# Supporting Information 

# Synthesis, structure, magnetic and magnetocaloric properties of a series of $\left\{\mathrm{Cr}_{4}{ }^{\text {III }} \mathrm{Ln}^{\mathrm{III}}\right\}$ complexes 

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## 1. Selected bond lengths and angles

Table S1. Selected bond lengths $(\AA)$ and angles $\left({ }^{\circ}\right)$ for $\mathbf{1 a}$.

| Bonds | Length ranges $(\AA)$ | Bonds | Length ranges $(\AA \mathbf{\AA})$ |
| :--- | :--- | :--- | :--- |
| $\mathbf{d}(\mathbf{C r} \cdots \mathbf{O})$ | $1.923(2)-1.982(2)$ | $\mathbf{d}(\mathbf{C r} \cdots \mathbf{N})$ | $2.041(3)-2.079(3)$ |
| $\mathbf{d}(\mathbf{D y} \cdots \mathbf{O})$ | $2.348(2)-2.431(2)$ |  |  |
|  |  |  |  |
| Interatomic distances $(\AA)$ |  | Selected angles $\left.\mathbf{(}^{\circ}\right)$ |  |
| $\mathbf{C r} \cdots \mathbf{C r}$ | $5.2198(7)-7.8419(8)$ | Dy1-O7-Cr2 | $133.57(10)$ |
| $\mathbf{C r} \cdots \mathbf{D y}$ | $3.4399(5)-3.9809(7)$ | Dy1-O17-Cr4 | $132.35(11)$ |
| Selected angles $\left.\mathbf{(}^{\circ}\right)$ |  | Dy1-O3-Cr1 | $105.681(8)$ |
| $\mathbf{C r 2} \cdots \mathbf{D y 1} \cdots \mathbf{C r} 4$ | Dy1-O18-Cr1 | $102.86(8)$ |  |
| $\mathbf{C r 1} \cdots \mathbf{D y 1} \cdots \mathbf{C r 3}$ | $162.923(14)$ | Dy1-O8-Cr3 | $103.62(9)$ |
|  | Dy1-O13-Cr3 | $105.40(9)$ |  |



Figure S1. Ellipsoidal representation of 1a, with a probability of the ellipsoids of $50 \%$.

Table S2. Selected bond lengths $(\AA)$ and angles $\left({ }^{\circ}\right)$ for $\mathbf{1 b}$.

| Bonds | Length ranges $(\AA)$ | Bonds | Length ranges $(\AA)$ |
| :--- | :--- | :--- | :--- |
| $\mathbf{d}(\mathbf{C r} \cdots \mathbf{O})$ | $1.911(2)-1.980(3)$ | $\mathbf{d}(\mathbf{C r} \cdots \mathbf{N})$ | $2.038(3)-2.080(3)$ |
| $\mathbf{d}(\mathbf{D y} \cdots \mathbf{O})$ | $2.348(2)-2.431(2)$ |  |  |
|  |  |  |  |
| Interatomic distances $(\AA)$ |  |  | Selected angles $\left.\mathbf{(}^{\circ}\right)$ |
| $\mathbf{C r} \cdots \mathbf{C r}$ | $5.1764(7)-7.8872(9)$ | Dy1-O7-Cr2 | $132.2(1)$ |
| $\mathbf{C r} \cdots \mathbf{D y}$ | $3.4479(6)-3.9837(8)$ | Dy1-O17-Cr4 | $132.5(1)$ |
| Selected angles $\left.\mathbf{(}^{\circ}\right)$ |  | Dy1-O3-Cr1 | $106.6(1)$ |
| $\mathbf{C r 2} \cdots \mathbf{D y 1} \cdots \mathbf{C r} 4$ | $165.96(2)$ | Dy1-O18-Cr1 | $104.2(1)$ |
| $\mathbf{C r 1} \cdots \mathbf{D y 1} \cdots \mathbf{C r 3}$ | $117.20(2)$ | Dy1-O8-Cr3 | $104.5(1)$ |
|  | Dy1-O13-Cr3 | $105.4(1)$ |  |



Figure S2. Top: Ellipsoidal representation of 1b, with a probability of the ellipsoids of $50 \%$. Bottom: Ball and stick representation with labelled atoms of the central metal core of complexes $\mathbf{1 a}$ and $\mathbf{1 b}$.

Table S3. Selected bond lengths $(\AA)$ and angles $\left(^{\circ}\right)$ for 2.

| Bonds | Length ranges ( $\AA$ ) | Bonds | Length ranges ( $\AA$ ) |
| :---: | :---: | :---: | :---: |
| $\mathrm{d}(\mathrm{Cr} \cdots \mathrm{O})$ | 1.916(5)-1.981(5) | $\mathrm{d}(\mathrm{Cr} \cdots \mathrm{N})$ | 2.032(7)-2.086(7) |
| $\mathbf{d}(\mathbf{T b} \cdots \mathbf{O}$ | 2.358(5)-2.426(4) |  |  |
| Interatomic distances ( $\AA$ ) |  | Selected angles ( ${ }^{\circ}$ ) |  |
| $\mathrm{Cr} \cdots \mathrm{Cr}$ | 5.1824(16)-7.8924(17) | Tb1-O7-Cr2 | 131.88(22) |
| $\mathbf{C r} \cdots$ Tb | 3.4553(13)-3.8949(12) | Tb1-O17-Cr4 | 131.61(22) |
| Selected angles ( ${ }^{\circ}$ ) |  | Tb1-O3-Cr1 | 106.98(20) |
| Cr2 $\cdots$ Tb1 $\cdots$ Cr4 | 166.187(27) | Tb1-O18-Cr1 | 104.02(19) |
| Cr1 $\cdots$ Tb1 $\cdots$ Cr3 | 106.992(29) | Tb1-O8-Cr3 | 104.11(19) |
|  |  | Tb1-O13-Cr3 | 105.39(20) |



Figure S3. Top: Ellipsoidal representation of 2, with a probability of the ellipsoids of $30 \%$. Bottom: Ball and stick representation with labelled atoms of the central metal core of complexes 2.

Table S4. Selected bond lengths $(\AA)$ and angles $\left(^{\circ}\right)$ for $\mathbf{3}$.

| Bonds | Length ranges $(\AA)$ | Bonds | Length ranges $(\AA)$ |
| :--- | :--- | :--- | :--- |
| $\mathbf{d}(\mathbf{C r} \cdots \mathbf{O})$ | $1.911(3)-1.979(3)$ | $\mathbf{d}(\mathbf{C r} \cdots \mathbf{N})$ | $2.044(3)-2.081(4)$ |
| $\mathbf{d}(\mathbf{G d} \cdots \mathbf{O})$ | $2.384(3)-2.449(3)$ |  |  |
|  |  |  |  |
| Interatomic distances $(\AA)$ |  | Selected angles $\left(^{\circ}\right)$ |  |
| $\mathbf{C r} \cdots \mathbf{C r}$ | $5.1863(9)-7.9177(10)$ | Gd1-O4-Cr1 | $131.63(13)$ |
| $\mathbf{C r} \cdots \mathbf{G d}$ | $3.4553(13)-3.8949(12)$ | Gd1-O13-Cr4 | $131.49(13)$ |
| Selected angles $\left(^{\circ}\right)$ |  | Gd1-O11-Cr3 | $105.68(12)$ |
| $\mathbf{C r 1} \cdots \mathbf{G d 1} \cdots \mathbf{C r} 4$ | $167.254(19)$ | Gd1-O12-Cr3 | $103.98(11)$ |
| $\mathbf{C r 2} \cdots \mathbf{G d 1} \cdots \mathbf{C r} 3$ | $106.971(19)$ | Gd1-O5-Cr2 | $103.86(11)$ |
|  |  | Gd1-O8-Cr2 | $106.87(12)$ |




Figure S4. Top: Ellipsoidal representation of 3, with a probability of the ellipsoids of $30 \%$. Bottom: Ball and stick representation with labelled atoms of the central metal core of complexes 3 .

## 2. Crystal packing and crystal structure representation



Figure S5. Representation of the highly distorted square anti-prism geometry around Dy1. Colour code: Dy: teal, O: Red.


Figure S6. Ball and stick representations of the octahedral Cr 1 (left) and Cr 2 (right) closed environment, in complexes 1a, 1b, and 2. Colour code: Cr: yellow, O : red, N : blue, C: grey.



Figure S7. Ball and stick representations of the octahedral Cr 2 (left) and Cr 3 (right) closed environment, in complexes 1a, 1b, and 2. Colour code: Cr: yellow, O: red, N: blue, C: grey.


Figure S8. Ball and stick representation of the unit cell of complex 1a (along a axis). Nitrate anions and water molecules are omitted for clarity, as well as the H atoms. C atoms are represented as sticks for clarity. $\mathrm{Cr}^{\mathrm{III}}$ : yellow, $\mathrm{Dy}^{\mathrm{III}}$ : teal, O : red, N : blue, C : grey.


Figure S9. Ball and stick representation of an extended unit cell of complex $\mathbf{1 a}$ (along a axis). Nitrate anions and water molecules are omitted for clarity, as well as the H atoms. C atoms are represented as sticks for clarity. $\mathrm{Cr}^{\text {III }}$ : yellow, $\mathrm{Dy}^{\text {III }}$ : teal, O : red, N : blue, C : grey.


Figure S10. Ball and stick representation of the unit cell of complex 1b. Nitrate anions and water molecules are omitted for clarity, as well as the H atoms. C atoms are represented as sticks for clarity. $\mathrm{Cr}^{\text {III }}$ : yellow, $\mathrm{Dy}^{\text {III }}$ : teal, O: red, N : blue, C : grey.

## 3. Coordination geometry of Dy (III) ion

Results of Continuous Shape measurements for LnCr 4 complexes. Legend and corresponding symmetry: HPY: C7v Heptagonal pyramid; HBPY: D6h Hexagonal bipyramid; CU: Oh, Cube; SAPR: D4d Square antiprism; TDD: D2d, Triangular dodecahedron; JGBF: D2d, Johnson gyrobifastigium; JETBPY: D3h, Johnson elongated triangular bipyramid; JBTPR: C2v, Biaugmented trigonal prism J50; BTPR: C2v, Biaugmented trigonal prism; JSD: D2d, Snub diphenoid; TT: Td, Triakis tetrahedron; ETBPY: D3h, Elongated trigonal bipyramid

Table S5 (two parts). Results of Continuous Shape measurements for Dy(III) ions in complex 1a.

| HPY | HBPY | CU | SAPR | TDD | JGBF |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 19.595 | 12.811 | 6.340 | 4.092 | 3.921 | 15.579 |


| JETBPY | JBTPR | BTPR | JSD | TT | ETBPY |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 23.226 | 6.321 | 5.726 | 8.109 | 7.189 | 20.569 |

