Electronic Supplementary Information

Nature of surface carbon species and pathways of their formation in the heterogeneously catalysed acetoxylation of toluene

S. Reining, E. V. Kondratenko*, U. Bentrup, J. Radnik, V. N. Kalevaru, A. Martin*

Leibniz-Institut für Katalyse e. V. an der Universität Rostock, Albert-Einstein-Str. 29 A D-18059 Rostock, Germany

*To whom correspondence should be addressed.

E-mails: evgenii.kondratenko@catalysis.de, andreas.martin@catalysis.de



Figure S1. Schematic representation of in situ TPO tests and detailed set-up scheme.



Figure S2. Profiles of CO₂ (solid line) and H₂O (dotted line) liberated and O₂ (dashed line) consumed upon TPO of Pd-Sb/TiO₂ initially pre-treated in a toluene/acetic acid/O₂ feed and subsequent treatment in toluene/acetic acid/O₂ feed at 463 K, 483 K or 503 K.



Figure S3. Profiles of CO₂ (solid line) and H₂O (dotted line) liberated and O₂ (dashed line) consumed upon TPO of Pd-Sb/TiO₂ initially pre-treated in a toluene/acetic acid/O₂ feed and subsequent treatment in toluene/O₂ feed at 463 K, 483 K or 503 K.



Figure S4. Profiles of CO₂ (solid line) and H₂O (dotted line) liberated and O₂ (dashed line) consumed upon TPO of Pd-Sb/TiO₂ initially pre-treated in a toluene/acetic acid/O₂ feed and subsequent treatment in acetic acid/O₂ feed at 463 K, 483 K or 503 K.



Figure S5. In-situ IR spectra recorded after dosing toluene/He over Pd-Sb/TiO₂ at 423 K, 473 K, 523 K, and gas phase toluene as reference.

Figure S5 shows the IR absorption spectra of toluene in the absence of gas phase O_2 at different temperatures. The vibration bands at 1651 and 1357 cm⁻¹ are assigned to the interaction of the methyl group with the catalyst. The band at 1651 cm⁻¹ indicates the formation of an adsorbed benzaldehyde species. The band at 1357 cm⁻¹ is assigned to the deformation of the methyl group caused by an interaction with the catalytic surface.



Figure S6. In situ IR spectra at 423 K, 473 K, 523 K, and ex situ IR spectrum after dosing toluene/O₂ at 503 K; gas phase toluene is given as reference.

The original profiles of CO₂, H₂O, and O₂ obtained from TPO experiments are shown in Figures S2-S4. The CO₂ and H₂O profiles were integrated and the calculated areas were divided by the calculated areas obtained from TPO after activation of the catalyst under toluene/acetic acid/O₂ feed for 6 h. The corresponding areas to the latter profiles are 0.264 for CO₂/Ar and 0.171 for H₂O/Ar.



Figure S7. Selectivity to benzyl acetate as a function of acetic acid conversion at 463 K (\Box), 483 K (Δ) or 503 K (\circ) and 2 bar under toluene/acetic acid/O₂ feed with a molar concentration of 1/4/3.