## **Electronic Supplementary Information**

# Titanium Oxide Morphology Controls Charge Collection Efficiency in Quantum Dot Solar Cells

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#### Fluorescence decay analysis of TiO<sub>2</sub>/CdS films

The emission decay plots of CdS, TiO<sub>2</sub> NSHs/CdS, TiO<sub>2</sub> PNPs/CdS, TiO<sub>2</sub> NPs/CdS and TiO<sub>2</sub> NWs/CdS films (in Figure 2d) were fitted into bi-exponential functions based on  $\chi^2$  values. The emission average life times, and fitted parameters are tabulated in the Table S1.

$$f(t) = B_1 e^{-t/\tau_1} + B_2 e^{-t/\tau_2}$$
(1)  
$$\langle \tau \rangle = \Sigma B_i \tau_i^2 / \Sigma B_i \tau_i$$
(2)

In (1) and (2),  $B_i$  is the decay amplitude, t is the time after excitation,  $\tau$  is the electron lifetime and  $\langle \tau \rangle$  is the average electron lifetime.

Table S1 Kinetic parameters of emission decay analysis of photosensitizer films deduced from double exponential fits.

Sample	<b>B</b> <sub>1</sub>	$\tau_1(ns)$	B <sub>2</sub>	$\tau_2(ns)$	$<\tau>$ (ns)	$\chi^2$
CdS	19.46	0.00944	80.54	4.77	4.77	1.04
TiO <sub>2</sub> NSHs/CdS	97.82	0.00615	2.18	1.41	1.182	1.13
TiO <sub>2</sub> PNPs/CdS	99.38	0.00623	0.62	1.78	1.145	0.95
TiO <sub>2</sub> NPs/CdS	20.45	0.0379	79.55	1.14	1.135	0.97
TiO <sub>2</sub> NWs/CdS	19.64	0.00354	80.36	0.553	0.552	1.04

## Surface area and porosity studies

Brunauer-Emmett-Teller (BET) surface area analysis and Barrett-Joyner-Halenda (BJH) pore size and volume analysis were performed under Nitrogen at 77.3 K, after degassing at 300 °C for 3 h. Figure S1 shows the BET analysis plots, and from the analysis it is deduced that  $TiO_2$  NSHs have the highest surface area, followed by  $TiO_2$  PNPs,  $TiO_2$  NPs and  $TiO_2$  NWs. The data for surface area, and pore volume of the  $TiO_2$  morphologies are listed in Table S2. CdS QDs are incorporated within these morphologies, depending on the size and volume of the pores, and the effective available surface area of  $TiO_2$ .



Figure S1 Adsorption-desorption isotherms of different TiO<sub>2</sub> morphologies. Table S2 BET and BJH analysis data for different TiO<sub>2</sub> morphologies.

Morphology	Surface area $(m^2 g^{-1})$	Pore volume (cm <sup>3</sup> g <sup>-1</sup> )
TiO <sub>2</sub> NPs	64.304	0.198
TiO <sub>2</sub> NWs	61.82	0.227
TiO <sub>2</sub> PNPs	102.359	0.3107
TiO <sub>2</sub> NSHs	111.968	0.42

# J-V characteristics

Table S3 Solar cell parameters of the 5-cells for each  $TiO_2$  morphology measured using carbon fabric as the counter electrode in a 0.1 M Na<sub>2</sub>S electrolyte (in 3:7 v/v of water:methanol), under 1 sun illumination (AM 1.5, 100 mW cm<sup>-2</sup>).

Cells	$J_{SC}$	V <sub>OC</sub> (V)	FF	Efficiency	
	$(\text{mA cm}^2)$			(η, %)	
TiO <sub>2</sub> NWs/CdS					
Cell 1	14.856	0.978	43.27	6.288	
Cell 2	14.856	0.978	43.18	6.275	
Cell 3	14.256	0.912	44.98	5.849	
Cell 4	13.731	0.973	43.23	5.776	
Cell 5	13.726	0.958	42.67	5.611	
Average	14.285	0.959	43.46	5.959	
TiO <sub>2</sub> NPs/CdS					
Cell 1	14.645	0.978	38.79	5.557	
Cell 2	14.219	0.988	38.67	5.433	
Cell 3	13.59	0.933	36.6	4.641	
Cell 4	13.279	0.943	37.03	4.637	
Cell 5	13.951	0.882	36.4	4.48	
Average	13.936	0.944	37.49	4.949	
TiO <sub>2</sub> PNPs/CdS					
Cell 1	16.439	0.857	38.33	5.401	
Cell 2	15.322	0.852	39.36	5.139	
Cell 3	14.947	0.862	39.43	5.081	
Cell 4	11.731	0.923	40.32	4.366	
Cell 5	11.00	0.938	41.5	4.282	
Average	13.8878	0.8864	39.788	4.854	
TiO <sub>2</sub> NSHs/CdS					
Cell 1	5.839	0.963	47.69	2.682	
Cell 2	5.713	0.928	47.26	2.506	
Cell 3	5.215	0.938	50.57	2.474	
Cell 4	5.109	0.953	50.42	2.455	
Cell 5	5.702	0.933	45.65	2.429	
Average	5.5156	0.943	48.318	2.509	

Electrolyte effect on photovoltaic performance of TiO<sub>2</sub> NPs based cells



Figure S2 J-V characteristics for cells with a  $TiO_2$  NPs/CdS photoanode and a carbon fabric as the counter electrode with two different electrolytes (under 1 sun illumination, AM 1.5).

Table S4 Solar cell parameters of cells with a  $TiO_2$  NPs/CdS photoanode and a carbon fabric as the counter electrode with different listed electrolytes. Exposed cell area is 0.06 to 0.08 cm<sup>2</sup> and measurements are performed under 1 sun illumination (100 mW cm<sup>-2</sup>).

Electrolyte	V <sub>OC</sub> (mV)	J <sub>SC</sub> (mA cm <sup>-2</sup> )	FF	$\eta_{champion}$ (%)	$\eta_{avg}$ (%)
0.1 M Na <sub>2</sub> S solution in water	651	14.49	0.44	4.18	3.72
0.1 M Na <sub>2</sub> S in MeOH:water (7:3)	978	14.645	0.38	5.557	4.949

Table S5 Comparison of  $J_{SC}$  and  $J_{IPCE}$  values for cells with different TiO<sub>2</sub> morphologies.

Photoanode	J <sub>SC</sub> (mA cm <sup>-2</sup> ) from J-V	J <sub>IPCE</sub> (mA cm <sup>-2</sup> ) from IPCE
TiO <sub>2</sub> NSHs/CdS	5.84	5.70
TiO <sub>2</sub> PNPs/CdS	16.43	15.97
TiO <sub>2</sub> NPs/CdS	14.22	14.1
TiO <sub>2</sub> NWs/CdS	14.85	14.4