

ESI for

Complexes of (η^5 -Cp*)Ir(III) with 1-Benzyl-3-Phenylthio/selenomethyl-1,3-Dihydrobenzoimidazole-2-Thione/Selenone: Catalysts for Oxidation and 1,2-Substituted Benzimidazole Synthesis

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List of content

1. Table S1 of the crystallographic and Refinement Data of **1-4**
2. TableS2 of Bond Lengths and Bond Angles of **1- 4**
3. Table S3 of Non-covalent Interactions C–H···F Distances (Å) of Complexes **1- 4**
4. Figure of Secondary C–H···F Interactions
5. Mass and NMR Spectra
6. NMR of compound (3a-3g, 5a-g, 7a-d)

Table S1. Crystal data and structural refinement parameters for complexes 1-4

Compounds	1	2	3	4
Empirical formula	C ₃₁ H ₃₃ ClF ₆ IrN ₂ PSe ₂	C ₃₁ H ₃₃ ClF ₆ IrN ₂ PSeS	C ₃₁ H ₃₃ ClF ₆ IrN ₂ PSSe	C ₃₁ H ₃₃ ClF ₆ IrN ₂ PS ₂
Formula wt.	964.15	917.25	917.25	870.35
Crystal size [mm]	0.34×0.25×0.22	0.43 × 0.21 × 0.19	0.38×0.24×0.17	0.31×0.28×0.13
Crystal system	Monoclinic	Monoclinic	Monoclinic	Monoclinic
Space group	<i>P</i> 21/ <i>n</i>	<i>P</i> 21/ <i>n</i>	<i>P</i> 21/ <i>n</i>	<i>P</i> 21/ <i>n</i>
Unit Cell dimension	<i>a</i> = 8.0384(18) <i>b</i> = 19.300(4) <i>c</i> = 22.282(5) α = 90.00° β = 91.219(4) γ = 90.00°	<i>a</i> = 7.980(4) <i>b</i> = 19.271(9) <i>c</i> = 22.304(10) α = 90.00 β = 91.000(9) γ = 90.00	<i>a</i> = 8.0837(16) <i>b</i> = 19.099(4) <i>c</i> = 22.104(4) α = 90.00 β = 91.680(3) γ = 90.00	<i>a</i> = 8.0195(14) <i>b</i> = 19.158(3) <i>c</i> = 22.138(4) α = 90.00 β = 91.279(3) γ = 90.00
Volume [Å ³]	3456.1(13)	3429(3)	3411.2(12)	3400.4(10)
<i>Z</i>	4	4	4	4
Density (Calc.) [Mg m ⁻³]	1.853	1.777	1.786	1.700
Absorption coeff. [mm ⁻¹]	6.150	5.197	5.225	4.232
<i>F</i> (000)	1856.0	1784.0	1784.0	1712.0
θ range [°]	2.30–19.50	2.30 – 19.04	2.13- 19.72	2.32- 22.31
Index ranges	-9 ≤ <i>h</i> ≤ 9 -22 ≤ <i>k</i> ≤ 22 -26 ≤ <i>l</i> ≤ -26	-9 ≤ <i>h</i> ≤ 9 -22 ≤ <i>k</i> ≤ 22 -26 ≤ <i>l</i> ≤ 26	-9 ≤ <i>h</i> ≤ 9 -22 ≤ <i>k</i> ≤ 22 -26 ≤ <i>l</i> ≤ 26	-9 ≤ <i>h</i> ≤ 9 -22 ≤ <i>k</i> ≤ 22 -26 ≤ <i>l</i> ≤ 26
Reflections collected	32943	31761	31700	31686
Independent reflections (<i>R</i> _{int.})	6092 (0.0760)	6045 (0.1110)	6020(0.0976)	6001(0.0831)
Max./min. Transmission	0.258 /0.172	0.620/0.493	0.371 / 0.279	0.358/0.447
Data/restraints/parameters	6092 /0/402	6045 /0/ 402	6020 /0/ 402	6001/0/402
Goodness-of-fit on <i>F</i> ²	1.275	1.179	1.189	1.156
Final <i>R</i> indices [<i>I</i> > 2σ(<i>I</i>)]	<i>R</i> ₁ = 0.0827 <i>wR</i> ₂ = 0.1363	<i>R</i> ₁ = 0.0818 <i>wR</i> ₂ = 0.1168	<i>R</i> ₁ = 0.1025 <i>wR</i> ₂ = 0.1922	<i>R</i> ₁ = 0.0617 <i>wR</i> ₂ = 0.1103
<i>R</i> indices (all data)	<i>R</i> ₁ = 0.0668, <i>wR</i> ₂ = 0.1303	<i>R</i> ₁ = 0.0818 <i>wR</i> ₂ = 0.1111	<i>R</i> ₁ = 0.0761 <i>wR</i> ₂ = 0.1824	<i>R</i> ₁ = 0.0503 <i>wR</i> ₂ = 0.1062
Largest diff. peak/hole [e.Å ⁻³]	1.267/-2.079	1.461 /-2.610	3.712/-1.739	1.681/-2.228

Table S2. Bond Lengths and Bond Angles of Complexes 1-4

Complex	Bond length [Å]	Bond angle [°]
1	Ir1— Se1 2.4663(12)	Se1— Ir1— Se2 95.99(4)
	Ir1— Se2 2.5159(12)	Cl1— Ir1— Se2 90.79(8)
	Ir1— Cl1 2.439(3)	Cl1— Ir1— Se1 78.31(8)
	Se2— C14 1.840(11)	C6— Se1— Ir1 113.8(3)
	Se1— C6 1.939(11)	C7— Se1— Ir1 102.1(3)
	Se1— C7 1.973(10)	C14— Se2— Ir1 105.6(3)
	C14— N1 1.369(13)	C14— N2— C15 123.4(9)
	C14— N2 1.352(13)	C14— N1— C7 123.2(9)
	N1— C8 1.393(13)	N1— C7— Se1 111.9(7)
		N2— C15— C16 113.3(8)
		C(6)—Se(1)—C(7) 96.6(5)
2.	Ir1— Se1 2.4663(13)	S1— Ir1— Se1 95.41(7)
	Ir1— S1 2.417(2)	Cl1— Ir1— S1 90.69(10)
	Cl1— Ir1 2.436(3)	Cl1— Ir1— Se1 78.08(7)
	C14— S1 1.692(10)	C6— Se1— Ir1 113.3(3)
	C14— N2 1.365(11)	C7— Se1— Ir1 100.7(3)
	C6— Se1 1.945(10)	C14— S1— Ir1 108.9(3)
	C7— Se1 1.945(10)	C14— N2— C15 124.2(8)
	C8— N1 1.365(11)	C14— N1— C7 122.7(8)
	C14— N1 1.363(11)	N1— C7— Se1 122.7(8)
		C16— C15— N2 113.6(8)
	C6—Se1—C7 95.4(4)	
3	Ir1— S1 2.355(4)	S1— Ir1— Se1 94.99(9)
	Ir1— Se1 2.5141(16)	Cl1— Ir1— Se1 90.85(12)
	Cl1— Ir1 2.427(4)	Cl1— Ir1— S1 79.62(14)
	C14— Se1 1.844(15)	C6— S1— Ir1 116.2(5)
	C14— N2 1.344(18)	C7— S1— Ir1 106.5(5)
	C6— S1 1.777(15)	C14— Se1— Ir1 104.8(4)
	C7— S1 1.823(15)	C14— N2— C15 125.6(12)
	C8— N1 1.386(17)	C14— N1— C7 123.2(12)
	C14— N1 1.351(17)	N1— C7— S1 111.9(9)
		C16— C15— N2 112.2(12)
		C6—S1—C7 98.6(7)

4	Ir1—S1	2.3469(18)	S1—Ir1—S2	94.61(6)
	Ir1—S2	2.4146(19)	S1—Ir1—Cl1	79.33(7)
	N1—C14	1.364(9)	S2—Ir1—Cl1	90.55(8)
	N1—C8	1.392(9)	C6—S1—Ir1	116.0(3)
	C14—N2	1.346(8)	C7—S1—Ir1	104.4(3)
	S2—C14	1.685(7)	C14—S2—Ir1	108.0(2)
	S1—C6	1.782(8)	C14—N1—C7	123.2(6)
	S1—C7	1.834(7)	C14—N2—C15	124.0(6)
			N1—C7—S1	112.3(5)
			N2—C15—C16	113.4(6)
			C6—S1—C7	98.9(3)

Table S3. Non-covalent Interactions C–H···F Distances (Å) of Complexes 1-4

Complex 1		Complex 2		Complex 3		Complex 4	
H30C-F4	2.453	H7B-F2	2.486	H9-F5	2.580	H28B-F2	2.632
H27A-F3	2.531	H30B-F5	2.581	H7B-F3	2.489	H9-F6	2.606
H15A-F3	2.606	H15b-F3	2.619	H15A-F1	2.546	H7B-F6	2.501
H7B-F2	2.530	H28B-F3	2.541	H28C-F6	2.612	H2-F5	2.668
H7B-F1	2.480	H12-Cl1	2.883	H7B-F3	2.489	H12-Cl1	2.865
				H12-Cl1	2.829		

Secondary Interaction-

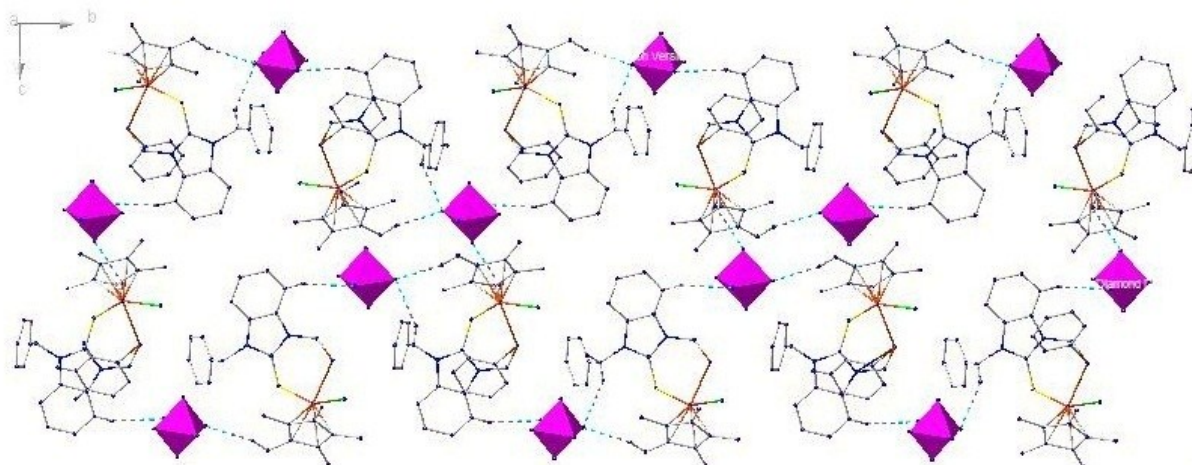


Figure S1 Three dimensional packing framework showing non-covalent C–H···F interactions in the crystal lattice contain PF₆ in polyhedral form of **2**

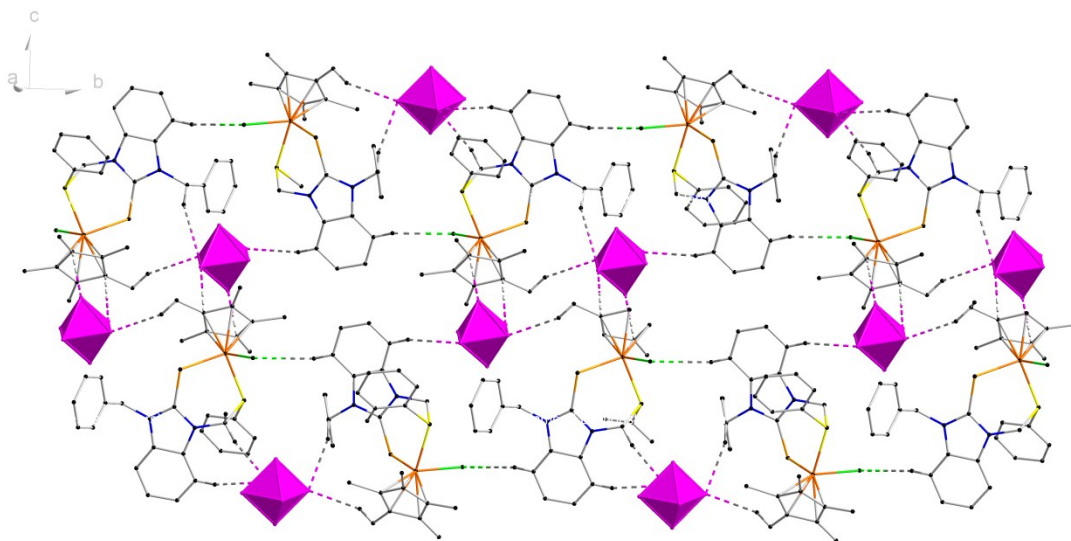


Figure S2 Three dimensional packing framework showing non-covalent C–H···F interactions in the crystal lattice contain PF_6 in polyhedral form of **3**

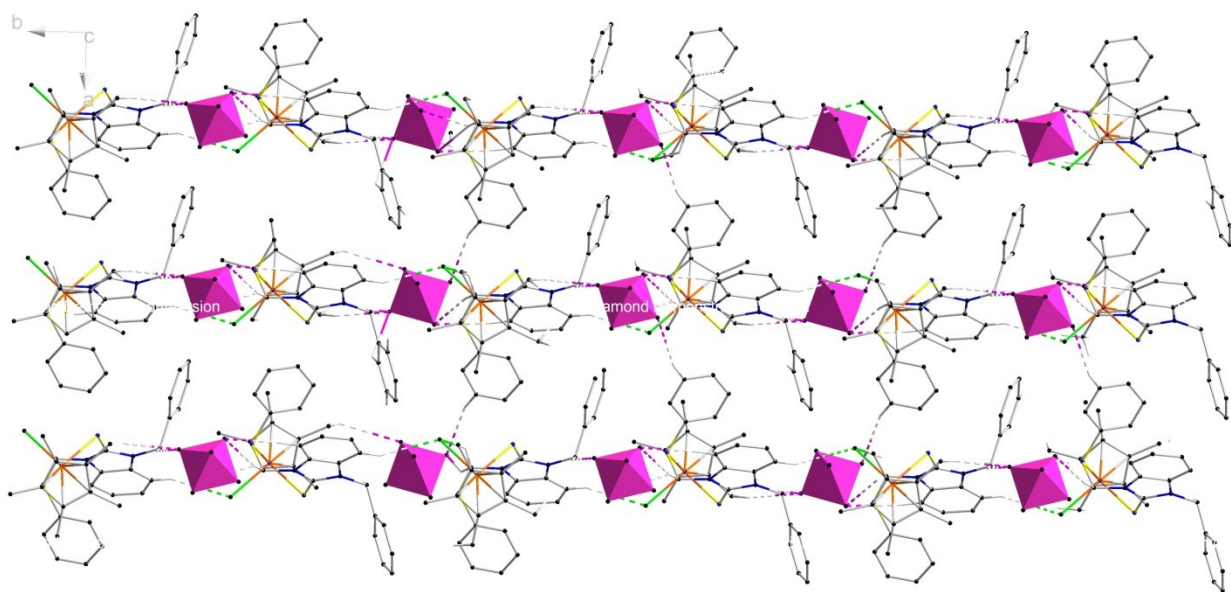


Figure S3 Three dimensional packing framework showing non-covalent C–H···F interactions in the crystal lattice contain PF_6 in polyhedral form of **4**

Mass Spectrum-

Mass Spectrum SmartFormula Report

Analysis Info		Acquisition Date	3/30/2016 11:23:18 AM
Analysis Name	D:\Data\MAR 2016\laks A2.d	Operator	IITD
Method	tune_wide.m	Instrument / Ser#	micrOTOF-Q II 10262
Sample Name	TM1:100	Comment	

Acquisition Parameter					
Source Type	ESI	Ion Polarity	Positive	Set Nebulizer	0.3 Bar
Focus	Active	Set Capillary	4500 V	Set Dry Heater	180 °C
Scan Begin	50 m/z	Set End Plate Offset	-500 V	Set Dry Gas	4.0 l/min
Scan End	3000 m/z	Set Collision Cell RF	500.0 Vpp	Set Divert Valve	Source



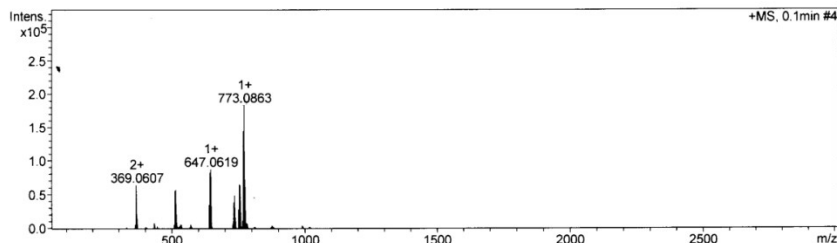
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821.0276	1	C ₃₁ H ₃₃ ClIrN ₂ Se ₂	821.0281	0.6	-0.8	16.5	ok	even	28.3	25.1	4.9	7.3	3.7	842.7		

Fig S4 Mass spectrum of complex 1

Mass Spectrum SmartFormula Report

Analysis Info		Acquisition Date	8/5/2014 10:12:34 AM
Analysis Name	Z:\Aug_2014\ALP1.d	Operator	Sharma/Singh
Method	tune_wide.m	Instrument	micrOTOF-Q II 228888.10262
Sample Name	TM1:100	Comment	

Acquisition Parameter					
Source Type	ESI	Ion Polarity	Positive	Set Nebulizer	0.3 Bar
Focus	Active	Set Capillary	4500 V	Set Dry Heater	180 °C
Scan Begin	50 m/z	Set End Plate Offset	-500 V	Set Dry Gas	4.0 l/min
Scan End	3000 m/z	Set Collision Cell RF	600.0 Vpp	Set Divert Valve	Source



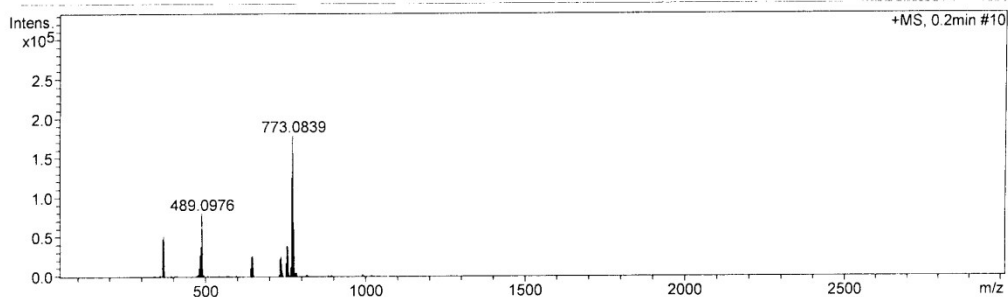
Meas. m/z	# Ion	Formula	m/z	err [ppm]	Mean err [ppm]	rdb	N-Rule	e ⁻ Conf	mSigma	Std I	Std I	Std I	Std I	Std I	Std I
773.086288	1	C ₃₁ H ₃₃ ClIrN ₂ Se	773.083333	-3.8	-1.9	16.5	ok	even	28.6	24.7	2.2	4.2	2.2	842.7	

Fig S5 Mass spectrum of complex 2

Mass Spectrum SmartFormula Report

Analysis Info
 Analysis Name: D:\Data\Aug_2014\ALP2.d
 Method: tune_wide.m
 Sample Name: TM1:100
 Comment:
 Acquisition Date: 8/5/2014 11:34:19 AM
 Operator: Sharma/Singh
 Instrument / Ser#: micrOTOF-Q II 10262

Acquisition Parameter
 Source Type: ESI Ion Polarity: Positive Set Nebulizer: 0.3 Bar
 Focus: Active Set Capillary: 4500 V Set Dry Heater: 180 °C
 Scan Begin: 50 m/z Set End Plate Offset: -500 V Set Dry Gas: 4.0 l/min
 Scan End: 3000 m/z Set Collision Cell RF: 600.0 Vpp Set Divert Valve: Source



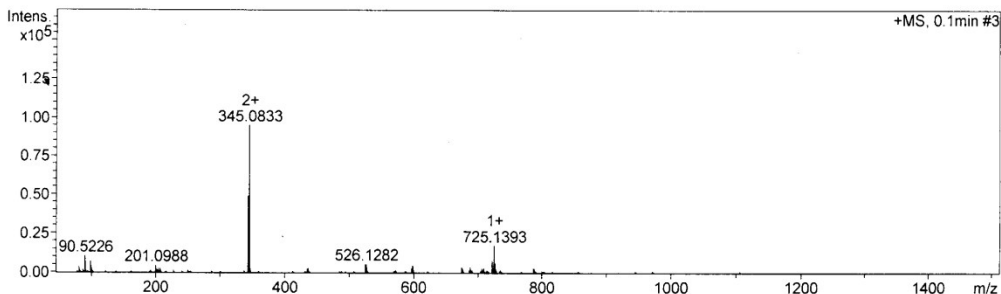
Meas. m/z	#	Formula	m/z	err [ppm]	Me an err [ppm]	rdb	N-R	e ⁻ Conf	mSigma	Std l	St d	St d l	St d m/	Std Com b
773.0839	1	C ₃₁ H ₃₃ ClIrN ₂ S ₂ Se	773.0833	-0.8	0.6	16.5	ok	even	46.0	35.2	1.9	5.6	2.7	842.7

Fig S6 Mass spectrum of complex 3

Mass Spectrum SmartFormula Report

Analysis Info
 Analysis Name: Z:\SEPT_2014\ALP1.d
 Method: tune_low.m
 Sample Name: Lactamase digest
 Comment:
 Acquisition Date: 9/5/2014 10:57:57 AM
 Operator: Sharma/Singh
 Instrument: micrOTOF-Q II 228888.10262

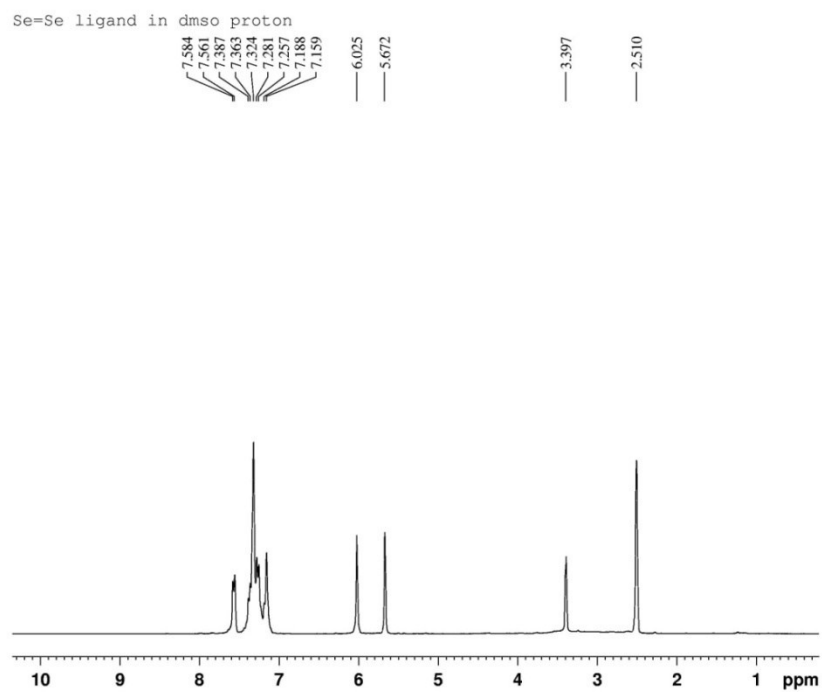
Acquisition Parameter
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 Focus: Active Set Capillary: 4500 V Set Dry Heater: 190 °C
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 Scan End: 1500 m/z Set Collision Cell RF: 100.0 Vpp Set Divert Valve: Source



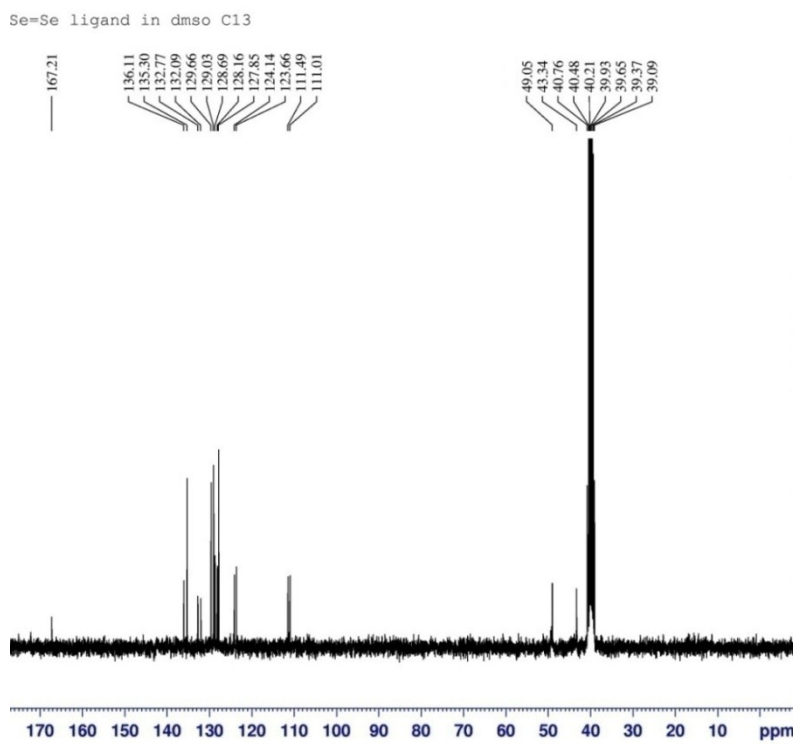
Meas. m/z	#	Ion Formula	Score	m/z	err [ppm]	Mean err [ppm]	mSigma	rdb	e ⁻ Conf	N-Rule
725.139260	1	C ₃₁ H ₃₃ ClIrN ₂ S ₂	100.00	725.138804	-0.6	0.6	18.2	16.5	even	ok

Fig S7 Mass spectrum of complex 4

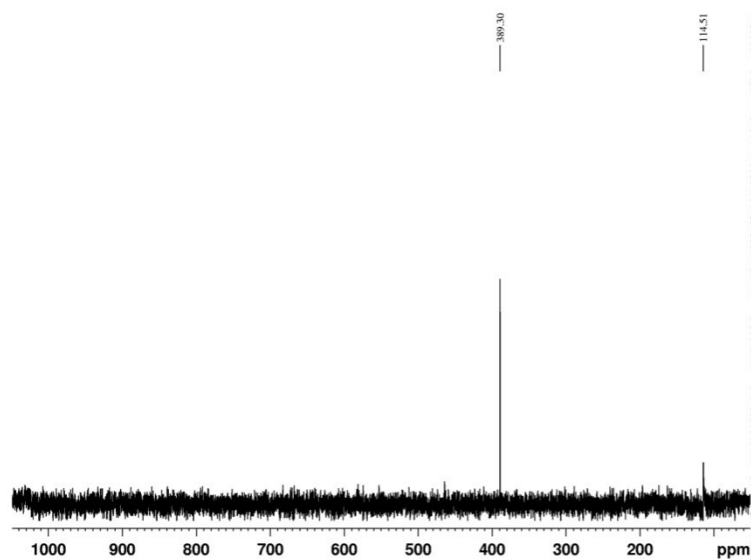
NMR Spectra-



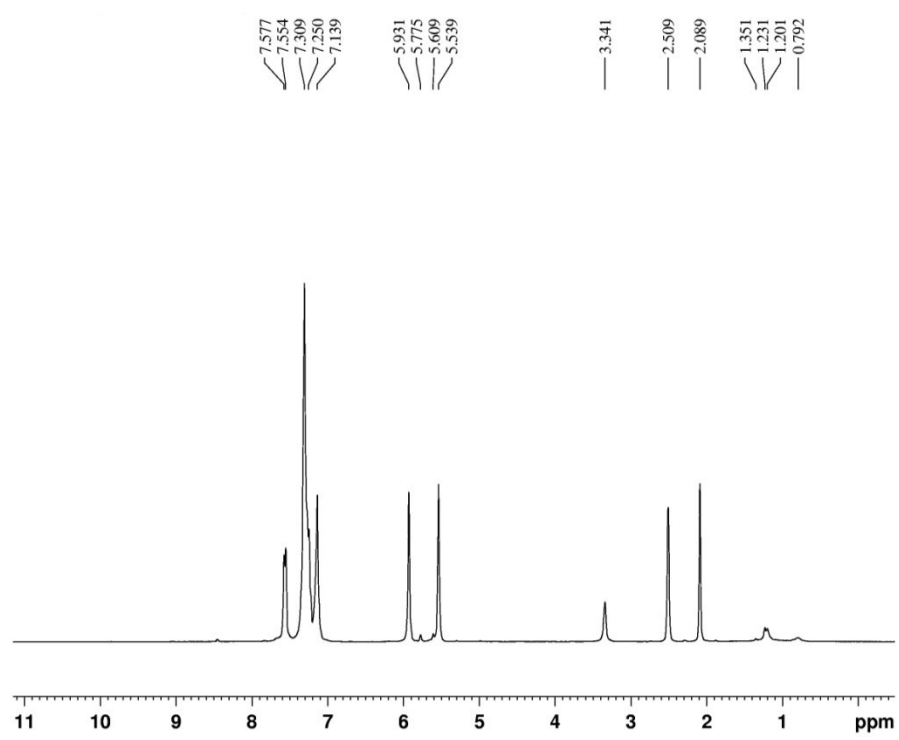
FigureS8 ^1H NMR spectrum of ligand L1



FigureS9 $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum of ligand L1

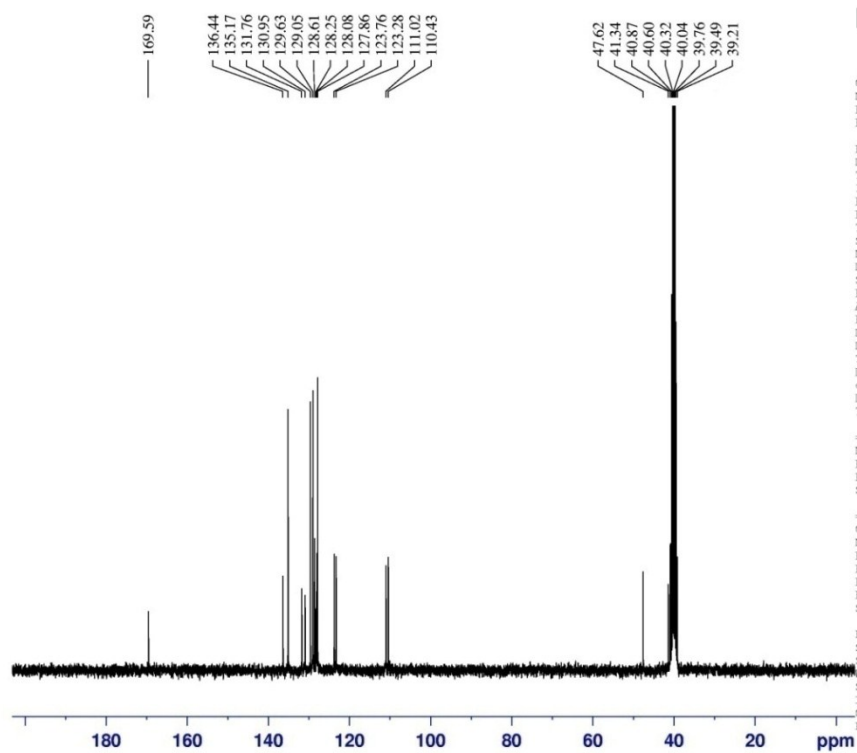


FigureS10 ⁷⁷Se{¹H} NMR spectrum of ligand L1



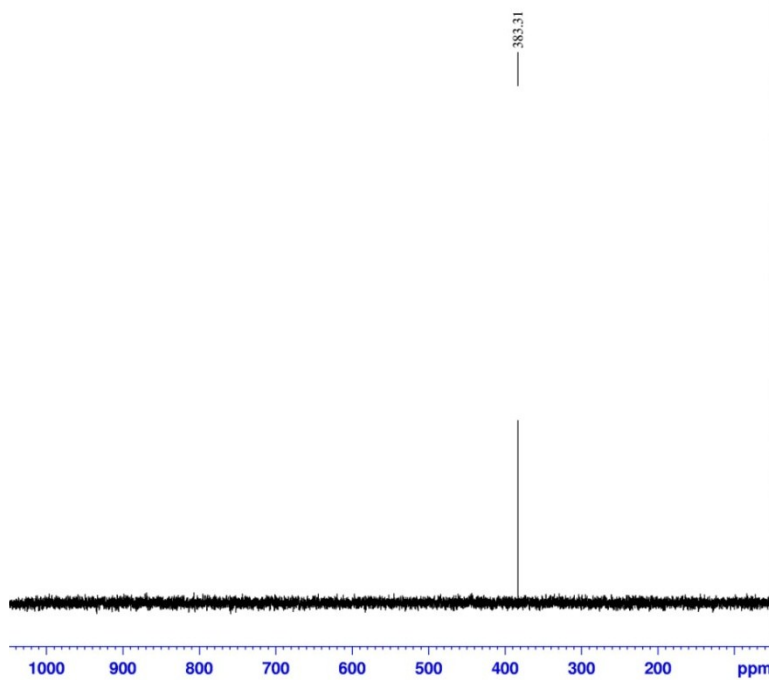
FigureS11 ¹H NMR spectrum of ligand L2

Se=S ligand in dmso C13

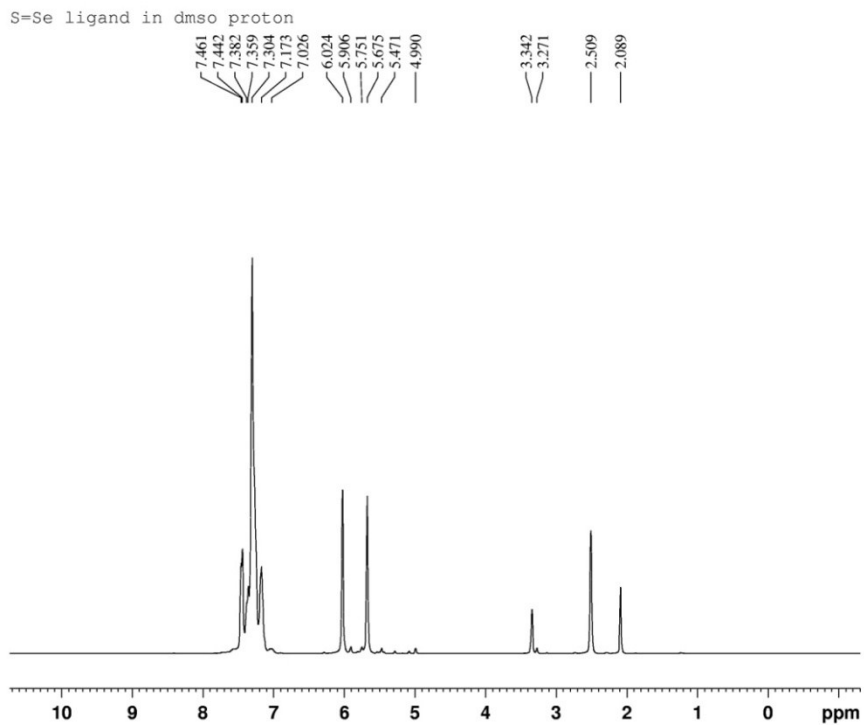


FigureS12 $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum of ligand L2

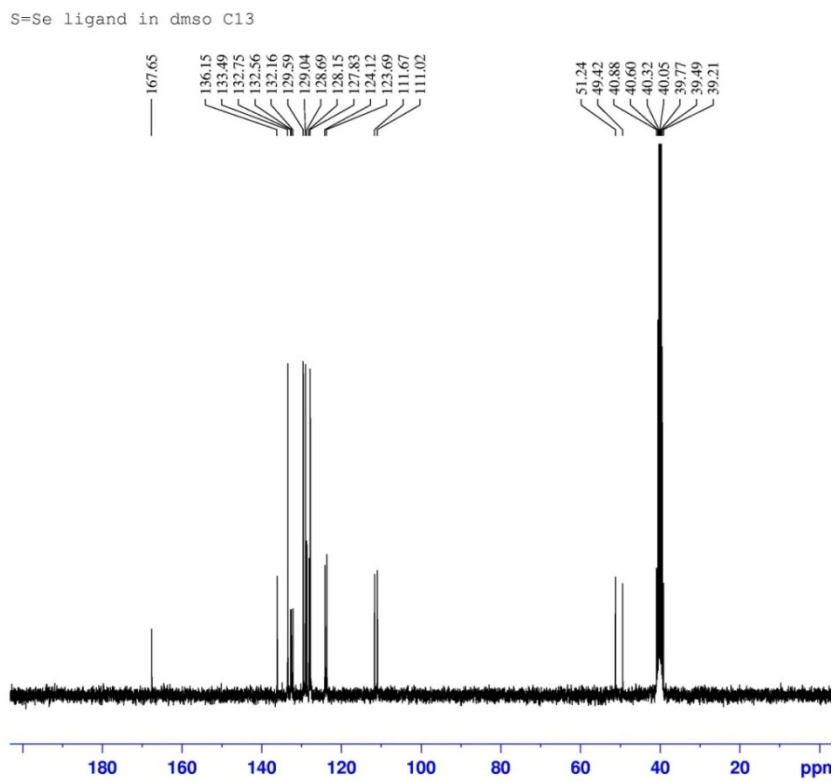
Se=S ligand in dmso Se77



FigureS13 $^{77}\text{Se}\{^1\text{H}\}$ NMR spectrum of ligand L2

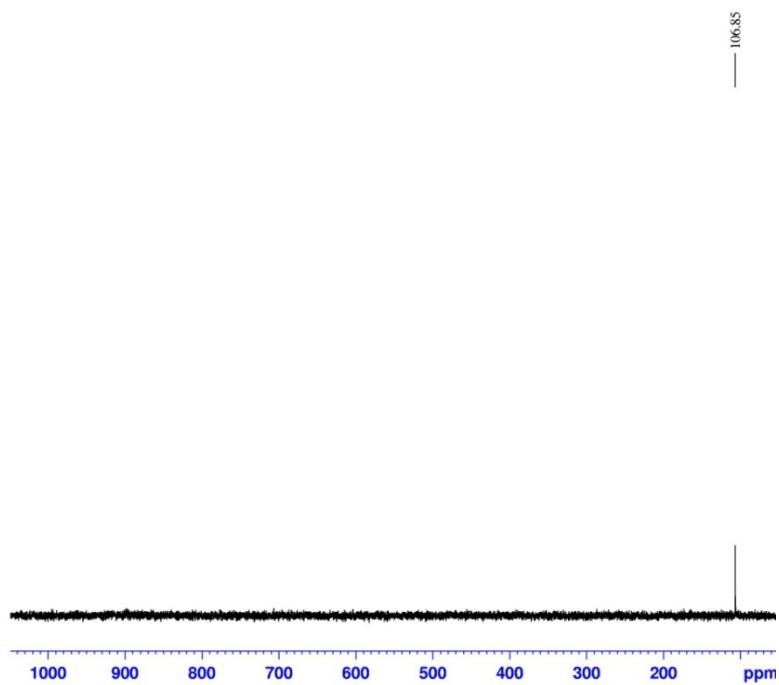


FigureS14 ^1H NMR spectrum of ligand L3



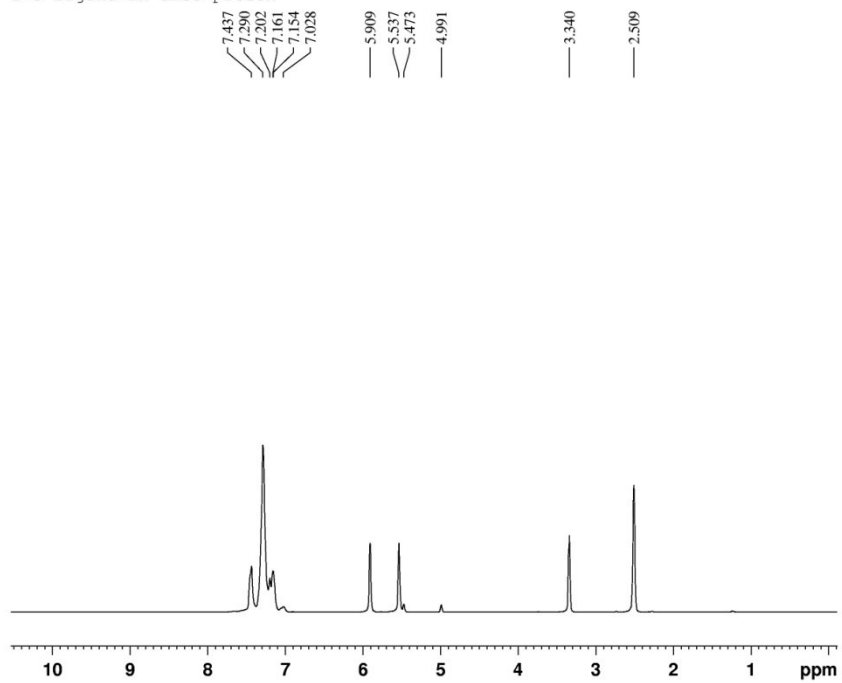
FigureS15 $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum of ligand L3

S=Se ligand in dmso se77



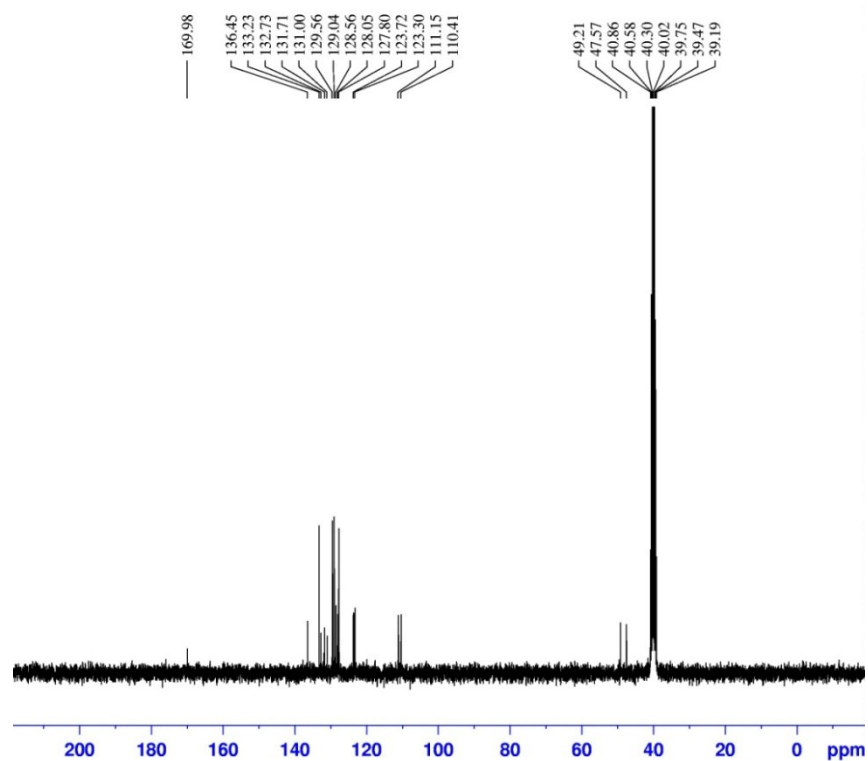
FigureS16 $^{77}\text{Se}\{^1\text{H}\}$ NMR spectrum of ligand **L3**

S=S ligand in dmso proton



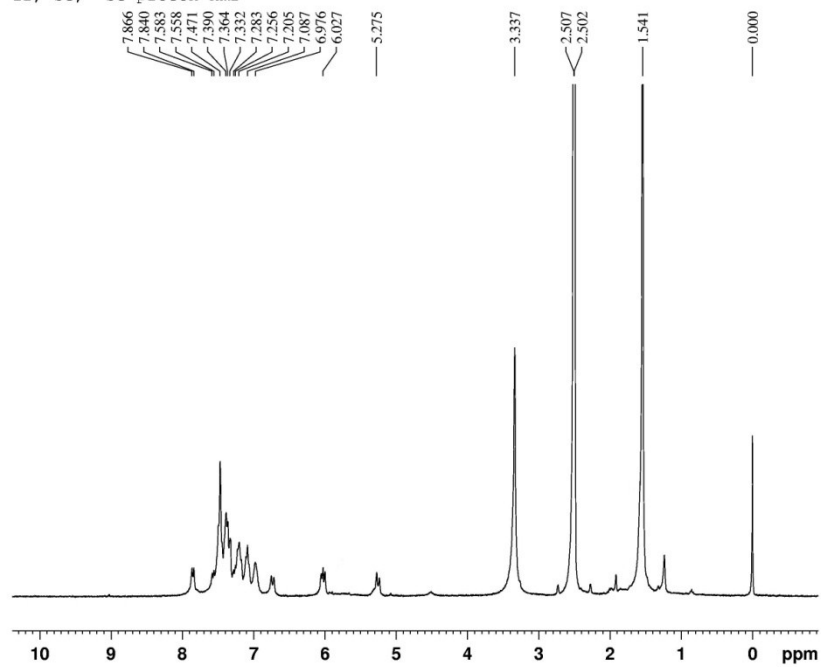
FigureS17 ^1H NMR spectrum of ligand **L4**

S=S ligand in dmsO C13

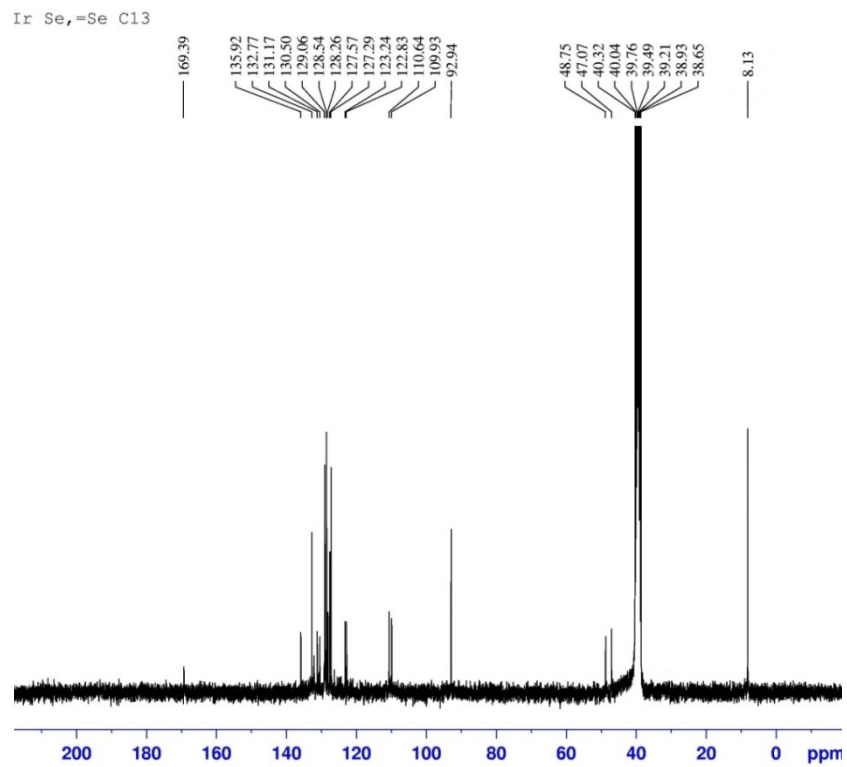


FigureS18 $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum of ligand L4

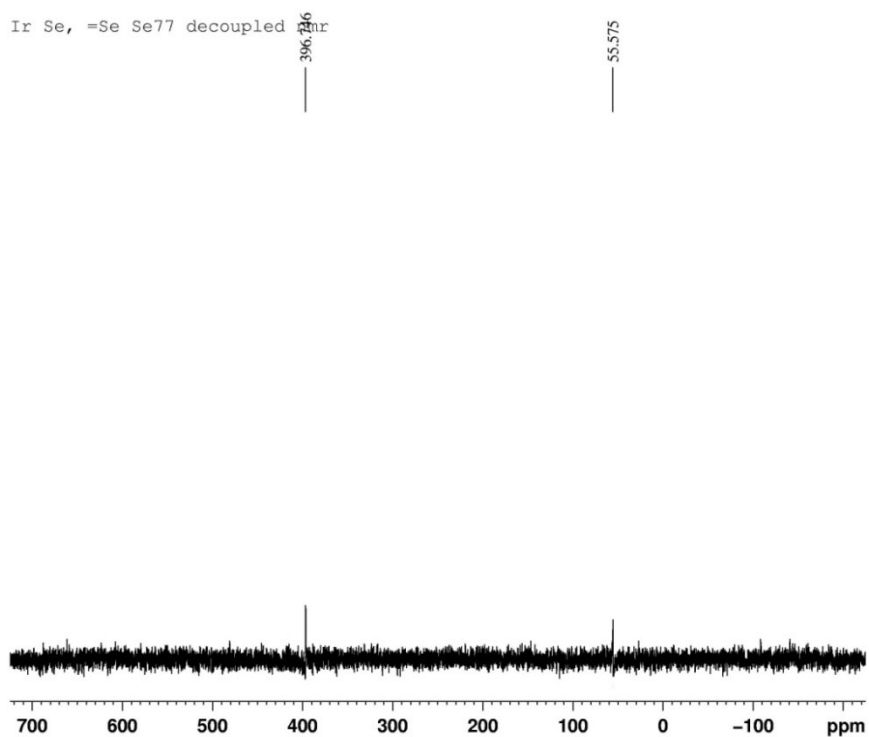
Ir, Se, =Se proton nmr



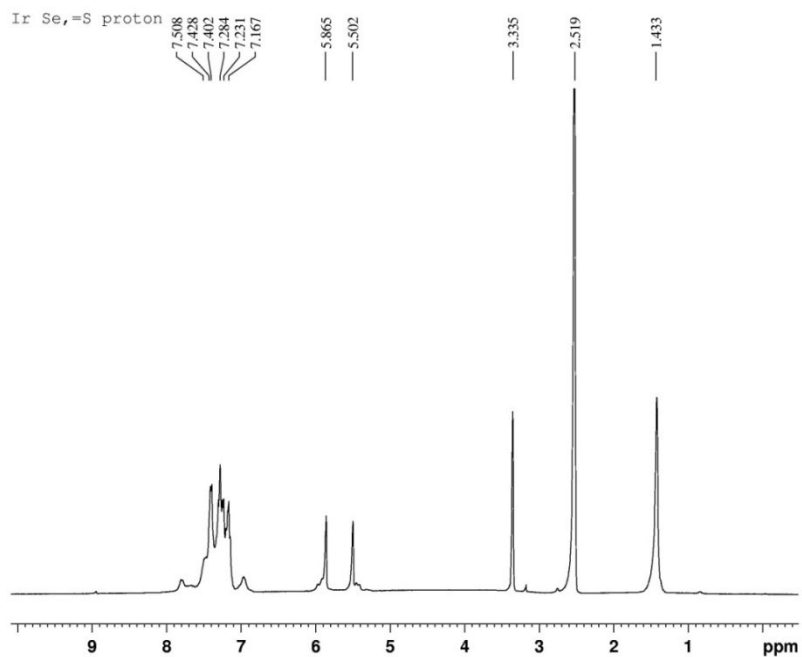
FigureS19 ^1H NMR spectrum of complex 1



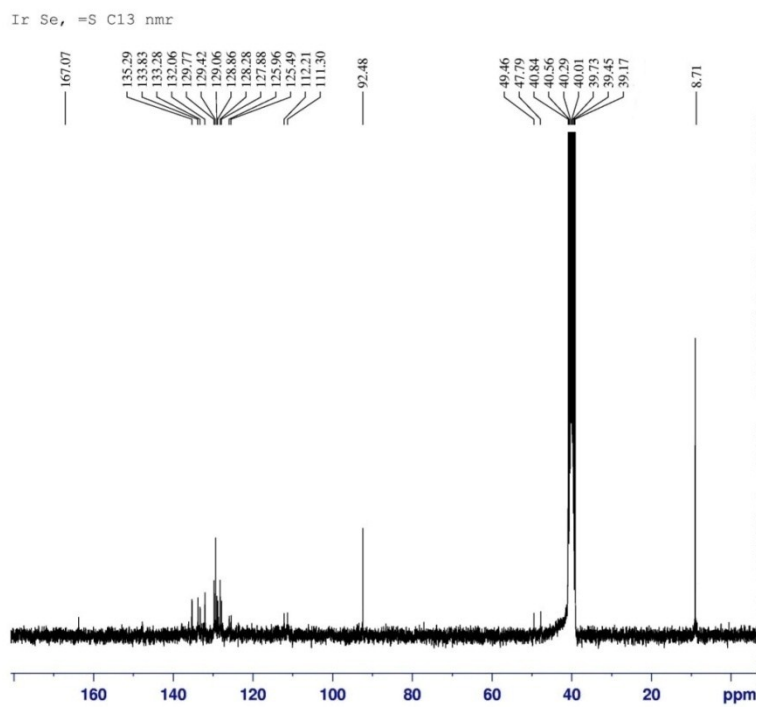
FigureS20 $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum of complex 1



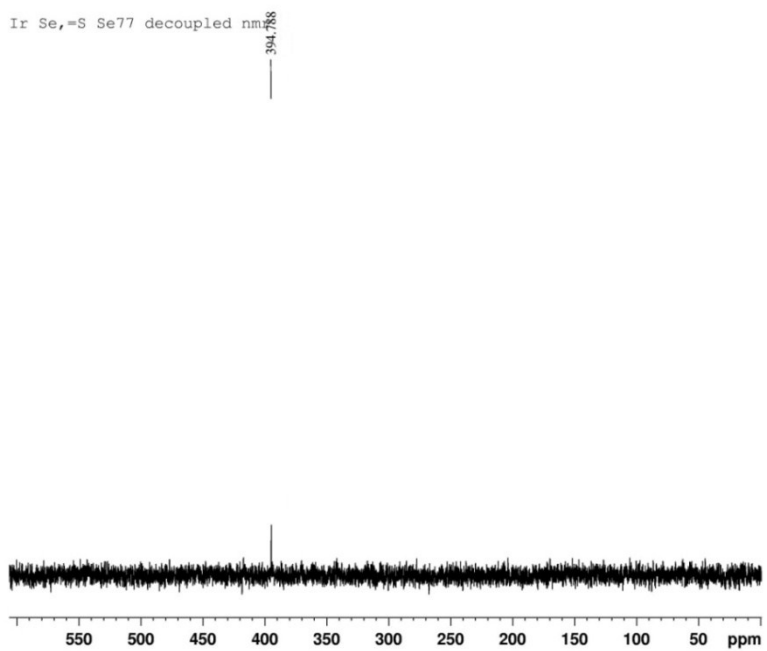
FigureS21 $^{77}\text{Se}\{^1\text{H}\}$ NMR spectrum of complex 1



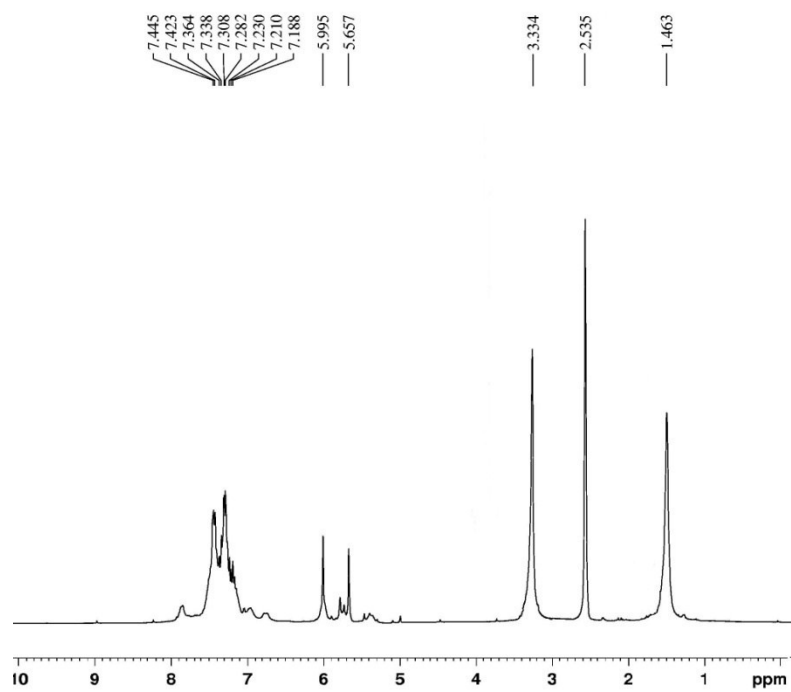
FigureS22 ^1H NMR spectrum of complex 2



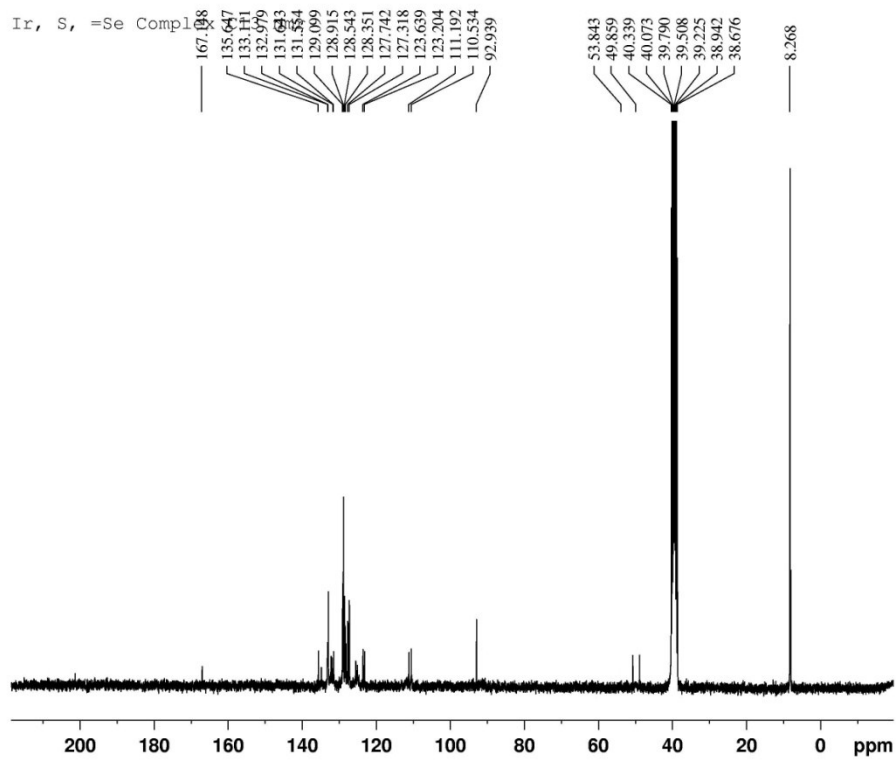
FigureS23 $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum of complex 2



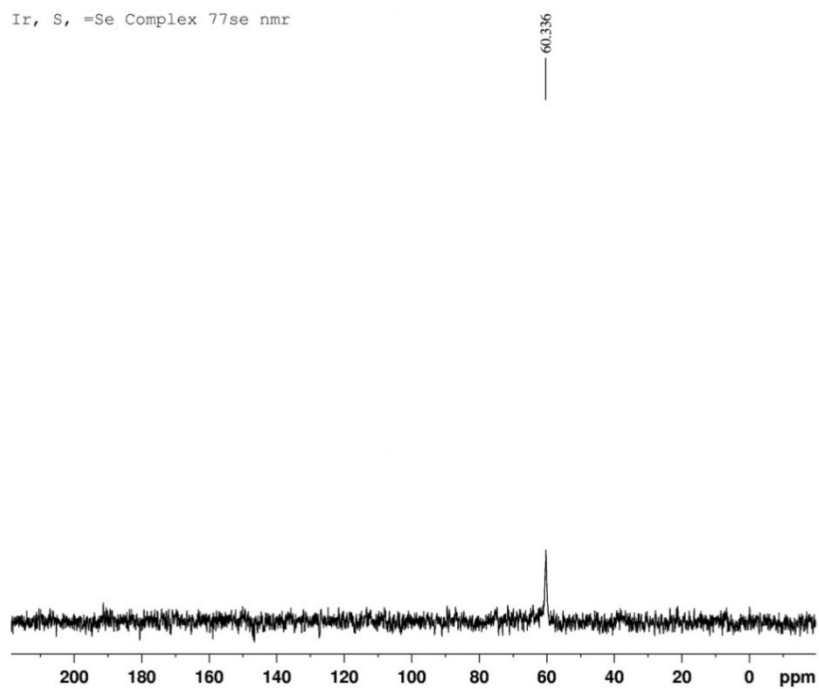
FigureS24 $^{77}\text{Se}\{^1\text{H}\}$ NMR spectrum of complex 2



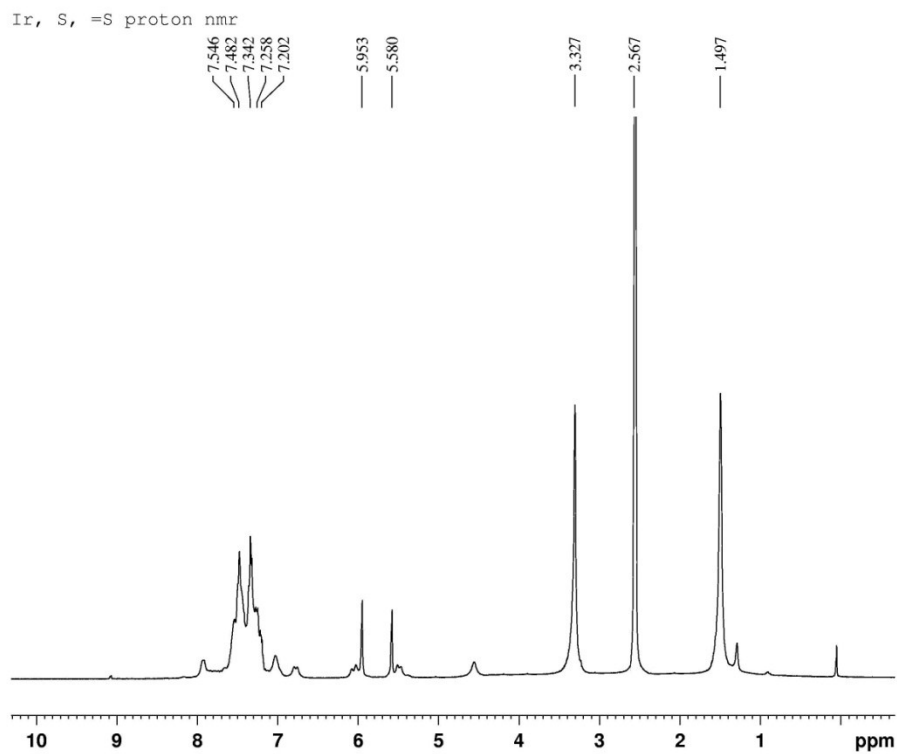
FigureS25 ^1H NMR spectrum of complex 3



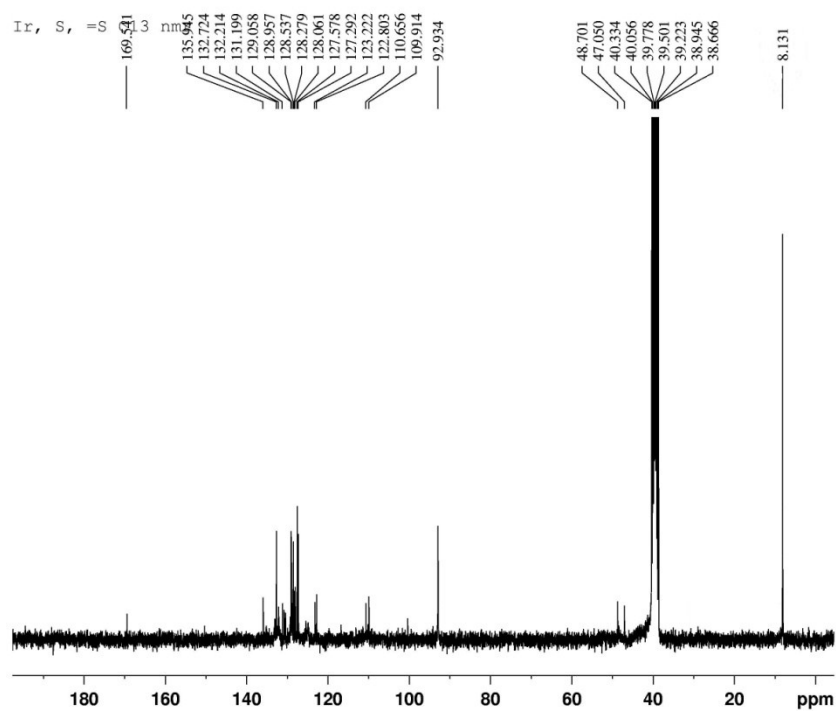
FigureS26 $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum of complex 3



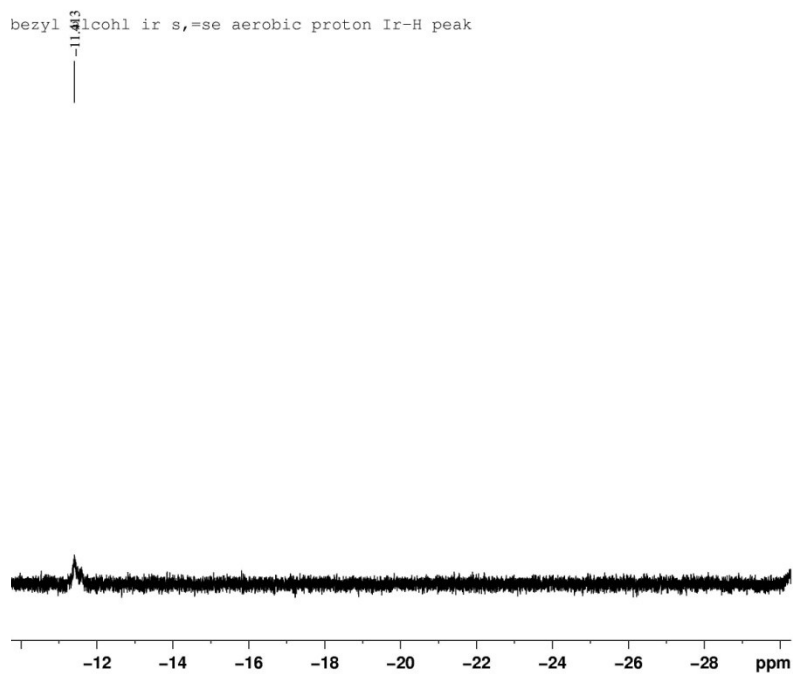
FigureS27 $^{77}\text{Se}\{^1\text{H}\}$ NMR spectrum of complex 3



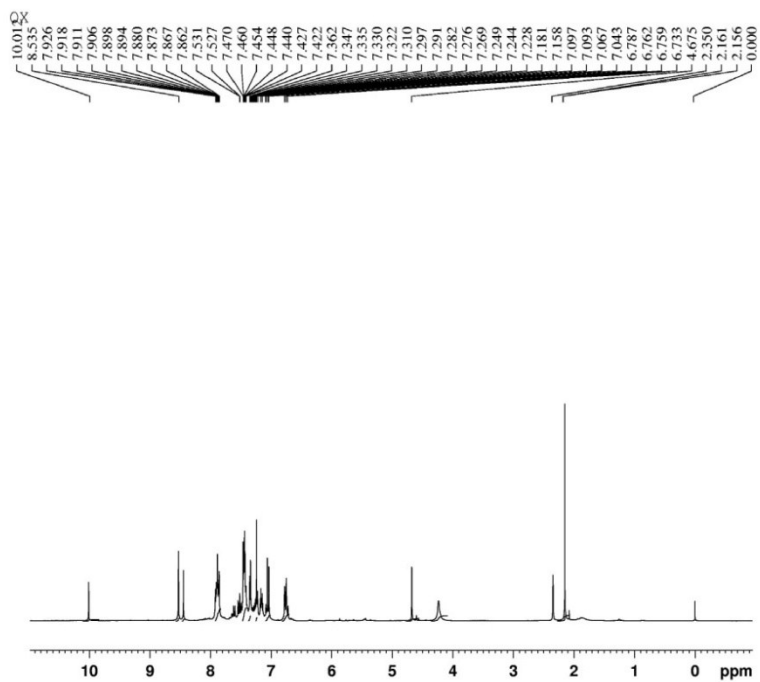
FigureS28 ^1H NMR spectrum of complex 4



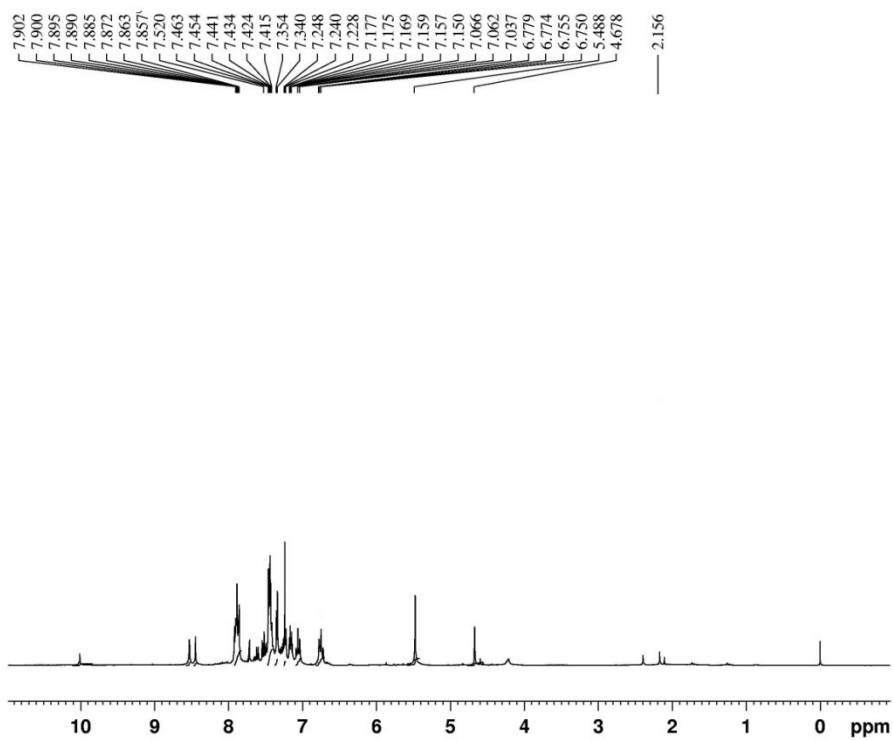
FigureS29 $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum of complex 4



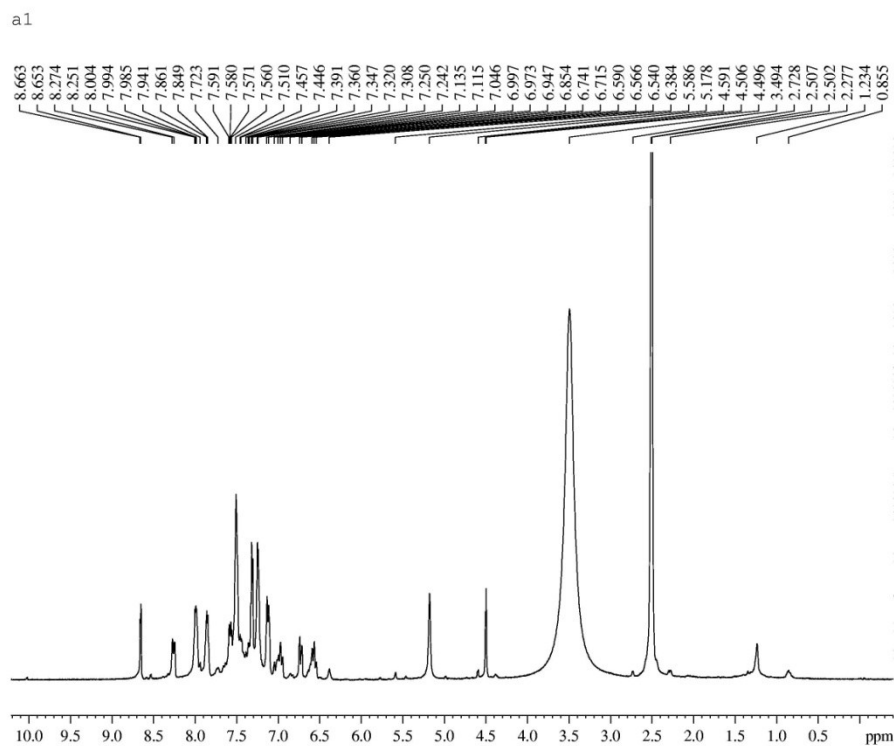
FigureS30 ^1H NMR spectrum for Ir-H species



FigureS31 ^1H NMR spectrum after 30 min of catalysis reaction progress (1,2-phenylenediamine and benzyl alcohol)



FigureS32 ^1H NMR spectrum after 90 min of catalysis reaction progress 1,2-phenylenediamine and benzyl alcohol



FigureS33 ^1H NMR spectrum of catalytic reaction mixture

NMR Spectra of compound (3a-3g)-

1-Benzyl-2-phenyl-1H-benzimidazole (3a)^{1,3}- White solid; ¹H NMR (CDCl₃, 25°C vs Me₄Si): δ (ppm): 7.87 (d, J = 6.1 Hz, 1H), 7.70–7.68 (m, 2H), 7.47–7.41 (m, 3H), 7.34–7.30 (m, 4H), 7.25–7.20 (m, 2H), 7.12–7.10 (m, 2H), 5.47 (s, 2H).

1-(4-Methylbenzyl)-2-(4-methylphenyl)-1H-benzimidazole (3b)^{1,2}- White solid; ¹H NMR (CDCl₃, 25°C vs Me₄Si): δ (ppm) 7.83 (d, J = 6.4 Hz, 1H), 7.60 (d, J = 6.2 Hz, 2H), 7.32–7.20 (m, 5H), 7.13 (d, J = 6.2 Hz, 2H), 6.98 (d, J = 6.1 Hz, 2H), 5.41 (s, 2H), 2.41 (s, 3H), 2.36 (s, 3H).

1-(4-Methoxybenzyl)-2-(4-methoxyphenyl)-1H-benzimidazole (3c)^{1,2}- White solid; ¹H NMR (CDCl₃, 25°C vs Me₄Si): δ (ppm) 7.67 (d, J = 6.2 Hz, 3H), 7.45 (d, J = 6.4 Hz, 1H), 7.27–7.18 (m, 2H), 7.06 (d, J = 5.2 Hz, 2H), 6.96 (d, J = 5.6 Hz, 2H), 6.85 (d, J = 6.5 Hz, 2H), 5.46 (s, 2H), 3.84 (s, 3H), 3.63 (s, 3H).

1-(3-Methoxybenzyl)-2-(3-methoxyphenyl)-1H-benzimidazole (3d)²- White solid; ¹H NMR (DMSO, 25°C vs Me₄Si): δ (ppm) 7.54 (d, J = 6.2 Hz, 1H), 7.47–7.32 (d, J = 6.8 Hz, 2H), 7.28–7.19 (m, 1H), 7.13–7.05 (m, 3H), 6.98–6.92 (d, J = 6.2 Hz, 2H), 6.82–6.72 (d, J = 6.2 Hz, 2H), 5.21 (s, 2H), 3.84 (s, 3H), 3.80 (s, 3H).

1-(2-Methoxybenzyl)-2-(2-methoxyphenyl)-1H-benzimidazole (3e)²- White solid; ¹H NMR (DMSO, 25°C vs Me₄Si): δ (ppm) 7.82 (d, J = 7.6 Hz, 1H), 7.63 (d, J = 7.2 Hz, 2H), 7.30 (d, J = 7.2 Hz, 1H), 7.24–7.22 (m, 1H), 7.19–7.17 (m, 3H), 7.11 (d, J = 6.2 Hz, 2H), 7.10 (d, J = 6.0 Hz, 2H), 5.41 (s, 2H), 2.32 (s, 3H), 2.27 (s, 3H).

2-(Furan-2-yl)-1-(furan-2-ylmethyl)-1H-benzimidazole (3f)^{1,2}. White solid; ¹H NMR (DMSO, 25°C vs Me₄Si): δ (ppm) 7.78–7.73 (m, 1H), 7.64 (s, 1H), 7.55–7.49 (m, 1H), 7.30–7.28 (m, 3H), 7.22 (d, J = 7.4 Hz, 1H), 7.63–7.58 (m, 1H), 6.31–6.25 (m, 2H), 5.66 (s, 2H).

1-(4-Nitrobenzyl)-2-(4-nitrophenyl)-1H-benzimidazole (3g)². Yellow solid; ¹H NMR (DMSO, 25°C vs Me₄Si): δ (ppm) 8.45–8.31 (m, 4H), 8.29 (d, J = 6.4 Hz, 2H), 8.17–8.09 (m, 1H), 7.36–7.23 (m, 2H), 7.21–7.15 (m, 1H), 7.13–7.12 (m, 1H), 6.79 (m, 1H), 6.56 (m, 1H), 5.40 (s, 2H).

NMR Spectra of compound (5a-g, 7a-d)-

Benzaldehyde (5a)³ - Colorless liquid; ¹H NMR (CDCl₃, 25°C vs Me₄Si) δ (ppm): 9.98 (s, 1H), δ 7.44-7.478 (m, 2H), 7.51-7.59 (m, 1H), 7.78-7.83 (m, 2H).

4-Methylbenzaldehyde (5b)³- Colorless liquid; ¹H NMR (CDCl₃, 25°C vs Me₄Si); δ (ppm): 2.43(s, 3H), 7.33(d, J = 6.3 Hz, 2H), 7.78 (d, J = 6.2 Hz, 2H), 9.96 (s, 1H).

4-Methoxybenzaldehyde(5c)³- Colorless liquid; ¹H NMR (CDCl₃, 25°C vs Me₄Si); δ (ppm): 9.82 (s, 1H), 7.82 (d, J = 6.6 Hz, 2H), 6.95 (d, J = 6.4 Hz, 2H), 3.91 (s, 3H).

3-Methoxybenzaldehyde (5d)³- Colorless liquid; ¹H NMR (CDCl₃, 25°C vs Me₄Si); δ (ppm): 3.82(s, 3H), 7.18(d, J = 6.2 Hz, 2H), 7.38(d, J = 6.2 Hz, 2H), 7.43(d, J = 6.8 Hz, 2H), 9.93(s, 1H).

2-Methoxybenzaldehyde (5e)⁵- Colorless liquid; ¹H NMR (CDCl₃, 25°C vs Me₄Si): δ (ppm): 3.92(s, 3H), 7.02-7.04(m, 2H), 7.53-7.58(m, 1H), 7.82-7.85(q, 1H), 10.49(s, 1H).

2-Furaldehyde (5f)⁴ – Pale yellow liquid; ¹H NMR (CDCl₃, 25°C vs Me₄Si); δ (ppm): 9.65 (s, 1H), 7.69-7.67 (m, 1H), 7.52-7.44 (m, 1H), 6.60-6.59 (m, 1H).

4-Nitrobenzaldehyde (5g)^{3,4}- Pale yellow solid; ¹H NMR (CDCl₃, 25°C vs Me₄Si); δ (ppm): 10.1 (s, 1H), 8.71 (s, 1H), 8.46-8.50 (m, 1H), 8.22-8.20 (m, 1H), 7.72-7.80 (m, 1H).

Acetophenone (7a)³- Colorless liquid; ¹H NMR (CDCl₃, 25°C vs Me₄Si); δ (ppm): 7.93-7.90 (m, 2H), 7.55-7.50 (m, 1H), 7.46-7.40 (m, 2H), 2.59 (s, 3H).

2-Pentanone (7b)⁶ - Colorless liquid; ¹H NMR (CDCl₃, 25°C vs Me₄Si); δ (ppm): 0.92 (t, J = 6.2 Hz, 3H), 1.61(m, 3H), 2.15(s, 3H), 2.40(t, J = 6.6 Hz, 2H).

2-octanone (7c)⁶ - Colorless liquid; ¹H NMR (CDCl₃, 25°C vs Me₄Si); δ (ppm): 0.89 (t, J = 6.4 Hz, 3H), 1.22-1.39 (m, 6H), 1.53-1.64 (m, 2H), 2.11(s, 3H), 2.42 (t, J = 8.4 Hz, 2H).

Cyclopentanone (7d)⁴ - Colorless liquid; ¹H NMR (CDCl₃, 25°C vs Me₄Si); δ (ppm): 1.90-1.78 (m, 4H), 2.16-2.00 (m, 4H).

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