

# **Enhanced activity and stability of Co<sub>3</sub>O<sub>4</sub>-decorated nitrogen-doped carbon hollow sphere catalysts for microbial fuel cells**

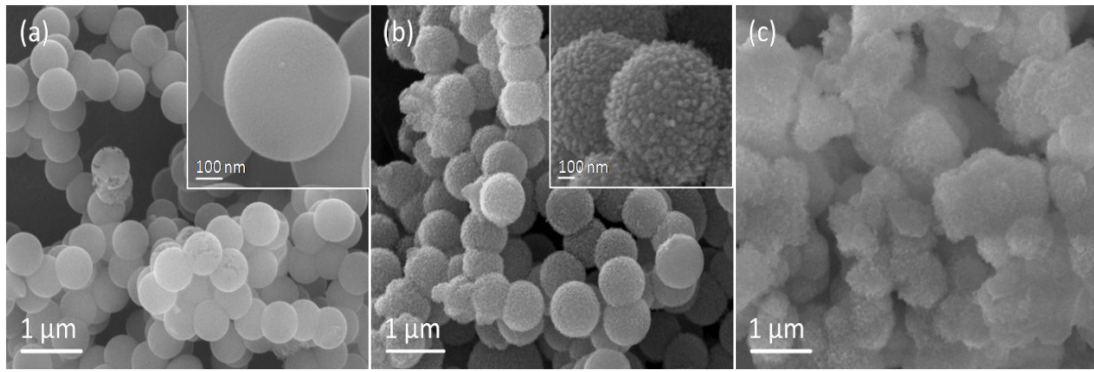
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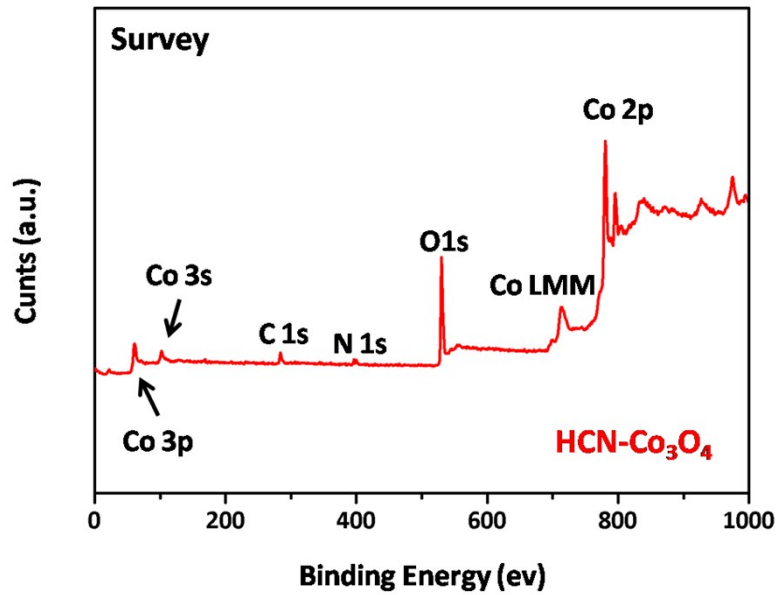
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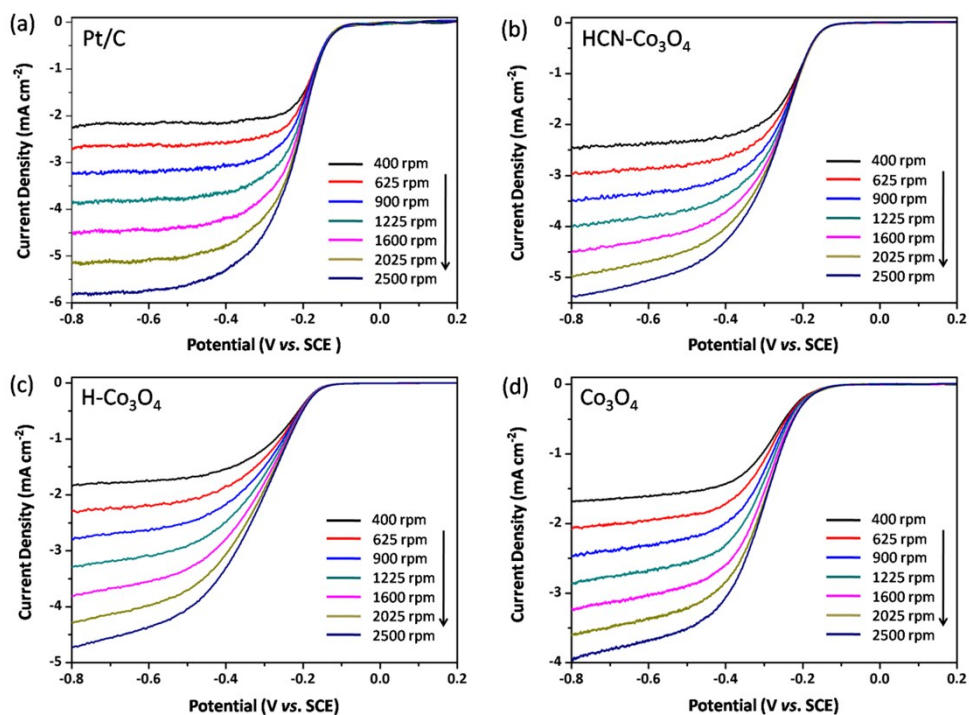
*E-mail: [nanli@gzhu.edu.cn](mailto:nanli@gzhu.edu.cn) (N. Li); [lzqgz@gzhu.edu.cn](mailto:lzqgz@gzhu.edu.cn) (Z. Q. Liu)*



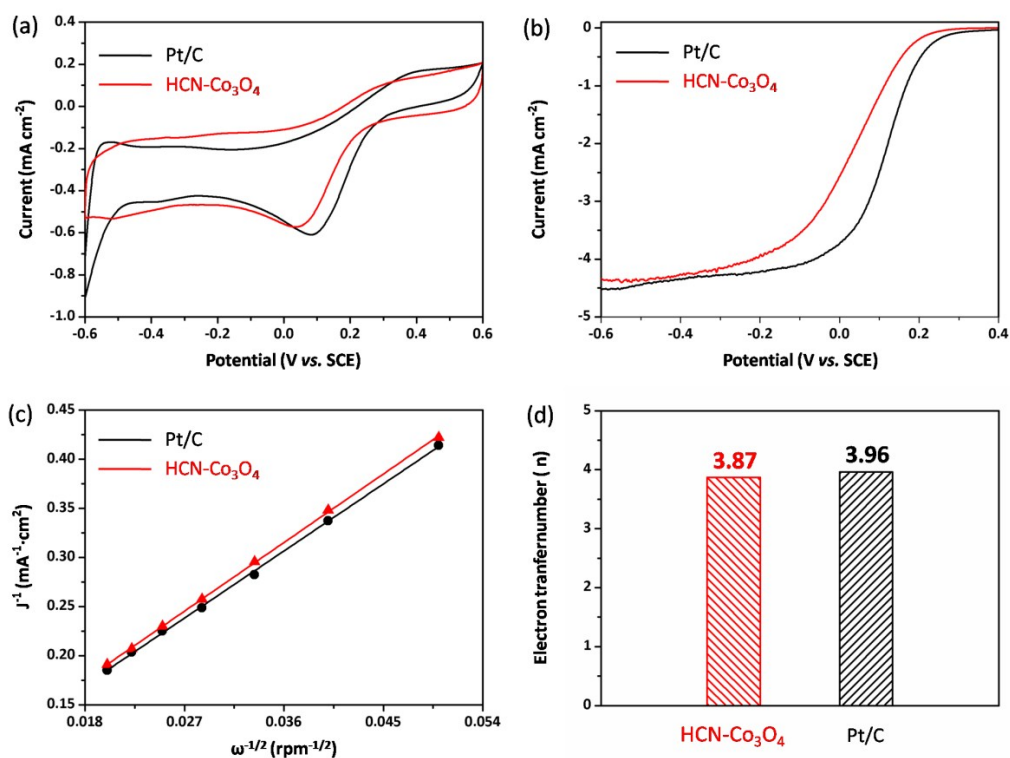
**Fig. S1.** (a) SEM images of nitrogen-doped carbon spheres; (b) SEM image of CN-Co<sub>3</sub>O<sub>4</sub>; (c) SEM image of Co<sub>3</sub>O<sub>4</sub>.



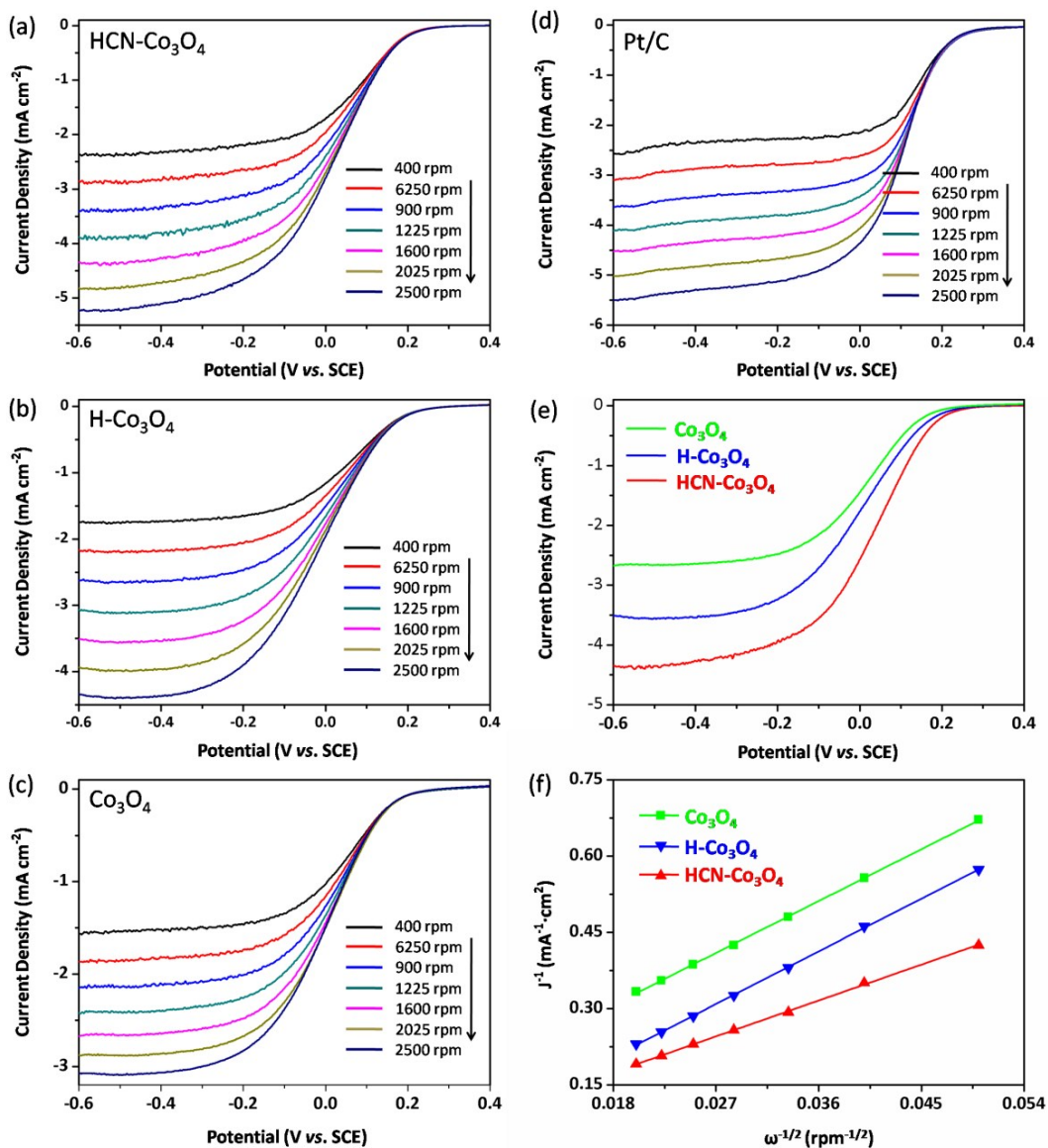
**Fig. S2.** XPS survey spectra of HCN-Co<sub>3</sub>O<sub>4</sub>.



**Fig. S3.** LSV curves under the different rotational speed from 400 to 2500 rpm at the 0.1 M KOH solution for different samples.



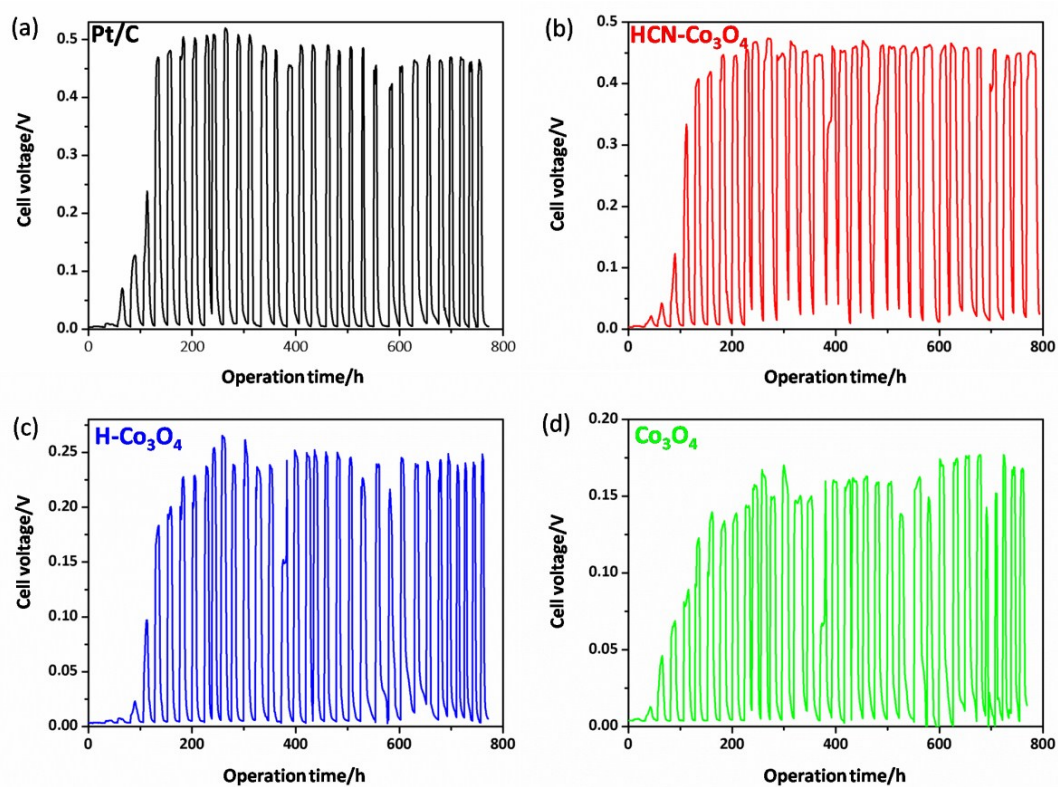
**Fig. S4.** (a) CV and (b) LSV curves of Pt/C and HCN-Co<sub>3</sub>O<sub>4</sub> obtained at 0.05 M PBS; (c) K-L curves and (d) electron transfer numbers of Pt/C and HCN-Co<sub>3</sub>O<sub>4</sub>.



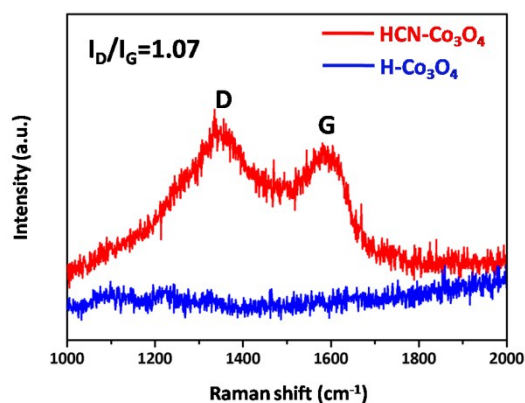
**Fig. S5.** (a-d) LSV curves under the different rotational speed from 400 to 2500 rpm at the 0.05 M PBS solution for different samples; (e) LSV and K-L curves of Co<sub>3</sub>O<sub>4</sub>, H-Co<sub>3</sub>O<sub>4</sub> and HCN-Co<sub>3</sub>O<sub>4</sub> obtained at 1600 rpm at the 0.05 M PBS.

**Table S1.** Main parameter from Nyquist curve.

Para	HCN-Co <sub>3</sub> O <sub>4</sub>	H-Co <sub>3</sub> O <sub>4</sub>	Co <sub>3</sub> O <sub>4</sub>
$R_s$ ( $\Omega$ )	8.07	8.37	8.61
$R_{ct}$ ( $\Omega$ )	1.02	1.98	1.24
$Z_w$ ( $\Omega$ S <sup>-1/2</sup> )	$3.19 \times 10^{-6}$	$1.32 \times 10^{-6}$	$9.45 \times 10^{-5}$
$C$ ( $\Omega^{-1}$ s <sup>n</sup> cm <sup>-2</sup> )	$8.38 \times 10^{-6}$	$3.63 \times 10^{-6}$	$5.93 \times 10^{-6}$
$n$	0.91	0.92	0.85



**Fig. S6.** (a-d) Voltage-Time profile of MFCs with different cathode catalysts.



**Fig. S7.** Raman spectra of HCN-Co<sub>3</sub>O<sub>4</sub> and H-Co<sub>3</sub>O<sub>4</sub>.

we have carefully characterized the HCN-Co<sub>3</sub>O<sub>4</sub> and H-Co<sub>3</sub>O<sub>4</sub> by Raman spectrum. The obtained spectrum have shown in Fig. S7, and the D band (1350 cm<sup>-1</sup>) and G band (1580 cm<sup>-1</sup>) respond to the disordered carbon and ordered carbon of the graphite carbon, respectively. There is not observable special spectra can be found in the H-Co<sub>3</sub>O<sub>4</sub> composite. On the contrary, HCN-Co<sub>3</sub>O<sub>4</sub> composite shows obvious D band and G hand in the Raman spectrum. The  $I_D/I_G$  value of 1.07 was illustrated that the disordered feature may be originated from the doped-nitrogen and Co<sub>3</sub>O<sub>4</sub> effect. This results are in great consistent with the XRD and SEM results.

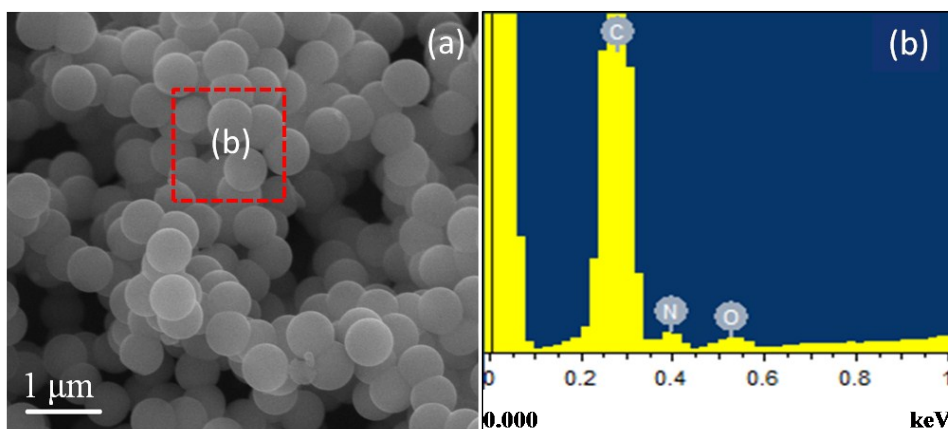


Fig. S8. (a) SEM image of the CN sphere and (b) the corresponding EDS pattern.

We have carefully characterized CN sphere by different methods. Although just a trace of N can be detected from XPS spectra of HCN- $\text{Co}_3\text{O}_4$  due to the coverage of dense  $\text{Co}_3\text{O}_4$  nanoparticles on the HCN surface, the SEM image and EDS pattern of the CN sphere still indicate the substantial doping of N species into the CN sphere framework (Fig. S8).

The MFCs performance is difficult to compare directly with other literatures since the different adopted parameters, such as the organic substrate, buffer system, inoculated bacterial strain, cell configuration, catalyst content, etc. Herein, similar MFCs device were fabricated for better comparison by using commercial Pt/C as air cathode under the same identical configuration and operation conduction. The MFCs with the HCN-Co<sub>3</sub>O<sub>4</sub> cathode catalyst has generated a maximum power density of  $553 \pm 10$  mW m<sup>-2</sup>, which is comparable to that of the Pt/C ( $535 \pm 11$  mW m<sup>-2</sup>). Moreover, the HCN-Co<sub>3</sub>O<sub>4</sub> also exhibits higher catalytic effect than that the previously reported catalysts with identical or similar mass loading. The detailed comparison was also shown in Table S2.

**Table S2.** The MFCs activity comparison with different cathodes

Reference	Mass loading [mg cm <sup>-2</sup> ]	Maximum power density	Maximum power density of Pt/C	Percentage to Pt/C (%)
<b>This work</b>	<b>3</b>	<b><math>553 \pm 10</math> mW m<sup>-2</sup></b>	<b><math>535 \pm 11</math> mW m<sup>-2</sup></b>	<b>103</b>
Mn-Co oxide [1]	0.5	$113 \pm 25$ mW m <sup>-2</sup>	$148 \pm 21$ mW m <sup>-2</sup>	76
CoNPc/C [2]	1.72	$64.7$ mW m <sup>-2</sup>	$81.3$ mW m <sup>-2</sup>	79.6
C-CoOx-FePc [3]	2.5	$654 \pm 32$ mW m <sup>-2</sup>	$850 \pm 42$ mW m <sup>-2</sup>	77.4
Co <sub>3</sub> O <sub>4</sub> /N-G [4]	1	$1340 \pm 10$ mW m <sup>-2</sup>	$1470 \pm 10$ mW m <sup>-2</sup>	91.1
CoTTP [5]	1	$158$ mW m <sup>-2</sup>	$171$ mW m <sup>-2</sup>	92.4
Bi-CoPc/C-NiO [6]	2.5	$400$ mW m <sup>-2</sup>	$450$ mW m <sup>-2</sup>	88.9
Porous Co <sub>3</sub> O <sub>4</sub> nanorods [7]	2	$503 \pm 16$ mW m <sup>-2</sup>	-	Comparable



$(\text{Cu}_{0.3}\text{Co}_{0.7})\text{Co}_2\text{O}_4$ <sup>[8]</sup>	0.5	567 mW m <sup>-2</sup>	647 mW m <sup>-2</sup>	87
N-G@CoNi/BCNT <sup>[9]</sup>	5	$2.0 \pm 0.1$ W m <sup>-2</sup>	$2.6 \pm 0.2$ W m <sup>-2</sup>	76.9
Urchin-like NiCo <sub>2</sub> O <sub>4</sub> /AC <sup>[10]</sup>	3	$1730 \pm 14$ mW m <sup>-2</sup>	-	Comparable

## References

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