Electronic Supporting Information

Regioisomerism in cationic sulfonyl-substituted [Ir(C^N)₂(N^N)]⁺ complexes: influence on photophysical properties and LEC performance

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Fig. S1 Normalized photoluminescence spectra of complexes $[Ir(C^N)_2(bpy)][PF_6]$ with $C^N = [1]^-$ to $[3]^-$ in the thin film configuration used in LEC devices. The complex is mixed with the ionic liquid $[Bmim][PF_6]$ in a complex:IL 4:1 molar ratio.



Fig. S2. Cyclic voltammograms showing the first oxidation and reduction processes of $[Ir(C^N)_2(bpy)][PF_6]$ (C^N = [1]⁻, [2]⁻ or [3]⁻) in MeCN solution referenced to Fc/Fc⁺ with 0.1 M [ⁿBu₄][PF₆] as supporting electrolyte and a scan rate of 0.1 V s⁻¹.



Fig. S3 Theoretical simulation of the absorption spectrum of complexes $[Ir(C^N)_2(bpy)][PF_6]$ with $C^N = [1]^-$ to $[3]^-$ obtained from TD-DFT/B3LYP/(6-31G** + LANL2DZ) calculations of the 40 lowestenergy singlet excited states. The spectra have been obtained as convoluted sums of Lorentzian curves. Each curve is centered on the wavelength value calculated for a singlet excited state, and its area is proportional to the oscillator strength.



Fig. S4 Normalized electroluminescence spectra of ITO/PEDOT:PSS/active layer/Al LEC devices measured by applying a block-wave pulsed current of 100 A m⁻² at a frequency of 1 kHz and duty cycles of 50%. Active layer: $[Ir(C^N)_2(bpy)][PF_6]$ (C^N = [1]⁻ to [3]⁻) : $[Bmim][PF_6]$ 4:1 molar ratio.

Table S1 Lowest singlet excited states calculated at the TD-DFT B3LYP/(6-31G**+LANL2DZ) level for complexes $[Ir(1)_2(bpy)]^+$, $[Ir(2)_2(bpy)]^+$ and $[Ir(3)_2(bpy)]^+$ in acetonitrile. Vertical excitation energies (*E*), oscillator strengths (*f*), dominant monoexcitations with contributions (within parentheses) greater than 20% and description of the excited state. H and L denote HOMO and LUMO, respectively.

Complex	State	<i>E</i> (eV/nm)	f	Monoexcitations	Description
[lr(1) ₂ (bpy)] ⁺	S ₁	2.81/441	0.000	$H \rightarrow L (98)$	¹ MLCT/LLCT
	S ₂	3.12/397	0.067	$H \rightarrow$ L+1 (97)	¹ LC
	S ₅	3.55/349	0.097	H-1 \rightarrow L (50)	¹ MLCT/LLCT
[lr(2) ₂ (bpy)] ⁺	S ₁	2.89/429	0.000	$H \rightarrow L (99)$	¹ MLCT/LLCT
	S ₂	3.30/375	0.058	$H \rightarrow$ L+1 (96)	¹ LC
	S ₅	3.55/350	0.084	$H-1 \rightarrow L(51)$	¹ MLCT/LLCT
[lr(3) ₂ (bpy)] ⁺	S ₁	2.79/445	0.000	$H \rightarrow L (98)$	¹ MLCT/LLCT
	S ₂	3.05/406	0.067	$H \rightarrow$ L+1 (97)	¹ LC
	S ₅	3.53/351	0.084	H-1 \rightarrow L (50)	¹ MLCT/LLCT