Electronic Supplementary Information to:

C_{β} -H stretching vibration as a new probe for conformation of n-propanol in gaseous and liquid States

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Figure S1. Geometry structures and labeling of atoms of n-propanol in the five conformers. The plane in n-propanol refers to C1-C5-C8-O11.

	Gg			T _g			G _g '					G _t	T _t		
Mode	$\upsilon_{cal}^{\ a}$	Activity	PED(%) ^b	Ucal ^a	Activity	PED(%) ^b	Ucal ^a	Activity	PED(%) ^b	U _{cal} ^a	Activity	PED(%) ^b	U _{cal} ^a	Activity	PED(%) ^b
Description		(ρ)			(ρ)			(ρ)			(ρ)			(ρ)	
β -CH ₂ bending	1442.9	7.7	958(C5H ₂)	1456.6	7.5	958(C5H ₂)	1437.8	7.7	95δ(C5H ₂)	1439.4	7.8	958(C5H ₂)	1459.9	7.7	958(C5H ₂)
β-CH ₂ -SS	2918.4	135.2	76v(C5H7)+ 23v(C5H6)	2923.1	118.8	88v(C5H7)+12v(C5H6)	2936.1	142.4	72u(C5H7)+28u(C5H6)	2939.9	134.7	63v(C5H7)+36v(C5H6)	2949.9	129.9	50v(C5H7)+ 50v(C5H6)
		(0.08)			(0.10)			(0.08)			(0.07)			(0.06)	
β-CH ₂ -AS	2951.6	85.3	76u(C5H6)-24u(C5H7)	2972.3	80.1	87u(C5H6)-12u(C5H7)	2974.1	77.8	71u(C5H6)-28u(C5H7)	2973.1	75.3	63v(C5H6)-37v(C5H7)	2983.4	72.3	50v(C5H6) - 50v(C5H7)
		(0.59)			(0.49)			(0.66)			(0.7)			(0.75)	
OH stretching	3661.2	74.8	100v(OH)	3663.1	75.9	100u(OH)	3674.8	57.5	100v(OH)	3681.3	112.6	100u(OH)	3678.7	125.6	100v(OH)
		(0.21)			(0.21)			(0.18)			(0.25)			(0.26)	

Table S1. The calculated vibrational frequencies (cm⁻¹) and Raman activities and depolarization ratio (ρ) as well as potential energy distribution (PED) analysis for five conformers of CD₃CH₂CD₂OH in the C-H bending (1400-1500 cm⁻¹) and C-H stretching (2800-3100cm⁻¹) and OH stretching (3600-3700cm⁻¹) regions.

^a Scale factors of 0.973 for C-H bending and stretching regions and of 0.956 for O-H stretching region. ^b potential energy terms higher than 5% are included.

The following symbols stand for: v - stretching; δ - CH₂ bending.

			Gg	Tg			G _g ,					G _t	T _t		
Mode	U _{cal} ^a	Raman	PED(%) ^b	U _{cal} ^a	Raman	PED(%) ^b	U _{cal} ^a	Raman	PED(%) ^b	U _{cal} ^a	Raman	PED(%)	Ucal ^a	Raman	PED(%) ^b
		activity			activity			activity			activity			activity	
β-CH ₂ bending	1441.6	0.6	868 (C5H ₂)	1449.6	18.6	70δ (C5H ₂)+ 18β ₃ (C1H ₃)	1437.1	10.2	898 (C5H ₂)	1438.1	10.7	908 (C5H ₂)	1453.1	18.0	66δ (C5H ₂)+26β ₂ (C1H ₃)
						+8δ (C8H ₂)									
out-of-plane	1454.1	12.4	71β ₂ (C1H ₃)+17δ (C8H ₂)	1460.8	7.2	89β ₂ (C1H ₃)	1470.3	2.2	52β ₂ (C1H ₃)-23β ₃ (C1H ₃) -	1454.5	11.0	78β ₂ (C1H ₃)+6δ(C8H ₂)+	1461.3	7.5	91β ₂ (C1H ₃)
γ -CH ₃ bending									168 (C8H ₂)			6β ₃ (C1H ₃)			
in-plane	1467.9	5.3	61β ₃ (C1H ₃)-22δ (C8H ₂)	1462.7	0.3	52 _{β3} (C1H ₃)-33δ(C8H ₂)	1456.2	10.3	37β ₂ (C1H ₃)+28δ (C8H ₂)	1468.4	4.0	$77\beta_3(C1H_3)-7\beta_2(C1H_3)$	1466.6	0.3	52β ₃ (C1H ₃)-19δ (C5H ₂)
γ -CH ₃ bending						-5δ (C5H ₂)			+25β ₃ (C1H ₃)						-19δ (C8H ₂)
α-CH ₂	1466.7	0.6	61δ (C8H ₂)-16β ₂ (C1H ₃)+	1475.4	3.2	57δ(C8H ₂)-	1463.3	6.5	55δ (C8H ₂) -34β ₃ (C1H ₃)	1476.8	4.8	886 (C8H ₂)	1482.6	4.5	74δ (C8H ₂)-13δ (C5H ₂)
bending			14β ₃ (C1H ₃)			228(C5H ₂)+13β ₃ (C1H ₃)									+5β ₂ (C1H ₃)
α-CH ₂ -SS	2907.2	69.1	69v(C8H10)- 17v(C5H6)-	2898.2	115.8	95v(C8H10)	2897.3	126.5	95u(C8H9)	2891.2	130.5	70u(C8H9)+29u(C8H10)	2889.2	127.1	50v(C8H10)+49v(C8H9)
			10u(C5H7)												
α-CH ₂ -AS	2988.5	57.6	86v(C5C8) -5v(C8H10)	2989.7	30.2	66v(C8H9)+15v(C1H3)-	2999.2	91.0	80v(C8H10)+6v(C1H2) -	2920.2	103.9	59v(C8H10)-26v(C8H9)	2914.9	105.3	48v(C8H9)-47v(C8H10)
						14u(C1H2)			5v(C1H4)			-9u(C5H6)-5u(C5H7)			
γ-CH ₃ -SS	2943.4	121.1	47u(C1H3)+25u(C1H4)+	2940.6	163.8	41v(C1H3)+ 36v(C1H2)+	2935.8	40.7	35u(C1H3)-21u(C5H7)+	2941.0	65.7	45u(C1H3)+25u(C1H4)-	2938.9	204.1	40v(C1H2)+39v(C1H3)
			13u(C1H2)-11u(C5H7)			21u(C1H4)			21u(C1H2)+16u(C1H4)+			15v(C5H7)+11v(C1H2)			+18v(C1H4)
									-6u(C5H6)						
β-CH ₂ -SS	2920.1	222.5	62v(C5H6)+22v(C8H10)	2921.4	153.7	87u(C5H7)+ 10u(C5H6)	2934.2	254.8	49u(C5H7)+22u(C5H6)+	2942.2	249.2	44u(C5H7)+26u(C5H6)	2948.8	80.9	48u(C5H7)+48u(C5H6)
			+9v(C5H7)+ 5v(C8H9)						13v(C1H3)+13v(C1H2)			+13v(C1H3)+9v(C8H10)			
β-CH ₂ -AS	2950.2	159.5	66u(C5H7)-	2966.1	112.0	73v(C5H6)-10v(C8H9) -	2968.5	111.2	54u(C5H6)-20u(C5H7)-	2971.0	86.5	54v(C5H6)-32v(C5H7)	2976.2	99.5	29v(C5H6)-29v(C5H7)
			14v(C5H6)+ 8v(C1H2)			6v(C5H7)			14v(C1H2)+9v(C1H3)			-6u(C5H7)-5u(C1H4)			+19v(C1H3)-19v(C1H2)
out-of-plane	3001.4	66.8	48v(C1H4) -42v(C1H3)	3001.9	43.3	31u(C1H2)-	3004.5	95.8	75u(C1H4)-12u(C1H3) -	3023.2	38.6	78v(C1H2)-14v(C1H4)	3001.1	5.38	31u(C1H2)-30u(C1H3)
γ-CH ₃ -AS						30v(C1H3)+18v(C8H9)			5v(C5H6)						-19v(C5H7)+19v(C5H6)

Table S2. The calculated vibrational frequencies (cm⁻¹) and Raman activities as well as potential energy distribution (PED) analysis for five conformers of $CH_3CH_2CH_2OH$ in the C-H bending (1400-1500 cm⁻¹) and C-H stretching (2800-3100 cm⁻¹) and OH stretching (3600-3700 cm⁻¹) regions.

in-plane	3018.3	41.4	74u(C1H2)-18u(C1H4) -	3008.0	80.7	77u(C1H4)-12u(C1H2)	2994.6	14.5	40u(C1H2)-	2998.9	63.2	54u(C1H4)-36u(C1H3)	3007.6	82.0	78v(C1H4)-10v(C1H3)
γ -CH ₃ -AS			5u(C1H3)			+10 v(C1H3)			31u(C1H3)+11u(C5H6)-			+8v(C5H6)			-10v(C1H2)
									9v(C8H10)+ -8v(C5H7)						
OH stretching	3724.1	73.3	100v(O11H12)	3725.8	74.5	100u(O11H12)	3737.5	56.1	100v(OH)	3744.2	112.6	100u(OH)	3741.5	125.6	100v(O11H12)

^{*a*} Scale factors of 0.973 for C-H bending and stretching regions and of 0.956 for O-H stretching region. ^{*b*} potential energy terms higher than 5% are included.

The following symbols stand for: v - stretching; δ - CH₂ bending; β_1 - CH₃ umbrella bending; β_2 - CH₃ out-of-plane bending; β_3 - CH₃ in-plane bending; the plane denotes the atoms of C1-C5-C8-O11 in CH₃CH₂CH₂OH.



Figure S2. The temperature-dependent Raman spectra of liquid CD₃CH₂CD₂OH in pure liquid state fitted with four Voigt profiles.



Figure S3. The temperature-dependent Raman spectra of liquid CD₃CH₂CD₂OH in 5% water solution fitted with four Voigt profiles.