Supporting Information

Synthesis, structure, magnetic and magnetocaloric properties of a series of {Cr₄^{III}Ln^{III}} complexes

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Table of Contents

1. Selected bond lengths and angles	2
2. Crystal packing and crystal structure representation	6
3. Coordination geometry of Dy(III) ion	9

1. Selected bond lengths and angles

Bonds	Length ranges (Å)	Bonds	Length ranges (Å)	
d(Cr…O)	1.923(2)-1.982(2)	d(Cr…N)	2.041(3)-2.079(3)	
d(Dy…O)	2.348(2)-2.431(2)			
Interatomic distances (Å)		Selected angles (°)		
Cr…Cr	5.2198(7)-7.8419(8)	Dy1–O7–Cr2	133.57(10)	
Сг…Ду	3.4399(5)-3.9809(7)	Dy1–O17–Cr4	132.35(11)	
Selected angles (°)		Dy1-O3-Cr1	105.681(8)	
Cr2…Dy1…Cr4	162.923(14)	Dy1–O18–Cr1	102.86(8)	
Cr1…Dy1…Cr3	110.041(16)	Dy1-O8-Cr3	103.62(9)	
		Dy1013Cr3	105.40(9)	

Table S1. Selected bond lengths (Å) and angles (°) for 1a.



Figure S1. Ellipsoidal representation of 1a, with a probability of the ellipsoids of 50%.

Bonds	Length ranges (Å)	Bonds	Length ranges (Å)	
d(Cr…O)	1.911(2)-1.980(3)	d(Cr…N)	2.038(3)-2.080(3)	
d(Dy…O)	2.348(2)-2.431(2)			
Interatomic distances (Å)		Selected angles (°)		
Cr…Cr	5.1764(7)-7.8872(9)	Dy1–O7–Cr2	132.2(1)	
Cr…Dy	3.4479(6)-3.9837(8)	Dy1–O17–Cr4	132.5(1)	
Selected angles (°)		Dy1–O3–Cr1	106.6(1)	
Cr2…Dy1…Cr4	165.96(2)	Dy1-O18-Cr1	104.2(1)	
Cr1…Dy1…Cr3	117.20(2)	Dy1-O8-Cr3	104.5(1)	
		Dy1-O13-Cr3	105.4(1)	

Table S2. Selected bond lengths (Å) and angles (°) for 1b.



Figure S2. Top: Ellipsoidal representation of **1b**, with a probability of the ellipsoids of 50%. Bottom: Ball and stick representation with labelled atoms of the central metal core of complexes **1a** and **1b**.

Bonds	Length ranges (Å)	Bonds	Length ranges (Å)	
d(Cr…O)	1.916(5)-1.981(5)	d(Cr…N) 2.032(7)-2.086(7)		
d(Tb…O)	2.358(5)-2.426(4)			
Interatomic distances (Å)		Selected angles (°)		
Cr…Cr	5.1824(16)-7.8924(17)	Tb1–O7–Cr2	131.88(22)	
Cr…Tb	3.4553(13)-3.8949(12)	Tb1–O17–Cr4	131.61(22)	
Selected angles (°)		Tb1-O3-Cr1	106.98(20)	
Cr2…Tb1…Cr4	166.187(27)	Tb1018Cr1	104.02(19)	
Cr1···Tb1···Cr3	106.992(29)	Tb108Cr3	104.11(19)	
		Tb1-O13-Cr3	105.39(20)	

Table S3. Selected bond lengths (Å) and angles (°) for 2.



Figure S3. Top: Ellipsoidal representation of **2**, with a probability of the ellipsoids of 30%. Bottom: Ball and stick representation with labelled atoms of the central metal core of complexes **2**.

Bonds	Length ranges (Å)	Bonds	Length ranges (Å)
d(Cr…O)	1.911(3)-1.979(3)	d(Cr…N)	2.044(3)-2.081(4)
d(Gd…O)	2.384(3)-2.449(3)		
Interatomic distances (Å)		Selected angles (°)	
Cr…Cr	5.1863(9)-7.9177(10)	Gd1-O4-Cr1	131.63(13)
Cr…Gd	3.4553(13)-3.8949(12)	Gd1-O13-Cr4	131.49(13)
Selected angles (°)		Gd1-O11-Cr3	105.68(12)
Cr1…Gd1…Cr4	167.254(19)	Gd1-O12-Cr3	103.98(11)
Cr2…Gd1…Cr3	106.971(19)	Gd1–O5–Cr2	103.86(11)
		Gd1–O8–Cr2	106.87(12)

Table S4. Selected bond lengths (Å) and angles (°) for 3.



Figure S4. Top: Ellipsoidal representation of **3**, with a probability of the ellipsoids of 30%. Bottom: Ball and stick representation with labelled atoms of the central metal core of complexes **3**.

2. Crystal packing and crystal structure representation



Figure S5. Representation of the highly distorted square anti-prism geometry around Dy1. Colour code: Dy: teal, O: Red.



Figure S6. Ball and stick representations of the octahedral Cr1 (left) and Cr2 (right) closed environment, in complexes 1a, 1b, and 2. Colour code: Cr: yellow, O: red, N: blue, C: grey.



Figure S7. Ball and stick representations of the octahedral Cr2 (left) and Cr3 (right) closed environment, in complexes 1a, 1b, and 2. Colour code: Cr: yellow, O: red, N: blue, C: grey.



Figure S8. Ball and stick representation of the unit cell of complex **1a** (along a axis). Nitrate anions and water molecules are omitted for clarity, as well as the H atoms. C atoms are represented as sticks for clarity. Cr^{III}: yellow, Dy^{III}: teal, O: red, N: blue, C: grey.



Figure S9. Ball and stick representation of an extended unit cell of complex **1a** (along a axis). Nitrate anions and water molecules are omitted for clarity, as well as the H atoms. C atoms are represented as sticks for clarity. Cr^{III} : yellow, Dy^{III} : teal, O: red, N: blue, C: grey.



Figure S10. Ball and stick representation of the unit cell of complex **1b**. Nitrate anions and water molecules are omitted for clarity, as well as the H atoms. C atoms are represented as sticks for clarity. Cr^{III}: yellow, Dy^{III}: teal, O: red, N: blue, C: grey.

3. Coordination geometry of Dy(III) ion

Results of Continuous Shape measurements for LnCr4 complexes. Legend and corresponding symmetry: HPY: C7v Heptagonal pyramid; HBPY: D6h Hexagonal bipyramid; CU: Oh, Cube; SAPR: D4d Square antiprism; TDD: D2d, Triangular dodecahedron; JGBF: D2d, Johnson gyrobifastigium; JETBPY: D3h, Johnson elongated triangular bipyramid; JBTPR: C2v, Biaugmented trigonal prism; JSD: D2d, Snub diphenoid; TT: Td, Triakis tetrahedron; ETBPY: D3h, Elongated trigonal bipyramid

Table S5 (two parts). Results of Continuous Shape measurements for Dy(III) ions in complex 1a.

HPY	HBPY	CU	SAPR	TDD	JGBF
19.595	12.811	6.340	4.092	3.921	15.579

JETBPY	JBTPR	BTPR	JSD	TT	ETBPY
23.226	6.321	5.726	8.109	7.189	20.569