**Electronic Supplementary Material (ESI)** 

## Clonidine: molecule structure from gas-phase electron diffraction and quantum chemical calculations

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#### 1 Information used for the GED model and results of the GED refinement

1) Nozzle-to-Plate distance = 362.3 mm. Scale factor = 0.53(1) electrons' wavelength  $\lambda$ = 0.049988 Å, s in Å<sup>-1</sup>

	$\lambda = 0.049988$	A, S IN A $^{1}$	1		
S	I(s)	B(s)	S	I(s)	B(s)
3.4	0.54730312	0.570601	11.0	0.52567541	0.52239242
3.6	0.54846914	0.58475209	11.2	0.52856442	0.52315327
3.8	0.54769755	0.59890317	11.4	0.5300357	0.52393669
4.0	0.54788661	0.61248457	11.6	0.52886426	0.52483926
4.2	0.56442034	0.62508736	11.8	0.52631748	0.5260438
4.4	0.59467537	0.63563348	12.0	0.52278873	0.52766829
4.6	0.6291482	0.6429818	12.2	0.5202981	0.52972297
4.8	0.65406861	0.64692225	12.4	0.52082432	0.53211993
5.0	0.67733225	0.64817669	12.6	0.52453115	0.53471968
5.2	0.68902711	0.64719805	12.8	0.52948715	0.53742547
5.4	0.69335453	0.64421686	13.0	0.53382516	0.54025662
5.6	0.69394545	0.63941965	13.2	0.53721985	0.54330471
5.8	0.68697468	0.63299379	13.4	0.54156688	0.54661308
6.0	0.66977066	0.62525168	13.6	0.54775475	0.55010715
6.2	0.64359303	0.61659328	13.8	0.55668414	0.5536795
6.4	0.61456521	0.60741479	14.0	0.56614603	0.55727578
6.6	0.5912247	0.59808432	14.2	0.57381137	0.56097917
6.8	0.57720399	0.58888526	14.4	0.57808351	0.56499504
7.0	0.5658096	0.58002125	14.6	0.58053404	0.56953932
7.2	0.55457021	0.57177393	14.8	0.5816826	0.57469046
7.4	0.5433938	0.56437109	15.0	0.58380968	0.58039482
7.6	0.53382693	0.55788334	15.2	0.58721797	0.58649277
7.8	0.53023379	0.55227457	15.4	0.59219154	0.59280621
8.0	0.53188583	0.54739219	15.6	0.59759865	0.59922051
8.2	0.53656742	0.54306882	15.8	0.60419018	0.60574829
8.4	0.54004752	0.53919843	16.0	0.61093058	0.61248118
8.6	0.54001631	0.53576737	16.2	0.61769314	0.6195769
8.8	0.53680761	0.53279072	16.4	0.62439813	0.62719816
9.0	0.53268169	0.53024341	16.6	0.63123236	0.63542161
9.2	0.53110734	0.52804532	16.8	0.63823386	0.64418063
9.4	0.53123715	0.52607946	17.0	0.64665926	0.65328236
9.6	0.53184809	0.52431116	17.2	0.65473316	0.66249116
9.8	0.53057416	0.52280441	17.4	0.66403168	0.67169997
10.0	0.52759149	0.52167931	10.8	0.52277844	0.5216808
10.2	0.52370299	0.5210249	11.0	0.52567541	0.52239242
10.4	0.52130825	0.52086396	11.2	0.52856442	0.52315327
10.6	0.52099621	0.52112485	17.4	0.66403168	0.67169997
10.8	0.52277844	0.5216808			

<u>S</u>	$\frac{1}{I(s)}$	$\frac{B(s)}{B(s)}$	s	I(s)	B(s)	S	I(s)	B(s)
6.8	0.619235	0.634517	15.6	0 473891	0 475779	24.4	0 417467	0 417091
7.0	0.610824	0.629312	15.8	0.47182	0.473736	24.6	0.415307	0.416434
7.2	0.603646	0.624107	16.0	0.470207	0.471789	24.8	0.414242	0.415786
7.4	0.590841	0.618937	16.2	0.468481	0.46992	25.0	0.413344	0.415141
7.6	0.582616	0.613858	16.4	0.466287	0.468097	25.2	0.412671	0.414505
7.8	0.579447	0.608859	16.6	0.463333	0.46629	25.4	0.412634	0.413892
8.0	0.584445	0.603869	16.8	0.459573	0.46449	25.6	0.412659	0.41331
8.2	0.590194	0.598788	17.0	0.4565	0.462714	25.8	0.411977	0.412766
8.4	0.594769	0.593595	17.2	0.454377	0.460979	26.0	0.411743	0.412276
8.6	0.595365	0.588318	17.4	0.452985	0.459298	26.2	0.411522	0.41184
8.8	0.588225	0.583029	17.6	0.452274	0.457677	26.4	0.411448	0.411451
9.0	0.580147	0.577884	17.8	0.452302	0.456117	26.6	0.411178	0.411107
9.2	0.575239	0.573026	18.0	0.452725	0.454604	26.8	0.411199	0.410813
9.4	0.575208	0.56849	18.2	0.453286	0.453126	27.0	0.411192	0.410571
9.6	0.573333	0.564215	18.4	0.454516	0.45169	27.2	0.410726	0.410382
9.8	0.570712	0.560148	18.6	0.455734	0.450313	27.4	0.410426	0.41025
10.0	0.563507	0.556241	18.8	0.456337	0.449007	27.6	0.410297	0.410177
10.2	0.556656	0.552464	19.0	0.456491	0.447766	27.8	0.410298	0.41016
10.4	0.549971	0.548773	19.2	0.454902	0.446555	28.0	0.410462	0.410195
10.6	0.544753	0.545145	19.4	0.45226	0.445334	28.2	0.410853	0.410281
10.8	0.542786	0.541592	19.6	0.448507	0.444074	28.4	0.411221	0.410417
11.0	0.541754	0.538111	19.8	0.444565	0.442759	28.6	0.411336	0.410606
11.2	0.540819	0.534691	20.0	0.440108	0.441385	28.8	0.41148	0.410859
11.4	0.538432	0.531318	20.2	0.436443	0.439967	29.0	0.411528	0.411193
11.6	0.532983	0.527988	20.4	0.433708	0.438527	29.2	0.411619	0.411627
11.8	0.525583	0.524709	20.6	0.432207	0.437091	29.4	0.41182	0.412176
12.0	0.516266	0.521484	20.8	0.431442	0.435685	29.6	0.41245	0.412846
12.2	0.507535	0.518322	21.0	0.431339	0.434326	29.8	0.41309	0.413626
12.4	0.502026	0.515243	21.2	0.430981	0.433012	30.0	0.414001	0.414496
12.6	0.499993	0.512264	21.4	0.429985	0.431735	30.2	0.414963	0.415425
12.8	0.500091	0.509391	21.6	0.428551	0.430498	30.4	0.415891	0.416383
13.0	0.49976	0.506615	21.8	0.427826	0.429309	30.6	0.417235	0.417339
13.2	0.4978	0.503922	22.0	0.427402	0.428161	30.8	0.417873	0.418272
13.4	0.495872	0.501305	22.2	0.427134	0.427044	31.0	0.418584	0.419195
13.6	0.496193	0.498761	22.4	0.426/19	0.425947	31.2	0.419861	0.420134
13.8	0.498582	0.49629	22.6	0.426062	0.42487	31.4	0.420568	0.421114
14.0	0.502867	0.493892	22.8	0.424895	0.42382	31.6	0.421972	0.422168
14.2	0.50482	0.491547	23.0	0.424082	0.422811	31.8	0.423271	0.423313
14.4	0.503315	0.489235	23.2	0.423231	0.421849	32.0	0.424834	0.424556
14.0	0.498681	0.486943	25.4	0.423087	0.420934	32.2	0.426183	0.425892
14.8	0.492100	0.484030	23.0	0.422528	0.420062	32.4	0.42/69	0.42/513
15.0	0.403003	0.4823/9	23.8	0.4213/4	0.41924	32.0	0.429/89	0.428/93
15.2	0.480684	0.480124		0.420352	0.4184/4	32.8	0.4311/3	0.430272
15.4	0.4/0/91	0.4//915	24.2	0.418/11	0.41//03			

2) Nozzle-to-Plate distance = 193.9 mm. Scale factor = 0.64(2) electrons' wavelength  $\lambda$ = 0.050388 Å, s in Å<sup>-1</sup>

	GED		GED.	n n n n n n n n n n n n n n n n n n n	GED	<u> </u>	arp.
Distance	GED	Calc <sup>o</sup>	GED <sup>c</sup>	Distance	GED	Calc <sup>o</sup>	GED <sup>c</sup>
	ra	u	и		ra	и	и
N3-H14	1.033	0.072	assumed	H14Cl23	4.528	0.353	assumed
N1-H13	1.033	0.072	assumed	N1C9	4.552	0.230	0.230(8)7
C9–H20	1.083	0.077	assumed	N6H22	4.563	0.114	assumed
C11–H22	1.083	0.077	assumed	C10H13	4.565	0.358	assumed
C10-H21	1.085	0.077	assumed	N3C12	4.586	0.133	0.133(8) <sup>7</sup>
C5–H17	1.109	0.078	assumed	N6H20	4.605	0.113	assumed
C4–H16	1.110	0.078	assumed	N3Cl23	4.615	0.272	$0.272(8)^7$
C5–H18	1.116	0.079	assumed	C2C11	4.628	0.142	0.142(8)7
C4–H15	1.117	0.079	assumed	C4C7	4.649	0.096	0.096(8) <sup>7</sup>
C2-N6	1.287	0.041	$0.042(2)^1$	C7H18	4.686	0.217	assumed
C2-N3	1.386	0.048	$0.049(2)^1$	C5C8	4.713	4.732	$0.232(8)^7$
N6-C7	1.389	0.048	$0.049(2)^1$	C12H14	4.758	0.213	assumed
N1-C2	1.390	0.049	$0.050(2)^1$	H21Cl23	4.823	0.116	assumed
C11–C12	1.394	0.047	$0.048(2)^1$	C8H18	4.834	0.331	assumed
C8–C9	1.396	0.047	$0.048(2)^1$	Cl19H21	4.836	0.117	assumed
C9–C10	1.396	0.046	$0.048(2)^1$	H14Cl19	4.866	0.286	assumed
C10-C11	1.398	0.046	$0.048(2)^1$	N1C11	4.873	0.312	$0.312(8)^7$
C7–C8	1.416	0.048	$0.050(2)^1$	C8H14	4.905	0.164	assumed
C7–C12	1.416	0.048	$0.050(2)^1$	C7H17	5.003	0.179	assumed
N3-C4	1.467	0.051	$0.052(2)^1$	C7H15	5.054	0.219	assumed
N1-C5	1.471	0.052	$0.053(2)^1$	C2C10	5.078	0.122	$0.122(8)^7$
C4–C5	1.548	0.054	$0.056(2)^1$	C4Cl19	5.084	0.342	$0.342(8)^7$
C12–Cl23	1.733	0.053	$0.050(3)^2$	H17Cl19	5.139	0.510	assumed
C8–C119	1.741	0.054	$0.051(3)^2$	N1C10	5.183	0.259	0.260(8) <sup>7</sup>
H15H16	1.803	0.125	assumed	N1H20	5.194	0.295	assumed
H17H18	1.804	0.125	assumed	C2H20	5.234	0.173	assumed
C2H14	2.055	0.103	assumed	N3Cl19	4.269	0.264	$0.264(8)^7$
C2H13	2.069	0.103	assumed	C5C7	4.289	0.131	0.132(8)7
N1H18	2.134	0.105	assumed	H20H22	4.292	0.132	assumed
N1H17	2.138	0.106	assumed	N1Cl23	4.297	0.452	$0.452(8)^7$
C12H22	2.140	0.100	assumed	C5Cl19	4.327	0.451	$0.451(8)^7$
C8H20	2.141	0.101	assumed	N6H16	4.402	0.113	assumed
N3H16	2.141	0.105	assumed	H13H20	4.409	0.377	assumed
N3H15	2.147	0.105	assumed	N6H17	4.428	0.113	assumed
C9H21	2.148	0.101	assumed	N3C8	4.437	0.136	$0.137(8)^7$
C11H21	2.150	0.101	assumed	H13Cl23	4.451	0.517	assumed
C10H20	2.157	0.100	assumed	C11H13	4.462	0.423	assumed
C10H22	2.159	0.101	assumed	C2C9	4.481	0.120	$0.121(8)^7$
C4H14	2.173	0.107	assumed	C9Cl23	4.498	0.072	$0.072(8)^7$
C5H13	2.174	0.106	assumed	C11Cl19	4.508	0.072	$0.073(8)^7$
C4H18	2.198	0.112	assumed	H14Cl23	4.528	0.353	assumed
C5H15	2.201	0.112	assumed	N1C9	4.552	0.230	$0.230(8)^7$
C4H17	2.217	0.111	assumed	N6H22	4.563	0.114	assumed
C5H16	2.228	0.111	assumed	C10H13	4.565	0.358	assumed
N1N3	2.246	0.057	$0.060(5)^3$	N3C12	4.586	0.133	0.133(8)7
N3C5	2.302	0.065	$0.068(5)^3$	N6H20	4.605	0.113	assumed
C2C5	2.318	0.060	$0.063(5)^3$	N3Cl23	4.615	0.272	$0.272(8)^7$

Table S2 Calculated and observed vibrational amplitudes in clonidine

C2C7	2.335	0.072	$0.075(5)^3$	C2C11	4.628	0.142	$0.142(8)^7$
C2C4	2.343	0.060	$0.063(5)^3$	C4C7	4.649	0.096	0.096(8) <sup>7</sup>
N3N6	2.344	0.060	$0.063(5)^3$	C7H18	4.686	0.217	assumed
N1C4	2.352	0.066	$0.068(5)^3$	C5C8	4.713	4.732	$0.232(8)^7$
C8C12	2.385	0.060	$0.063(5)^3$	C12H14	4.758	0.213	assumed
C10C12	2.408	0.058	$0.061(5)^3$	H21Cl23	4.823	0.116	assumed
C8C10	2.410	0.058	$0.061(5)^3$	C8H18	4.834	0.331	assumed
N1N6	2.414	0.061	$0.063(5)^3$	Cl19H21	4.836	0.117	assumed
C9C11	2.415	0.058	$0.061(5)^3$	H14Cl19	4.866	0.286	assumed
H15H17	2.429	0.182	assumed	N1C11	4.873	0.312	0.312(8)7
N6C12	2.433	0.067	$0.070(5)^3$	C8H14	4.905	0.164	assumed
H16H18	2.438	0.182	assumed	C7H17	5.003	0.179	assumed
C7C9	2.468	0.059	$0.062(5)^3$	C7H15	5.054	0.219	assumed
C7C11	2.469	0.059	$0.062(5)^3$	C2C10	5.078	0.122	$0.122(8)^7$
N6C8	2.477	0.067	$0.070(5)^3$	C4Cl19	5.084	0.342	$0.342(8)^7$
H20H21	2.486	0.164	assumed	N1C11	4.873	0.312	0.312(8)7
H21H22	2.488	0.164	assumed	C8H14	4.905	0.164	assumed
H13H18	2.507	0.192	assumed	C7H17	5.003	0.179	assumed
H14H15	2.542	0.188	assumed	C7H15	5.054	0.219	assumed
N6H14	2.573	0.142	assumed	C2C10	5.078	0.122	$0.122(8)^7$
H14H16	2.645	0.158	assumed	C4Cl19	5.084	0.342	$0.342(8)^7$
H13H17	2.649	0.158	assumed	H17Cl19	5.139	0.510	assumed
C11Cl23	2.683	0.070	$0.070(3)^4$	N1C10	5.183	0.259	$0.260(8)^7$
C9Cl19	2.697	0.070	$0.070(3)^4$	N1H20	5.194	0.295	assumed
N6H13	2.712	0.137	assumed	C2H20	5.234	0.173	assumed
C7Cl23	2.712	0.070	$0.070(3)^4$	C5C12	5.241	0.288	$0.291(9)^8$
C7Cl19	2.715	0.072	$0.072(3)^4$	N6H21	5.302	0.101	assumed
C7H13	2.718	0.217	assumed	H13H22	5.330	0.499	assumed
C9C12	2.772	0.066	$0.066(3)^4$	H15Cl23	5.362	0.567	assumed
C8C11	2.775	0.067	$0.066(3)^4$	C4C8	5.370	0.167	$0.170(9)^8$
N3H18	2.787	0.167	assumed	Cl19Cl23	5.414	0.095	$0.098(9)^8$
H22Cl23	2.798	0.149	assumed	C8H17	5.418	0.265	assumed
H16H17	2.804	0.160	assumed	C2H22	5.458	0.192	assumed
Cl19H20	2.812	0.150	assumed	C4Cl23	5.491	0.417	$0.419(9)^8$
H13Cl19	2.830	0.544	assumed	H13H21	5.492	0.392	assumed
N1H15	2.842	0.168	assumed	C4C12	5.545	0.470	$0.473(9)^8$
C7C10	2.852	0.066	$0.066(3)^4$	C5Cl23	5.545	0.228	$0.231(9)^{8}$
C8H13	2.857	0.289	assumed	C7H16	5.553	0.127	assumed
C2H18	2.858	0.152	assumed	H20Cl23	5.568	0.102	assumed
C2H15	2.891	0.153	assumed	Cl19H22	5.578	0.102	assumed
N1C7	2.896	0.137	$0.142(14)^5$	H16Cl19	5.647	0.416	assumed
N6Cl23	2.973	0.119	$0.125(14)^{5}$	N1H22	5.690	0.387	assumed
N6Cl19	3.052	0.125	$0.131(14)^{5}$	C12H15	5.716	0.364	assumed
H15H18	3.083	0.132	assumed	C12H17	5.785	0.384	assumed
N3H13	3.141	0.104	assumed	N3C9	5.787	0.129	$0.132(9)^8$
N1H14	3.148	0.102	assumed	C12H18	5.803	0.290	assumed
C2C8	3.166	0.121	$0.127(14)^5$	C5C9	5.885	0.263	$0.265(9)^8$
C2H17	3.220	0.104	assumed	N3C11	5.902	0.125	0.128(9)8
C2H16	3.241	0.104	assumed	H15Cl19	5.907	0.334	assumed
N3H17	3.249	0.103	assumed	C8H15	5.945	0.220	assumed
C5H14	3.275	0.097	assumed	H17Cl23	5.984	0.613	assumed
N1H16	3.295	0.104	assumed	C9H18	5.998	0.364	assumed

C2Cl19	3.297	0.212	$0.217(14)^5$	C11H14	6.123	0.203	assumed
C4H13	3.313	0.098	assumed	C2H21	6.136	0.149	assumed
C2C12	3.369	0.147	$0.153(14)^5$	N1H21	6.168	0.296	assumed
C12H21	3.384	0.097	assumed	C8H16	6.173	0.227	assumed
C8H21	3.385	0.097	assumed	C9H14	6.240	0.156	assumed
C11H20	3.394	0.097	assumed	C5C11	6.319	0.318	$0.320(9)^8$
C9H22	3.394	0.097	assumed	H18Cl23	6.331	0.410	assumed
N1C8	3.398	0.219	$0.214(14)^5$	H18H20	6.376	0.453	assumed
N1Cl19	3.399	0.404	$0.399(14)^{5}$	N3C10	6.410	0.106	$0.109(9)^8$
C7H20	3.427	0.098	assumed	C5H20	6.419	0.348	assumed
C7H22	3.429	0.098	assumed	C9H17	6.464	0.315	assumed
C4N6	3.555	0.068	$0.073(14)^5$	N3H20	6.495	0.195	assumed
C5N6	3.569	0.066	$0.072(14)^5$	H16Cl23	6.496	0.390	assumed
N3C7	3.616	0.069	$0.075(14)^{5}$	C12H16	6.530	0.206	assumed
C12H13	3.626	0.396	assumed	C5C10	6.600	0.278	$0.281(9)^8$
N6C11	3.719	0.068	$0.074(14)^{5}$	C4C9	6.660	0.178	$0.181(9)^{8}$
C2Cl23	3.726	0.250	$0.256(14)^{5}$	N3H22	6.677	0.187	assumed
N6C9	3.749	0.068	$0.074(14)^{5}$	C10H14	6.758	0.160	assumed
H14H18	3.771	0.187	assumed	C11H17	6.774	0.427	assumed
N1C12	3.814	0.300	$0.306(14)^5$	C4C11	6.802	0.236	$0.239(9)^8$
H13H15	3.826	0.186	assumed	C11H18	6.810	0.333	assumed
C12H20	3.845	0.098	assumed	H14H22	6.822	0.263	assumed
C8H22	3.848	0.098	assumed	C10H18	6.896	0.345	assumed
C9H13	3.861	0.312	assumed	H17H20	6.980	0.410	assumed
C7H21	3.926	0.098	assumed	C11H15	6.981	0.380	assumed
C7H14	3.936	0.146	assumed	H14H20	7.007	0.203	assumed
C10Cl23	3.982	0.070	$0.076(4)^{6}$	C10H17	7.081	0.360	assumed
C8Cl23	3.986	0.070	$0.077(4)^{6}$	C5H22	7.114	0.406	assumed
C12Cl19	3.991	0.071	$0.078(4)^6$	C9H15	7.173	0.248	assumed
C10Cl19	3.994	0.070	$0.077(4)^{6}$	C4C10	7.288	0.187	$0.190(9)^8$
N6H15	4.002	0.183	assumed	C4H20	7.309	0.255	assumed
N6H18	4.018	0.179	assumed	N3H21	7.471	0.135	assumed
H13H14	4.022	0.129	assumed	H17H22	7.485	0.547	assumed
H18Cl19	4.062	0.548	assumed	C9H16	7.492	0.234	assumed
H14H17	4.149	0.143	assumed	C4H22	7.537	0.320	assumed
H13H16	4.176	0.147	assumed	C5H21	7.566	0.325	assumed
N6C10	4.231	0.071	$0.077(4)^{6}$	H15H22	7.595	0.475	assumed
N3Cl19	4.269	0.264	$0.264(8)^7$	C10H15	7.629	0.307	assumed
C5C7	4.289	0.131	$0.132(8)^7$	H18H22	7.689	0.393	assumed
H20H22	4.292	0.132	assumed	C11H16	7.790	0.214	assumed
N1Cl23	4.297	0.452	$0.452(8)^7$	H14H21	7.823	0.178	assumed
C5Cl19	4.327	0.451	$0.451(8)^7$	H18H21	7.836	0.390	assumed
N6H16	4.402	0.113	assumed	H15H20	7.906	0.288	assumed
H13H20	4.409	0.377	assumed	H17H21	7.987	0.420	assumed
N6H17	4.428	0.113	assumed	H16H20	8.068	0.323	assumed
N3C8	4.437	0.136	$0.137(8)^7$	C10H16	8.214	0.199	assumed
H13Cl23	4.451	0.517	assumed	C4H21	8.317	0.224	assumed
C11H13	4.462	0.423	assumed	H16H22	8.554	0.291	assumed
C2C9	4.481	0.120	0.121(8) <sup>7</sup>	H15H21	8.642	0.342	assumed
C9Cl23	4.498	0.072	$0.072(8)^7$	H16H21	9.244	0.233	assumed
C11Cl19	4.508	0.072	0.073(8)7				

<sup>a</sup>Values of distances ( $r_a$ ), and amplitudes (u) are in Å.

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<sup>b</sup> Calculated with B3LYP/6-31G(d,p) quadratic force constants.

<sup>c</sup>The same numeric superscripts indicate that these amplitudes were refined in one group; differences between their values were assumed from theoretical calculation. Amplitudes without superscript were constrained to the calculated values.

**Table S3** Matrix of correlation for GED refinement. Parameters 10 and 20 are the scale factors for LD and SD, respectively. The definition of all other parameters can be found in Tab. 4 and S2.

	10	20	b	с	d	e	f	g
10	1.0000							
20	0.5036	1.0000						
b	-0.1958	-0.1676	1.0000					
c	0.3205	0.3627	-0.3938	1.0000				
d	-0.4115	-0.3600	0.3804	-0.4945	1.0000			
e	0.0397	0.0012	-0.0583	-0.0359	-0.0210	1.0000		
f	-0.3584	-0.1979	0.3160	-0.0823	0.4296	-0.5576	1.0000	
g	0.0721	0.0881	-0.0323	0.3002	-0.0904	0.0108	-0.0302	1.0000
h	0.1669	0.2227	-0.3445	0.1841	-0.2652	0.0694	-0.2023	-0.0002
j	0.0479	0.0937	-0.1006	-0.1784	-0.1775	0.0140	-0.0478	-0.1331
k	-0.0643	0.0087	0.2971	-0.0003	0.1269	0.0224	0.0861	0.6005
1	-0.1441	-0.0349	0.2301	-0.0729	0.2315	-0.1068	0.2050	0.0769
n	-0.1064	0.0070	0.0929	-0.1659	0.0664	-0.0208	0.1158	-0.0201
1	0.1477	0.3451	0.2747	-0.5926	0.3120	-0.0087	-0.0420	-0.1218
2	0.1818	0.4141	0.0034	-0.1164	0.0157	0.0195	-0.0582	-0.0557
3	0.4015	0.5379	-0.3673	0.6451	-0.4254	-0.0914	-0.0230	0.0699
4	0.1240	0.2930	-0.1506	0.0013	-0.1412	0.0058	-0.1184	0.1013
5	-0.0791	-0.1271	-0.0003	0.0628	0.0975	-0.0883	0.0863	-0.0011
6	0.1835	0.2804	-0.0688	0.0800	-0.1039	-0.0295	-0.0291	-0.0289
7	0.1652	0.1674	-0.0467	0.0839	-0.1236	-0.0333	-0.0485	-0.0342
8	0.1340	0.1197	-0.0504	0.0294	-0.0891	0.0312	-0.1151	-0.0025
	h	j	k	1	n	1	2	3
h	1							
j	0.5573	1.0000						
k	-0.0431	-0.0700	1.0000					
1	-0.2790	-0.3505	0.2774	1.0000				
n	0.2167	0.4505	0.1955	0.5436	1.0000			
1	-0.0280	0.1626	0.0653	0.0740	0.1378	1.0000		
2	0.0434	0.1123	-0.0136	0.0118	0.0655	0.4221	1.0000	
3	0.2891	0.3151	-0.2645	-0.2403	0.0339	-0.1887	0.1221	1.0000
4	0.0214	-0.0959	0.0775	0.4103	0.2268	0.1830	0.1757	-0.0244
5	-0.4071	-0.4779	-0.1695	-0.3273	-0.5764	-0.1294	-0.0865	-0.1305
6	0.1462	0.0903	-0.0402	0.0592	0.1378	0.1143	0.1182	0.1625
7	-0.1929	-0.0809	-0.1604	-0.1738	-0.3372	0.0324	0.0737	0.1478

8	0.0504	0.0568	-0.0115	-0.1386	-0.1137	0.0515	0.0434	0.0660
	4	5	6	7	8			
4	1.0000							
5	-0.0565	1.0000						
6	0.1242	-0.1237	1.0000					
7	-0.0275	0.1509	-0.0169	1.0000				
8	-0.0106	0.0178	0.0202	0.0812	1.0000			

Table S4 Cartesian coordinates of clonidine obtained from GED refinement

Atom	Х	Y	Ζ
Ν	-1.879695122	0.061381849	1.046286814
С	-1.621525359	-0.354747009	-0.255075602
Ν	-2.824103123	-0.723237773	-0.836023741
С	-3.882436265	-0.818042142	0.176752308
С	-3.335955311	0.193049119	1.215110806
Ν	-0.493527753	-0.443144532	-0.867288801
С	0.698482964	-0.011279429	-0.301246257
С	1.050676824	1.349209024	-0.116305465
С	2.305338525	1.754715559	0.343552
С	3.273135751	0.795193792	0.643510373
С	2.980065167	-0.5615769	0.481778341
С	1.720528173	-0.941189517	0.019220849
Н	-1.242813675	0.773862716	1.445451949
Н	-2.766308792	-1.41919726	-1.600798203
Н	-3.955401568	-1.841794321	0.6194092
Н	-4.874326819	-0.535179136	-0.235412588
Н	-3.644956147	-0.054532861	2.252877592
Н	-3.675717398	1.229133812	0.973877846
Cl	-0.128489424	2.572976726	-0.496357716
Н	2.514285096	2.813438122	0.457540652
Н	4.252667327	1.101996622	0.998625505
Н	3.717659478	-1.323849668	0.710777044
Cl	1.394457007	-2.634182214	-0.16004164

#### 2 Results of quantum-chemical calculations

2.1 Geometries and thermochemical properties for clonidine conformers obtained by the B3LYP/6-31G(d,p) calculations

Atom	Х	Y	Z
Ν	-1.834760	0.104895	1.0228940
С	-1.578059	-0.288618	-0.2824940
Ν	-2.789733	-0.595301	-0.8710440
С	-3.835911	-0.666455	0.1438730
С	-3.273808	0.298651	1.2035580
Ν	-0.458636	-0.411442	-0.9024850
С	0.746536	-0.041067	-0.3282130
С	1.157555	1.300537	-0.1495600
С	2.422200	1.650600	0.3180220
С	3.339482	0.651186	0.6316690
С	2.984920	-0.688035	0.4748990
С	1.717418	-1.015492	0.0049410
Н	-1.193364	0.768044	1.4349150
Н	-2.758955	-1.258796	-1.6319920
Н	-3.946332	-1.678452	0.5603010
Н	-4.803203	-0.339542	-0.2462790
Н	-3.591481	0.039230	2.2167030
Н	-3.579877	1.332048	0.9883980
Cl	0.033293	2.601616	-0.5498240
Н	2.677543	2.698523	0.4275070
Н	4.328041	0.913991	0.9933760
Н	3.683195	-1.481720	0.7151070
Cl	1.301080	-2.712687	-0.1739270

Table S5 Cartesian coordinates of I conformer found using B3LYP/6-31G(d,p) calculations

Thermochemical data for I conformer of clonidine calculated with B3LYP/6-31G(d,p) at 420 K using ideal gas, rigid rotor, harmonic normal mode approximations

	E (Thermal), kcal/mol	c <sub>v,</sub> cal/(mol·K)	S, cal/(mol·K)
Total	119.693	64.247	134.34
Electronic	0.000	0.000	0.000
Translational	1.252	2.981	43.890
Rotational	1.252	2.981	33.451
Vibrational	117.189	58.286	56.998

Ee =-1433.0377755 Ha ZPE = 0.167018 Ha

Atom	Х	Y	Z
Ν	2.717484	-1.031707	-0.705422
С	1.719774	-0.353013	-0.267563
Ν	1.900095	0.391994	0.899259
С	3.361222	0.444376	1.088512
С	3.819613	-0.764846	0.22795
Ν	0.478482	-0.403357	-0.902015
С	-0.747155	-0.039116	-0.318684
С	-1.132630	1.305098	-0.142978
С	-2.362776	1.660219	0.405748
С	-3.272194	0.666381	0.757701
С	-2.951330	-0.674679	0.561767
С	-1.705928	-1.010924	0.039484
Н	-0.062537	2.599212	-0.667955
Н	-2.606439	2.708996	0.528579
Н	-4.236096	0.937640	1.175223
Н	-3.650189	-1.461179	0.821503
Н	-1.322751	-2.709367	-0.186574
Н	0.436961	-1.216493	-1.504457
Cl	1.414041	1.278206	0.951055
Н	3.776317	1.388275	0.707863
Н	3.627201	0.351252	2.145115
Н	4.752687	-0.563496	-0.308518
Cl	3.991394	-1.650066	0.856247

Table S6 Cartesian coordinates of A1 conformer found using B3LYP/6-31G(d,p) calculations

Thermochemical data for A1 conformer of clonidine calculated with B3LYP/6-31G(d,p) using ideal gas, rigid rotor, harmonic normal mode approximations

	E (Thermal),	c <sub>v,</sub>	S,
	kcal/mol	cal/(mol·K)	cal/(mol·K)
Total	119.566	64.196	134.701
Electronic	0.000	0.000	0.000
Translational	1.252	2.981	43.890
Rotational	1.252	2.981	33.468
Vibrational	117.062	58.234	57.343

Ee = -1433.0251552 Ha ZPE = 0.166791 Ha

Atom	Х	Y	Z
Ν	-2.775005	-0.836195	-0.817439
С	-1.728716	-0.295096	-0.305932
Ν	-1.883486	0.425299	0.877165
С	-3.212368	0.048088	1.38456
С	-3.879388	-0.491107	0.088066
Ν	-0.484689	-0.381520	-0.922886
С	0.741161	-0.032825	-0.333924
С	1.693146	-1.014431	0.015584
С	2.935574	-0.692397	0.553835
С	3.254751	0.645073	0.777123
С	2.350627	1.647361	0.43513
С	1.122379	1.312048	-0.133993
Н	1.302720	-2.708333	-0.241525
Н	3.631496	-1.484631	0.803788
Н	4.215655	0.907217	1.207513
Н	2.597878	2.692827	0.57794
Н	0.071735	2.608809	-0.655606
Н	-0.459998	-1.147181	-1.583716
Cl	-1.110844	0.413767	1.531224
Н	-3.139001	-0.733771	2.154231
Н	-3.733289	0.908198	1.814394
Н	-4.518602	-1.358834	0.28056
Cl	-4.509750	0.280950	-0.375272

Table S7 Cartesian coordinates of A2 conformer found using B3LYP/6-31G(d,p) calculations

Thermochemical data for A2 conformer of clonidine calculated with B3LYP/6-31G(d,p) using ideal gas, rigid rotor, harmonic normal mode approximations

	E (Theorem al)		C
	E (Thermal),	$c_{v,}$	5,
	kcal/mol	cal/(mol·K)	cal/(mol·K)
Total	119.448	64.389	136.112
Electronic	0.000	0.000	0.000
Translational	1.252	2.981	43.890
Rotational	1.252	2.981	33.460
Vibrational	116.944	58.427	58.012

Ee = -1433.023614 Ha ZPE = 0.166456 Ha

Atom	Х	Y	Z
Ν	-1.776220	0.088168	1.057201
С	-1.640778	-0.413308	-0.110393
Ν	-2.773691	-0.945156	-0.741931
С	-3.916229	-0.473755	0.064113
С	-3.208547	-0.004551	1.372157
Ν	-0.445274	-0.516830	-0.826291
С	0.763495	-0.027707	-0.298769
С	1.827273	-0.890941	0.021911
С	3.050509	-0.426740	0.496104
С	3.233426	0.941470	0.677109
С	2.210570	1.831449	0.357214
С	1.002797	1.351517	-0.142879
Н	1.614279	-2.627499	-0.190331
Н	3.837199	-1.134663	0.729053
Н	4.177713	1.315886	1.05825
Н	2.346401	2.900455	0.472103
Н	-0.210122	2.514419	-0.632533
Н	-0.352885	-1.413825	-1.288882
Cl	-2.827371	-0.756901	-1.736849
Н	-4.435525	0.355498	-0.433729
Н	-4.639267	-1.276735	0.233888
Н	-3.593162	0.956262	1.730699
Cl	-3.352574	-0.731487	2.181889

Table S8 Cartesian coordinates of A3 conformer found using B3LYP/6-31G(d,p) calculations

Thermochemical data for A3 conformer of clonidine calculated with B3LYP/6-31G(d,p) using ideal gas, rigid rotor, harmonic normal mode approximations

	E (Thermal), kcal/mol	$c_{v_{i}}$	S,
	Keal/III01		
Total	119.558	64.177	134.756
Electronic	0.000	0.000	0.000
Translational	1.252	2.981	43.890
Rotational	1.252	2.981	33.459
Vibrational	117.055	58.215	57.407

Ee = -1433.0251604 Ha ZPE = 0.166740 Ha

**Table S9** Equilibrium distribution of clonidine conformers at 420 K by means of B3LYP/6-31G(d,p) calculations

Conformer	$\Delta E_{\rm e}$	$\Delta H$	$S_{420}$	p <sub>420</sub>	
Ι	0.0	0.0	135.7	100	
A1	8.1	7.8	136.1	0	
A2	8.9	8.5	136.7	0	
A3	7.9	7.7	136.1	0	

 $\Delta E_{\rm e}$  (kcal /mol) is the relative electronic energy;  $H = E_{\rm e} + ZPE$  (kcal·mol<sup>-1</sup>), where ZPE is the zero point energy correction calculated from scaled vibrational frequencies;  $S_{420}$  (cal/(K·mol)) is the entropy value at 420 K obtained using scaled frequencies including the entropy of mixing (*R*ln2), since all conformers present as an equimolar mixture of the two optical isomers;  $\Delta G_T(i) = \Delta H(i) + T \Delta S_T(i)$  is the Gibbs free energy difference (kcal·mol<sup>-1</sup>);  $p_T(i) = [\exp(-\Delta G_T(i) / RT)]/\Sigma[\exp(-G_T(i) / RT)]$  is the mole fraction of conformers (%).

2.2 Geometries and thermochemical properties for clonidine conformers obtained by the B3LYP/cc-pVTZ calculations

Atom	Х	Y	Z
Ν	-1.837584	0.099649	1.019084
С	-1.574738	-0.290820	-0.280993
Ν	-2.780756	-0.594687	-0.87445
С	-3.834237	-0.662430	0.130695
С	-3.275758	0.286861	1.199668
Ν	-0.456123	-0.412155	-0.887614
С	0.747940	-0.040914	-0.323059
С	1.153143	1.296983	-0.148103
С	2.413520	1.649357	0.311271
С	3.332506	0.657697	0.619436
С	2.982183	-0.677169	0.466814
С	1.718884	-1.008789	0.006219
Н	-1.199492	0.752658	1.442601
Н	-2.751845	-1.248831	-1.638572
Η	-3.954964	-1.673101	0.535458
Η	-4.791094	-0.325922	-0.264353
Η	-3.592305	0.015381	2.2051
Η	-3.579869	1.318798	0.998034
Cl	0.027552	2.589641	-0.541703
Η	2.665268	2.694194	0.418086
Η	4.318330	0.923037	0.974215
Η	3.681704	-1.465265	0.703337
Cl	1.310381	-2.701715	-0.167987

Table S10 Cartesian coordinates of I conformer found using B3LYP/cc-pVTZ calculations

Thermochemical data for I conformer of clonidine calculated with B3LYP/cc-pVTZ using ideal gas, rigid rotor, harmonic normal mode approximations

	E (Thermal),	c <sub>v,</sub>	S,
	kcal/mol	cal/(mol·K)	cal/(mol·K)
Total	119.708	64.114	134.197
Electronic	0.000	0.000	0.000
Translational	1.252	2.981	43.890
Rotational	1.252	2.981	33.436
Vibrational	117.204	58.152	56.870

Ee = -1433.2785142 Ha ZPE = 0.167101 Ha

Atom	Х	Y	Ζ
Ν	2.702478	-1.071461	-0.703435
С	1.711656	-0.401551	-0.253611
Ν	1.913664	0.342737	0.904596
С	3.376220	0.401693	1.066751
С	3.821267	-0.805957	0.206931
Ν	0.461704	-0.470267	-0.855977
С	-0.750214	-0.041521	-0.301932
С	-1.080953	1.313926	-0.144956
С	-2.297625	1.722032	0.381738
С	-3.249899	0.775460	0.726933
С	-2.984493	-0.573621	0.543435
С	-1.752522	-0.964936	0.04522
Н	0.396796	-1.296828	-1.431156
Н	1.421757	1.219457	0.975582
Н	3.776513	1.341464	0.673392
Н	3.660386	0.314602	2.114182
Н	4.738663	-0.607213	-0.347407
Н	4.002410	-1.686901	0.831058
Cl	0.037844	2.556928	-0.671212
Н	-2.497768	2.777262	0.491498
Н	-4.203520	1.089488	1.126904
Н	-3.717762	-1.325003	0.795237
Cl	-1.442728	-2.673782	-0.16812

Table S11 Cartesian coordinates of A1 conformer found using B3LYP/cc-pVTZ calculations

Thermochemical data for A1 conformer of clonidine calculated with B3LYP/cc-pVTZ using ideal gas, rigid rotor, harmonic normal mode approximations

	E (Thermal),	c <sub>v,</sub>	S,
	kcal/mol	cal/(mol·K)	cal/(mol·K)
Total	119.603	64.062	134.453
Electronic	0.000	0.000	0.000
Translational	1.252	2.981	43.890
Rotational	1.252	2.981	33.454
Vibrational	117.099	58.100	57.108

Ee = -1433.26655 Ha ZPE = 0.166928 Ha

Atom	Х	Y	Ζ
Ν	-2.766112	-0.835771	-0.830016
С	-1.721916	-0.325703	-0.29842
Ν	-1.882708	0.362402	0.897673
С	-3.227767	0.007335	1.374566
С	-3.880366	-0.508829	0.066494
Ν	-0.473589	-0.423956	-0.892249
С	0.743693	-0.033137	-0.321861
С	1.723250	-0.982023	0.019328
С	2.955041	-0.624089	0.543376
С	3.236762	0.716848	0.760375
С	2.305563	1.687199	0.422571
С	1.088008	1.317271	-0.133133
Н	-0.433982	-1.189183	-1.547126
Н	-1.124640	0.324904	1.561588
Н	-3.181492	-0.777186	2.136452
Н	-3.738347	0.871149	1.797688
Н	-4.515353	-1.378723	0.235237
Н	-4.504215	0.265296	-0.391321
Cl	1.383780	-2.680871	-0.234802
Н	3.673355	-1.392303	0.787228
Н	4.189630	1.006718	1.180308
Н	2.523875	2.735962	0.557223
Cl	0.006480	2.580013	-0.656075

Table S12 Cartesian coordinates of A2 conformer found using B3LYP/cc-pVTZ calculations

Thermochemical data for A2 conformer of clonidine calculated with B3LYP/cc-pVTZ using ideal gas, rigid rotor, harmonic normal mode approximations

	E (Thermal),	c <sub>v,</sub>	S,
	kcal/mol	cal/(mol·K)	cal/(mol·K)
Total	119.498	64.243	135.208
Electronic	0.000	0.000	0.000
Translational	1.252	2.981	43.890
Rotational	1.252	2.981	33.446
Vibrational	116.994	58.281	57.871

Ee = -1433.265027 Ha ZPE = 0.166595 Ha

Atom	Х	Y	Ζ
Ν	-1.779415	0.096718	1.046757
С	-1.637112	-0.410272	-0.110422
Ν	-2.767057	-0.943544	-0.737214
С	-3.911095	-0.475241	0.065868
С	-3.208925	0.008832	1.365589
Ν	-0.442735	-0.519677	-0.816568
С	0.764883	-0.028074	-0.296983
С	1.822352	-0.889216	0.024649
С	3.040023	-0.428076	0.498409
С	3.223384	0.934393	0.679018
С	2.206685	1.821483	0.356752
С	1.004150	1.345063	-0.142841
Н	-0.351007	-1.410574	-1.282866
Н	-2.822315	-0.771764	-1.730744
Н	-4.433572	0.342375	-0.437295
Н	-4.623639	-1.279623	0.243057
Н	-3.589096	0.971605	1.70916
Н	-3.350935	-0.704141	2.181802
Cl	1.608568	-2.618474	-0.190067
Н	3.822890	-1.133698	0.732311
Н	4.163707	1.306590	1.060468
Н	2.343311	2.886413	0.470024
Cl	-0.200389	2.505827	-0.635638

Table S13 Cartesian coordinates of A3 conformer found using B3LYP/cc-pVTZ calculations

Thermochemical data for A3 conformer of clonidine calculated with B3LYP/cc-pVTZ using ideal gas, rigid rotor, harmonic normal mode approximations

	E (Thermal),	c <sub>v,</sub>	S,
	kcal/mol	cal/(mol·K)	cal/(mol·K)
Total	119.631	64.003	134.483
Electronic	0.000	0.000	0.000
Translational	1.252	2.981	43.890
Rotational	1.252	2.981	33.443
Vibrational	117.127	58.041	57.149

Ee = -1433.2669916 Ha ZPE = 0.166946 Ha

**Table S14** Equilibrium distribution of clonidine conformers at 420 K by means of B3LYP/ccpVTZ calculations

Conformer	$\Delta E_{\rm e}$	$\Delta H$	$S_{420}$	p <sub>420</sub>
Ι	0.0	0.0	135.6	100
A1	7.5	7.4	135.8	0
A2	8.5	8.1	136.6	0
A3	7.2	7.1	135.9	0

 $\Delta E_{\rm e}$  (kcal /mol) is the relative electronic energy;  $H = E_{\rm e} + ZPE$  (kcal·mol<sup>-1</sup>), where ZPE is the zero point energy correction calculated from scaled vibrational frequencies;  $S_{420}$  (cal/(K·mol)) is the entropy value at 420 K obtained using scaled frequencies including the entropy of mixing (*R*ln2), since all conformers present as an equimolar mixture of the two optical isomers;  $\Delta G_T(i) = \Delta H(i) + T \Delta S_T(i)$  is the Gibbs free energy difference (kcal·mol<sup>-1</sup>);  $p_T(i) = [\exp(-\Delta G_T(i) / RT)]/\Sigma[\exp(-G_T(i) / RT)]$  is the mole fraction of conformers (%).

2.3 Geometries and thermochemical properties for clonidine conformers obtained by the MP2/cc-pVTZ calculations

Atom	Х	Y	Z
Ν	-1.704501	0.033546	1.012663
С	-1.531031	-0.333559	-0.309834
Ν	-2.775533	-0.613675	-0.832894
С	-3.735403	-0.716932	0.257116
С	-3.124615	0.269875	1.24871
Ν	-0.444154	-0.447567	-0.985336
С	0.732290	-0.040802	-0.372932
С	1.066400	1.312414	-0.170734
С	2.291776	1.714500	0.352618
С	3.234857	0.751318	0.695077
С	2.948992	-0.599145	0.5186
С	1.717141	-0.976231	-0.004802
Н	-3.415553	1.292491	0.995475
Н	-0.093472	2.531244	-0.602937
Н	2.493932	2.768109	0.478878
Н	4.191392	1.052663	1.097785
Н	3.665727	-1.362514	0.783666
Н	1.366858	-2.658191	-0.203222
Cl	-1.032723	0.690252	1.380081
Η	-2.785023	-1.285400	-1.583937
Η	-3.765782	-1.720264	0.690506
Н	-4.733448	-0.426960	-0.059435
C1	-3.399204	0.054978	2.27773

Table S15 Cartesian coordinates of I conformer found using MP2/cc-pVTZ calculations

Thermochemical data for I conformer of clonidine calculated with MP2/cc-pVTZ using ideal gas, rigid rotor, harmonic normal mode approximations

	E (Thermal),	c <sub>v,</sub>	S,
	kcal/mol	cal/(mol·K)	cal/(mol·K)
Total	119.173	64.596	134.194
Electronic	0.000	0.000	0.000
Translational	1.252	2.981	43.890
Rotational	1.252	2.981	33.354
Vibrational	116.669	58.634	56.950

Ee = -1428.6038924 Ha ZPE = 0.166112 Ha

Atom	Х	Y	Ζ
Ν	2.716807	-1.003898	-0.710601
С	1.679541	-0.375440	-0.289469
Ν	1.778860	0.304168	0.924179
С	3.228998	0.442378	1.11966
С	3.738224	-0.766006	0.319394
Ν	0.469206	-0.408683	-0.968866
С	-0.732570	-0.038134	-0.35724
С	-1.077540	1.309336	-0.162806
С	-2.279375	1.684536	0.42932
С	-3.193071	0.705502	0.804285
С	-2.906469	-0.639251	0.594172
С	-1.686206	-0.994549	0.030938
Н	0.412736	-1.235612	-1.547303
Н	1.251326	1.163430	0.980417
Н	3.594499	1.376855	0.68516
Н	3.491419	0.406345	2.17355
Н	4.712422	-0.590609	-0.130697
Н	3.813288	-1.648879	0.959296
Cl	0.004619	2.551900	-0.700327
Н	-2.495248	2.733741	0.566331
Н	-4.133485	0.991637	1.253219
Н	-3.606367	-1.412481	0.874252
Cl	-1.323671	-2.669298	-0.208894

Table S16 Cartesian coordinates of A1 conformer found using MP2/cc-pVTZ calculations

Thermochemical data for A1 conformer of clonidine calculated with MP2/cc-pVTZ using ideal gas, rigid rotor, harmonic normal mode approximations.

		11	
	E (Thermal),	c <sub>v,</sub>	S,
	kcal/mol	cal/(mol·K)	cal/(mol·K)
Total	118.988	64.657	135.713
Electronic	0.000	0.000	0.000
Translational	1.252	2.981	43.890
Rotational	1.252	2.981	33.380
Vibrational	116.484	58.696	58.443

Ee = -1428.5909445 Ha ZPE = 0.165716 Ha

Atom	Х	Y	Ζ
Ν	2.778703	0.742923	-0.781844
С	1.698947	0.204115	-0.342338
Ν	1.770890	-0.518963	0.845111
С	3.031772	-0.072566	1.450845
С	3.807534	0.355806	0.194732
Ν	0.499157	0.263099	-1.037095
С	-0.721785	0.030553	-0.389787
С	-1.568268	1.084569	-0.005863
С	-2.797895	0.863149	0.604339
С	-3.197511	-0.443255	0.866781
С	-2.389245	-1.513877	0.499468
С	-2.389245	-1.513877	0.499468
Н	0.503451	1.041772	-1.681472
Н	0.947383	-0.485739	1.430146
Н	2.870080	0.779634	2.116541
Н	3.512755	-0.875948	2.002476
Н	4.492163	1.178447	0.386064
Н	4.384703	-0.481943	-0.204904
Cl	-1.055760	2.710483	-0.308249
Н	-3.416039	1.705677	0.876866
Н	-4.145944	-0.627832	1.351017
Н	-2.697683	-2.533267	0.678381
Cl	-0.233906	-2.622258	-0.666264

Table S17 Cartesian coordinates of A2 conformer found using MP2/cc-pVTZ calculations

Thermochemical data for A2 conformer of clonidine calculated with MP2/cc-pVTZ using ideal gas, rigid rotor, harmonic normal mode approximations.

	E (Thermal), kcal/mol	c <sub>v,</sub> cal/(mol·K)	S, cal/(mol·K)
Total	118.928	64.696	135.395
Electronic	0.000	0.000	0.000
Translational	1.252	2.981	43.890
Rotational	1.252	2.981	33.371
Vibrational	116.424	58.734	58.134

Ee = -1428.5900023 Ha ZPE = 0.165589 Ha

Atom	Х	Y	Ζ
Ν	2.778701	0.742939	-0.781838
С	1.698948	0.204122	-0.342339
Ν	1.770894	-0.518970	0.845104
С	3.031777	-0.072566	1.450842
С	3.807535	0.355813	0.19473
Ν	0.499154	0.263106	-1.037098
С	-0.721784	0.030552	-0.389789
С	-1.568273	1.084565	-0.005859
С	-2.797902	0.863138	0.604336
С	-3.197517	-0.443268	0.866773
С	-2.389244	-1.513884	0.499461
С	-1.174845	-1.276833	-0.14011
Н	0.503444	1.041798	-1.681453
Н	0.947389	-0.485748	1.430144
Н	2.870078	0.779634	2.116537
Н	3.512758	-0.875951	2.00247
Н	4.492167	1.178452	0.386066
Н	4.384700	-0.481934	-0.204915
Cl	-1.055771	2.710483	-0.308241
Н	-3.416045	1.705666	0.876866
Н	-4.145953	-0.627852	1.351001
Н	-2.697672	-2.533278	0.678371
Cl	-0.233892	-2.622256	-0.666261

Table S18 Cartesian coordinates of A3 conformer found using MP2/cc-pVTZ calculations

Thermochemical data for A3 conformer of clonidine calculated with MP2/cc-pVTZ using ideal gas, rigid rotor, harmonic normal mode approximations.

	E (Thermal),	C <sub>v,</sub>	S,
	kcal/mol	cal/(mol·K)	cal/(mol·K)
Total	118.928	64.696	135.393
Electronic	0.000	0.000	0.000
Translational	1.252	2.981	43.890
Rotational	1.252	2.981	33.371
Vibrational	116.424	58.734	58.132

Ee = -1428.5900019 Ha ZPE = 0.165590 Ha