Supporting Information:

## Synthesis of Cu-Sb-S Nanocrystals: insight into the mechanism of

## composition and crystal phase selection

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Figure S1. XRD patterns of nanocrystals synthesized using different amount of DDT (a) 3.6 mmol;
(b) 4 mmol; (c) 5.5 mmol; (d) 6.5 mmol; (e) 20.5 mmol; (f) 25.6 mmol with fixed Cu (Ac)<sub>2</sub>: SbCl<sub>3</sub> ratio of 1:1 and reacted at 200 °C for 1h.



Figure S2. SEM-EDX spectra of Cu<sub>3</sub>SbS<sub>4</sub>, Cu<sub>12</sub>Sb<sub>4</sub>S<sub>13</sub>, CuSbS<sub>2</sub> and Cu<sub>3</sub>SbS<sub>3</sub> nanocrystals



Figure S3. X-ray photoelectron spectra of as-prepared Cu<sub>3</sub>SbS<sub>4</sub> NCs.



Figure S4. X-ray photoelectron spectra of as-prepared Cu<sub>12</sub>Sb<sub>4</sub>S<sub>13</sub> NCs.



Figure S5. X-ray photoelectron spectra of as-prepared CuSbS<sub>2</sub> NCs.



Figure S6. X-ray photoelectron spectra of as-prepared Cu<sub>3</sub>SbS<sub>3</sub> NCs.



Figure S7. XRD patterns of nanocrystals synthesized with different Cu (Ac)<sub>2</sub>: SbCl<sub>3</sub> ratio. (a) 1:3; (b) 1:2; (c) 1:1; (d) 1:0.3.



Figure S8. (a) XRD patterns of as-prepared Cu<sub>2</sub>S nanocrystals before and after Sb<sup>3+</sup> injection. (b) TEM image of Cu<sub>2</sub>S seeds before Sb<sup>3+</sup> injection. (c) TEM image of Cu<sub>2</sub>S nanocrystals after Sb<sup>3+</sup> injection.



Figure S9. XRD pattern of nanocrystals synthesized with  $Cu_7S_4$  and  $Sb_2S_3$  as seeds and heated at 200 °C for 1h.



Figure S10. XRD pattern of  $Sb_2S_3$  nanocrystals.



Figure S11. XRD and TEM images of Cu<sub>2-x</sub>S nanocrystals with different reactivity of sulfur precursors: heating at (a,b) 150 °C; (c,d) 200 °C for 20 min. Both of the reactions were carried out at 200 °C for 1h with 1mL DDT.



Figure S12. UV-vis –NIR absorption spectroscopy of as-prepared (a)  $Cu_3SbS_4$ , (b)  $Cu_{12}Sb_4S_{13}$ , (c)  $CuSbS_2$ , and (d)  $Cu_3SbS_3$  nanocrystals.