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Flexible Transparent Colorimetric Wrist Strap Sensor

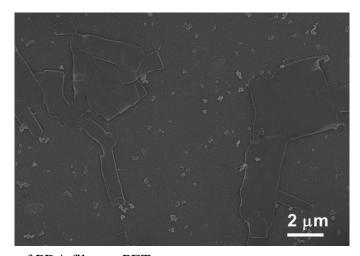


Fig. S1 SEM image of PDA film on PET.

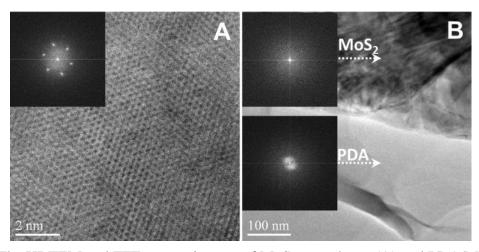


Fig. S2 The HRTEM and FFT pattern images of MoS₂ nanosheets (A) and PDA/MoS₂ (B).

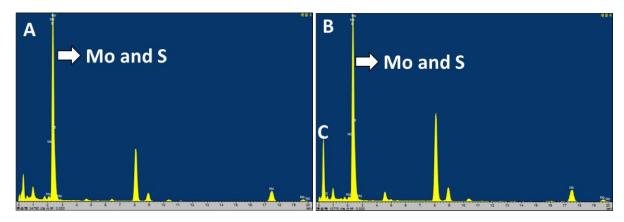


Fig. S3 EDS spectra of MoS_2 (A) and PDA/ MoS_2 (B).

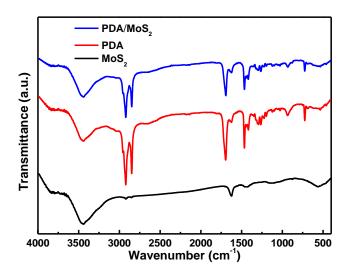


Fig. S4 FTIR of MoS₂, PDA, and PDA/MoS₂ composite.

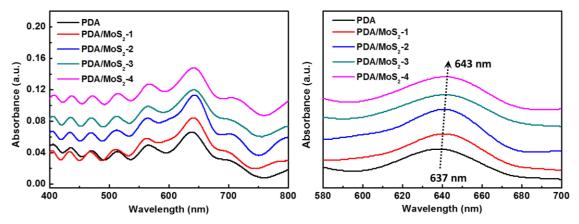


Fig. S5 UV-vis spectra of PDA, and PDA/MoS $_2$ films.

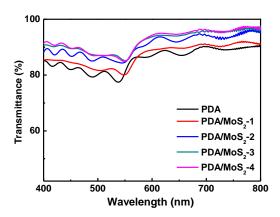


Fig. S6 Transmittance spectra of PDA, PDA/MoS₂-1, PDA/MoS₂-2, PDA/MoS₂-3, and PDA/MoS₂-4 films after exposed to DMF vapor.

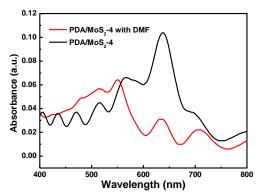


Fig. S7 UV-vis spectra of PDA/MoS₂-4 in the absence and presence of DMF vapor.

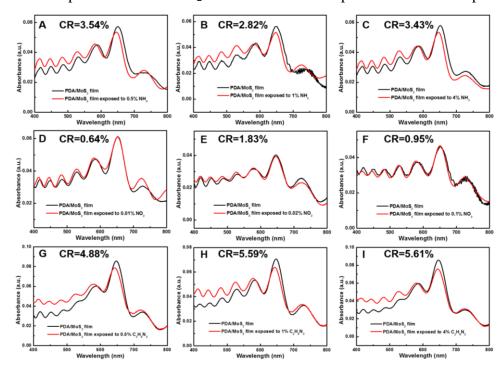


Fig. S8. UV-vis spectra of PDA/MoS₂ film exposed to NH₃ (A, B, C), NO₂ (D, E, F), and

 $C_2H_8N_2$ (Ethylenediamine) (G, H, I). The concentrations of NH_3 and $C_2H_8N_2$ were 0.5%, 1%, and 4%. The concentrations of NO_2 were 0.01%, 0.02%, and 0.1%.

We have conducted the detection of N-containing gas species including NH₃, NO₂, C₂H₈N₂ (Ethylenediamine) using PDA/MoS₂ film sensor. The detecting procedures were the same with that for DMF vapor. It was found that no obvious color changes were observed. The UV-vis spectra of PDA/MoS₂ film before and after exposed to gases were collected. Figure S8 indicated that PDA/MoS₂ film showed *CR* value less than 6% for NH₃, NO₂, and C₂H₈N₂. As proven by previous research, the colorimetric transition of PDA was generally ascribed to the deformation of PDA side chains by the interaction with suitable organic solvents (Davis, B. W.; Burris, A. J.; Niamnont, N.; Hare, C. D.; Chen, C.-Y.; Sukwattanasinitt, M.; Cheng, Q., Langmuir 2014, 30, 9616-9622). For NH₃, NO₂ gases, and C₂H₈N₂, despite of their affinity to MoS₂, their weak interaction with PDA failed to induce the color change of PDA.