

Supplementary Information

The Coordination Structure of the Extracted Nickel(II) Complex
with a Synergistic Mixture Containing Dinonylnaphthalene
Sulfonic Acid and 2-Ethylhexyl-4-Pyridine carboxylate Ester

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¹H-NMR Spectra

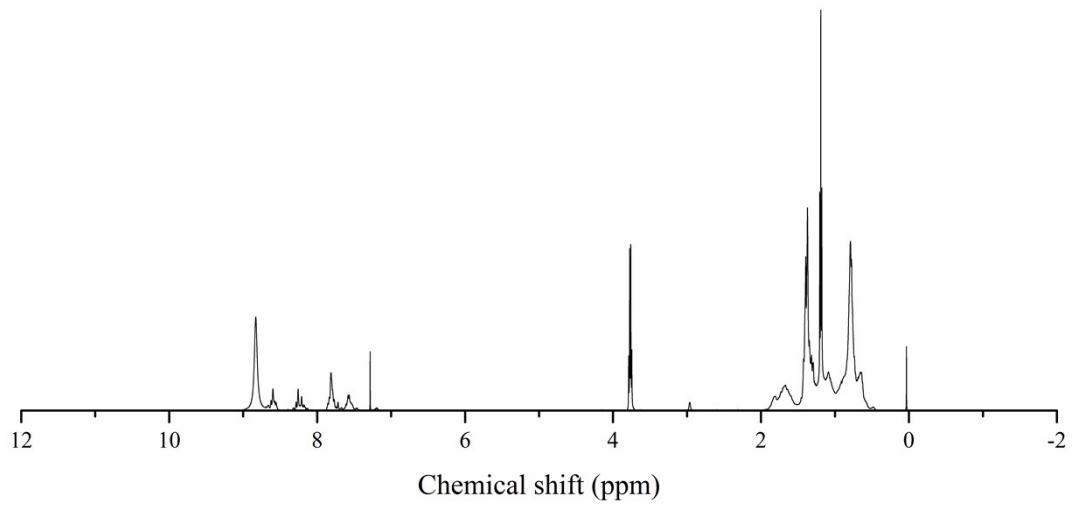


Figure S1 ¹H-NMR spectrum of purified HDNNS

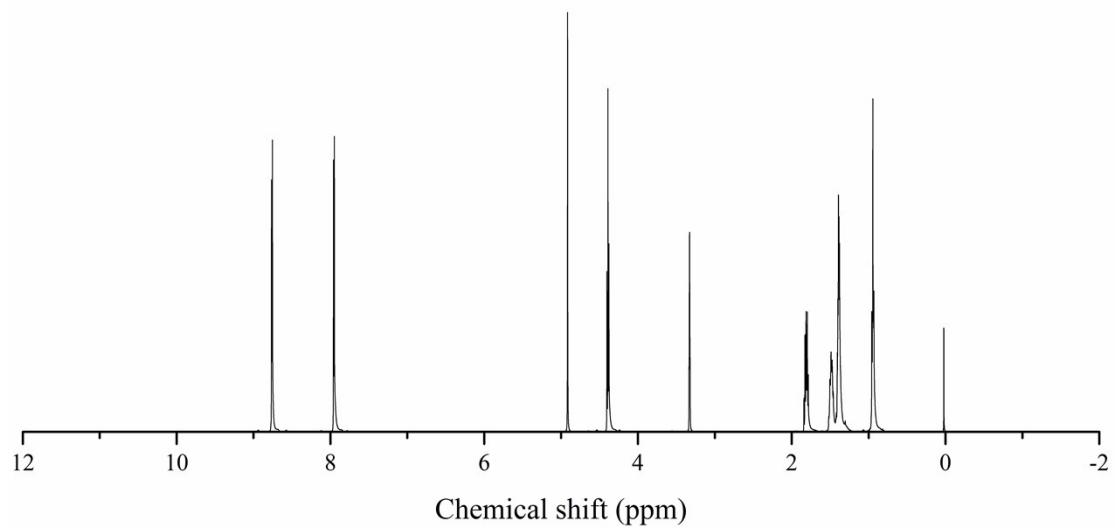


Figure S2 ¹H-NMR spectrum of n-hexyl-4-pyridinecarboxylate ester (L¹)

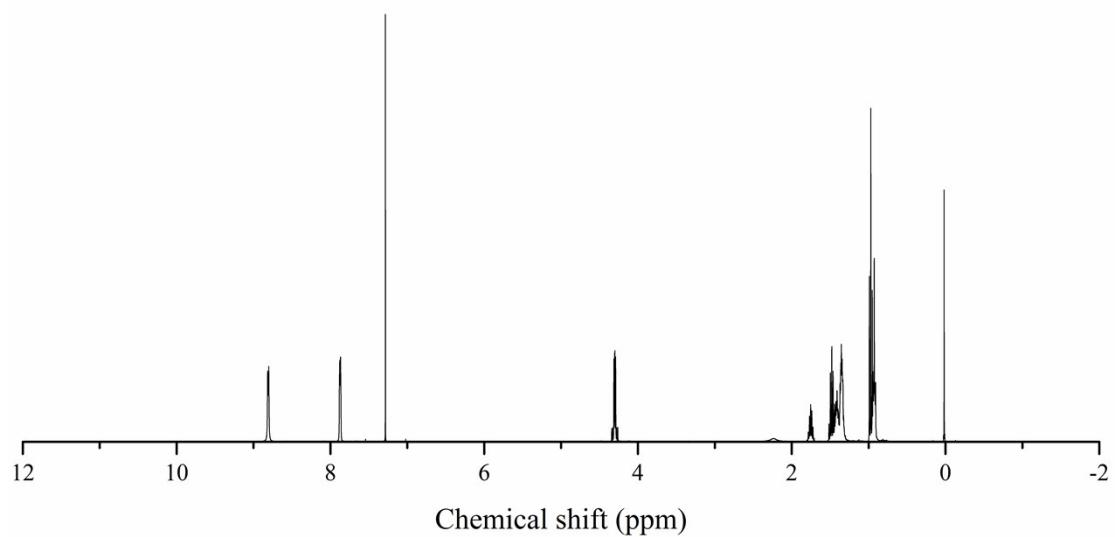


Figure S3 ¹H-NMR spectrum of 2-ethylhexyl 4-pyridinecarboxylate ester (4PC, L^{II})

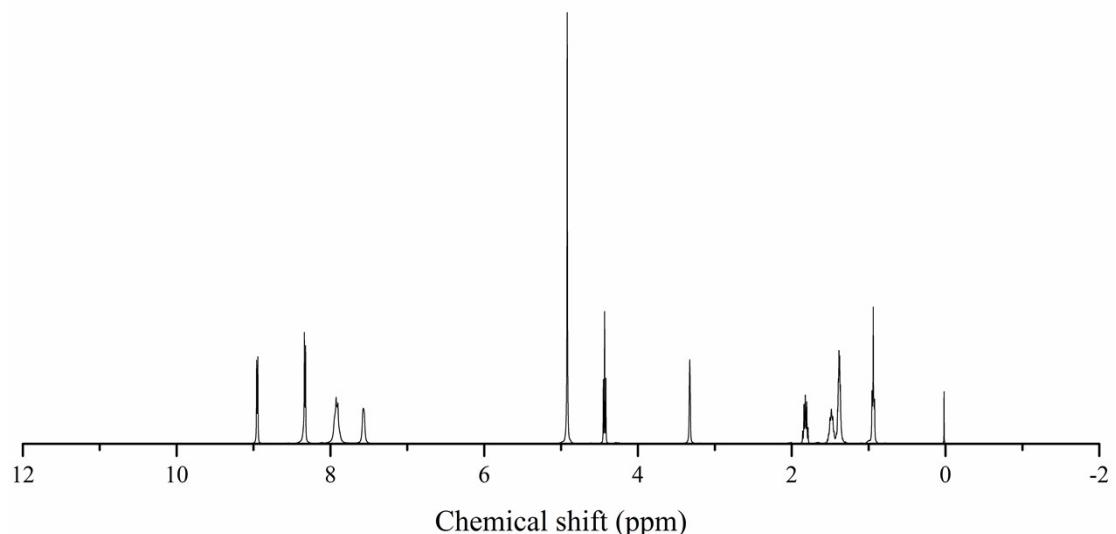


Figure S4 ¹H-NMR spectrum of the nickel synergist complex

Crystallographic details

Table S1 Important bond lengths (Å) and angles (°)

Ni1-O1W	2.063(3)	Ni1-O2W	2.071(3)
Ni1-N1	2.104(3)		
O1W-Ni1-O1W ^a	180.0	O2W-Ni1-N1	88.66(11)
O1W-Ni1-O2W	91.35(11)	O2W ^a -Ni1-N1	91.34(11)
O1W ^a -Ni1-O2W	88.65(11)	O1W-Ni1-N1 ^a	88.18(10)
O1W-Ni1-O2W ^a	88.65(11)	O1W ^a -Ni1-N1 ^a	91.82(10)
O1W ^a -Ni1-O2W ^a	91.35(11)	O2W-Ni1-N1 ^a	91.34(11)
O2W ^a -Ni1-O2W	180.0	O2W ^a -Ni1-N1 ^a	88.66(11)
O1W-Ni1-N1	91.82(10)	N1-Ni1-N1 ^a	180.00(5)
O1W ^a -Ni1-N1	88.18(10)		

Symmetry code: ^a-x, -y+1, -z+2.

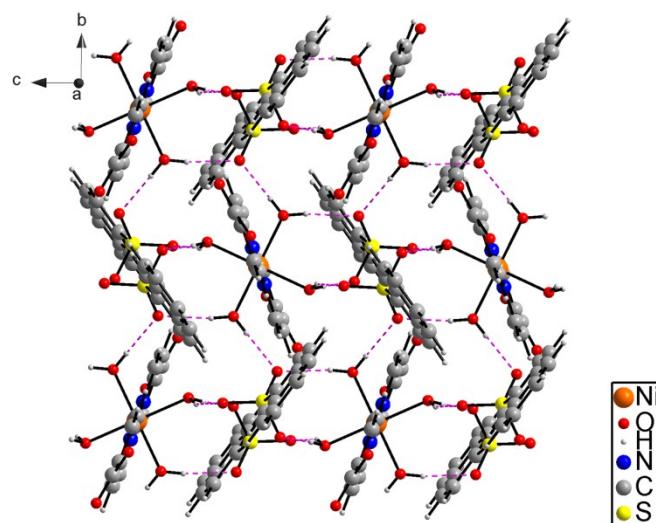


Figure S5 2D plane for alternating rows of $\text{Ni}(\text{H}_2\text{O})_4(\text{L}')_2^{2+}$ cations linked by naphthalene-2-sulfonate anions spacers in the nickel synergist complex as viewed along the a-axis

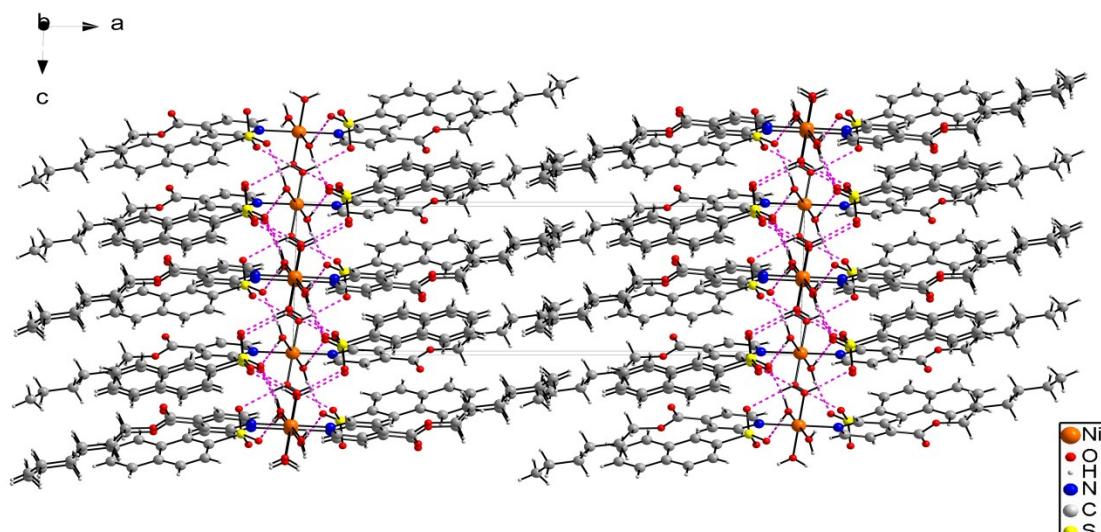


Figure S6 The packing structure of the nickel synergist complex viewed down the b-axis.

Table S2 Summary of hydrogen bonding (\AA and $^{\circ}$) for the nickel synergist complex

D-H...A	Symmetry operation on A	D-H (\AA)	H...A (\AA)	D...A (\AA)	D-H...A ($^{\circ}$)
O1W-H11...O5	-x, -y+1, -z+1	0.815	1.940	2.755	177.30
O1W-H12...O6	x, y, z	0.816	1.998	2.795	165.44
O2W-H21...O7	-x, y+1/2, -z+3/2	0.822	1.975	2.798	179.16
O2W-H22...O7	-x, -y+1, -z+1	0.815	2.031	2.829	166.22

FI-IR details**Table S3** Infrared frequencies and assignments of L^I, HNS, the nickel synergist complex, L^{II}, HDNNNS and the extracted Ni(II) complex in the non-polar organic phase

Vibrational modes	L ^I	HNS	The nickel synergist complex	L ^{II}	HDNNNS	The extracted Ni(II) complex
v _{as} (CH ₃)	2957		2956	2960	2960	2959
v _{as} (CH ₂)	2858		2861	2866	2870	2868
v _s (CH ₃)	2928		2928	2929	2928	2929
v(C=O)	1728		1731	1730		1731
δ_{as} (CH ₃)	1462		1465	1463	1464	1462
δ_s (CH ₃)	1383		1383	1383	1382	1382
v(py)	1407		1417	1407		1417
v(C=C)	1638	1644	1638	1636	1631	1634
	1561	1501	1562	1562	1502	1562
			1503			1502
v(C-O)	1279		1287	1281		1280
v _s (S=O)		1043	1034		1051	1040

ESI-MS Spectra

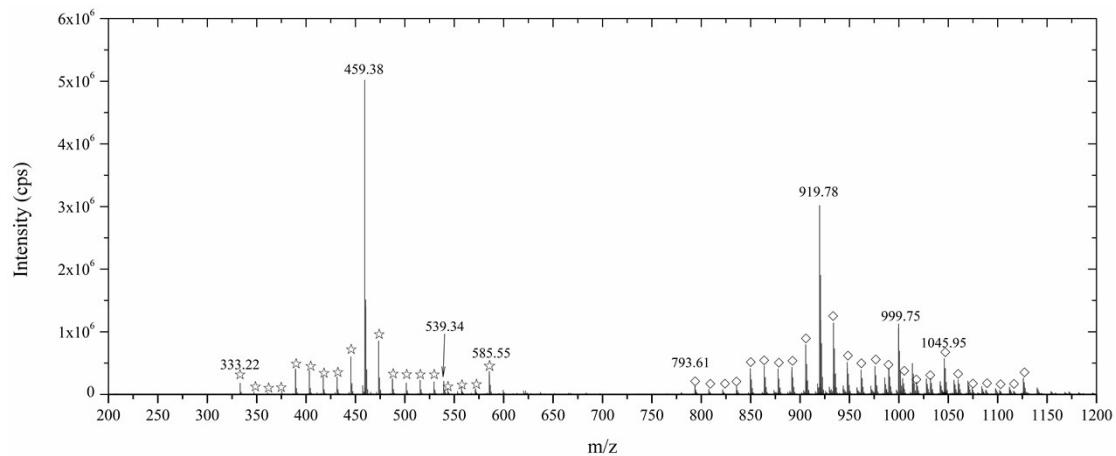


Figure S7 The ESI-MS spectrum of the purified HDNNS in n-hexane with a final concentration of 10^{-3} mol L⁻¹; \star represents species of monomer HDNNS homologues and \diamond represents species of dimer $(\text{HDNNS})_2$ where one of HDNNS was replaced by HDNNS homologues.

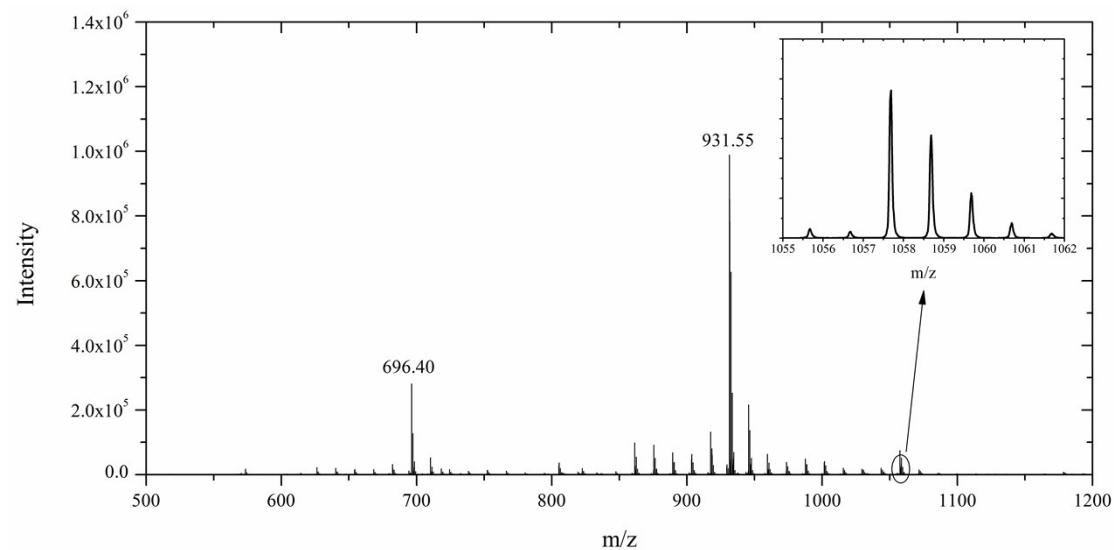


Figure S8 The ESI-MS spectrum of the synergistic extractants in the non-polar organic phase.