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Supplementary Information for:

Interplay between hydrophobicity and basicity toward the catalytic activity of isoreticular organocatalyst MOFs

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Figure S1. FT-IR spectra of the ligands, L1 and RL1.



Figure **S2**. FT-IR spectra of the ligands, L2 and RL2.



Figure **S3**. Mass spectra of the ligands, L2 and RL2.



Figure S4. Mass spectra of the ligands, L1 and RL1.



Figure S5 (a). 1H-NMR spectra of L1 and L2



Figure S5 (b). 1H-NMR spectra of RL1 and RL2.



Figure S6. Comparative FT-IR spectra of TMU-6(L1) synthesized with different method and the recycled one after the condensation reaction.

The FT-IR spectrum of the MOFs shows the same characteristic bands of the dicarboxylate groups of the H_2 oba ligands at ca. 1606 and 1404 cm⁻¹ for the asymmetric and symmetric vibrations, respectively, which indicate coordination of carboxylate ligands to metal ions. The absence of the expected characteristic bands at 1730–1690 cm⁻¹ for the protonated carboxylate groups indicates the complete deprotonation of H_2 oba ligand in the reaction with Zn ions. This point exists in FT-IR spectrum of all prepared MOFs.



Figure S7. Comparative FT-IR spectra of TMU-21(L2) synthesized with different method and the recycled one after the condensation reaction.

The FT-IR spectrum of the MOFs shows the same characteristic bands of the dicarboxylate groups of the H₂oba ligands at ca. 1609 and 1402 cm⁻¹ for the asymmetric and symmetric vibrations, respectively, which indicate coordination of carboxylate ligands to metal ions.



Figure S8. Comparative FT-IR spectra of TMU-6(RL1) and the recycled one after the condensation reaction.



Figure S9. Comparative FT-IR spectra of TMU-21(RL2) synthesized and the recycled one after the condensation reaction.



Figure S10. Thermogravimetric analysis of TMU-6(L1) (blue line) and TMU-21(L2) (red line)



Figure S11. Thermogravimetric analysis of TMU-6(RL1) (blue line) and TMU-21(RL2) (red line)



Figure S12. PXRD patterns of TMU-6(L1), TMU-6(RL1), TMU-21(L2) and TMU-21(RL2) revealed that these four MOFs are isoreticular framework.



Figure S13. PXRD patterns of simulated, as-synthesized, mechano-synthesized, activated, after reaction and water stability of TMU-6(L1). The recycling PXRD of the MOFs were recorded after 7th cycle of reusability test.



Figure S14. PXRD patterns of simulated, as-synthesized, mechano-synthesized, activated, after reaction and water stability of TMU-21(L2). The recycling PXRD of the MOFs were recorded after 7th cycle of reusability test.



Figure S15. PXRD patterns of simulated, mechano-synthesized, after reaction and water stability of TMU-6(L1) and TMU-6(RL1). The recycling PXRD of the MOFs were recorded after 7th cycle of reusability test.



Figure S16. PXRD patterns of simulated, mechano-synthesized, after reaction and water stability of TMU-21(L2) and TMU-21(RL2). The recycling PXRD of the MOFs were recorded after 7th cycle of reusability test.



Table S1. Catalytic performance of TMU-6(L1), TMU-6(RL1), TMU-21(L2) and TMU-21(RL2), obtained by mechanochemical synthesis, within aldol condensation of malonitrile with 2-cyclohexen-1-one, 24h.

Identification of the product: 2-(cyclohex-2-enylidene)malononitrile

FT-IR data (KBr pellet, v/cm-1): 436(m), 570(m), 806(w), 1024(m), 1095(m), 1260(w), 1443(w), 1637(s), 2203(s), 2925(s), 3432(m)

¹H NMR (500 MHz, CDCl₃): 1.86 (m, 2H), 2.44 (m, 2H), 2.55 (t, 2H), 6.13 (m, 1H), 6.70 (d, 1H)

$ \begin{array}{c} $						
Time	TMU-6(RL1)	TMU-6(L1)	TMU-21(L2)	TMU-21(RL2)		
12 h	54	47.8	40.2	25		
24 h	93.7	89	76.3	60		
36 h	97.3	99.2	93	80		
60 h			99.5	95.4		

Table S2. Time in depended catalytic performance of TMU-6(L1), TMU-6(RL1), TMU-21(L2) and TMU-21(RL2), obtained by mechanochemical synthesis within aldol condensation of malonitrile with 2-cyclopenten-1-one.

Identification of the product: 2-(cyclopent-2-enylidene)malononitrile FT-IR data (KBr pellet, v/cm⁻¹): 566(m), 1097(m), 1304(m), 1439(m), 1577(s), 1642(vs), 2204(vs), 2948(m), 3234(m), 3350(m)

¹H NMR (500 MHz, CDCl₃): 1.61 (m, 2H), 2.06 (t, 2H), 5.09 (m, 1H), 5.82 (d, 1H)

$ \begin{array}{c} $						
Time	TMU-21(L2)	TMU-6(L1)	TMU-6(RL1)	TMU-21(RL2)		
12 h	53.5	42.8	38.5	27.6		
24 h	72.7	53.4	48	31.3		
36 h	87	72.4	69	63		
60 h	91.4	80.8	77.8	70.3		

Table S3. Time in depended catalytic performance of TMU-6(L1), TMU-6(RL1), TMU-21(L2) and TMU-21(RL2), obtained by mechanochemical synthesis, within aldol condensation of malonitrile with 4,4-dimethylcyclohexen-1-one.

Identification of the product: 2-(4,4-dimethylcyclohex-2-enylidene)malononitrile

FT-IR data (KBr pellet, v/cm⁻¹): 682(m), 1022(m), 1219(m), 1456(m), 1553(m), 1605(s), 2192(s), 2959(m), 3209(m), 3347(m)

¹H NMR (500 MHz, CDCl₃): 1.14 (s, 6H), 1.71 (t, 2H), 2.83 (t, 2H), 6.45 (d, 1H), 6.62 (d, 1H)



Figure S17. The images of the catalysts in H_2O solvents after addition of toluene (up), in H_2O solvents after addition of dichloromethane (down) (a) TMU-21(L2), (b) TMU-21(RL2), (c) TMU-6(L1) and (d) TMU-6(RL1).



Figure S18. Yield-versus-time profile of aldol-type condensation reaction of (a) 2-cyclopenten-1-one and (b) 4,4-dimethyl-2-cyclohexen-1-one catalyzed by four MOF structures in the reaction conditions indicated in Table 1 of the manuscript.



Figure S19 Comparison of the catalyst reactivity of various substrates in the presence of different basic MOFs.



Figure S20. Reusability of TMU-6(RL1) (gray) and TMU-21(L2) (orange) in aldol-type condensation reaction of 2-cyclopenten-1-one and 4,4-dimethyl-2-cyclohexen-1-one (0.6 mmol), respectively. (Conditions: catalyst (5 mol%), malononitrile (0.9 mmol), MeOH, 60 °C, 24h)