

## Supporting Information

### **The effect of nanofibrous Al<sub>2</sub>O<sub>3</sub> aspect ratio on Fischer-Tropsch synthesis over cobalt catalysts**

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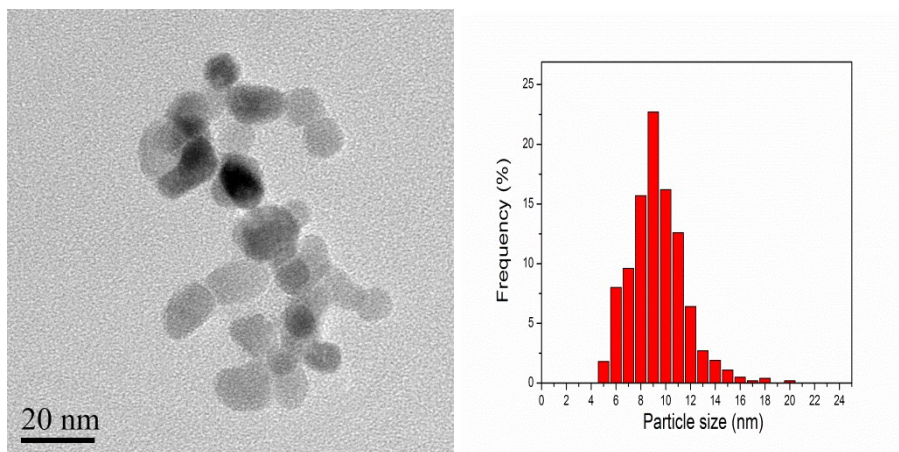
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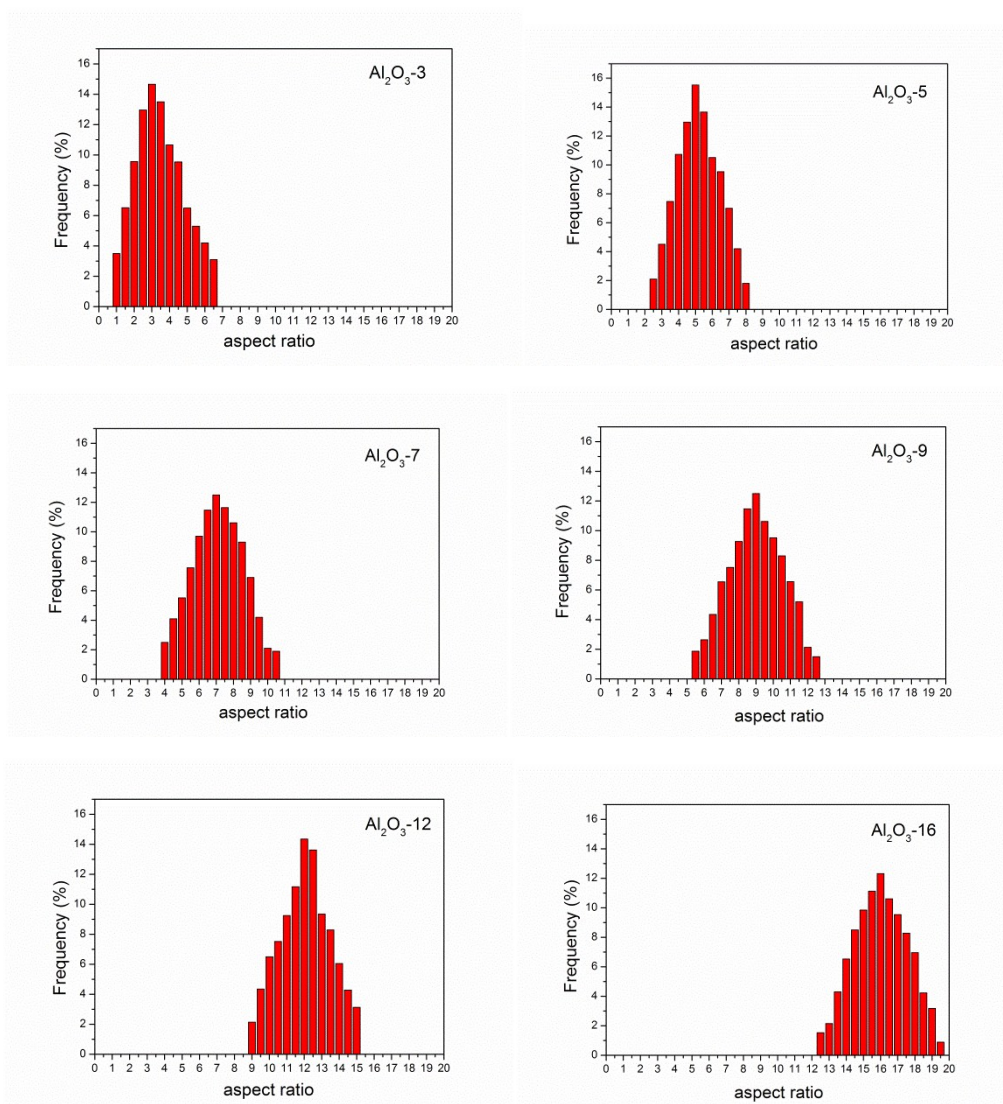
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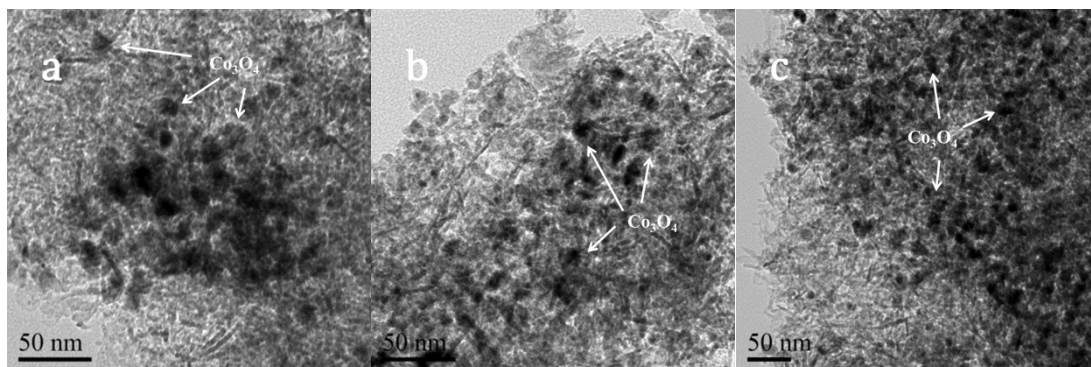
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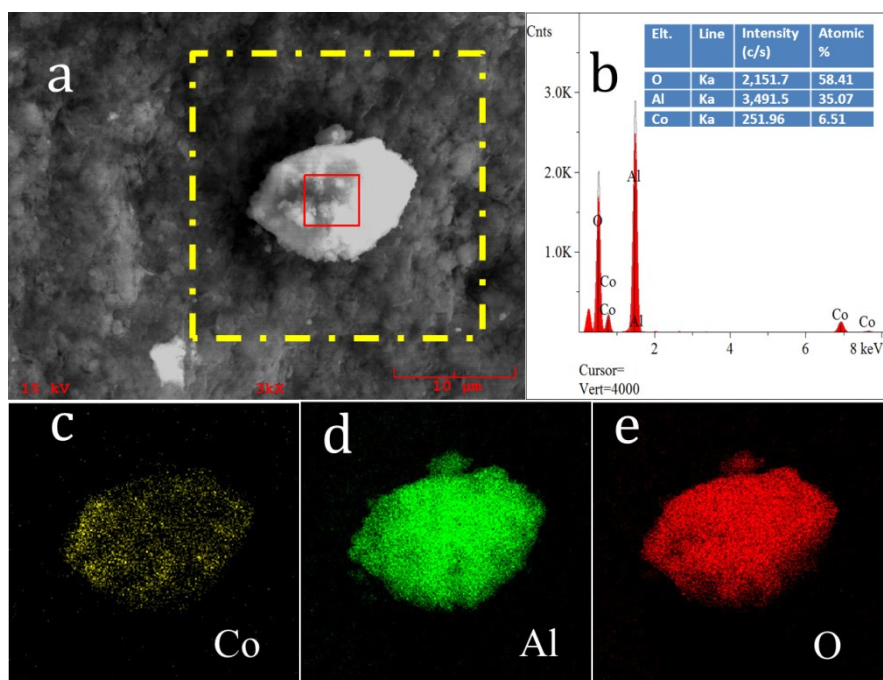
**Figure S1.** TEM images  $\text{Co}_3\text{O}_4$  nanoparticles and the histograms of the  $\text{Co}_3\text{O}_4$  particle size.



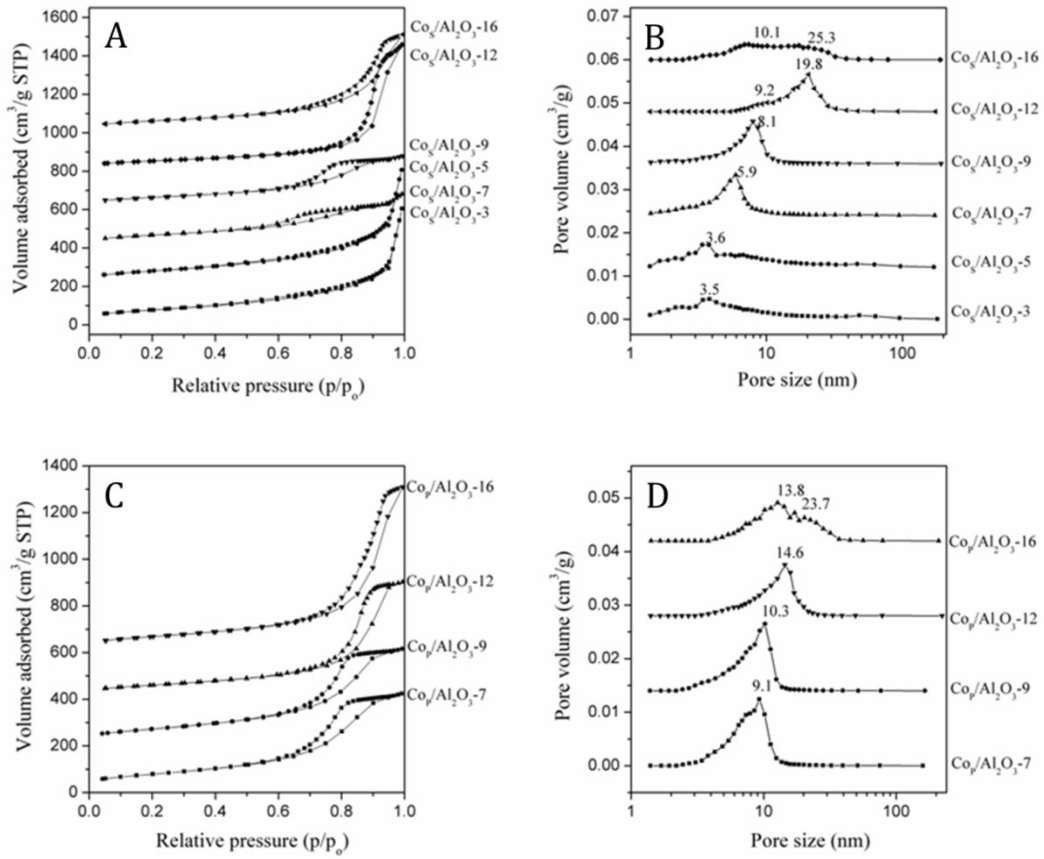
**Figure S2.** The histograms of the average aspect ratio of nanofibrous alumina.



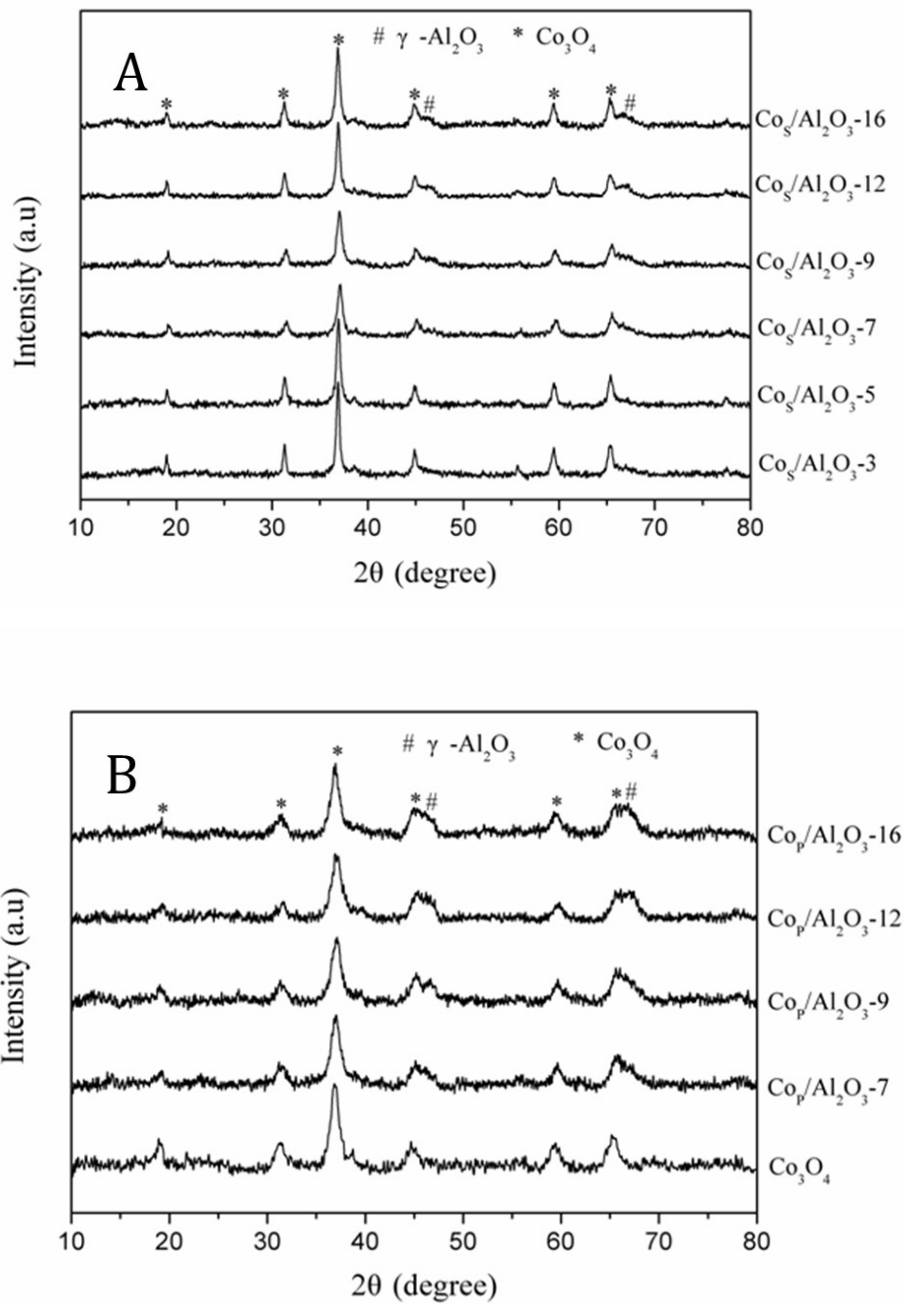
**Figure S3.** TEM images of the catalysts ( $\text{Co}_3\text{O}_4$  highlighted by white arrows): (a)  $\text{Co}_p/\text{Al}_2\text{O}_3$ -9; (b)  $\text{Co}_p/\text{Al}_2\text{O}_3$ -12 and (c)  $\text{Co}_p/\text{Al}_2\text{O}_3$ -16.



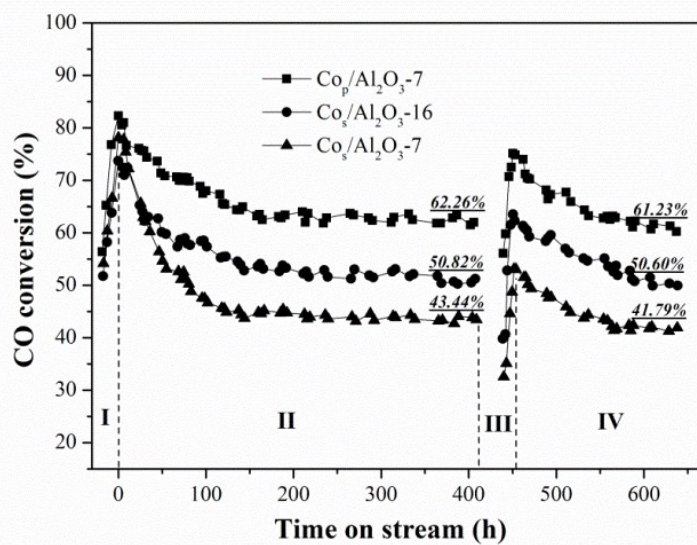
**Figure S4.** SEM-EDS elemental mapping images for Co, Al, and O of the corresponding region (yellow frame) of  $\text{Co}_p/\text{Al}_2\text{O}_3$ -16.



**Figure S5.** Nitrogen adsorption/desorption isotherms (A for impregnated catalysts, C for rearranged catalysts) and pore size distributions of the catalysts (B for impregnated catalysts, D rearranged catalysts).



**Figure S6.** Powder XRD patterns of the catalysts and prepared  $\text{Co}_3\text{O}_4$  nanoparticles.



**Figure S7.** CO conversion as a function of time on stream. After reduced, the reactor temperature was increased slowly to 503 K (Period I), run at 1.0MPa, 503K, GHSV = 3.2 SL•g<sup>-1</sup>•h<sup>-1</sup>, 2.9 SL•g<sup>-1</sup>•h<sup>-1</sup> and 6.5 SL•g<sup>-1</sup>•h<sup>-1</sup> for catalyst Co<sub>s</sub>/Al<sub>2</sub>O<sub>3</sub>-7, Co<sub>s</sub>/Al<sub>2</sub>O<sub>3</sub>-16 and Co<sub>p</sub>/Al<sub>2</sub>O<sub>3</sub>-7, respectively (Period II and IV), re-reduced treatment in flowing N<sub>2</sub>/H<sub>2</sub> (1:1, 6 NL h<sup>-1</sup> g<sup>-1</sup>) for 10 h at 503 K.

**Table S1.** The synthesis condition of nanofibers alumina with various aspect ratios.

Sample	Amount of acetic acid added (mL)	Aging temperature (K)	Average aspect ratio of sample
Al <sub>2</sub> O <sub>3</sub> -3	1.0	333	3
Al <sub>2</sub> O <sub>3</sub> -5	1.0	353	5
Al <sub>2</sub> O <sub>3</sub> -7	0.5	353	7
Al <sub>2</sub> O <sub>3</sub> -9	0	353	9
Al <sub>2</sub> O <sub>3</sub> -12	0	373	12

**Table S2.** Nitrogen adsorption results and crystallite size of the nanofibrous Al<sub>2</sub>O<sub>3</sub> supports.

sample	BJH pore size (nm)	Average pore size <sup>a</sup> (nm)	S <sub>BET</sub> (m <sup>2</sup> ·g <sup>-1</sup> ) <sup>1)</sup>	Pore volume (cm <sup>3</sup> ·g <sup>-1</sup> )	Crystallites size <sup>b</sup> (nm)
Al <sub>2</sub> O <sub>3</sub> -3	3.8, 6.5	16.4	426	1.75	3.5
Al <sub>2</sub> O <sub>3</sub> -5	4.9, 9.2	18.2	419	1.90	3.8
Al <sub>2</sub> O <sub>3</sub> -7	7.8, 12.3	24.1	330	1.99	4.6
Al <sub>2</sub> O <sub>3</sub> -9	9.6, 17.7	27.7	309	2.14	4.9
Al <sub>2</sub> O <sub>3</sub> -12	13.2, 20.5	26.1	279	1.82	6.0
Al <sub>2</sub> O <sub>3</sub> -16	15.6, 25.9	27.7	261	1.80	6.7

<sup>a</sup> Average pore size = 4(Pore volume/S<sub>BET</sub>)

<sup>b</sup> Average diameter of Al<sub>2</sub>O<sub>3</sub> crystallite, obtained by XRD measurement.