Electronic Supplementary Material (ESI) for Dalton Transactions. This journal is © The Royal Society of Chemistry 2016

Supporting Information

Central-metal Exchange, Improved Catalytic Activity, Photoluminescence Properties of A New Family of d¹⁰ Coordination Polymers Based on 5,5'-(1H-2,3,5-triazole-1,4diyl)diisophthalic Acid Ligand

Huarui Wang^{a,b}, Chao Huang^a, Yanbing Han^a, Zhichao Shao^a, Hongwei Hou^{a,*}, Yaoting Fan^a

^a College of Chemistry and Molecular Engineering, Zhengzhou University, Zhengzhou, Henan, 450052, P. R. China.

E-mail: <u>houhongw@zzu.edu.cn</u>; Fax :(86)371-67761744.

^b College of Chemistry and Chemical Engineering, Luoyang Normal University, Luoyang, Henan, 471022, P. R. China.

Contents

- 1. PXRD analysis of 1-5.
- 2. TGA analysis of 1-5.
- 3. Photoluminescence Properties of H_4L and 1-5.
- 4. Selected Bond Distances (Å) and Angles (deg) for 1-5.
- 5. PXRD analysis of 2 and exchange products 2a-2c.
- 6. Reusability experiment of 2a.
- 7. ¹H NMR spectra for 7a and 7b.

1. PXRD analysis of 1-5.



Fig. S1 Experimental and simulated PXRD patterns of 1-5 (as-synthesized).



Fig. S2 TGA curves of 1-5.

3. Photoluminescence Properties of H_4L and 1-5.



Fig. S3 Emission decay trace and biexponential fit for the free ligand H₄L at room temperature.



Fig. S4 Emission decay trace and biexponential fit for 1 (λ_{em} = 439 nm) at room temperature.



Fig. S5 Emission decay trace and biexponential fit for 2 (λ_{em} = 423 nm) at room temperature.



Fig. S6 Emission decay trace and triexponential fit for 3 (λ_{em} = 441 nm) at room temperature.



Fig. S7 Emission decay trace and biexponential fit for 4 (λ_{em} = 442 nm) at room temperature.



Fig. S8 Emission decay trace and biexponential fit for 5 (λ_{em} = 448 nm) at room temperature.

1						
Cd(1)-O(1)	2.252(4)	Cd(2)-O(5)#4	2.163(2)			
Cd(1)-O(6)#1	2.263(3)	Cd(2)-O(5)#1	2.163(2)			
Cd(1)-O(9)	2.290(4)	Cd(2)-O(2)	2.352(3)			
Cd(1)-O(8)#2	2.298(3)	Cd(2)-O(2)#5	2.352(3)			
Cd(1)-O(3)#3	2.407(3)	Cd(2)-O(3)#6	2.358(3)			
Cd(1)-O(7)#2	2.590(3)	Cd(2)-O(3)#3	2.358(3)			
Cd(1)-O(4)#3	2.592(4)	O(1)-Cd(1)-O(6)#1	131.28(12)			
O(1)-Cd(1)-O(9)	89.44(16)	O(1)-Cd(1)-O(4)#3	75.07(15)			
O(6)#1-Cd(1)-O(9)	82.59(14)	O(6)#1-Cd(1)-O(4)#3	129.92(15)			
O(1)-Cd(1)-O(8)#2	133.88(11)	O(9)-Cd(1)-O(4)#3	146.41(14)			
O(6)#1-Cd(1)-O(8)#2	93.43(11)	O(8)#2-Cd(1)-O(4)#3	84.48(12)			
O(9)-Cd(1)-O(8)#2	85.42(14)	O(3)#3-Cd(1)-O(4)#3	51.44(11)			
O(1)-Cd(1)-O(3)#3	89.55(13)	O(7)#2-Cd(1)-O(4)#3	69.03(12)			
O(6)#1-Cd(1)-O(3)#3	83.20(10)	O(5)#4-Cd(2)-O(5)#1	180.00(13)			
O(9)-Cd(1)-O(3)#3	160.13(11)	O(5)#4-Cd(2)-O(2)	88.25(11)			
O(8)#2-Cd(1)-O(3)#3	109.25(13)	O(5)#1-Cd(2)-O(2)	91.75(11)			
O(1)-Cd(1)-O(7)#2	81.18(12)	O(5)#4-Cd(2)-O(2)#5	91.75(11)			
O(6)#1-Cd(1)-O(7)#2	142.59(11)	O(5)#1-Cd(2)-O(2)#5	88.25(11)			
O(9)-Cd(1)-O(7)#2	79.38(12)	O(2)-Cd(2)-O(2)#5	180.000(1)			
O(8)#2-Cd(1)-O(7)#2	52.81(12)	O(5)#4-Cd(2)-O(3)#6	89.13(10)			
O(3)#3-Cd(1)-O(7)#2	120.02(10)	O(5)#1-Cd(2)-O(3)#6	90.87(10)			
O(2)-Cd(2)-O(3)#6	98.10(11)	O(2)-Cd(2)-O(3)#3	81.90(11)			

Table S1 Selected Bond Distances (Å) and Angles (deg) for 1-5.

O(2)#5-Cd(2)-O(3)#6	81.90(11)	O(2)#5-Cd(2)-O(3)#3	98.10(11)
O(5)#4-Cd(2)-O(3)#3	90.87(10)	O(3)#6-Cd(2)-O(3)#3	180.000(1)
O(5)#1-Cd(2)-O(3)#3	89.13(10)		
		2	
Cd(1)-O(2)#1	2.354(5)	N(2)-Cd(1)-O(4)	91.8(2)
Cd(1)-N(1)	2.369(4)	O(2)#1-Cd(1)-O(5)	141.48(17)
Cd(1)-N(3)	2.364(9)	N(1)-Cd(1)-O(5)	85.37(17)
Cd(1)-N(2)	2.376(4)	N(3)-Cd(1)-O(5)	82.0(3)
Cd(1)-O(4)	2.391(5)	N(2)-Cd(1)-O(5)	92.07(17)
O(2)#1-Cd(1)-N(3)	136.5(3)	O(4)-Cd(1)-O(5)	54.79(17)
N(1)-Cd(1)-N(3)	93.6(3)	O(2)#1-Cd(1)-O(3)#1	54.51(18)
O(2)#1-Cd(1)-N(2)	90.52(19)	N(1)-Cd(1)-O(3)#1	87.85(19)
N(1)-Cd(1)-N(2)	177.44(17)	N(3)-Cd(1)-O(3)#1	82.6(3)
N(3)-Cd(1)-N(2)	86.0(3)	N(2)-Cd(1)-O(3)#1	94.6(2)
O(2)#1-Cd(1)-O(4)	86.72(18)	O(4)-Cd(1)-O(3)#1	140.67(19)
N(1)-Cd(1)-O(4)	86.8(2)	O(5)-Cd(1)-O(3)#1	162.69(19)
N(3)-Cd(1)-O(4)	136.6(3)		
		3	
Cd(1)-O(11)	2.265(7)	Cd(2)-O(1)	2.230(7)
Cd(1)-O(8)	2.295(7)	Cd(2)-O(13)	2.262(7)
Cd(1)-O(9)	2.323(9)	Cd(2)-O(14)	2.339(10)
Cd(1)-O(6)#1	2.369(7)	Cd(2)-O(3)#2	2.350(7)
Cd(1)-O(10)	2.374(9)	Cd(2)-O(4)#2	2.436(7)
Cd(1)-O(5)#1	2.434(7)	Cd(2)-O(12)	2.444(10)
Cd(1)-O(7)	2.502(7)	O(11)-Cd(1)-O(8)	139.2(3)
O(11)-Cd(1)-O(9)	109.1(3)	O(1)-Cd(2)-O(13)	124.6(3)
O(8)-Cd(1)-O(9)	84.8(3)	O(1)-Cd(2)-O(14)	102.6(3)

O(11) Cd(1) O(6)#1	125 6(3)	O(13) Cd(2) O(14)	80 8(3)	
O(11)-Cu(1)-O(0)#1	125.0(5)	0(13)-Cu(2)-0(14)	67.6(3)	
O(8)-Cd(1)-O(6)#1	91.1(2)	O(1)-Cd(2)-O(3)#2	91.2(3)	
O(9)-Cd(1)-O(6)#1	90.6(3)	O(13)-Cd(2)-O(3)#2	142.4(3)	
O(11)-Cd(1)-O(10)	84.7(3)	O(14)-Cd(2)-O(3)#2	93.0(3)	
O(8)-Cd(1)-O(10)	78.9(3)	O(1)-Cd(2)-O(4)#2	145.2(3)	
O(9)-Cd(1)-O(10)	163.6(3)	O(13)-Cd(2)-O(4)#2	90.1(3)	
O(6)#1-Cd(1)-O(10)	88.1(3)	O(14)-Cd(2)-O(4)#2	76.6(3)	
O(11)-Cd(1)-O(5)#1	78.7(3)	O(3)#2-Cd(2)-O(4)#2	54.4(2)	
O(8)-Cd(1)-O(5)#1	142.1(2)	O(1)-Cd(2)-O(12)	91.5(3)	
O(9)-Cd(1)-O(5)#1	81.5(3)	O(13)-Cd(2)-O(12)	89.9(3)	
O(6)#1-Cd(1)-O(5)#1	54.0(2)	O(14)-Cd(2)-O(12)	163.1(3)	
O(10)-Cd(1)-O(5)#1	110.6(3)	O(3)#2-Cd(2)-O(12)	77.1(3)	
O(11)-Cd(1)-O(7)	86.8(3)	O(4)#2-Cd(2)-O(12)	86.4(3)	
O(8)-Cd(1)-O(7)	54.8(2)	O(10)-Cd(1)-O(7)	84.4(3)	
O(9)-Cd(1)-O(7)	87.5(3)	O(5)#1-Cd(1)-O(7)	157.8(2)	
O(6)#1-Cd(1)-O(7)	145.9(3)			
		4		
Zn(1)-O(1)	1.953(4)	Zn(1)-O(5)#2	1.972(4)	
Zn(1)-O(7)#1	1.957(4)	Zn(1)-O(3)#3	2.022(4)	
O(1)-Zn(1)-O(5)#2	129.27(18)	O(1)-Zn(1)-O(3)#3	108.04(19)	
O(7)#1-Zn(1)-O(5)#2	99.95(19)	O(7)#1-Zn(1)-O(3)#3	101.44(19)	
O(1)-Zn(1)-O(7)#1	116.21(18)	O(5)#2-Zn(1)-O(3)#3	97.43(18)	
		5		
Zn(1)-O(7)	1.940(3)	Zn(1)-O(6)#2	2.001(3)	
Zn(1)-O(3)#1	1.961(3)	Zn(1)-N(4)	2.018(3)	
O(7)-Zn(1)-O(3)#1	101.82(11)	O(7)-Zn(1)-N(4)	120.75(13)	
O(7)-Zn(1)-O(6)#2	111.98(12)	O(3)#1-Zn(1)-N(4)	111.16(12)	

Symmetry transformations used to generate equivalent atoms:

1: #1 x+1,-y+2,z+1/2; #2 x+1/2,y+1/2,z; #3 x,-y+2,z-1/2; #4 -x,y,-z+3/2; #5 -x+1,-y+2,-z+2; #6 -x+1,y,-z+5/2.

2: #1 -x+1/2,y+1/2,-z+1/2.

3: #1 x,-y+1/2,z+1/2; #2 -x,y-1/2,-z+1/2.

4: #1 x+1,y+1,z; #2 -x+2,y+1/2,-z+1/2; #3 -x+2,-y+2,-z.

5: #1 x+1,y,z+1; #2 -x+2,-y+2,-z+2; #3 x-1,y,z-1; #4 -x+4,-y+1,-z+2.

5. PXRD analysis of 2 and exchange products 2a-2c.



Fig. S9 Experimental and simulated PXRD patterns of 2 and 2a-2c (as-synthesized).



Fig. S10 Comparison of the PXRD patterns of 2a before and after catalysis.

6. Reusability experiment of 2a.



Fig. S11 Recyclability test of complex 2a for the synthesis of 7b.

7. 1H NMR spectra for 7a and 7b.

2-Phenyl-1,4,5,6-tetrahydropyrimidine (7a).

¹H NMR (300 MHz, CDCl3): δ = 7.62-7.60 (d, J = 6.6 Hz, 2 H), 7.36–7.29 (m, 3 H), 3.41-3.37 (t, J = 5.7 Hz, 4 H), 1.80–1.73 (m,

2 H).

2-(Pyridin-4-yl)-1,4,5,6-tetrahydropyrimidine (7b).

¹H NMR (300 MHz, CDCl3): δ = 8.61-8.59 (d, J = 8.1 Hz, 2 H), 7.54-7.52 (d, J = 8.4 Hz, 2 H), 3.51-3.47 (t, J = 5.7 Hz, 4 H),

1.87–1.81 (m, 2 H).

¹H NMR data of **7a-b** are matched with those in previous papers.¹



¹H NMR spectrum of compound **7a**



References:

(1) (*a*) S. J. An, B. Yin, P. Liu, L X. G. i, C. Li, J. L. Li, Z. Shi, *Synthesis*, 2013, **45**, 2525–2532. (*b*) M. Ishihara, H. Togo, *Tetrahedron*, 2007, **63**, 1474–1480.