

Support information

Light driven charge transfer in nano-Fe(III) complexes facilitates water oxidation

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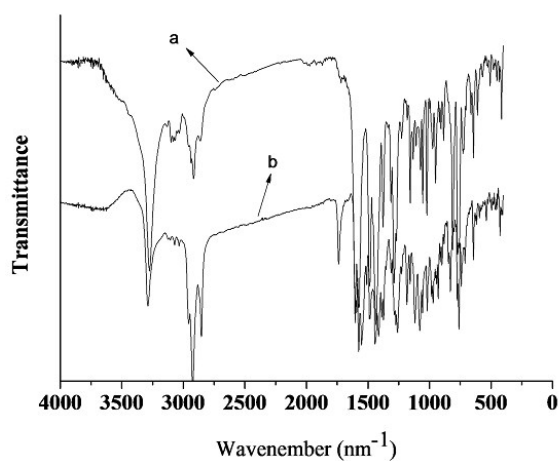


Figure S1. IR spectra of PPM@Fe (a) and BODIPY@Fe₁ (b).

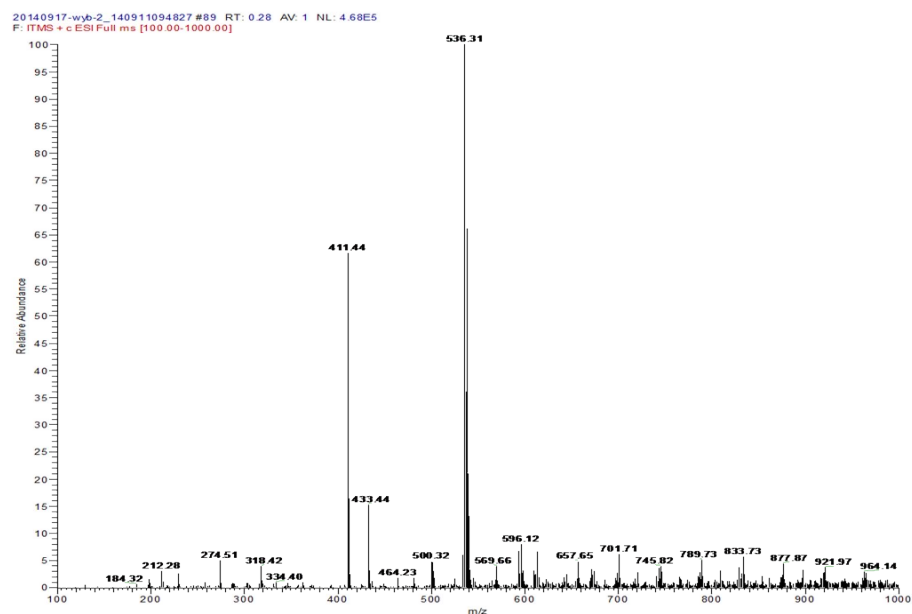


Figure S2. MS of PPM@Fe in MeCN. The main peak at $m/z=536.31$ corresponds to $[(\text{PPMdpa})\text{Fe}(\text{Cl})_2]^+$, indicating the ratio of PPMdpa : Fe in PPM@Fe is 1:1. The peak at $m/z = 411.44$ corresponds to $[(\text{PPMdpa})\text{H}]^+$.

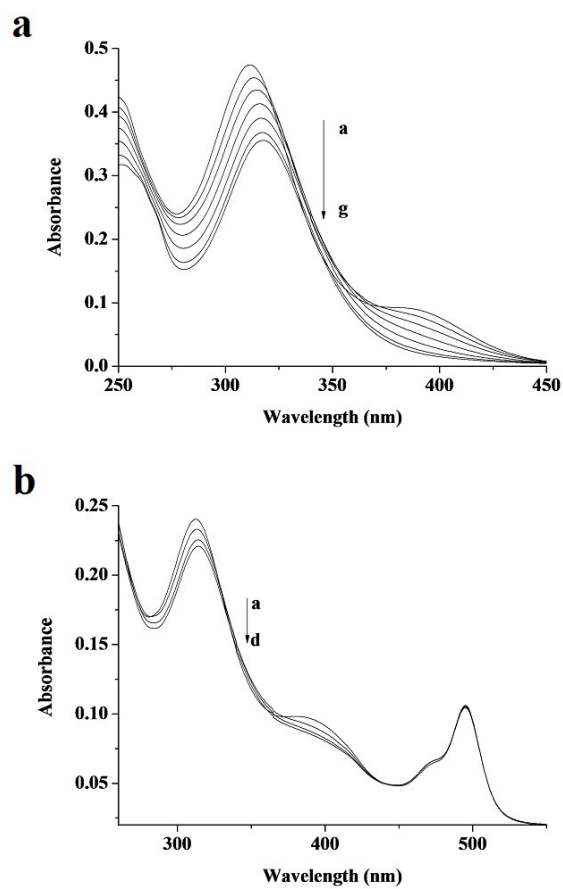


Figure S3. a) UV-vis absorption spectra of PPM@Fe (20 μM, MeCN) in the presence of various amount of H₂O (a – g, C_{H₂O} = 0, 0.28, 0.56, 0.83, 1.1, 1.39, 1.67 M). b) UV-vis absorption spectra of BODIPY@Fe (10 μM, MeCN) in the presence of various amount of H₂O (a – d, C_{H₂O} = 0, 0.56, 1.11, 1.67 M).

