Electronic Supplementary Information

Ionic Liquid-Assisted Solvothermal Synthesis of Three-Dimensional Hierarchical Copper Sulfide Microflowers at Low Temperature with Enhanced Photocatalytic Performance

Yingxue Cui,^a Caiying Wei,^a Jiaqin Yang,^a Jing Zhang^a and Wenjun Zheng^{*ab}

^aDepartment of Chemistry, and Key Laboratory of Advanced Energy Materials Chemistry (MOE), TKL of Metal and Molecule-based Material Chemistry, College of Chemistry, Nankai University, Tianjin, 300071, P. R. China.

E-mail: zhwj@nankai.edu.cn

^bCollaborative Innovation Center of Chemical Science and Engineering, Nankai University, Tianjin, 300071,

P. R. China.



Fig. S1 Structural view of (a) [BMIm]Cl and (b) [OMIm]Cl.



Fig. S2 High-magnification FE-SEM image of 3D hierarchical CuS microflowers (S-4) obtained in [BMIm]Cl-MeOH mixed solvent with 4.58 M [BMIm]Cl at 65 °C for 20 min.



Fig. S3 FE-SEM images of the samples obtained at different reaction durations in [BMIm]Cl-MeOH mixed solvent with 4.58 M [BMIm]Cl at 65 °C: (a) 1 min, (b) 3 min, and (c) 10 min.



Fig. S4 (a) EDS and (b-c) typical XPS spectra of the sample obtained in [BMIm]Cl-MeOH mixed solvent with 4.58 M [BMIm]Cl at 65 °C for 10 min: (b) survey spectra (c) Cu_{2p} region, and (d) S_{2p} region.

Table S1 The atomic ratios of Cu and S elements from the EDS and XPS spec
--

	EDS	XPS
Cu At%	49.05	48.98
S At%	50.95	51.02



Fig. S5 XRD patterns of samples obtained in different molar concentration of [BMIm]Cl at 65 °C for 20 min: (a) S-0, (b) S-1, (c) S-2, and (d) S-3.



Fig. S6 (a) XRD and (b) FE-SEM image of compact flower-like CuS superstructures obtained in [OMIm]Cl-MeOH mixed solvent with 4.58 M [OMIm]Cl at 65 °C for 20 min.



Fig. S7 Time-dependent UV-Vis absorption spectra applying 3D hierarchical CuS microflowers (S-4) obtained in [BMIm]Cl-MeOH mixed solvent with 4.58 M [BMIm]Cl at 65 °C for 20 min as photocatalyst.



Fig. S8 Visible-light irradiation photocatalytic degradation curves for MB (10 ppm) in the first three cycles applying 3D hierarchical CuS microflowers (S-4) obtained in [BMIm]Cl-MeOH mixed solvent with 4.58 M [BMIm]Cl at 65 °C for 20 min as photocatalyst.

materials	catalysts (mg)	MB concentration (ppm)	MB volume (mL)	time (min)	photodegradation degree (ξ , %)	reference
3D hierarchical CuS microflowers	20	10	100	25	99	this work
CuS hierarchical structures	30	20	40	90	87	S 1
CuS plates	10	3	30	30	80	S2
CuS nanoparticles	20	0.16	50	90	94	S3
hollow CuS microspheres	20	11.3	106	48	74	S4
CuS microflowers	5	10	30	25	98	S5
CdS hollow nanospheres	50	14	50	60	87	S6
Fe ₂ O ₃ /ZnO hollow spheres	50	5	100	50	95.2	S 7

Table S2 Comparison of photocatalytic activity of different materials for degradation of MB.

Table S3 The total organic carbon (TOC) results of MB.

Irradiation Time	TOC concentration (mg L ⁻¹)
Before degradation (-60 min)	8.2570
After degradation (25 min)	0.4885

Supplementary references

- S1 F. Li, J. Wu, Q. Qin, Z. Li and X. Huang, *Powder Technol.*, 2010, **198**, 267.
- M. Basu, A. K. Sinha, M. Pradhan, S. Sarkar, Y. N. Govind and T. Pal, *Environ. Sci. Technol.*, 2010, 44, 6313.
- S3 A. K. Sahoo and S. K. Srivastava, J. Nanopart. Res., 2013, 15, 1.
- S4 M. Tanveer, C. Cao, Z. Ali, I. Aslam, F. Idrees, W. S. Khan, F. K. But, M. Tahir and N. Mahmood, *CrystEngComm*, 2014, **16**, 5290.
- S5 Z. K. Yang, L. X. Song, Y. Teng and J. Xia, J. Mater. Chem. A, 2014, 2, 20004.
- S6 G. Lin, J. Zheng and R. Xu, J. Phys. Chem. C, 2008, 112, 7363.
- S7 Y. Liu, L. Yu, Y. Hu, C. Guo, F. Zhang and X. W. (David) Lou, *Nanoscale*, 2012, 4, 183.