

The effect of GME topology on the multicomponent adsorption in ZIFs

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Supporting Information

Table 1: atom atom interaction parameters

atom	ZIF atoms		atom	adsorbate atoms		
	$\sigma(A)$	$\varepsilon/k_B(K)$		$\sigma(A)$	$\varepsilon/k_B(K)$	q(e)
Carbon	3.43	40.27	Carbon CO_2^a	2.76	28.13	0.576
Oxygen	3.12	21.75	Oxygen CO_2^a	3.03	80.51	-0.288
Hydrogen	2.57	14.14	Nitrogen N_2^b	3.31	36.00	-0.482 (0.964) ^{com}
Nitrogen	3.26	22.72	Oxygen O_2^c	3.01	49.05	-0.123 (0.246) ^{com}
Zinc	2.46	53.4	CH_4	3.73	148.00	
Cl	3.52	99.23	$O(H_2O)^d$	3.15	76.58	-0.834 (0.417) ^H
Br	3.73	126.31				

^a C-O distance (A) :1.18; ^b N-N distance (A) :1.1; ^c O-O distance (A) :1.21;

^d O-H distance (A) :0.9572, H-O-H angle 104.52°; ^{com} center of mass charge;

^H hydrogen atom charge;

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Table 2: Atomic charges (e) on $Im - Zn - nIm$ fragments for ZIFs 68-78

ZIF68		ZIF69		ZIF70		ZIF78	
C1	0.225591	Zn1	0.7317525	C1	0.605978456	C1	-0.0106
C2	0.350384	O11	-0.478251	C2	0.001007456	C2	0.128964
C3	0.179233	N2	-0.1558705	C3	0.013118456	C3	0.001442
C4	0.174222	N8	-0.3714605	C4	-0.056530044	C4	-0.219686
C6	-0.1565715	N98	0.731731	H1	0.111105956	C5	-0.121936
C8	-0.143245	C6	0.061975	H2	0.046464456	C6	0.043451
H1	-0.015756	C4	-0.074958	H3	0.073645456	C7	-0.015263
H4	0.013269625	C3	0.092504	N1	0.497581456	C8	0.258017
H6	0.0190095	C13	0.092504	N2	-0.604204044	C9	-0.021719
H8	0.128247	C89	0.350589	N3	-0.201820044	H1	0.1944375
N1	-0.284354	C14	-0.17425	O1	-0.336702544	H2	0.050911
N2	-0.3847325	C58	-0.078663	Zn1	0.863852789	H3	-0.010385
N5	0.588478	C85	0.032752			H4	0.083533
O1	-0.513472	C82	0.0008315			H5	0.152612
Zn1	0.78809075	H86	0.0677			N1	0.549019
		H10	0.08801225			N2	0.637037
		H17	0.100886			N3	-0.1615285
		H15	0.02653			N4	-0.335407
		H5	0.025733			N5	-0.142997
		C17	-0.153309			O1	-0.469039
						O2	-0.417138
						Zn1	0.690143

Table 3: Atomic charges (e) on $Im - Zn - nIm$ fragments for ZIFs 79-82

	ZIF79	ZIF80	ZIF81	ZIF82
				
C1 0.032906518	C1 0.006586474	Br1 -0.05360366	C1 0.025438825	
C10 -0.060596982	C2 -0.112881026	C1 -0.00844216	C2 0.040290825	
C2 0.022409018	C3 0.398487974	C2 0.08614634	C3 -0.086898175	
C3 -0.022760982	C4 0.032632474	C3 -0.15867266	C4 0.019024825	
C4 -0.294764982	Cl1 -0.104059276	C4 -0.18855966	C5 0.346642825	
C5 -0.262450982	H1 0.160282974	C5 -0.09332866	C6 0.330535825	
C6 0.142721018	H2 0.074763974	C6 0.03654734	H1 0.122317325	
C7 0.246174018	N1 0.428308974	C7 0.33524434	H2 0.092760825	
C8 -0.040575982	N2 -0.390102026	C8 0.09322034	H3 0.053276825	
C9 0.144435018	N3 -0.051962026	C9 0.14520034	N1 0.549978825	
H1 0.100044018	O1 -0.396090026	H1 0.07414009	N2 -0.369172675	
H10 0.033027185	Zn1 0.696742974	H2 0.12699134	N3 -0.128532175	
H2 0.075918018		H3 0.05665434	N4 -0.498546175	
H3 0.009615018		H4 0.12364134	O1 -0.413424675	
H4 0.037820018		H5 0.11818534	Zn1 0.679680325	
H5 0.164021018		N1 0.72366234		
N1 0.525830018		N2 -0.38018566		
N2 -0.315941482		N3 -0.23377816		
N3 -0.119002982		O1 -0.48827466		
O1 -0.423880482		Zn1 0.72175234		
Zn1 0.664874018				

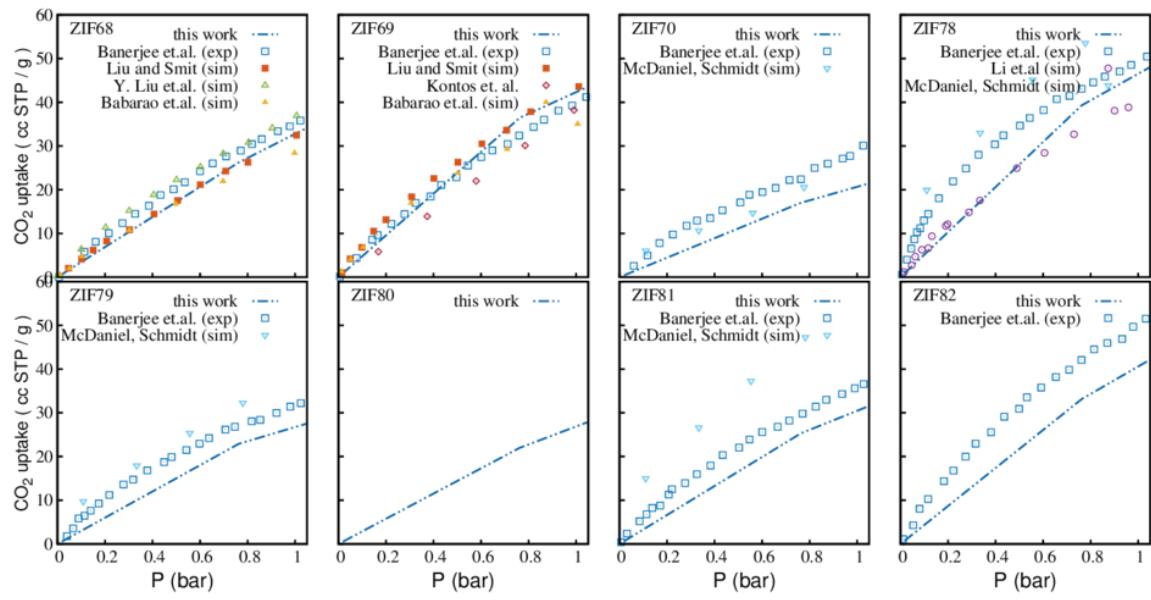


Figure 1: CO_2 adsorption isotherms computed at 298K in this work, compared with a set of experimental and simulated isotherms provided in the literature. The citing articles are included in the reference list of the manuscript

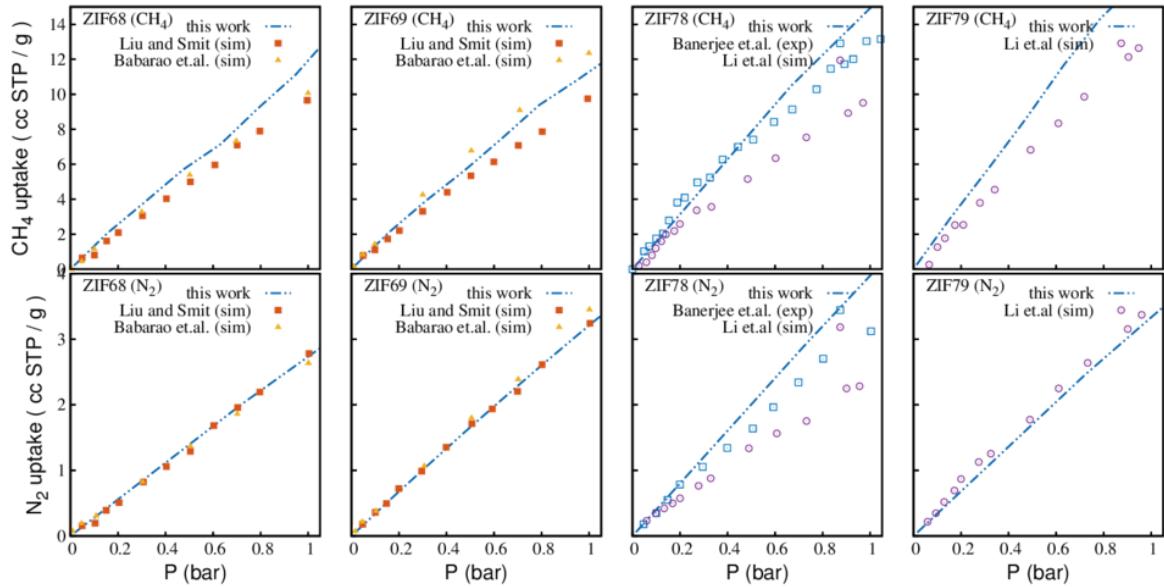


Figure 2: CH_4 and N_2 adsorption isotherms computed at 298K in this work, for ZIF68, ZIF69, ZIF78 and ZIF79, compared with a set of experimental and simulated isotherms provided in the literature. The citing articles are included in the reference list of the manuscript

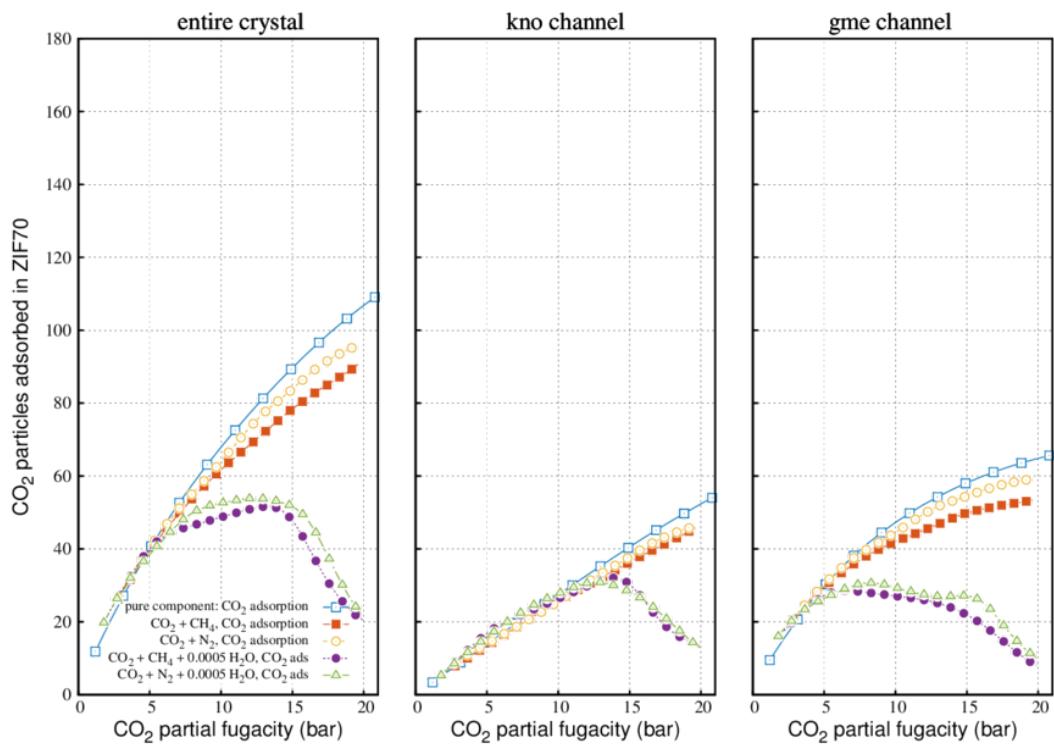


Figure 3: CO_2 adsorption in the unit cell, the *kno* and *gme* channels of ZIF70, for single and multicomponent systems.

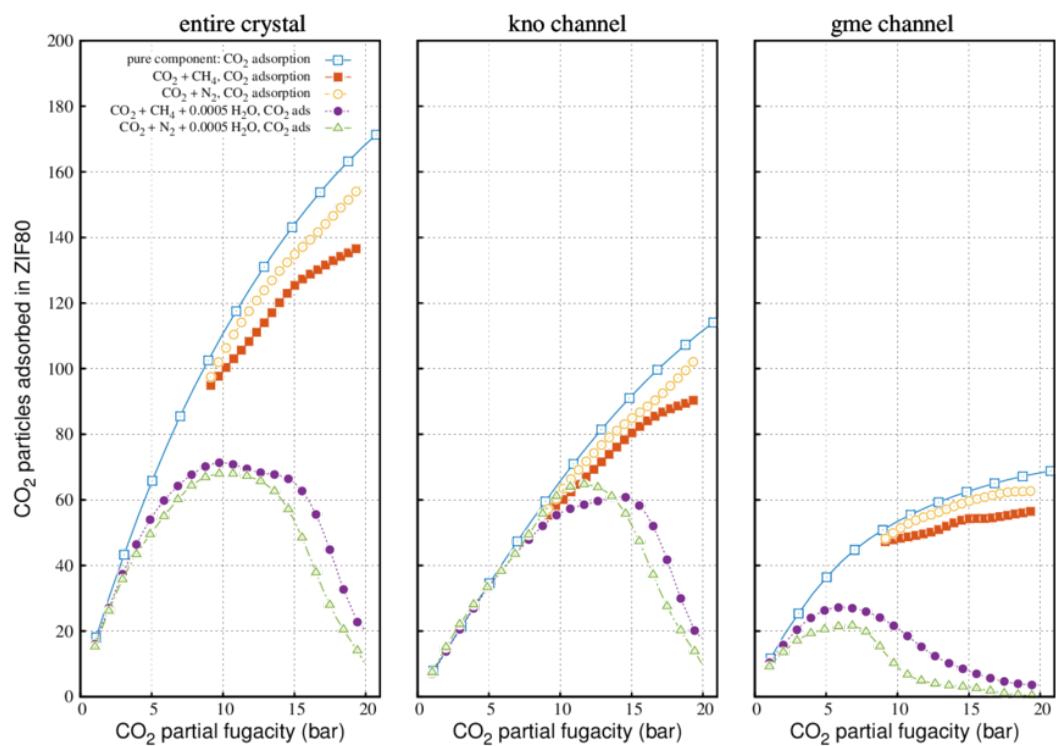


Figure 4: CO_2 adsorption in the unit cell, the *kno* and *gme* channels of ZIF80, for single and multicomponent systems.

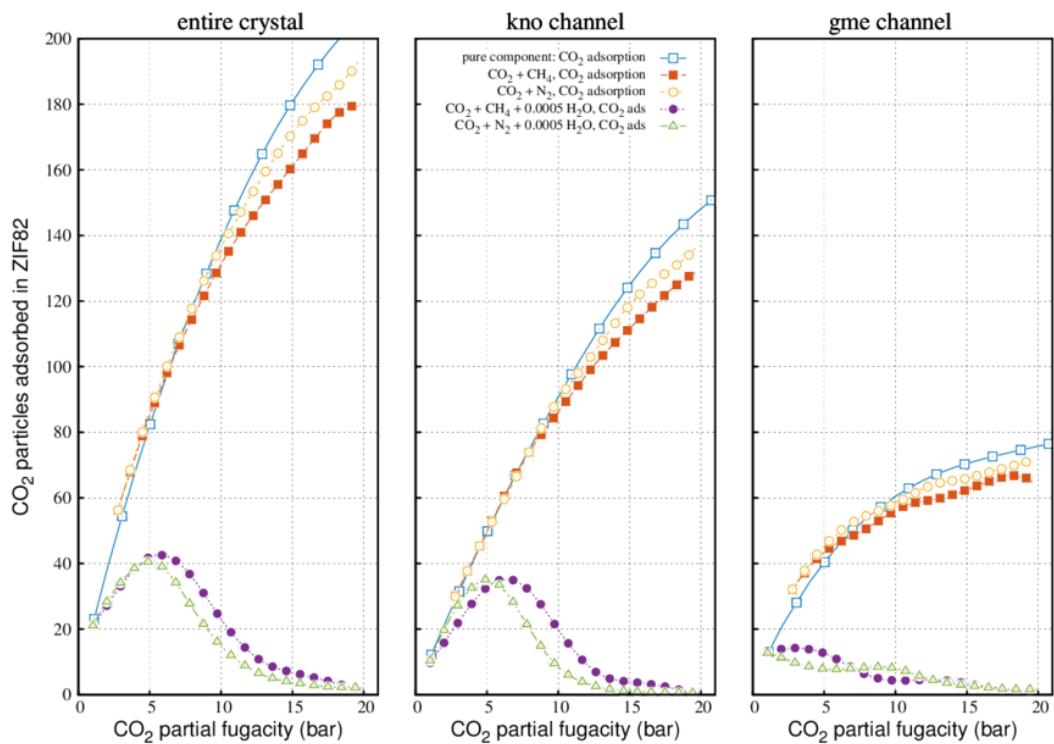


Figure 5: CO_2 adsorption in the unit cell, the *kno* and *gme* channels of ZIF82, for single and multicomponent systems.