

### Supporting Information

Nr.	Atom	X	Y	Z
1	C	-1.199125	0.195487	0.007724
2	C	-1.204282	-1.197737	0.007724
3	C	-0.000000	-1.898098	0.006451
4	C	1.204282	-1.197737	0.007724
5	C	1.199125	0.195487	0.007724
6	C	-0.000000	0.912818	0.005155
7	H	-2.139587	0.735525	0.012573
8	H	-2.145145	-1.733580	0.012494
9	H	-0.000000	-2.980597	0.009400
10	H	2.145145	-1.733580	0.012494
11	H	2.139587	0.735525	0.012573
12	C	-0.000000	2.415831	-0.030217
13	H	0.882490	2.820789	0.462959
14	H	-0.000000	2.778826	-1.059173
15	H	-0.882490	2.820789	0.462959

Table S1. Optimized geometry data of toluene by MP2/6-311++G(3df,2p) in Cartesian coordinates are shown in unit of angstrom. Number of atoms and Cartesian coordinates are shown in Fig. S1.

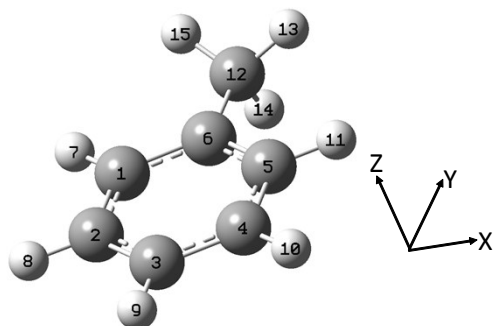


Fig. S1. Geometrical structure of toluene ( $C_7H_8$ ).

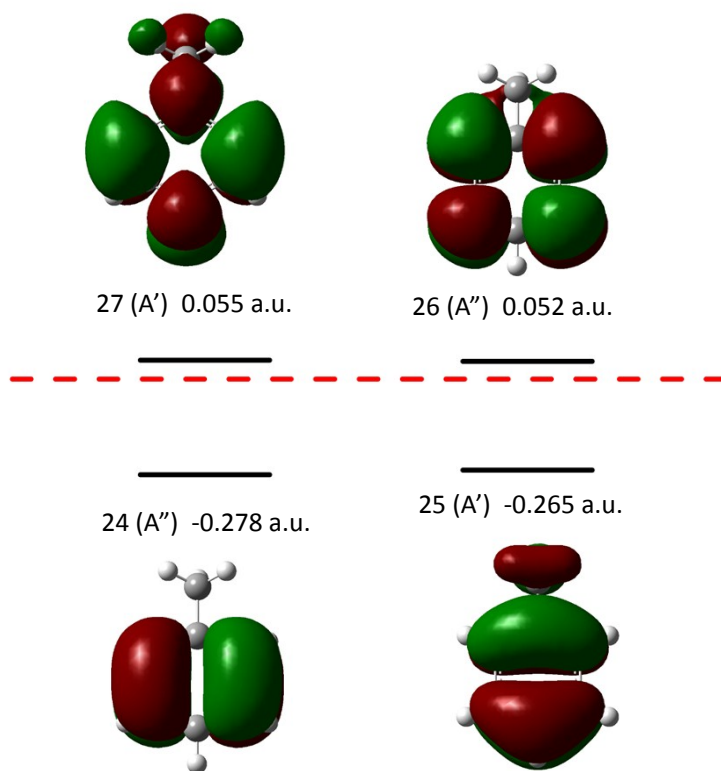


Fig. S2. Active orbitals with orbital energies, which were obtained by the CASSCF calculation, are shown.

CASSCF wavefunctions:

$\Phi_{S_0}(A') \equiv |\phi_{24}\bar{\phi}_{24}\phi_{25}\bar{\phi}_{25}|$  for the ground state configuration.  $\Psi_{S_0}(A') = 0.975\Phi_{S_0(A')} + L$  for the ground state.

The lower four singlet excited states are approximately expressed as a linear combination of the one-electron excited configurations as

$$\Psi_{S_1(A')} = 0.524 \left( |\phi_{24}\bar{\phi}_{24}\phi_{25}\bar{\phi}_{26}| - |\phi_{24}\bar{\phi}_{24}\bar{\phi}_{25}\phi_{26}| \right) + 0.469 \left( |\phi_{24}\bar{\phi}_{27}\phi_{25}\bar{\phi}_{25}| - |\bar{\phi}_{24}\phi_{27}\phi_{25}\bar{\phi}_{25}| \right)$$

$$\Psi_{S_2(A')} = 0.511 \left( |\phi_{24}\bar{\phi}_{24}\phi_{25}\bar{\phi}_{27}| - |\phi_{24}\bar{\phi}_{24}\bar{\phi}_{25}\phi_{27}| \right) + 0.487 \left( |\phi_{24}\bar{\phi}_{26}\phi_{25}\bar{\phi}_{25}| - |\bar{\phi}_{24}\phi_{26}\phi_{25}\bar{\phi}_{25}| \right)$$

$$\Psi_{S_3(A')} = -0.458 \left( |\phi_{24}\bar{\phi}_{24}\phi_{25}\bar{\phi}_{26}| - |\phi_{24}\bar{\phi}_{24}\bar{\phi}_{25}\phi_{26}| \right) + 0.513 \left( |\phi_{24}\bar{\phi}_{27}\phi_{25}\bar{\phi}_{25}| - |\bar{\phi}_{24}\phi_{27}\phi_{25}\bar{\phi}_{25}| \right)$$

$$\Psi_{S_4(A')} = -0.473 \left( |\phi_{24}\bar{\phi}_{24}\phi_{25}\bar{\phi}_{27}| - |\phi_{24}\bar{\phi}_{24}\bar{\phi}_{25}\phi_{27}| \right) + 0.498 \left( |\phi_{24}\bar{\phi}_{26}\phi_{25}\bar{\phi}_{25}| - |\bar{\phi}_{24}\phi_{26}\phi_{25}\bar{\phi}_{25}| \right)$$

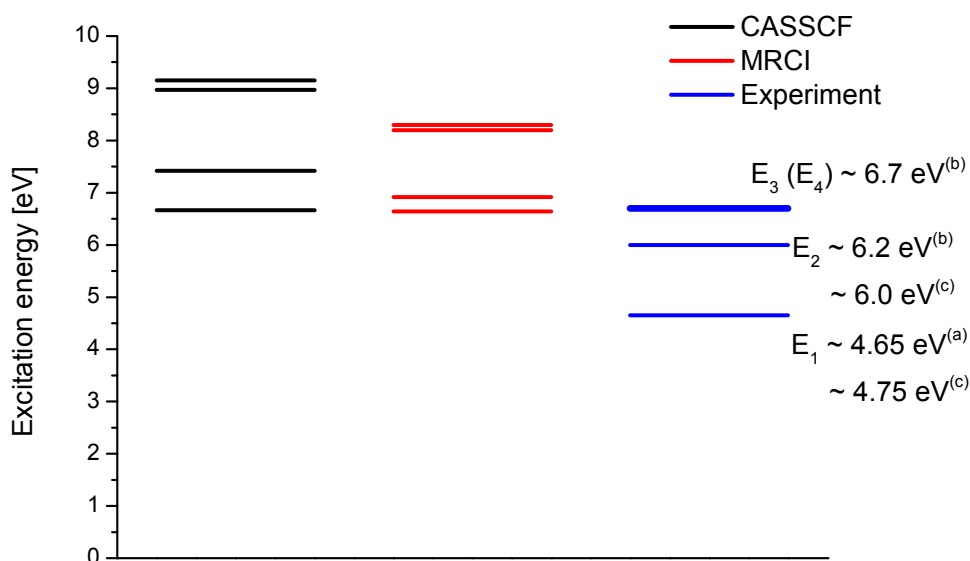


Fig. S3. Comparison of the excitation energies between the computational levels:

CASSCF/6-311++G(3df,2p), MRCI/6-311++G(3df,2p) and experimental data<sup>(a), (b), (c)</sup>.

Table S2. Data of the excitation energies calculated by CASSCF/6-311++G(3df,2p), MRCI/6-311++G(3df,2p) and experimental data<sup>(a), (b), (c)</sup>.

	E <sub>1</sub> [eV]	E <sub>2</sub> [eV]	E <sub>3</sub> [eV]	E <sub>4</sub> [eV]
CASSCF(4, 4)	6.63	7.42	8.97	9.15
MRCI	6.64	6.91	8.20	8.30
Exp.	4.65 <sup>(a)</sup> 4.75 <sup>(c)</sup>	6.2 <sup>(b)</sup> 6.0 <sup>(c)</sup>	6.7 <sup>(b)</sup>	6.7 <sup>(b)</sup>

## References

- (a) N. Ginsburg, W.W. Robertson, and F.A. Matsen, *J. Chem. Phys.*, 9, 511 (1946).
- (b) M.B. Robin, *Higher Excited states of Polyatomic Molecules*. (Academia New York, 1975), Vol. 2.
- (c) B.D. Mistry, *A Handbook of Spectroscopic Data Chemistry (UV, IR, PMR, 13CNMR and Mass Spectroscopy)*, Oxford Book Company, Jaipur, India, 2009). On pp. 16-17 4.75 eV for S<sub>1</sub>, and 6.0 eV for S<sub>2</sub> in Methanol (2%). There is no data for S<sub>3</sub> and S<sub>4</sub>.