

Supporting Information

Unravelling the promoting effect of ultrathin TaC/RGO nanosheets hybrid for enhanced catalytic activity of Pd nanoparticles

Chunyong He^{*a,b}, Juzhou Tao^{*a,b}, Guoqiang He^c Pei Kang Shen^{*,c} Yongfu Qiu^d

Dr. C. He, Prof. J. Tao, Prof. G. He, Prof. P. K. Shen

^a Institute of High Energy Physics, Chinese Academy of Sciences (CAS), Beijing 100049, China

^b Dongguan Neutron Science Center, Dongguan 523803, China

^c Collaborative Innovation Center of Sustainable Energy Materials, Guangxi University, Nanning, Guangxi, 530004, PR China

^d College of Chemistry and Environmental Engineering, Dongguan University of Technology, Guangdong 523808, P. R. China

E-mail: taoj@ihep.ac.cn (J. Tao); hechunyong@ihep.ac.cn (C. He); pkshen@gxu.edu.cn (P.K. Shen)

Keywords: Two-Dimensional, Transition Metal Carbides, Pd nanoparticles, Methanol oxidation reaction, Direct methanol fuel cells

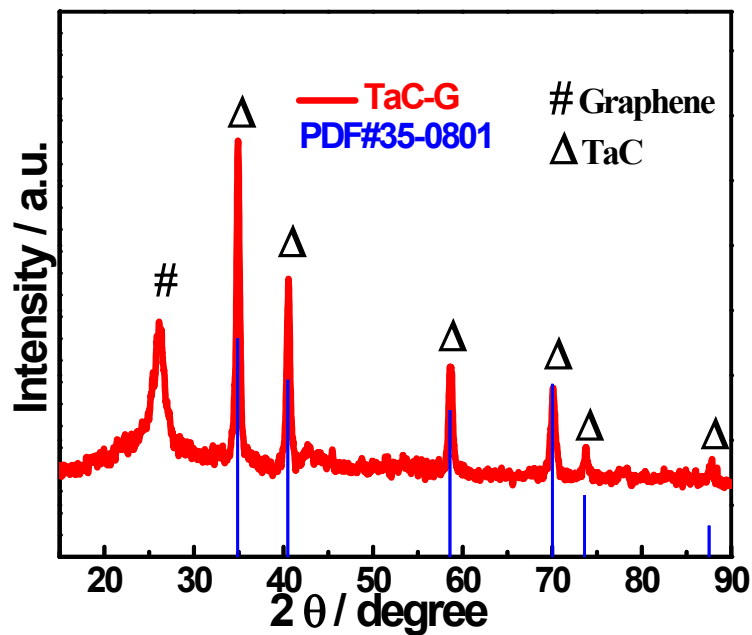


Fig. S1 The XRD pattern of TaC-G. The diffraction peak at $2\theta = 26.2^\circ$ is the characteristic of the graphite (002) plane, demonstrating the reduce of the GO after annealing. The distinct diffraction peaks at $2\theta = 34.94^\circ, 40.58^\circ, 58.74^\circ, 70.12^\circ, 73.76^\circ$ and 87.86° are indexed as the (111), (200), (220), (311), (220), (400) planes of TaC (Cubic, Fm-3m(225)). The blue vertical lines indicate the peaks of the cubic TaC reflections (PDF#35-0801)

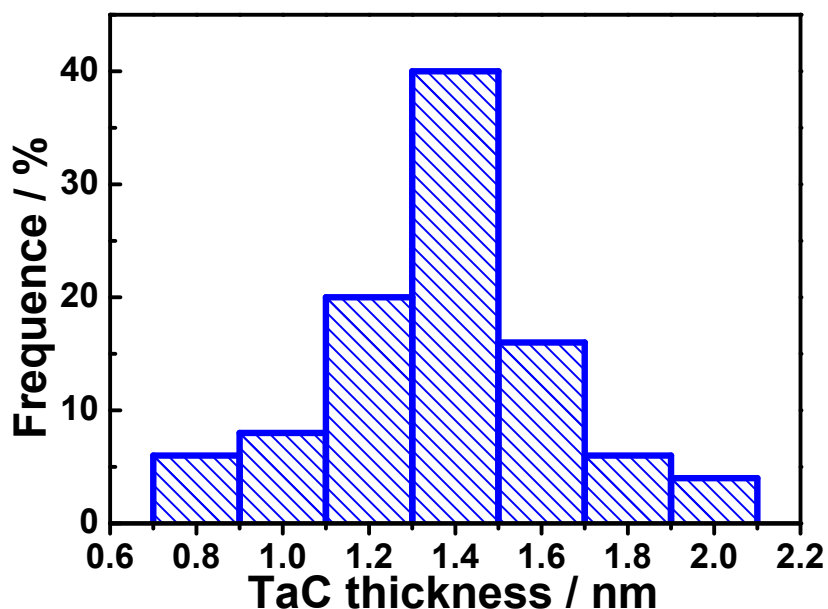


Fig. S2 The thickness distribution histograms of the ultrathin TaC nanosheets.

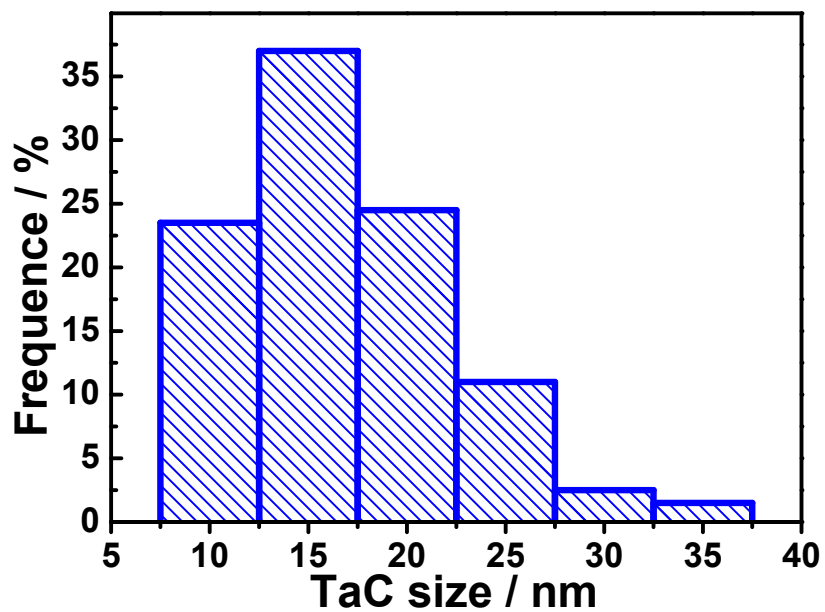


Fig. S3 Size distribution histograms of the ultrathin TaC nanosheets.

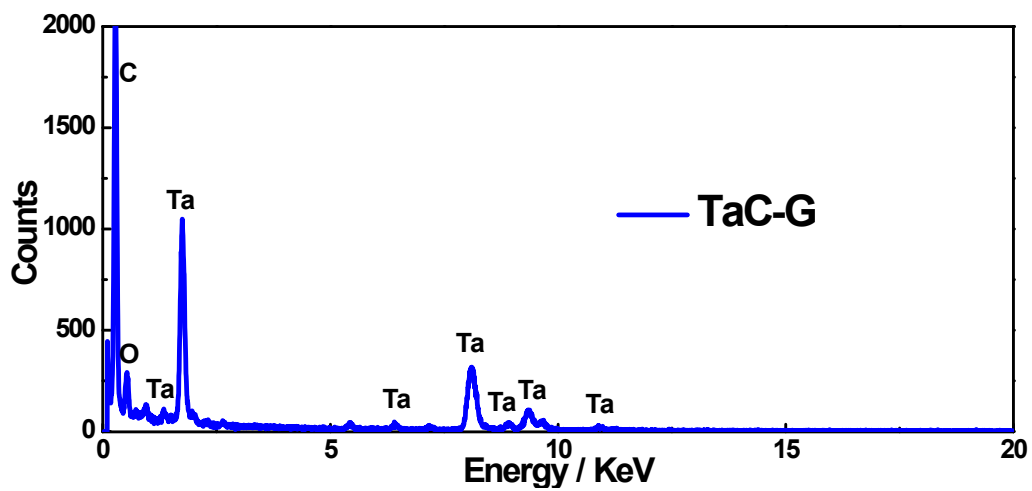


Fig. S4 The EDS pattern of TaC-G, showing the coexist of carbon, oxygen and tantalum elements.

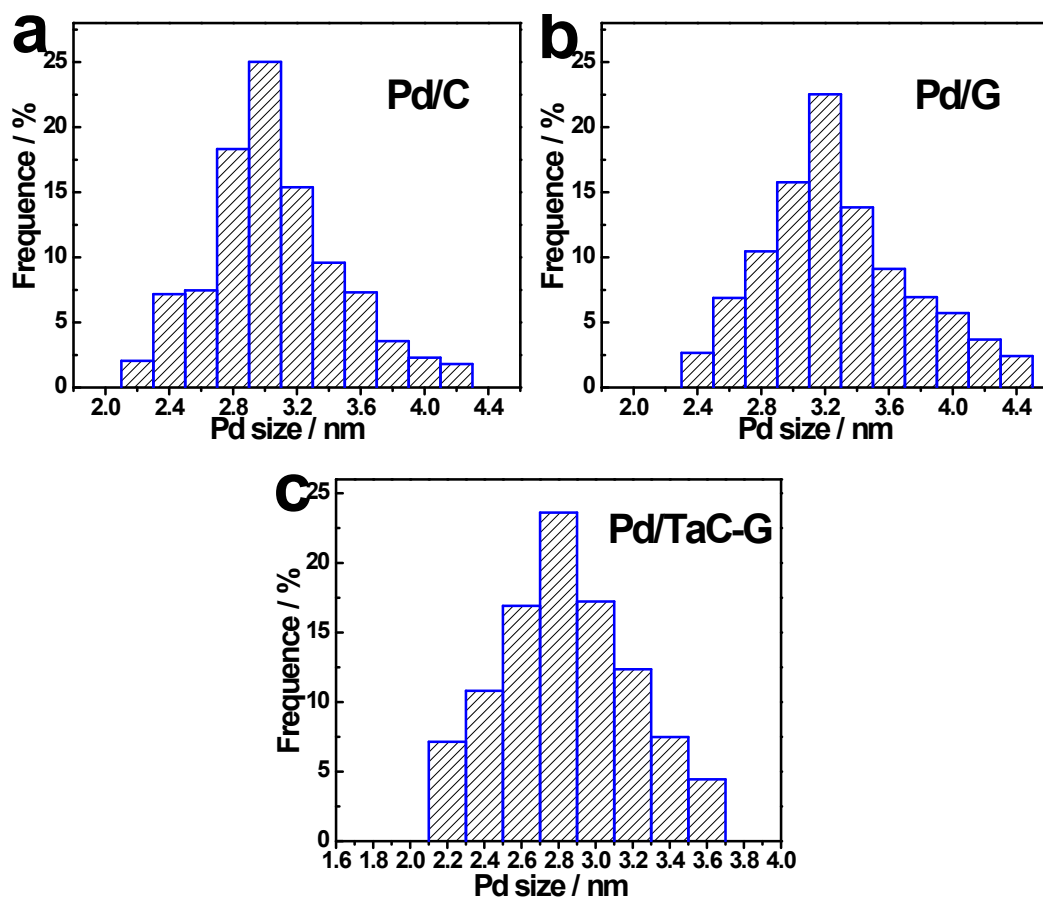


Fig. S5 Size-distribution histograms of the Pd/C (a), Pd/G (b) and Pd/TaC-G (c).

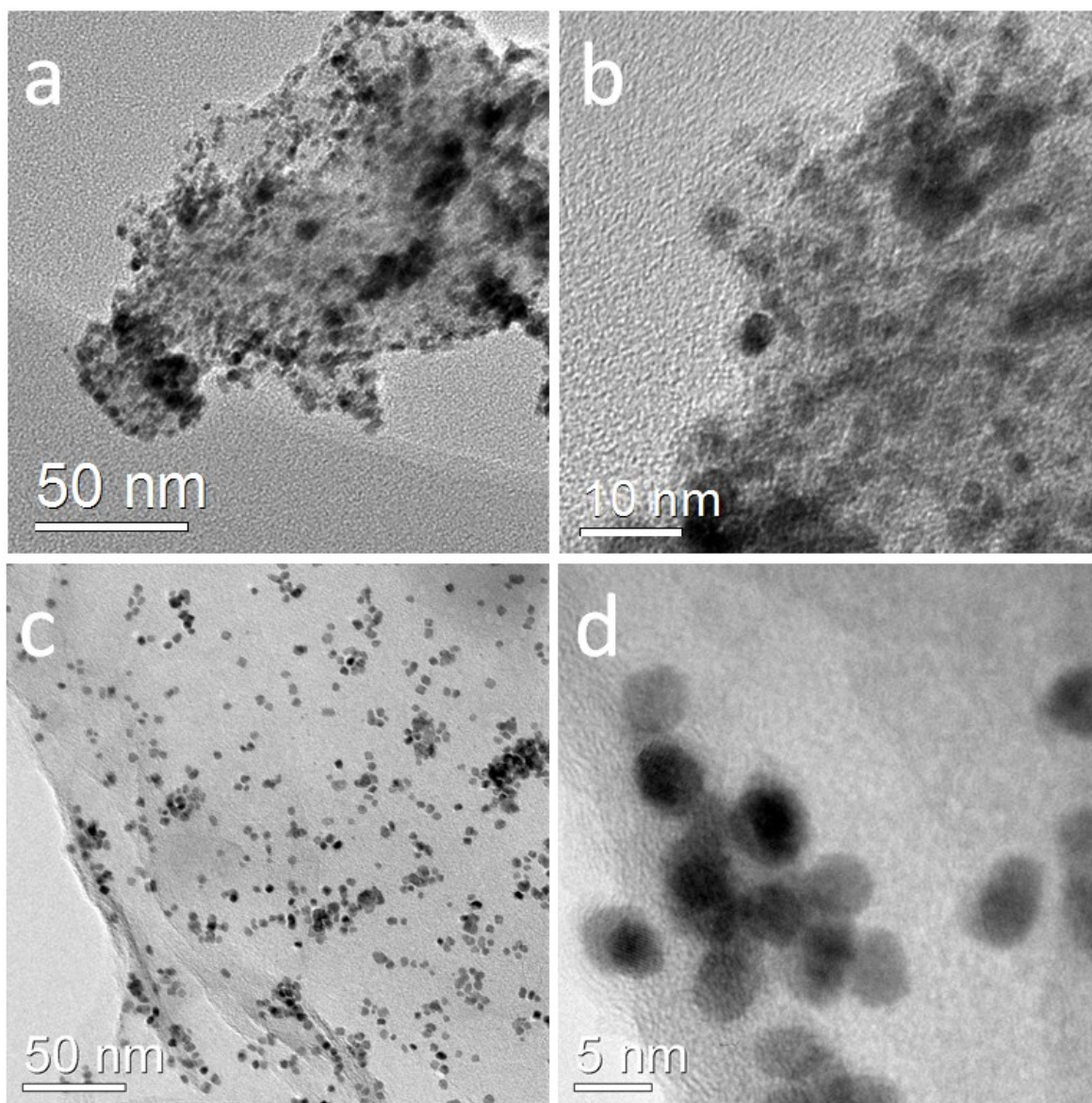


Fig. S6 (a,b) TEM images of Pd/C, (c,d) TEM images of Pd/G,

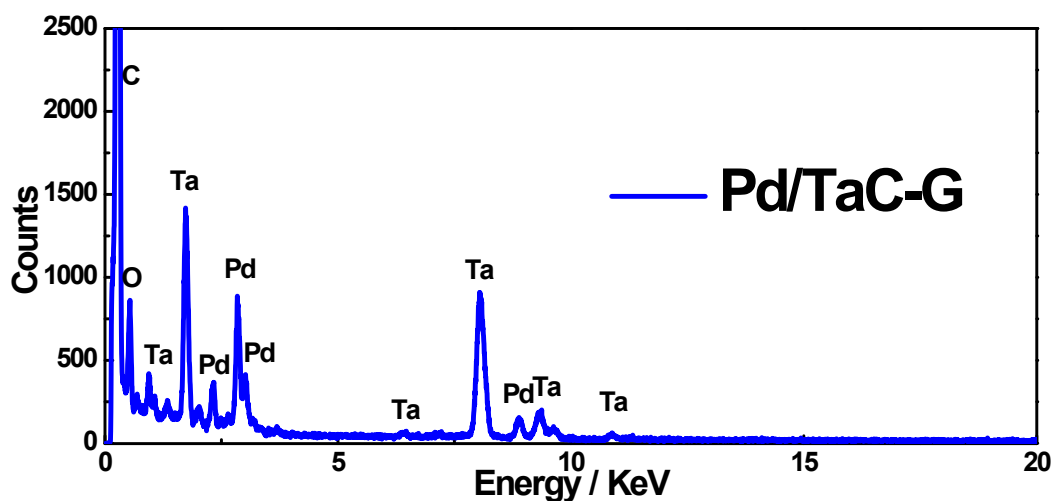


Fig. S7 The EDS pattern of Pd/TaC-G, showing the coexist of carbon, oxygen, tantalum and palladium elements.

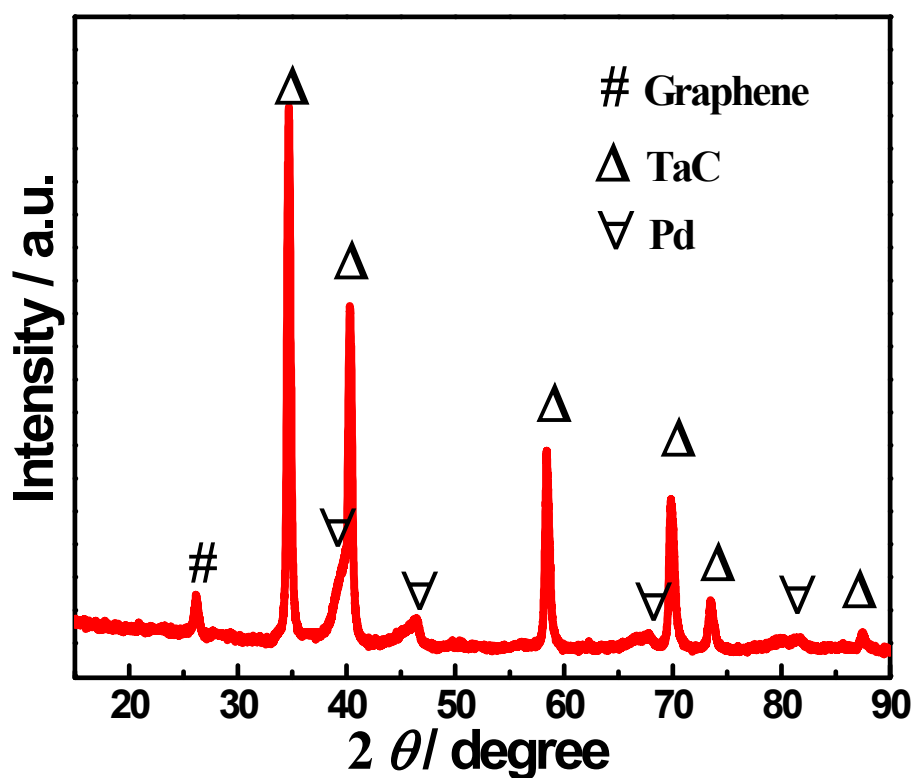


Fig. S8 The XRD pattern of Pd/TaC-G, showing the Pd and TaC phases unambiguously. The peaks at 2θ of 40.06° , 46.23° , 67.83° and 81.67° correspond to the (111), (200), (220) and (311) facets of Pd, showing a typical face-centered cubic (fcc) structure.

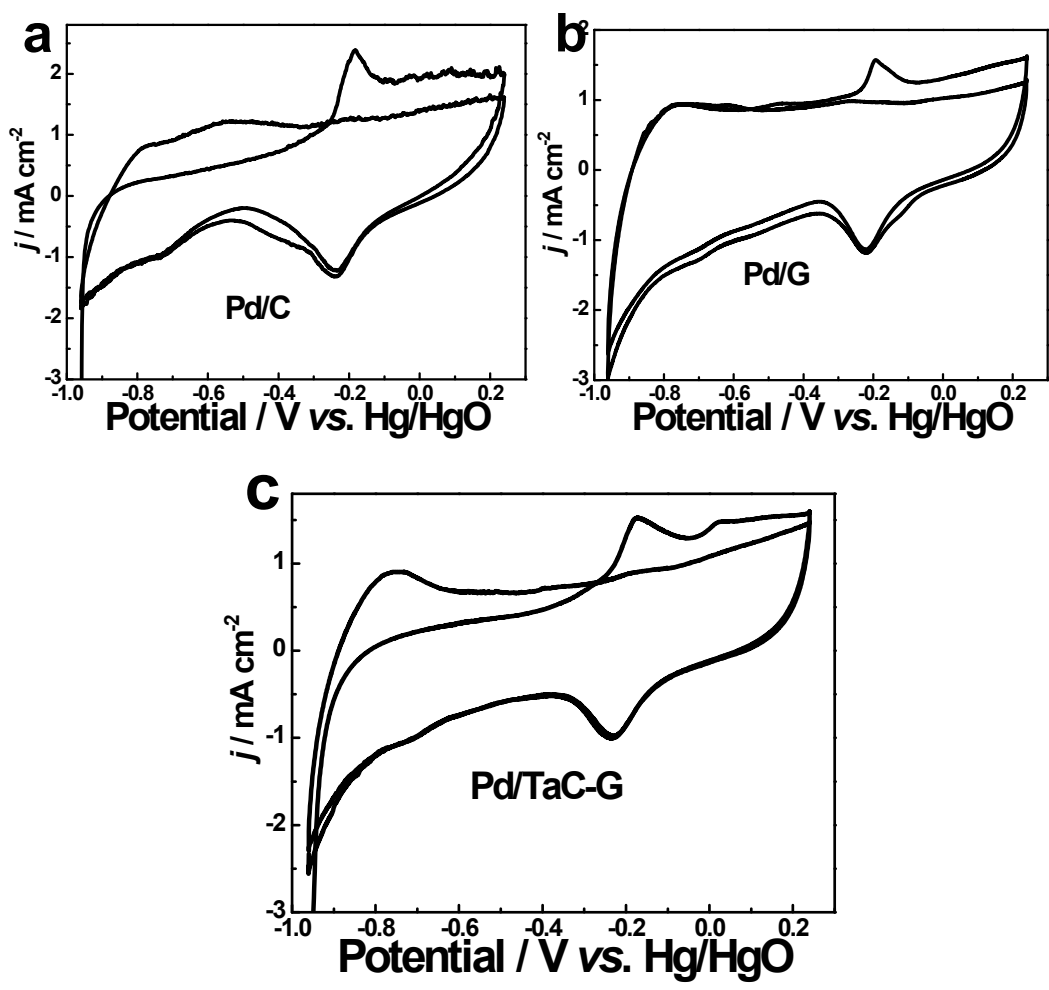


Fig. S9 CO-stripping of Pd/C, Pd/TaC-G and Pd/TaC-G in 1.0 mol L⁻¹ KOH solution of Pd/C (a), Pd/G (b) and Pd/TaC-G (c).

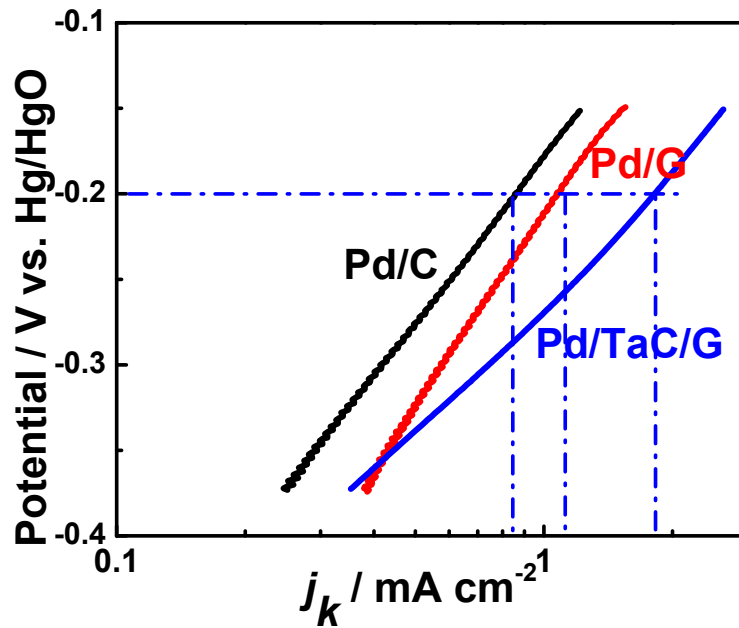


Fig. S10 shows the comparison of specific activities (j_k , calculated from normalizing the electrode current to the ECSA that obtained from the CO stripping) of Pd/C, Pd/G and Pd/TaC-G. The Pd/TaC-G (1.81 mA cm^{-2}) exhibits about 2.1-fold and 1.7-fold enhancement in specific activity compared to that of Pd/C (0.85 mA cm^{-2}) and Pd/G (1.08 mA cm^{-2}) at -0.2 V (vs. Hg/HgO), respectively.

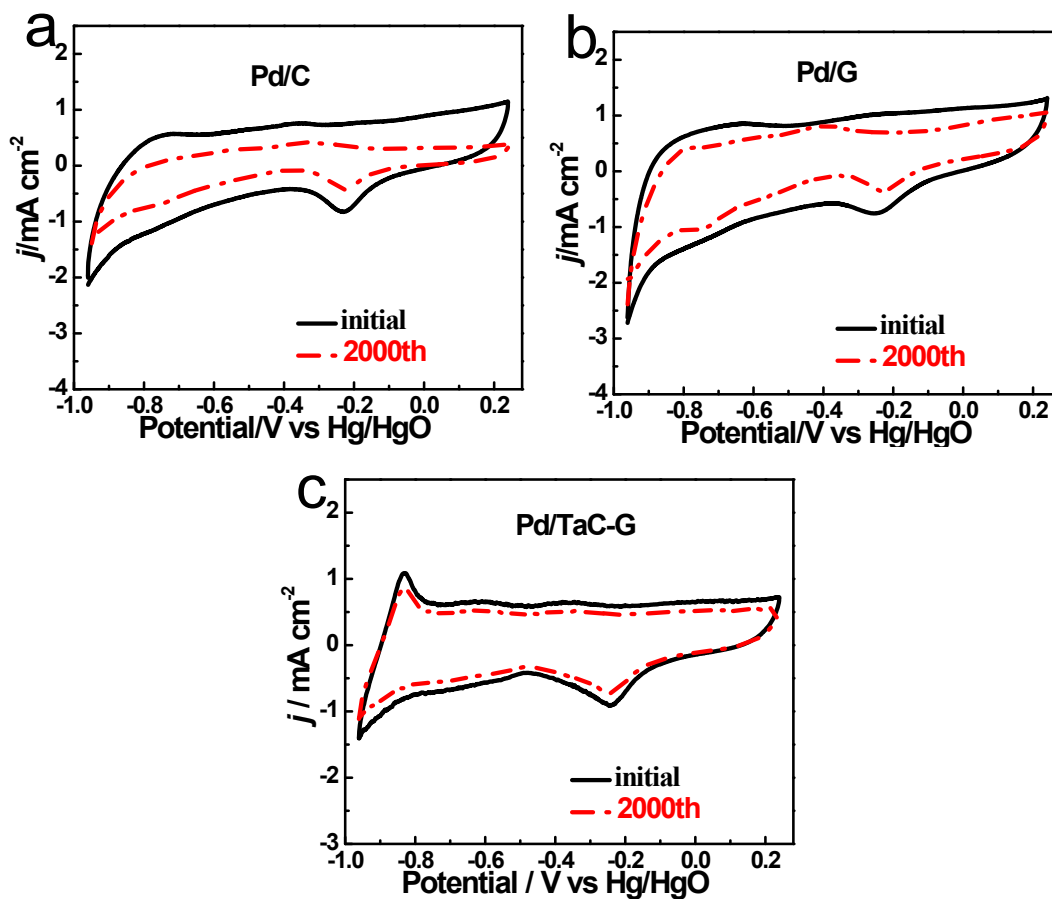


Fig. S11 CVs of Pd/C (a), Pd/G (b) and Pd/TaC-G (c) catalysts before and after the continuous cycling test in N_2 -saturated in 1.0 mol L^{-1} KOH solution.

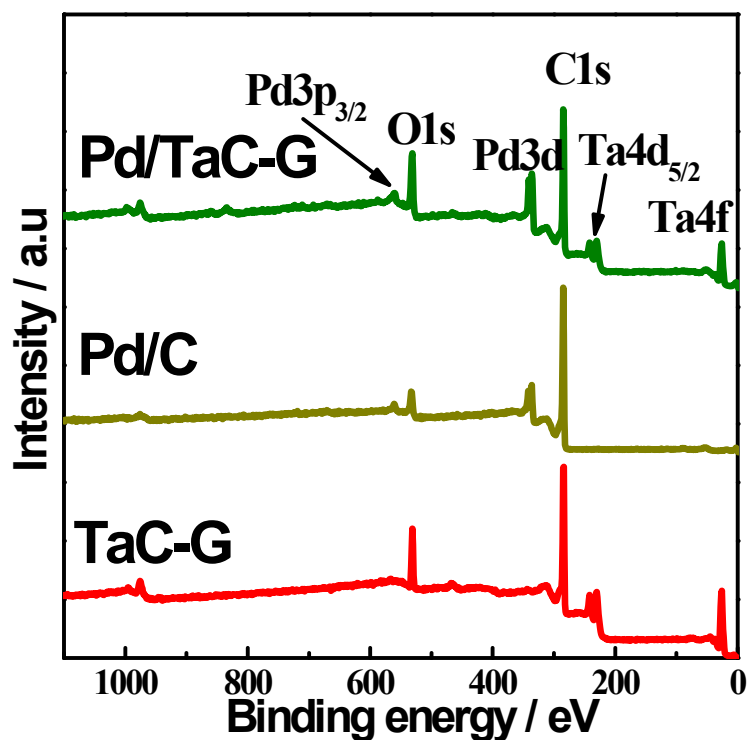


Fig. S12 XPS spectra of the TaC-G, Pd/C and Pd/TaC-G.