

Supplementary Information

**Improvement of hydrogen storage property of three-component
 $\text{Mg}(\text{NH}_2)_2\text{-LiNH}_2\text{-LiH}$ composites by additives**

Huai-Jun Lin^{a,*}, Hai-Wen Li^{a,c}, Biswajit Paik^a, Jianhui Wang^{a,d} and Etsuo Akiba^{a,b,c}

^a International Research Center for Hydrogen Energy, Kyushu University, Fukuoka, Japan

^b Department of Engineering, Kyushu University, Fukuoka, Japan

^c International Institute of Carbon-Neutral Energy Research (WPI-I²CNER), Kyushu University,
Fukuoka, Japan

^d Department of Chemical System Engineering, The University of Tokyo, Tokyo, Japan

* Corresponding author: lin.huaijun.916@m.kyushu-u.ac.jp

Table S1 Values of various functions of α in some commonly used reaction equations together with those extracted from Mg(NH₂)₂-LiNH₂-4LiH and the Mg(NH₂)₂-LiNH₂-3.9LiH-0.1KH samples.

α	Value for $t/t_{0.5}$								Mg(NH ₂) ₂ -LiNH ₂ -4LiH		Mg(NH ₂) ₂ -LiNH ₂ -3.9LiH-0.1KH		
	D ₁ (α)	D ₂ (α)	D ₃ (α)	D ₄ (α)	F ₁ (α)	R ₂ (α)	R ₃ (α)	A ₂ (α)	A ₃ (α)	140 °C	180 °C	140 °C	180 °C
0.1	0.040	0.033	0.028	0.032	0.152	0.174	0.165	0.390	0.533	0.20	0.24	0.66	0.63
0.2	0.160	0.140	0.121	0.135	0.322	0.362	0.349	0.567	0.685	0.38	0.44	0.76	0.75
0.3	0.360	0.328	0.295	0.324	0.515	0.556	0.544	0.717	0.801	0.58	0.59	0.83	0.84
0.4	0.640	0.609	0.576	0.595	0.737	0.768	0.762	0.858	0.903	0.78	0.79	0.90	0.92
0.5	1.000	1.000	1.000	1.000	1.000	1.000	1.000	1.000	1.000	1.00	1.00	1.00	1.00
0.6	1.440	1.521	1.628	1.541	1.322	1.253	1.277	1.150	1.097	1.22	1.45	1.04	1.07
0.7	1.960	2.207	2.568	2.297	1.737	1.543	1.607	1.318	1.198	1.51	2.31	1.12	1.14
0.8	2.560	3.115	4.051	3.378	2.322	1.887	2.014	1.524	1.322	1.96	4.00	1.35	1.24
0.9	3.240	4.363	6.747	5.028	3.322	2.334	2.602	1.822	1.492	3.28	8.65	1.95	1.49

Table S2 selected compositions of the Mg(NH₂)₂-LiNH₂-LiH system in this study.

Mg(NH ₂) ₂	LiNH ₂	LiH	KH
1	1	4	—
1	1	3.9	0.1
2	1	4	—
2	1	3.9	0.1
1	2	4	—
1	2	3.9	0.1
1	1	5	—
1	1	4.9	0.1
2	1	5	—
2	1	4.9	0.1
1	2	5	—
1	2	4.9	0.1
1	1	6	—
1	1	5.9	0.1
2	1	6	—
2	1	5.9	0.1
1	2	6	—
1	2	5.9	0.1

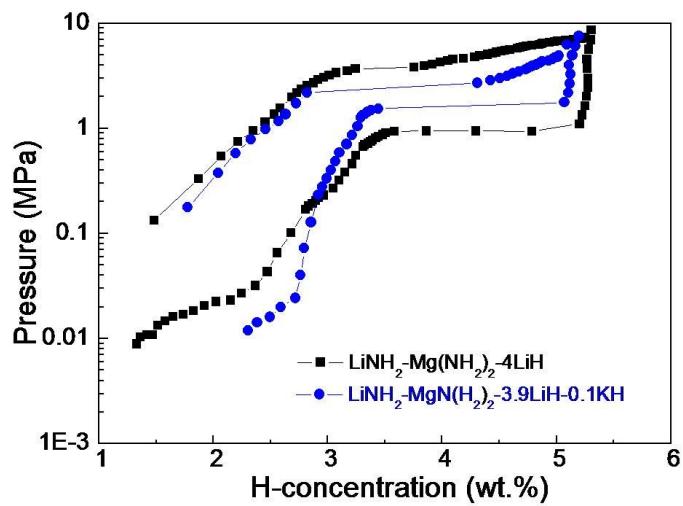


Fig. S1 experimental PCIs of the $\text{Mg}(\text{NH}_2)_2\text{-LiNH}_2\text{-}4\text{LiH}$ and $\text{Mg}(\text{NH}_2)_2\text{-LiNH}_2\text{-}3.9\text{LiH-}0.1\text{KH}$ samples at 180°C. The hysteresis of the PCIs is significantly reduced to 0.82 MPa for the $\text{Mg}(\text{NH}_2)_2\text{-LiNH}_2\text{-}3.9\text{LiH-}0.1\text{KH}$ sample compared with 2.78 MPa for the $\text{Mg}(\text{NH}_2)_2\text{-LiNH}_2\text{-}4\text{LiH}$ sample.

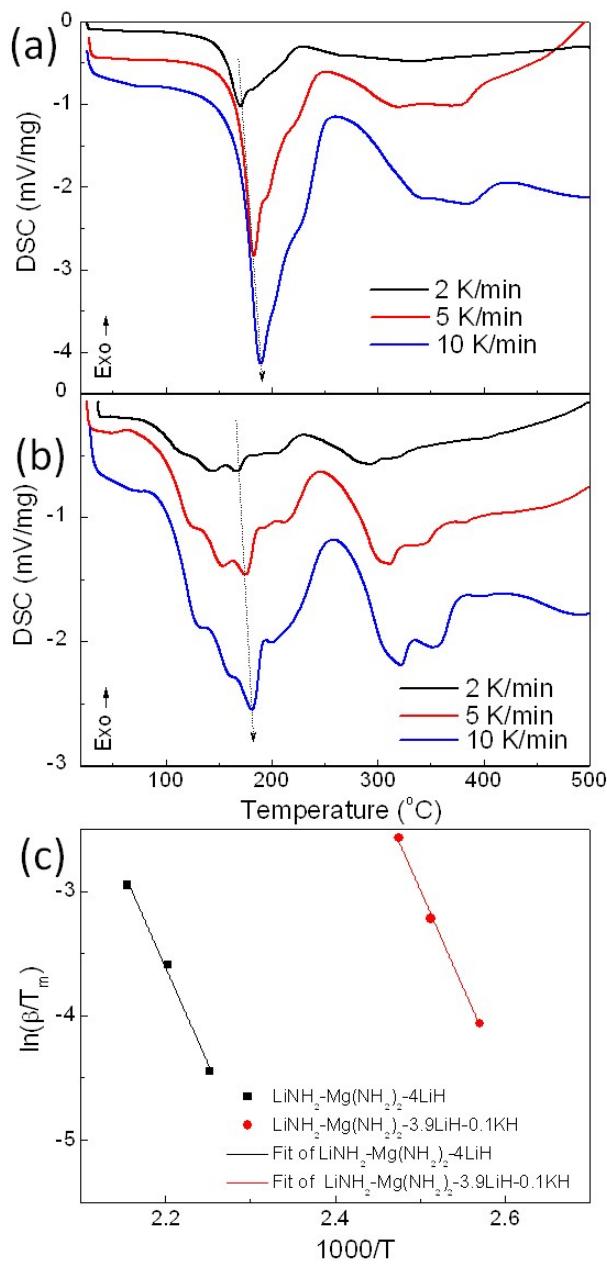


Fig. S2 DSC traces of (a) $\text{Mg}(\text{NH}_2)_2\text{-LiNH}_2\text{-4LiH}$ and (b) $\text{Mg}(\text{NH}_2)_2\text{-LiNH}_2\text{-3.9LiH-0.1KH}$ samples at different heating rates. (c) Kissinger's plots for the above two samples.

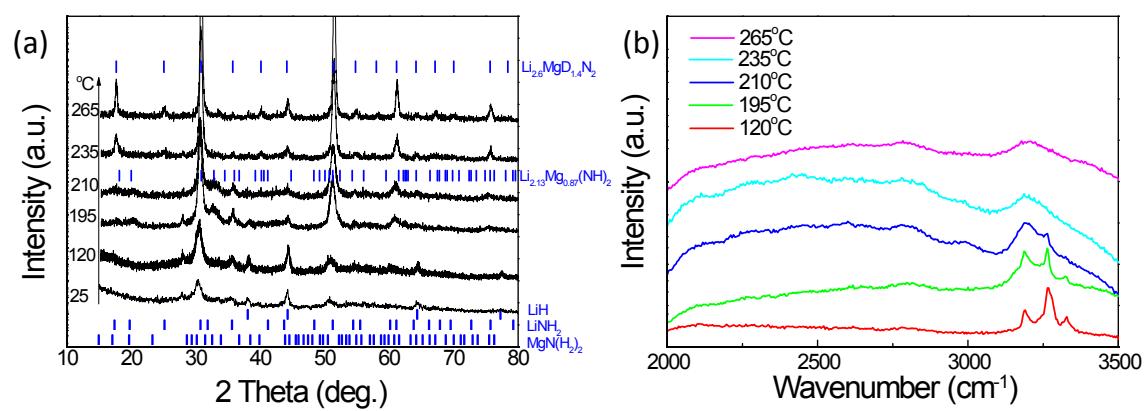


Fig. S3 Evolution of (a) XRD patterns and (b) Raman spectra of the $\text{Mg}(\text{NH}_2)_2\text{-LiNH}_2\text{-4LiH}$ composite during dehydrogenation.

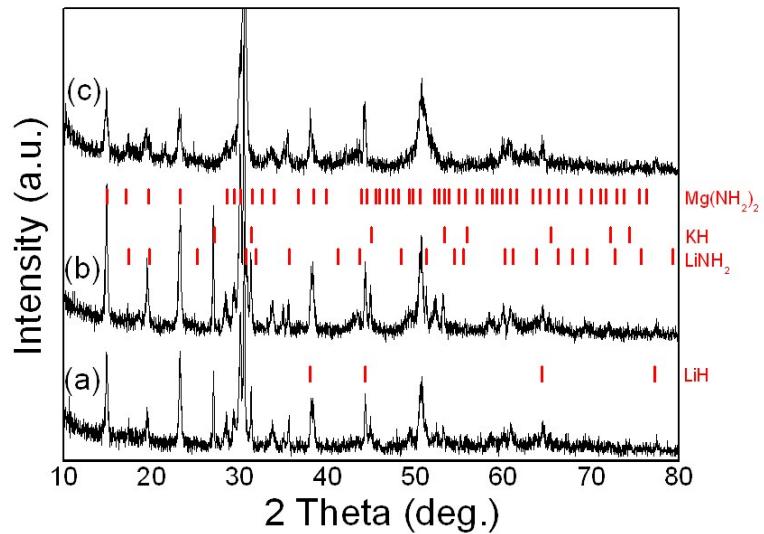


Fig. S4 XRD patterns of (a) 1st cycled and (b) 12th cycled $\text{Mg}(\text{NH}_2)_2\text{-LiNH}_2\text{-3.9LiH}\text{-0.1KH}$, and (c) 12th cycled $\text{Mg}(\text{NH}_2)_2\text{-LiNH}_2\text{-4LiH}$, all in hydrogenated state.