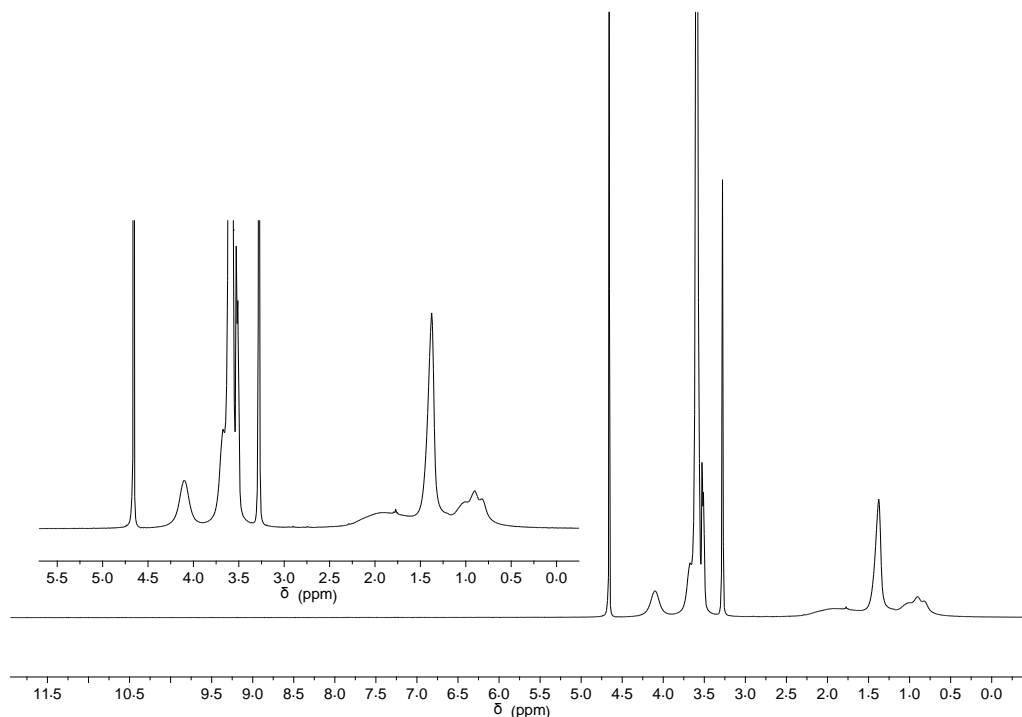


Fig. S31 ^1H NMR spectrum (300 MHz) of grafted PPEGMA₄₇₅-*co*-PAA in DMSO-*d*₆. The polymer was collected from Fe₃O₄@SiO₂@PPEGMA₄₇₅-*co*-PAA after treated with HF to remove the Fe₃O₄@SiO₂ core.

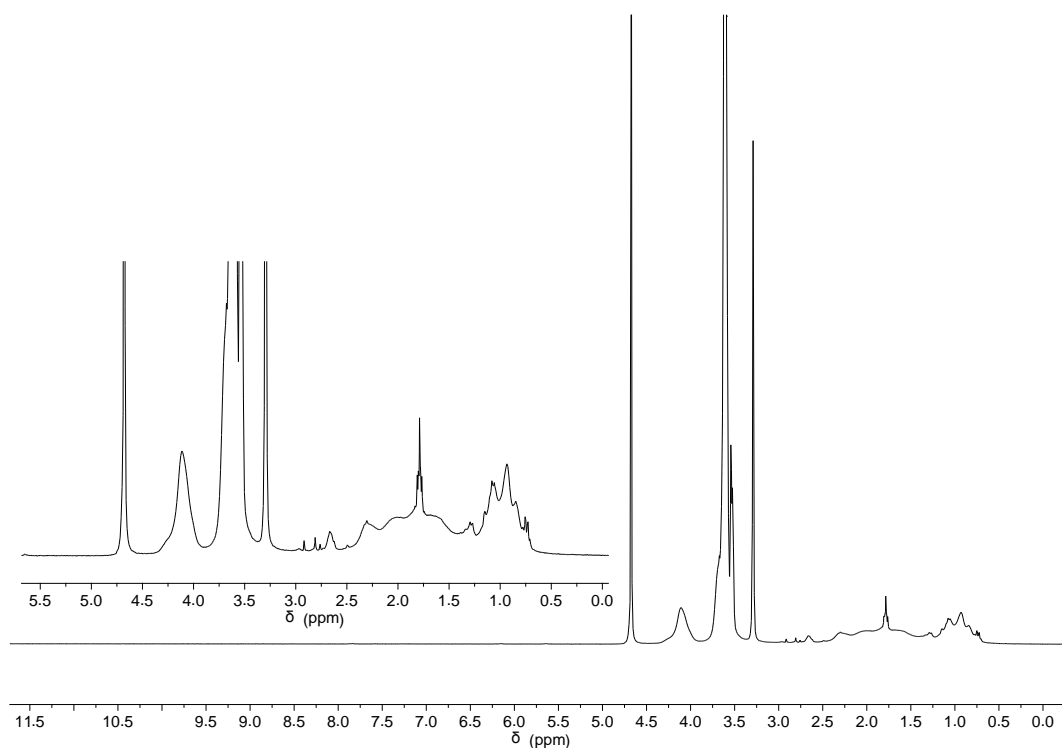


Fig. S32 ^1H NMR spectrum (300 MHz) of grafted PPEGMA₄₇₅-*co*-PAA in DMSO-*d*₆. The polymer was collected from Fe₃O₄@SiO₂@PPEGMA₄₇₅-*co*-PAA after treated with HF to remove the Fe₃O₄@SiO₂ core.

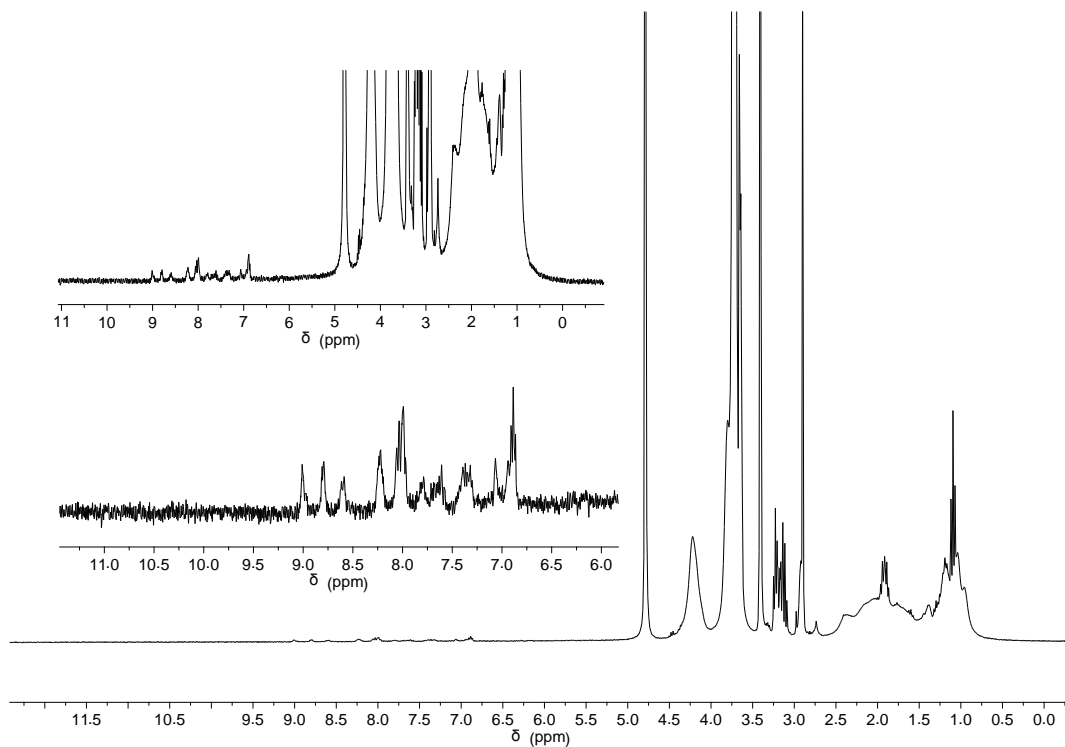


Fig. S33 ^1H NMR spectrum of grafted polymer in $\text{DMSO-}d_6$. The polymer was collected from $\text{Fe}_3\text{O}_4@\text{SiO}_2@\text{PPEGMA}_{475}\text{-}co\text{-PAA}@$ Oxazine-2 after treated with HF to remove the $\text{Fe}_3\text{O}_4@\text{SiO}_2$ core.

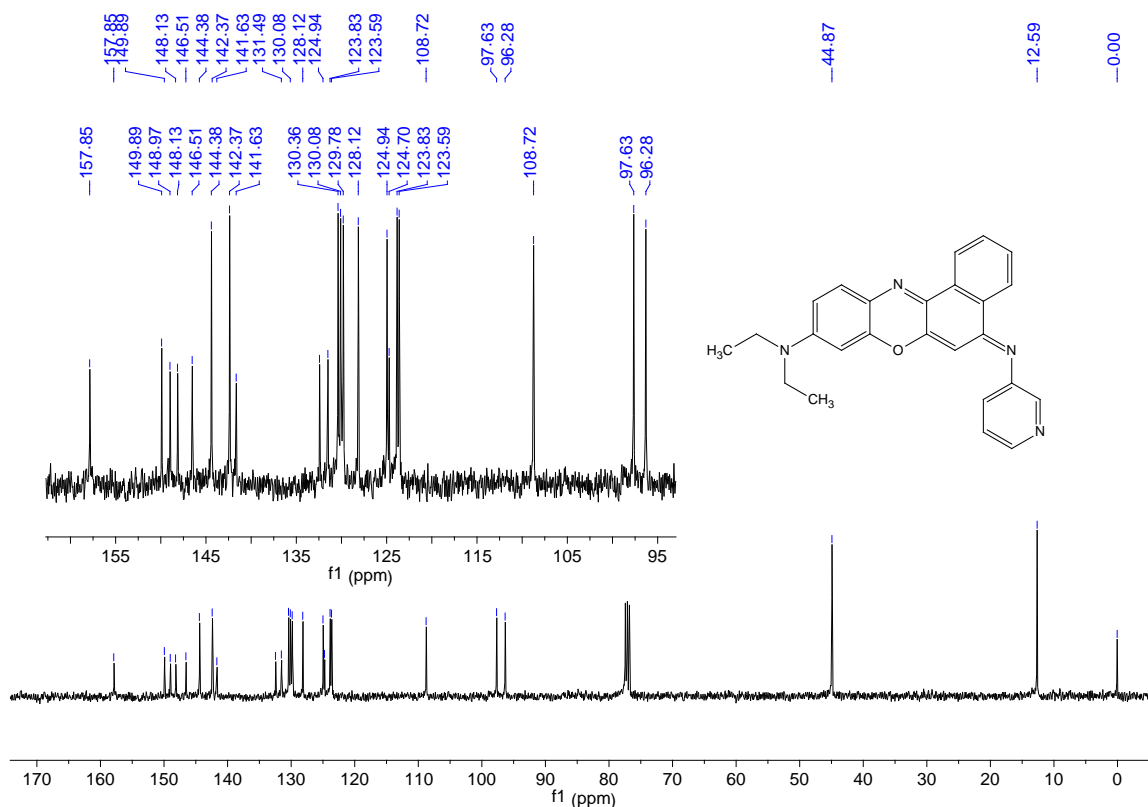


Fig. S34 ^{13}C NMR (75 MHz) of **2a** in CDCl_3 .

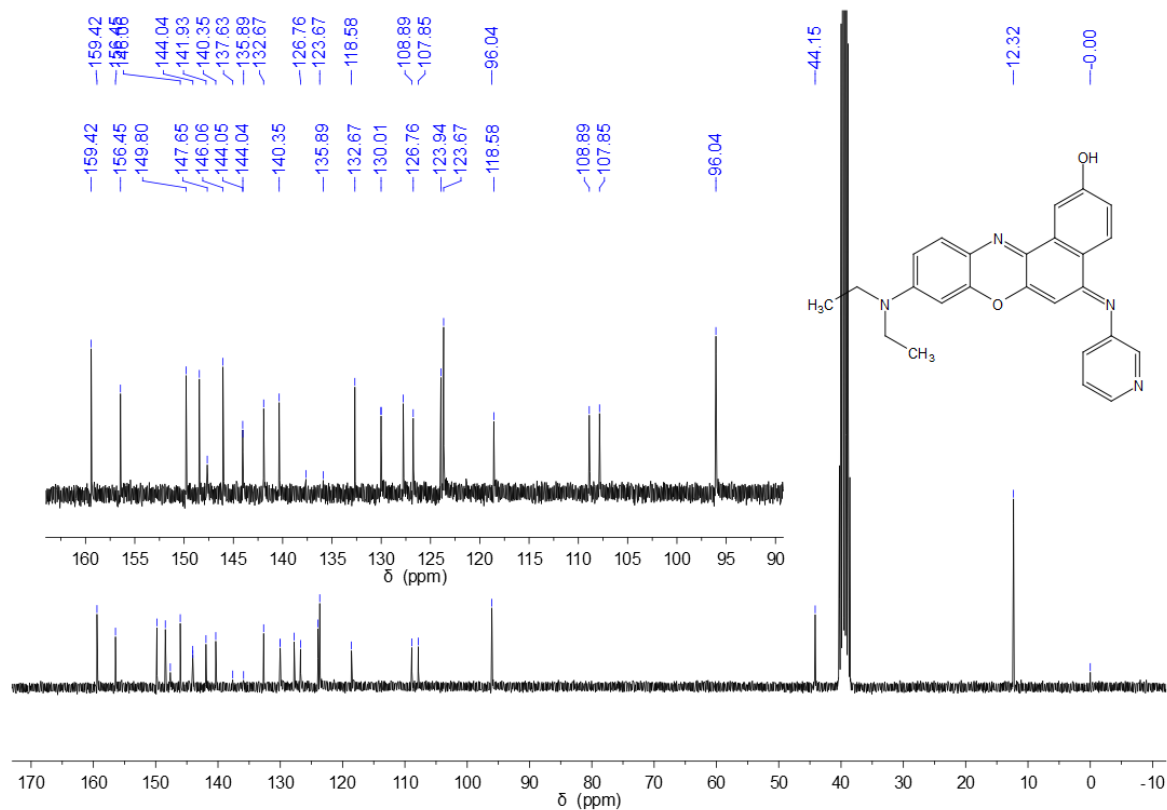


Fig. S35 ^{13}C NMR (75 MHz) of **2b** in $\text{DMSO-}d_6$.

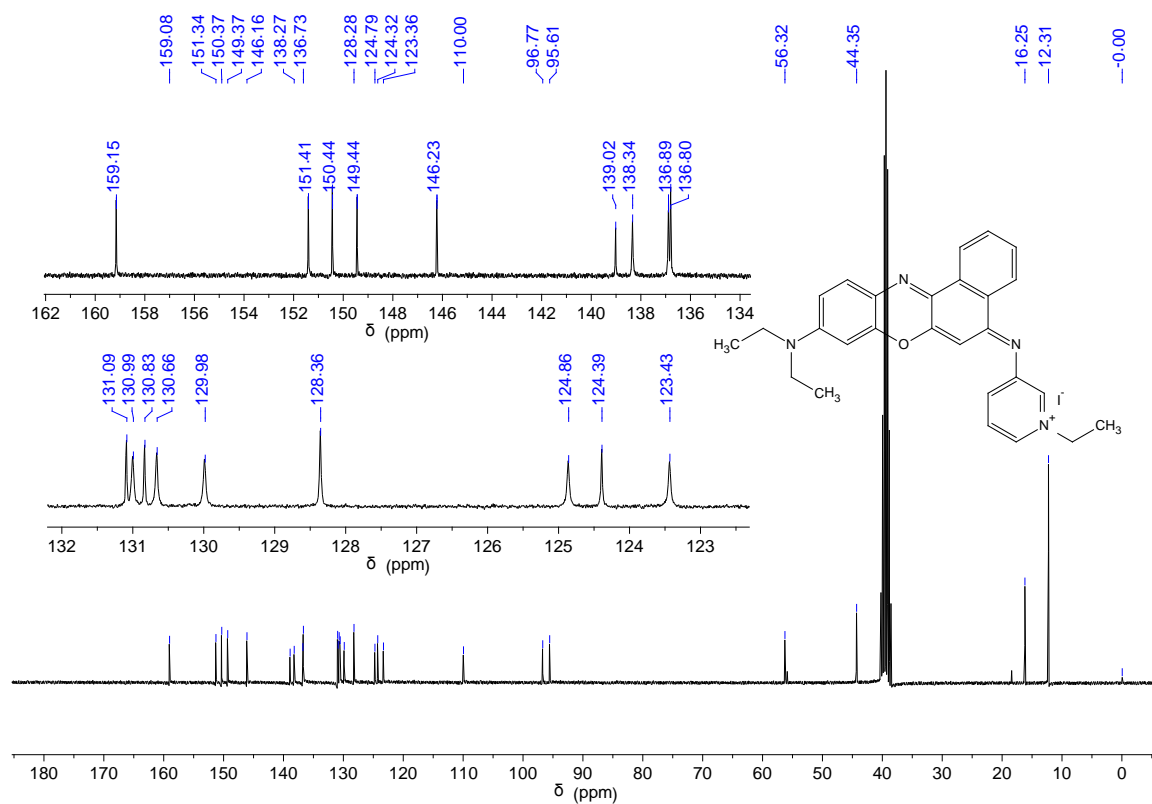


Fig. S36 ^{13}C NMR (75 MHz) of **3a** in $\text{DMSO-}d_6$.

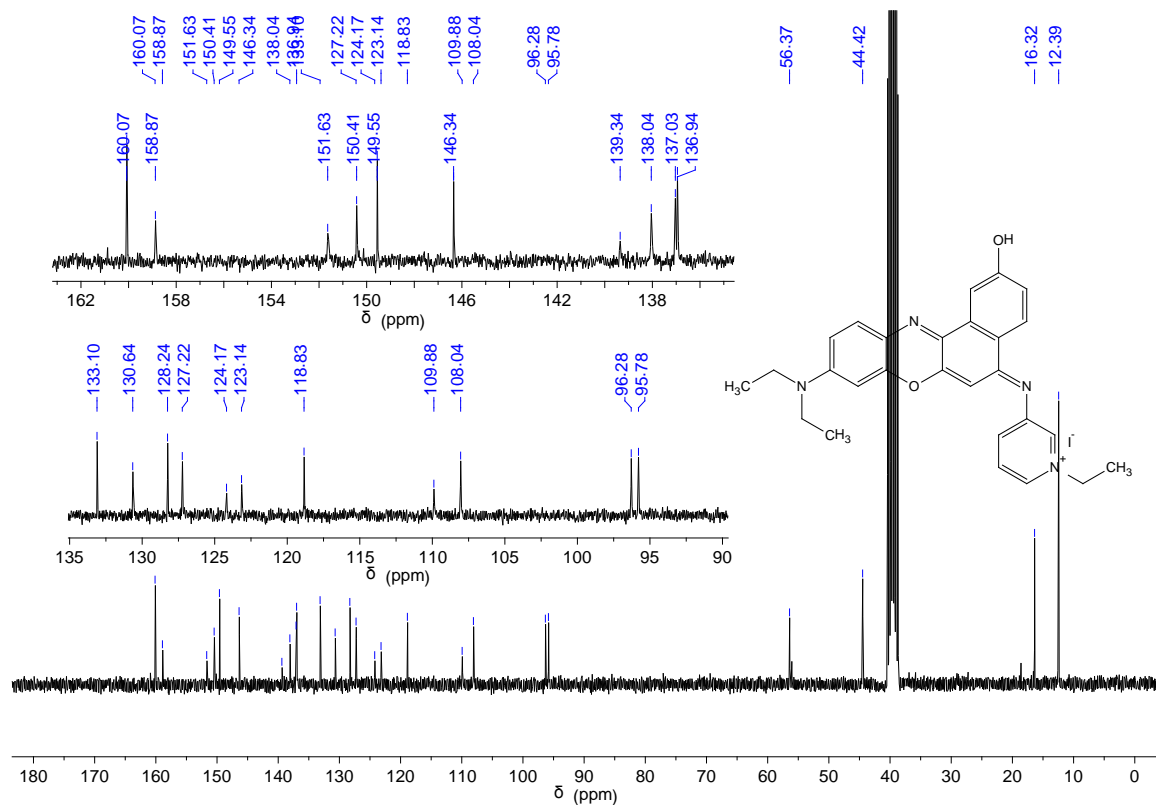


Fig. S37 ¹³C NMR (75 MHz) of **3b** in DMSO-*d*₆.

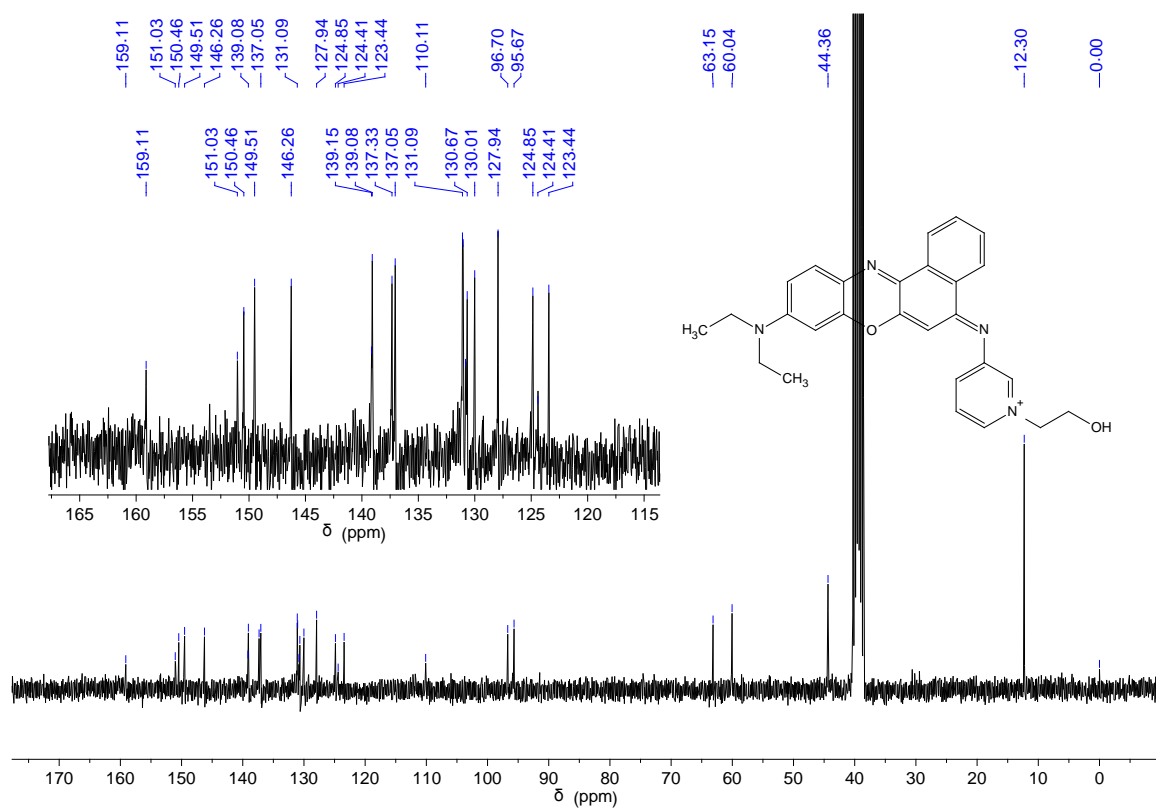


Fig. S38 ¹³C NMR (75 MHz) of **3c** in DMSO-*d*₆.

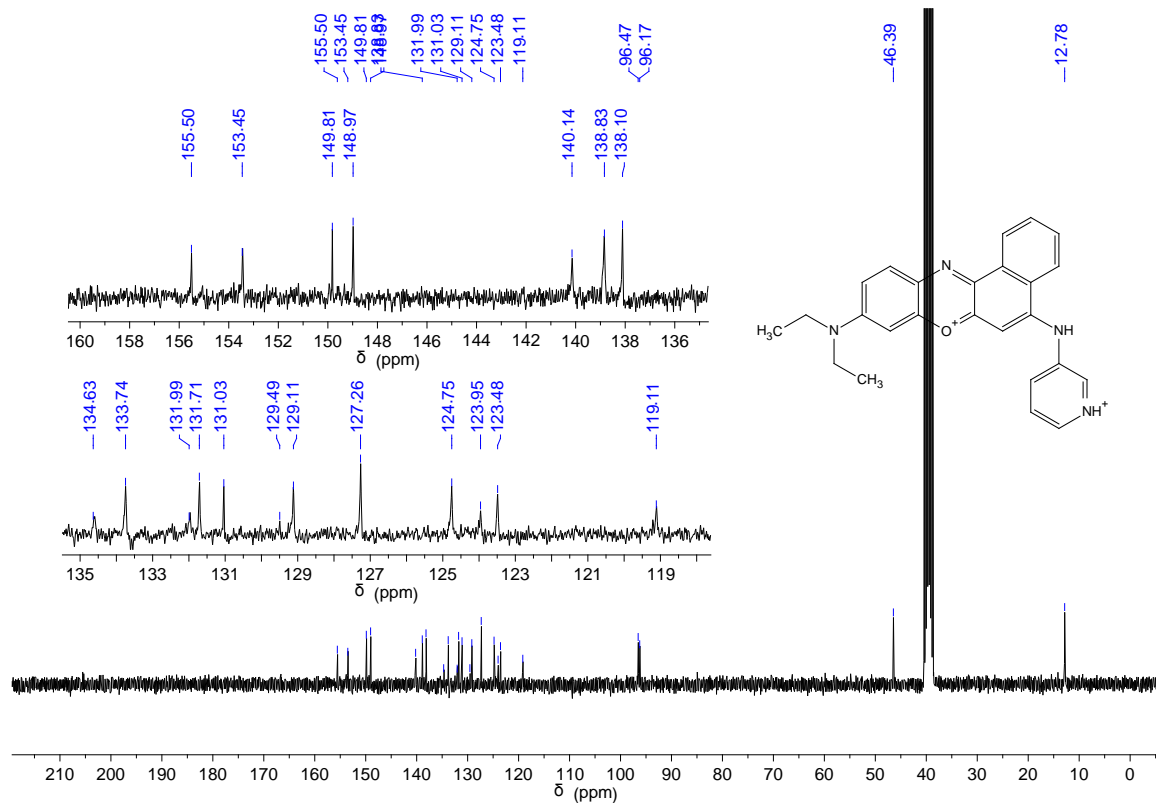


Fig. S39 ¹³C NMR (75 MHz) of **2a+H⁺** in DMSO-*d*₆.

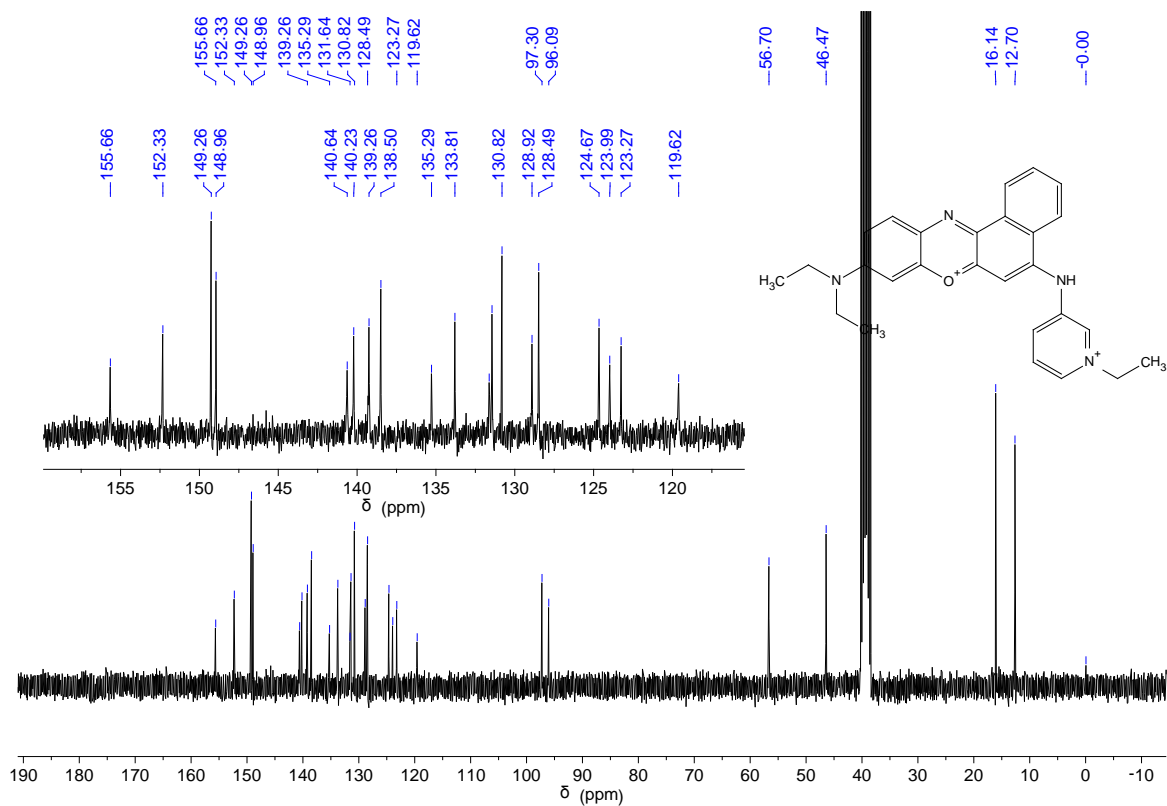


Fig. S40 ¹³C NMR (75 MHz) of **3a+H⁺** in DMSO-*d*₆.

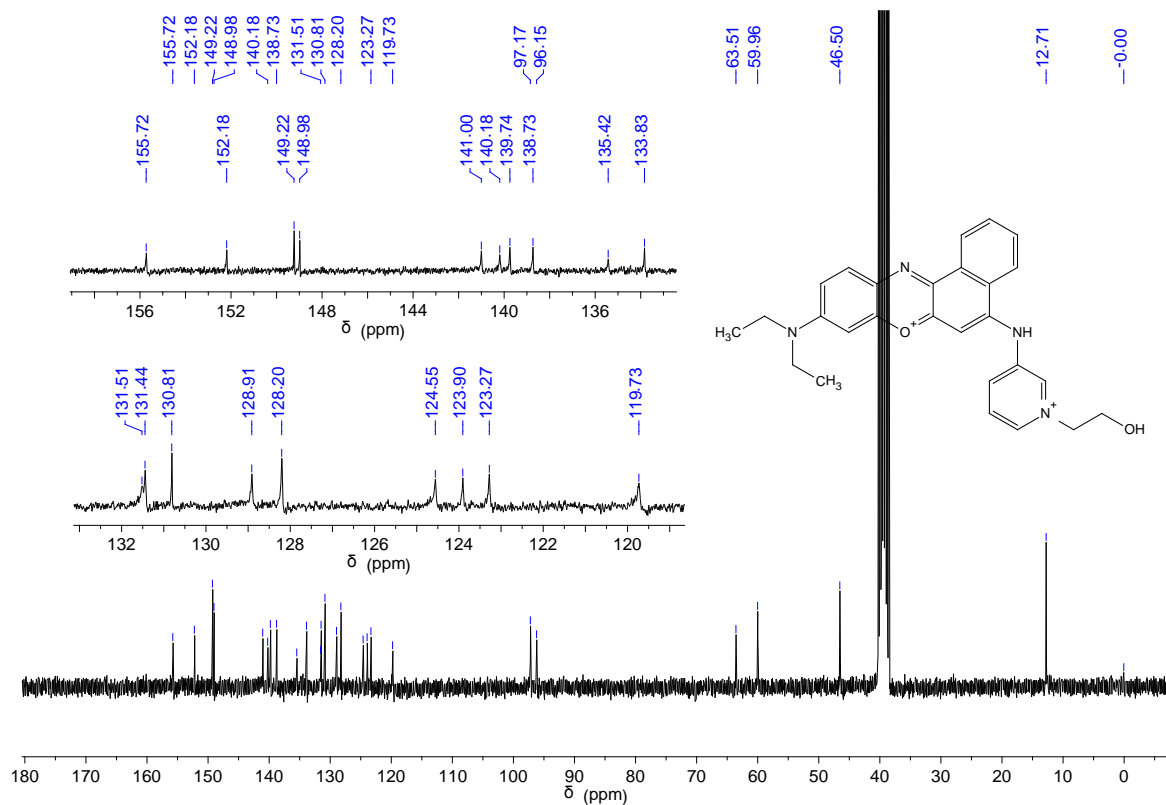


Fig. S41 ¹³C NMR (75 MHz) of **3c+H⁺** in DMSO-d₆.

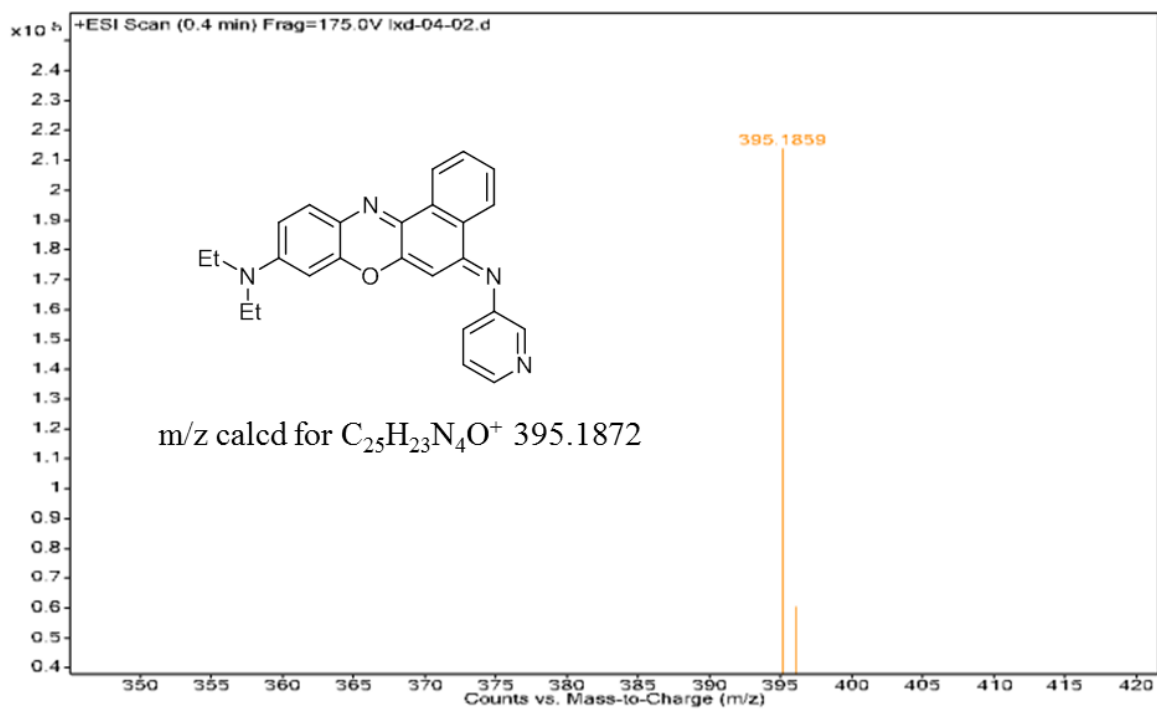


Fig. S42 HRMS of **2a**.

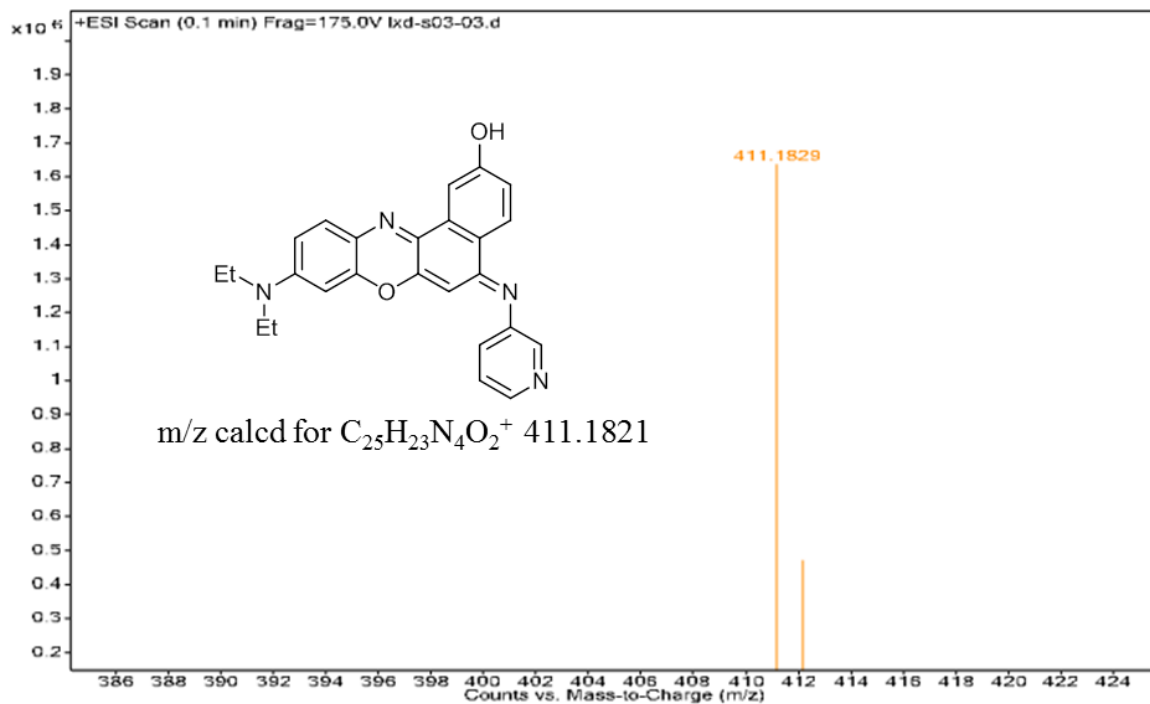


Fig. S43 HRMS of 2b.

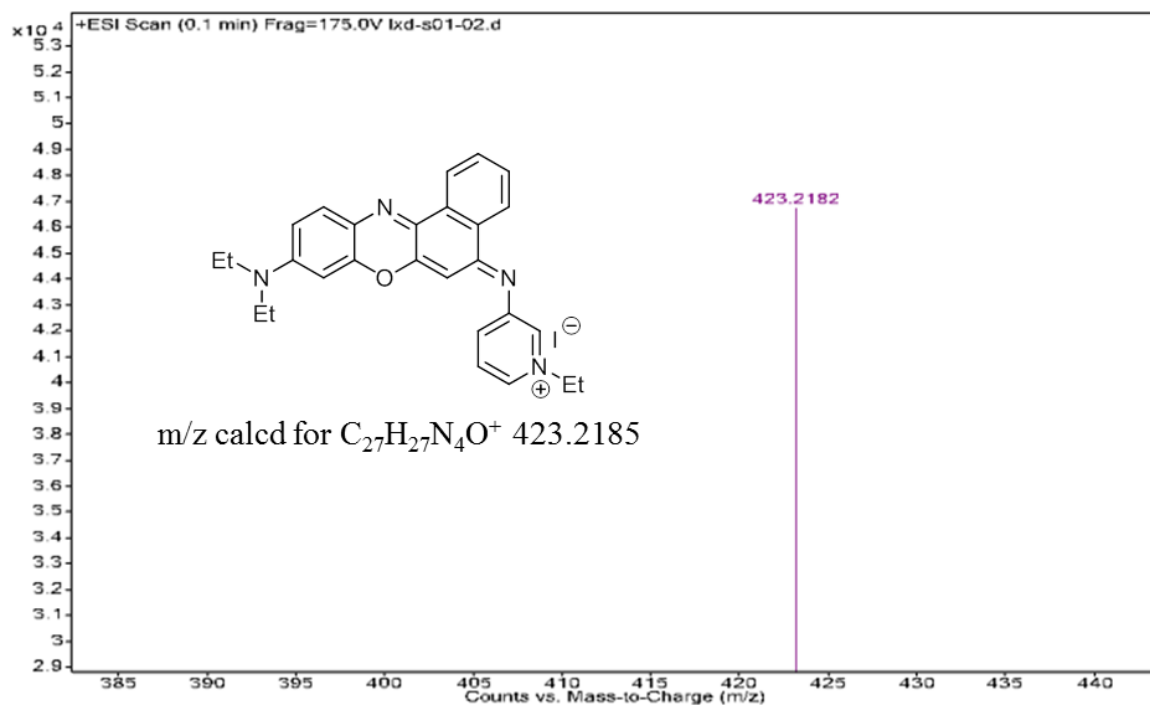


Fig. S44 HRMS of 3a.

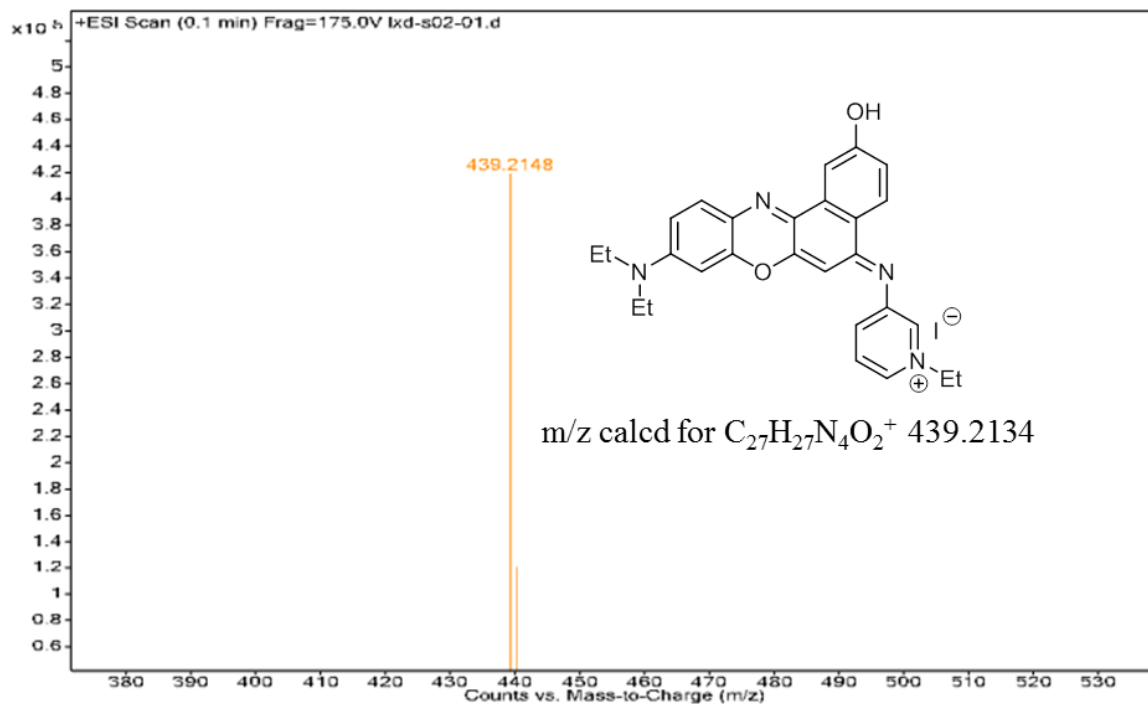


Fig. S45 HRMS of 3b.

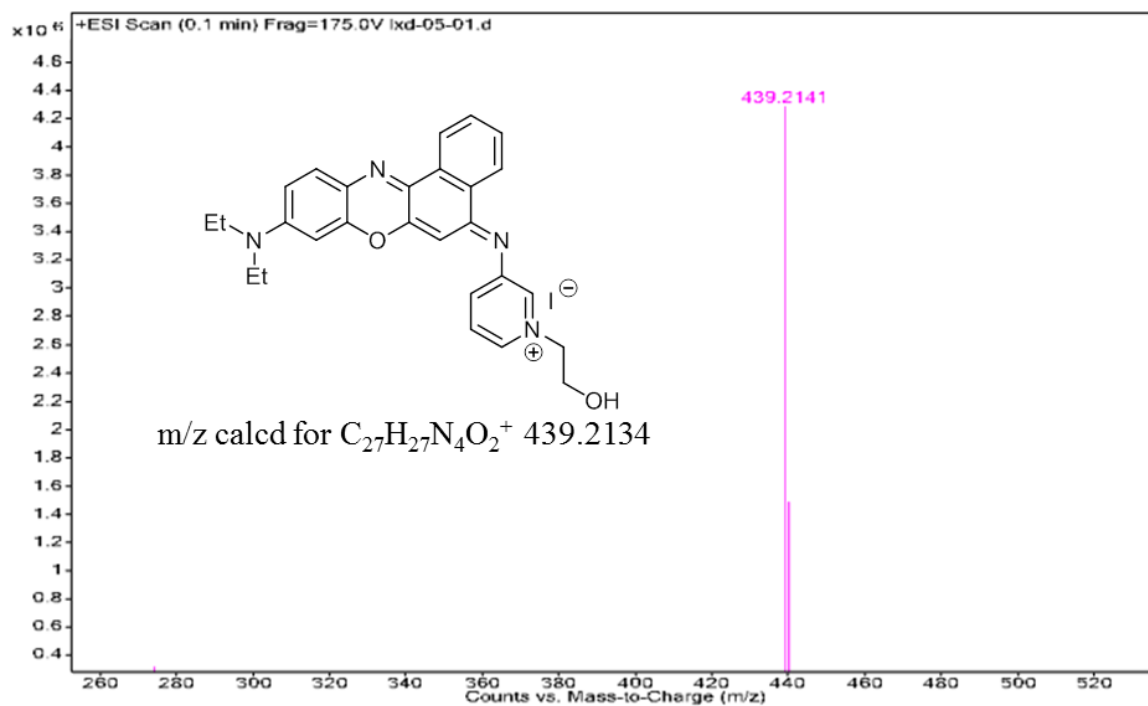


Fig. S46 HRMS of 3c.

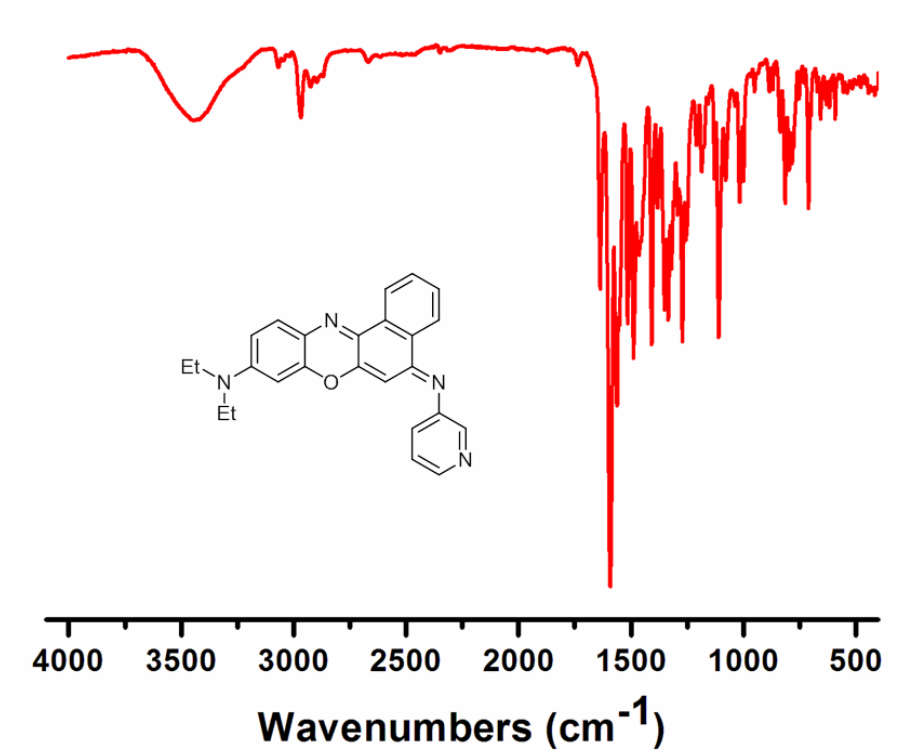


Fig. S47 FRTC of 2a.

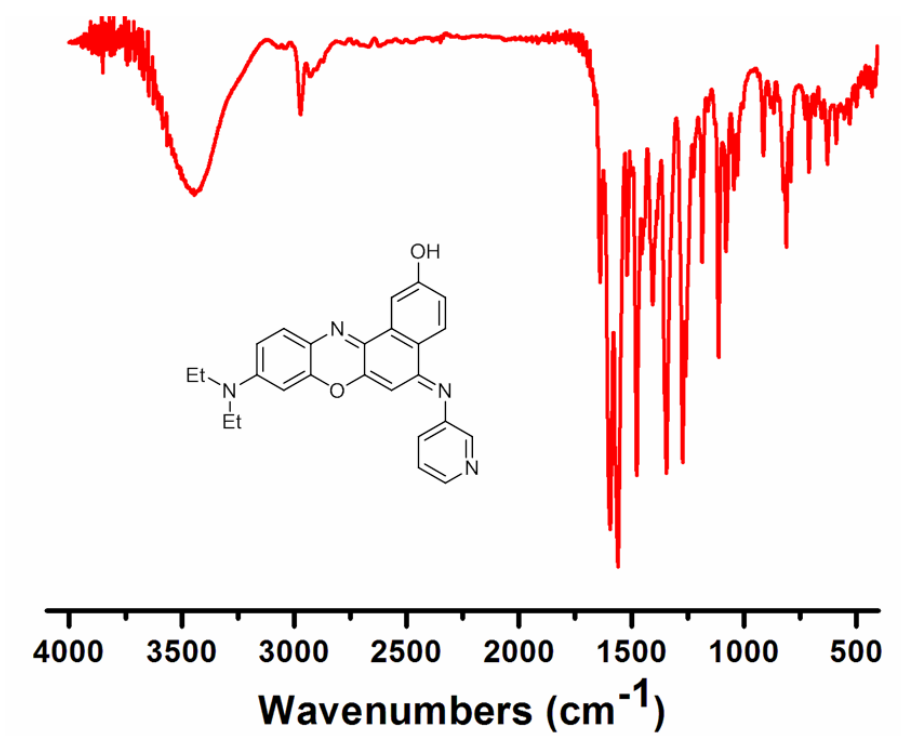


Fig. S48 FRTC of 2b.

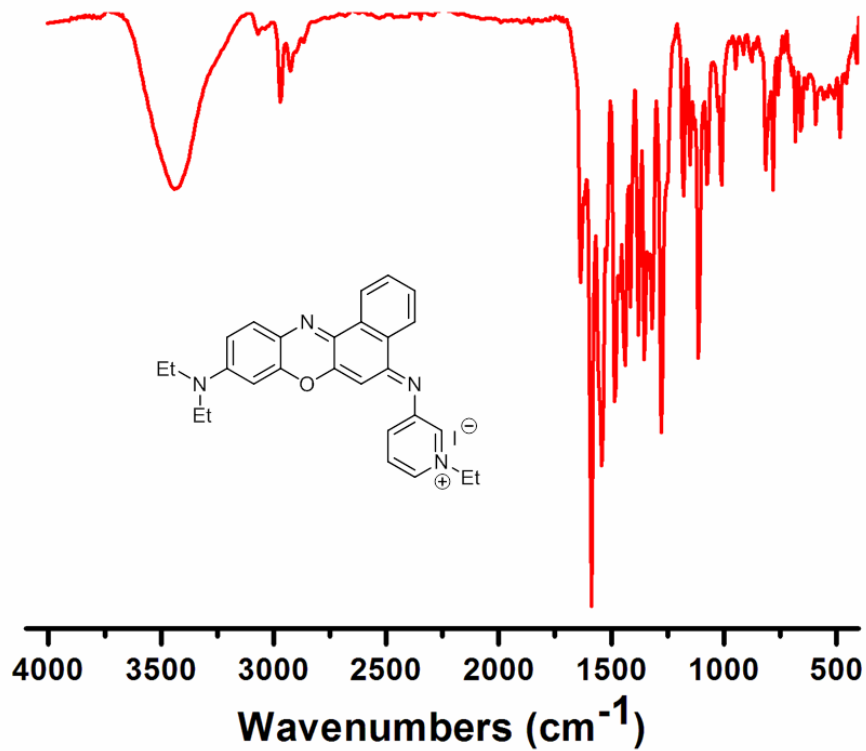


Fig. S49 FRTC of 3a.

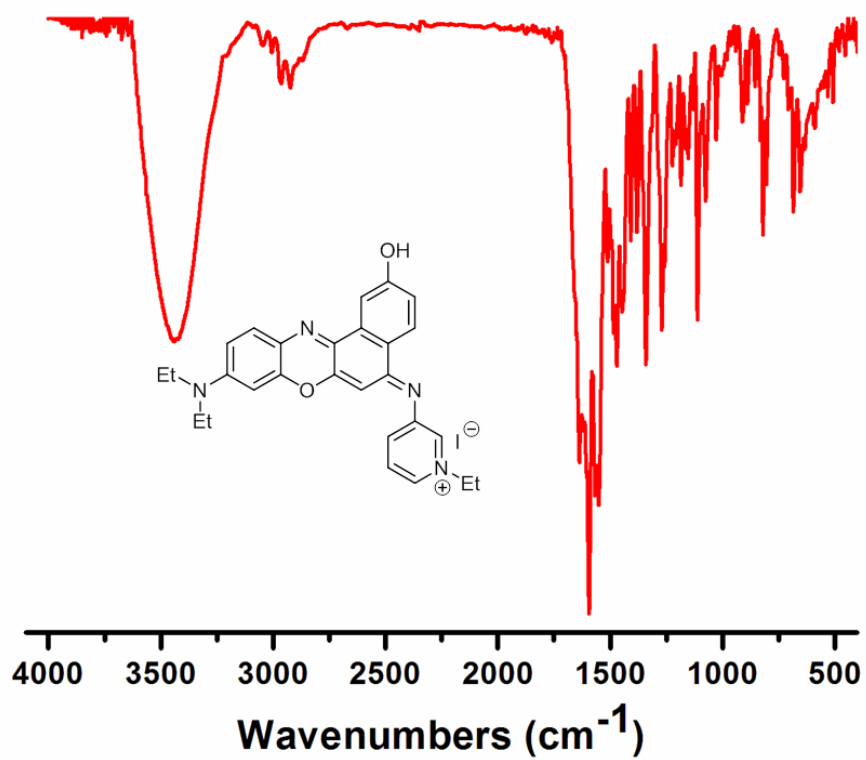


Fig. S50 FRTC of 3b.

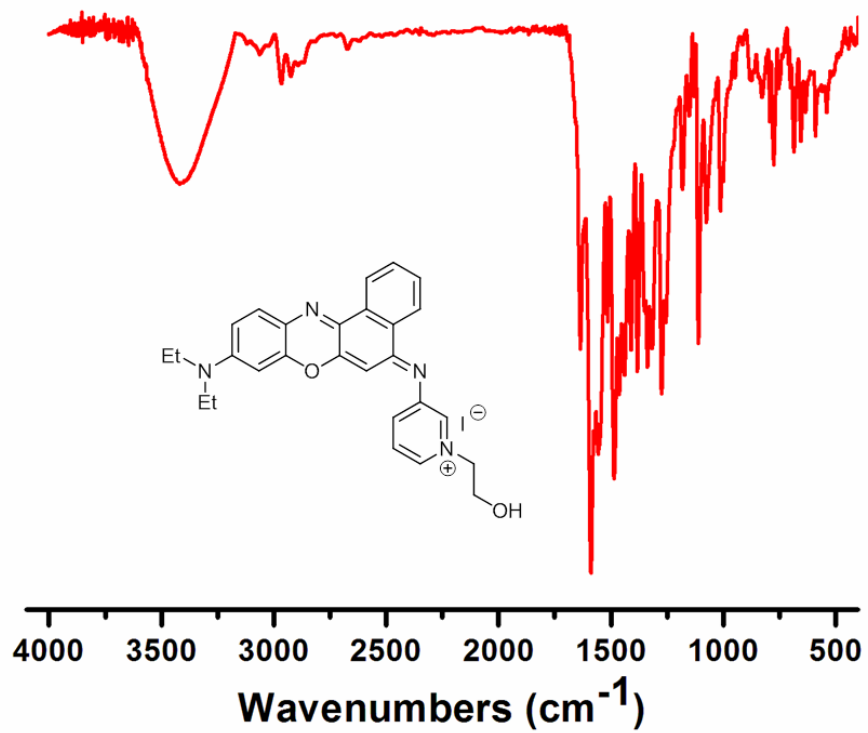


Fig. S51 FRTC of 3c.