

## Supporting Information

Table S1: The elemental analysis. The Br/I atom ration was detected by the EDS.

CsPbBr <sub>3</sub> perovskite		CsPbBr <sub>1.5</sub> I <sub>1.5</sub> perovskite	
Atom	Atom ratio (%)	Atom	Atom ratio (%)
Br	1.73	Br	0.74
Pb	0.49	I	0.53

Table S2: Detailed information of different composition samples.

Samples	Emission peak /nm	FWHM /nm	QY
CsPbBr <sub>3</sub>	517	26	46.91%
CsPbBr <sub>2.7</sub> I <sub>0.3</sub>	522	25	31.47%
CsPbBr <sub>2.4</sub> I <sub>0.6</sub>	531	24	20.50%
CsPbBr <sub>2.1</sub> I <sub>0.9</sub>	544	24	24.31%
CsPbBr <sub>1.8</sub> I <sub>1.2</sub>	566	26	19.19%
CsPbBr <sub>1.5</sub> I <sub>1.5</sub>	566	30	26.73%
CsPbBr <sub>1.2</sub> I <sub>1.8</sub>	631	38	30.49%
CsPbBr <sub>0.9</sub> I <sub>2.1</sub>	655	40	48.97%
CsPbBr <sub>0.6</sub> I <sub>2.4</sub>	675	38	56.42%
CsPbBr <sub>0.3</sub> I <sub>2.7</sub>	688	37	62.23%
CsPbI <sub>3</sub>	693	38	69.41%

Table S3: Detailed polarization of different composition samples.

Sample	I <sub>max</sub>	I <sub>min</sub>	P
CsPbBr <sub>3</sub>	8.03 e5	6.89 e5	0.08
CsPbBr <sub>2.7</sub> I <sub>0.3</sub>	7.94 e5	6.62 e5	0.09
CsPbBr <sub>2.4</sub> I <sub>0.6</sub>	5.32 e5	4.16 e5	0.12
CsPbBr <sub>2.1</sub> I <sub>0.9</sub>	5.72 e5	5.10 e5	0.06
CsPbBr <sub>1.8</sub> I <sub>1.2</sub>	5.15 e5	4.66 e5	0.05
CsPbBr <sub>1.5</sub> I <sub>1.5</sub>	4.75 e5	4.15 e5	0.07
CsPbBr <sub>1.2</sub> I <sub>1.8</sub>	7.47 e5	4.90 e5	0.21
CsPbBr <sub>0.9</sub> I <sub>2.1</sub>	8.17 e5	4.50 e5	0.29
CsPbBr <sub>0.6</sub> I <sub>2.4</sub>	5.81 e5	2.99 e5	0.32
CsPbBr <sub>0.3</sub> I <sub>2.7</sub>	6.22 e5	2.85 e5	0.37
CsPbI <sub>3</sub>	7.39 e5	3.48 e5	0.36

Table S4: The polarization properties of different volume concentration in the hexane solution for the sample of  $\text{CsPbBr}_{0.3}\text{I}_{2.7}$ .

Sample	$\phi\text{B}$ (VB/V)	$I_{\max}$	$I_{\min}$	P
$\text{CsPbBr}_{0.3}\text{I}_{2.7}$	100 $\mu\text{L}/6 \text{ mL}$	6.22 e5	2.85 e5	0.37
$\text{CsPbBr}_{0.3}\text{I}_{2.7}$	50 $\mu\text{L}/6 \text{ mL}$	7.67 e5	3.59 e5	0.36
$\text{CsPbBr}_{0.3}\text{I}_{2.7}$	25 $\mu\text{L}/6 \text{ mL}$	5.06 e4	2.22 e4	0.39
$\text{CsPbBr}_{0.3}\text{I}_{2.7}$	10 $\mu\text{L}/6 \text{ mL}$	8.61 e4	3.06 e4	0.48

Table S5: The polarization properties of the  $\text{CsPbI}_3$  film and  $\text{CsPbBr}_3$  film.

Sample	$I_{\max}$	$I_{\min}$	P
$\text{CsPbBr}_3$ film	2.43 e6	2.41 e6	0.0041
$\text{CsPbI}_3$ film	4.63 e6	1.96 e6	0.41

Figure S1: Energy dispersive spectrometer (EDS).

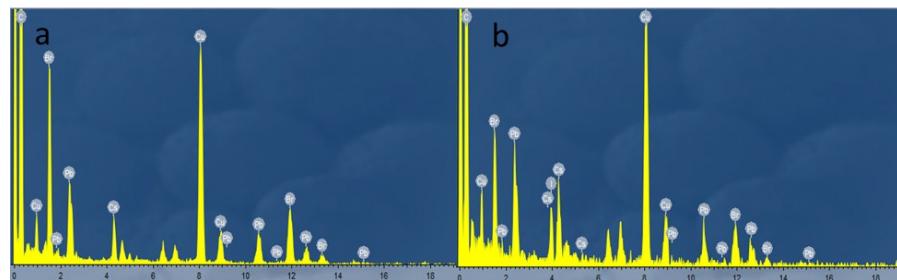


Figure S2: The UV/vis absorption spectra of perovskites samples in hexane solution.

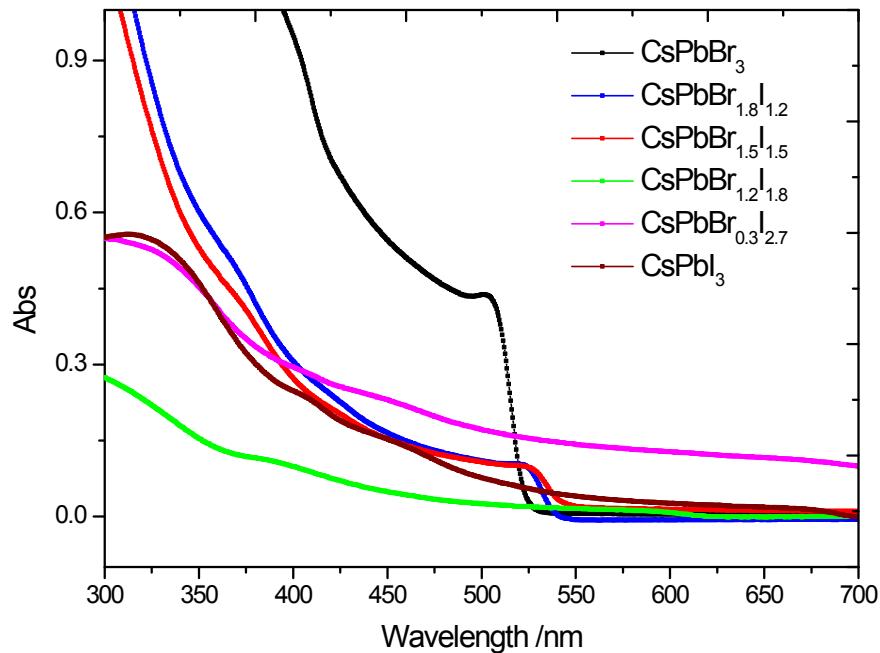
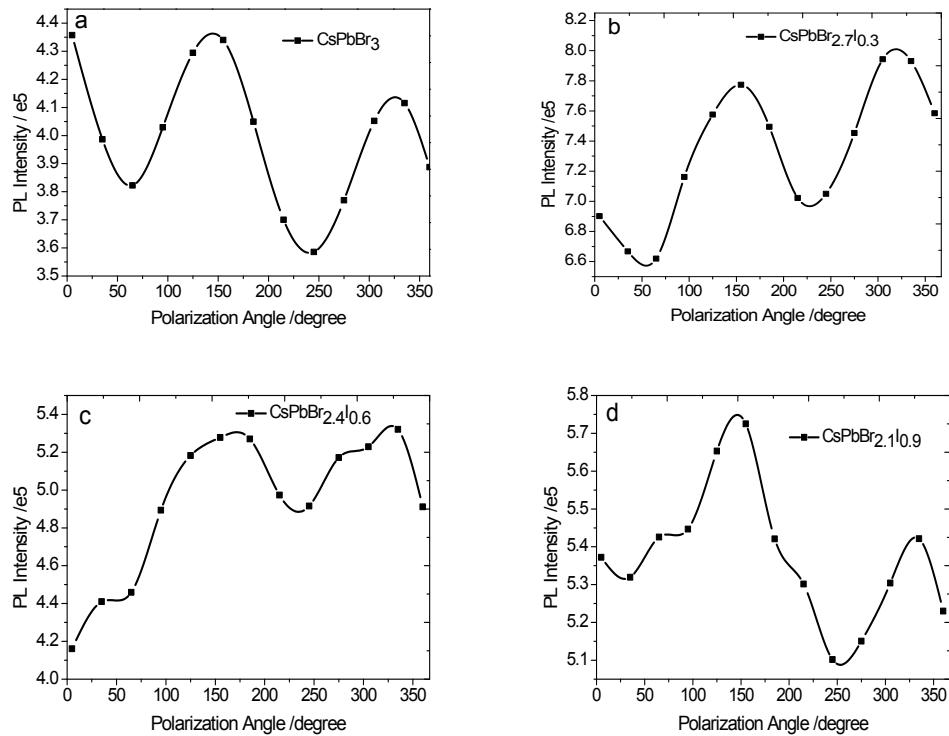


Figure S3: The polarization properties of  $\text{CsPbX}_3$  or the Br/I perovskite samples in hexane solution.



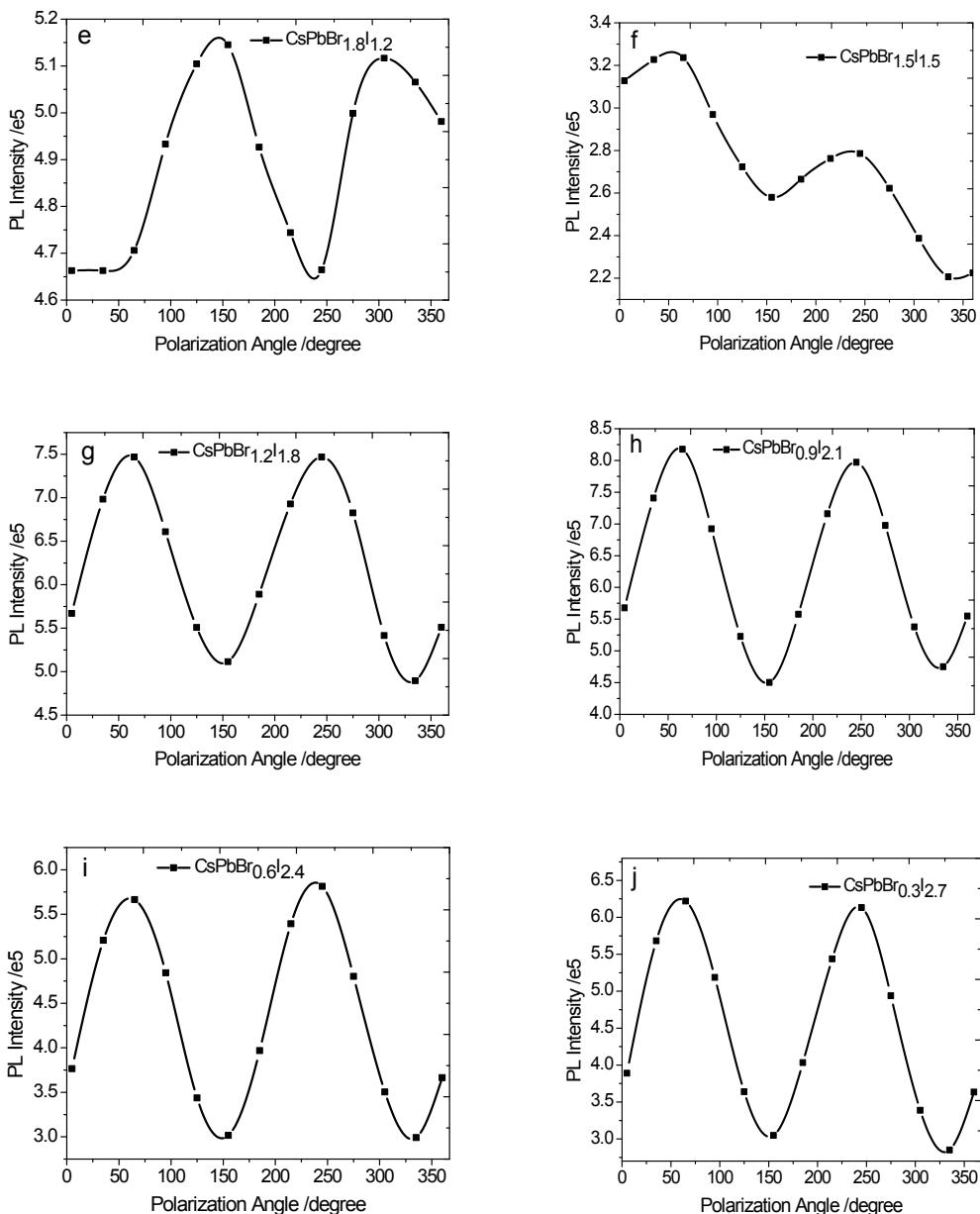
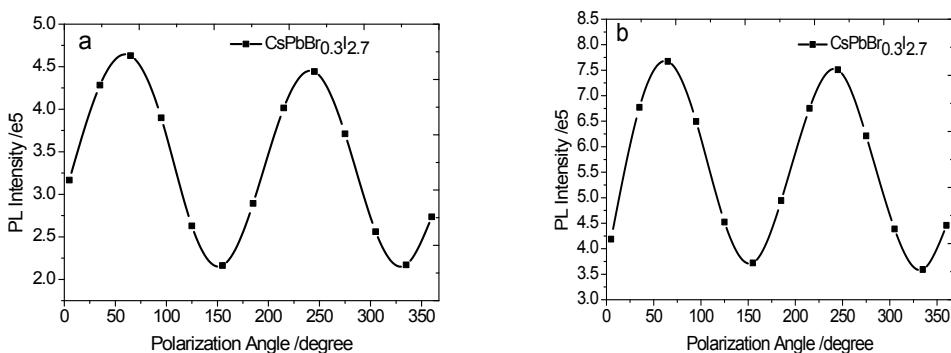


Figure S4: The polarization properties of different volume concentration in the hexane solution for the sample of CsPbBr<sub>0.3</sub>I<sub>2.7</sub> (VB/V: a: 100  $\mu$ L/6 mL; b: 50  $\mu$ L/6 mL; c: 25  $\mu$ L/6 mL; d: 10  $\mu$ L/6 mL).



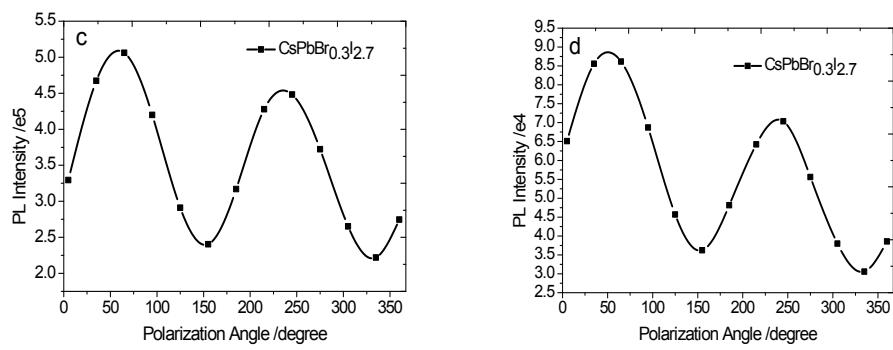


Figure S5: The polarization properties of perovskite films e: a:  $\text{CsPbBr}_3$  film; b:  $\text{CsPbI}_3$  film.

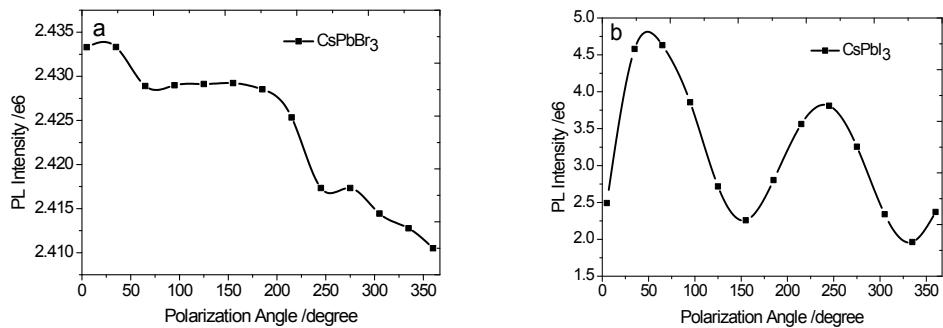


Figure S6: The TEM images of CdSe/ZnS QDs (Left) and  $\text{CsPbBr}_3$  (Right).

