

Electronic Supplementary Information (ESI)

For *Org. Biomol. Chem.*

K₂CO₃-Promoted formation of aryl esters from primary aryl amides by acyl-acyl exchange process

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Supporting information

Part I Experimental Section

^1H NMR and ^{13}C NMR spectra were recorded on Bruker ARX-300 or 500 MHz spectrometer with TMS as internal standard. Toluene and THF were distilled over sodium. 1,2-dichloroethane (DCE) and CH_3CN were distilled over calcium hydride. Other solvents were distilled under nitrogen. All bases were commercially available and used as received.

1. Table 1. Optimization of The Acyl-Acyl Exchange Reaction Conditions In the Presence of Copper Salts.^a

Entry	[Cu](10 mol %)	Ligand	Base	Yield (%) ^b
1	CuI	1,10-phenanthroline	$\text{K}_3\text{PO}_4 \cdot 3\text{H}_2\text{O}$	16
2	CuBr	1,10-phenanthroline	$\text{K}_3\text{PO}_4 \cdot 3\text{H}_2\text{O}$	23
3	CuCl	1,10-phenanthroline	$\text{K}_3\text{PO}_4 \cdot 3\text{H}_2\text{O}$	trace
4	$\text{Cu}(\text{OTf})_2$	1,10-phenanthroline	$\text{K}_3\text{PO}_4 \cdot 3\text{H}_2\text{O}$	trace
5	$\text{Cu}(\text{acac})_2$	1,10-phenanthroline	$\text{K}_3\text{PO}_4 \cdot 3\text{H}_2\text{O}$	7
6	CuCl_2	1,10-phenanthroline	$\text{K}_3\text{PO}_4 \cdot 3\text{H}_2\text{O}$	10
7	$\text{Cu}(\text{OAc})_2$	1,10-phenanthroline	$\text{K}_3\text{PO}_4 \cdot 3\text{H}_2\text{O}$	<5
8	CuSO_4	1,10-phenanthroline	$\text{K}_3\text{PO}_4 \cdot 3\text{H}_2\text{O}$	11
9	CuI	2,2'-dipyridyl	$\text{K}_3\text{PO}_4 \cdot 3\text{H}_2\text{O}$	15
10	CuI	1,10-phenanthroline	K_2CO_3	58
11	-	-	-	0
12	CuI	1,10-phenanthroline	-	0
13	-	-	K_2CO_3	60

^a Conditions: **1a** (0.25 mmol), **2a** (0.5 mmol), [Cu] (0.025 mmol), ligand (0.05 mmol), base (2 equiv), 120°C for 12 h in toluene, unless otherwise noted. ^b Isolated yield..

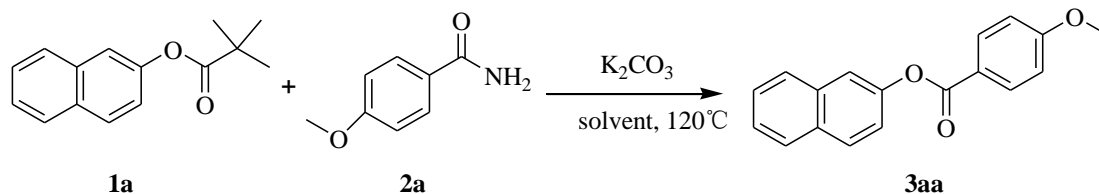
2. Table 2. Screening of Various Bases for This Acyl-Acyl Exchange Reaction without the Transition Metal^a

Entry	Base	Yield (%) ^b
1	$\text{K}_3\text{PO}_4 \cdot 3\text{H}_2\text{O}$	18
2	<i>t</i> -BuOK	33
3	LiOAc	trace

4	NaOAc	0
5	Cs ₂ CO ₃	43 ^c
6	KF	0
7	CsF	trace
8	NaOTf	0
9	DBU	0
10	NaCO ₃	0
11	KOH	33
12	<i>t</i> -BuOLi	52
13	CsOH	50
14	Ag ₂ CO ₃	0
15	K ₂ CO ₃	60
16	K ₂ CO ₃	72 ^d
17	K ₂ CO ₃	79 ^e
18	K ₂ CO ₃	82 ^f

^a Conditions: **1a** (0.25 mmol), **2a** (0.5 mmol), base (2 equiv), 120°C for 12 h in toluene, unless otherwise noted. ^b Isolated yield. ^c 1 equiv of Cs₂CO₃ was used. ^d 1 equiv of K₂CO₃ was used. ^e 0.5 equiv of K₂CO₃ was used. ^f 20 mol% of K₂CO₃ was used

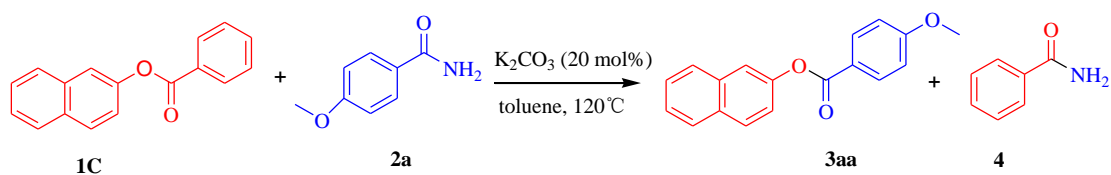
3. Table 3. Screening of Various Solvents for This Acyl-Acyl Exchange Reaction without the Transition Metal^a



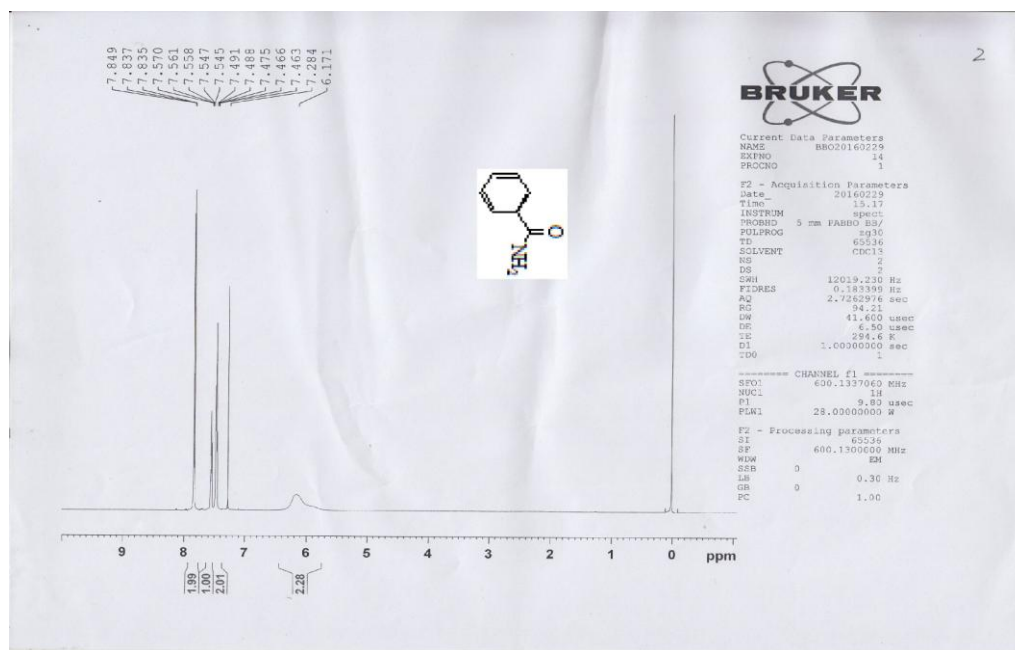
Entry	Solvent	Yield (%) ^b
1	toluene	82
2	1,4-dioxane	trace
3	DCE	0
4	CH ₃ CN	0
5	THF	0
6	DMF	<5
7	anisole	0
8	DMAc	0
9	DMSO	0

^a Conditions: **1a** (0.25 mmol), **2a** (0.5 mmol), K₂CO₃ (20 mol%), 120°C for 12 h in solvent, unless otherwise noted. ^b Isolated yield.

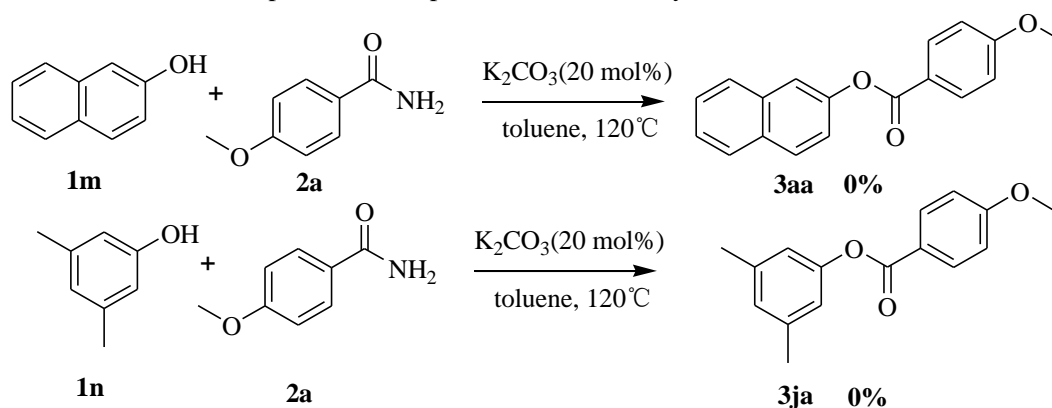
4. Isolation of the primary amide products for the evidence of the proposed acyl-acyl exchange reaction.



A mixture of 2-naphthyl benzoate **1c** (0.25 mmol), 4-methoxybenzamide **2a** (0.5 mmol) and K_2CO_3 (20 mol%) in toluene was stirred at 120 °C for 12 h. After the reaction was achieved, the primary benzamide product **4** was isolated by preparative TLC (silica gel, EtOAc as eluent) in 33% yield (10 mg) as a white solid: $R_f = 0.56$ (in EtOAc); $^1\text{H NMR}$ (CDCl_3 , 600 MHz): δ (ppm): 7.84 (d, $J = 7.2$ Hz, 2H), 7.56 (t, $J = 7.2$ Hz, 1H), 7.48 (t, $J = 7.2$ Hz, 2H), 6.17 (s, 2H).

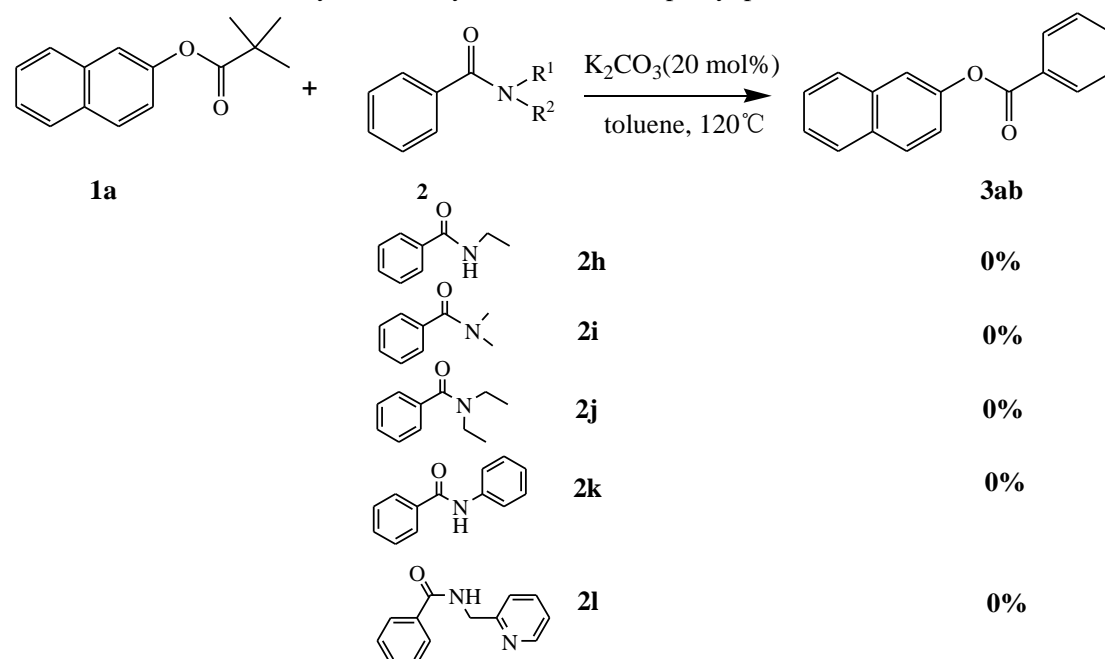


5. The reaction of unprotected 2-naphthol with 4-methoxybenzamide



Scheme 1. The unprotected 2-naphthol and 3, 5-dimethylphenol was tried for this reaction. A mixture of **1m** or **1n** (0.25 mmol), 4-methoxybenzamide **2a** (0.50 mmol) and K_2CO_3 (20 mol%) in toluene was stirred at 120 °C for 12 h;

6. The reaction of secondary and tertiary amides with 2-naphthyl pivalate

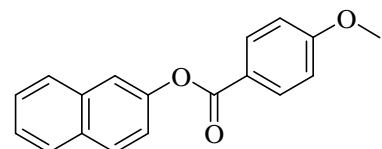


Scheme 2. Several secondary and tertiary amides were tried for this reaction under the standard condition.

General Procedure for This Acyl-Acyl Exchange Reaction of esters with primary

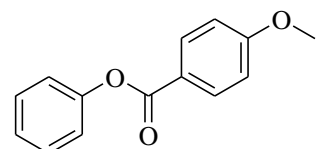
amides: A mixture of esters **1** (0.25 mmol), primary amides **2** (0.5 mmol) and K_2CO_3 (20 mol%) in toluene was stirred in a sealed tube under air without a reflux condenser attached at $120^\circ C$ for 12 h. After the reaction was achieved, the crude mixture was purified by column chromatography (silica gel, EtOAc / petroleum ether) to afford the desired products **3**. All products have been reported previously and identical with that described in literature.

naphthalen-6-yl 4-methoxybenzoate **3aa** ^[1]



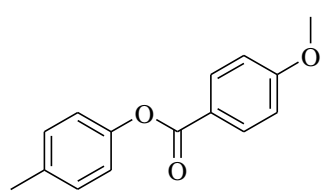
The general procedure was followed with 2-naphthyl pivalate (0.25 mmol), 4-methoxybenzamide (0.5 mmol) and K_2CO_3 (20 mol%); Yield: 57 mg, 82%; 1H NMR ($CDCl_3$, 300 MHz): δ (ppm): 8.22 (d, $J = 8.7$ Hz, 2H), 7.92-7.83 (m, 3 H), 7.69 (d, $J = 1.5$ Hz, 1 H), 7.54-7.47 (m, 2 H), 7.38-7.35 (m, 1 H) 7.02 (d, $J = 8.7$ Hz, 2 H), 3.92 (s, 3H); MS (EI) m/z (%) 278.1 (M+, 17.5), 135.0(100), 107.0 (26.8), 77.0 (31.0).

phenyl 4-methoxybenzoate **3ba** ^[2]



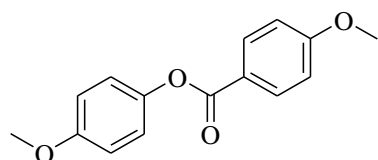
Yield: 41.6 mg, 73%; 1H NMR ($CDCl_3$, 300 MHz): δ (ppm): 8.19-8.15 (m, 2 H), 7.46-7.41 (m, 2 H), 7.30-7.20 (m, 3 H), 7.02-6.98 (m, 2 H), 3.91 (s, 3H); ^{13}C NMR ($CDCl_3$, 75 MHz): δ (ppm): 164.8, 163.9, 150.9, 132.2, 129.5, 125.6, 121.8, 121.7, 113.8, 55.4.

p-tolyl 4-methoxybenzoate **3ca** ^[3]



Yield: 44.7 mg, 74%; ¹H NMR (CDCl₃, 300 MHz): δ (ppm): 8.19-8.14 (m, 2 H), 7.21 (d, *J* = 8.4 Hz, 2 H), 7.10 (d, *J* = 8.4 Hz, 2 H), 7.03-7.01 (m, 2 H), 3.91 (s, 3H), 2.40 (s, 3H); ¹³C NMR (CDCl₃, 75 MHz): δ (ppm): 165.0, 163.8, 148.8, 135.0, 132.2, 130.0, 122.2, 121.5, 113.9, 55.4, 20.9.

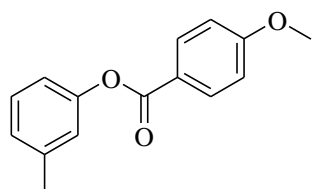
4-methoxyphenyl 4-methoxybenzoate **3da** ^[4]



Yield: 48.3 mg, 75%; ¹H NMR (CDCl₃, 300 MHz): δ (ppm): 8.17-8.14 (m, 2 H), 7.14-7.10 (m, 2 H), 7.00-6.91 (m, 4 H), 3.90 (s, 3H), 3.83 (s, 3H); ¹³C NMR (CDCl₃, 125 MHz): δ (ppm): 165.3, 163.8, 157.2, 144.5, 132.2, 122.5, 122.0, 114.5,

113.8, 55.6, 55.5.

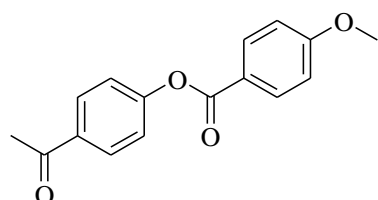
m-tolyl 4-methoxybenzoate **3ea** ^[3]



Yield: 36.7 mg, 61%; ¹H NMR (CDCl₃, 300 MHz): δ (ppm): 8.19-8.14 (m, 2 H), 7.34-7.27 (m, 1 H), 7.10-6.97 (m, 5 H), 3.90 (s, 3H), 2.40 (s, 3H); ¹³C NMR (CDCl₃, 125 MHz): δ (ppm): 165.1, 163.9, 151.0, 139.6, 132.3, 129.2, 126.6, 122.4, 122.0, 118.8, 113.8,

55.5, 21.4.

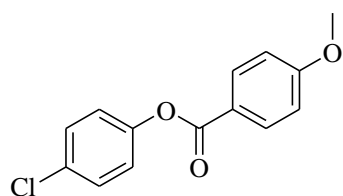
4-acetylphenyl 4-methoxybenzoate **3fa** ^[5]



Yield: 26 mg, 39%; ¹H NMR (CDCl₃, 300 MHz): δ (ppm): 8.16 (d, *J* = 8.4 Hz, 2 H), 8.03 (d, *J* = 8.1 Hz, 2 H), 7.32 (d, *J* = 8.4 Hz, 2 H), 7.00 (d, *J* = 8.1 Hz, 2 H), 3.91 (s, 3H), 2.63 (s, 3H); MS (EI) *m/z* (%) 270.0 (M⁺, 0.15), 135.9(6), 134.9 (100),

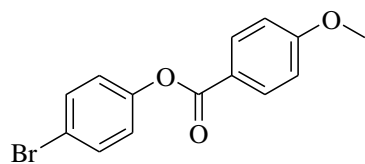
120.9(8), 106.9(6).

4-chlorophenyl 4-methoxybenzoate **3ga** ^[2]



Yield: 49.4 mg, 76%; ¹H NMR (CDCl₃, 300 MHz): δ (ppm): 8.17-8.12 (m, 2 H), 7.40 (d, *J* = 8.8 Hz, 2 H), 7.15 (d, *J* = 8.8 Hz, 2 H), 7.03-6.96 (m, 2 H), 3.90 (s, 3H); ¹³C NMR (CDCl₃, 75 MHz): δ (ppm): 163.9, 149.5, 132.2, 131.0, 129.4, 123.0, 121.3, 113.8, 55.4.

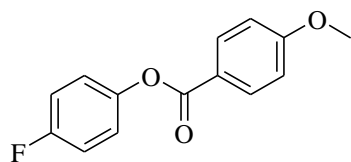
4-bromophenyl 4-methoxybenzoate **3ha** ^[6]



Yield: 55.8 mg, 73%; ¹H NMR (CDCl₃, 300 MHz): δ (ppm): 8.17-8.12 (m, 2 H), 7.56-7.51 (m, 2 H), 7.13-7.08 (m, 2 H), 7.02-6.97 (m, 2 H), 3.90 (s, 3H); ¹³C NMR (CDCl₃, 125 MHz):

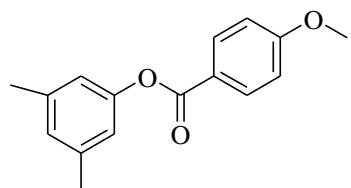
δ (ppm): 164.6, 164.1, 150.1, 132.5, 132.3, 123.6, 121.4, 118.8, 113.9, 55.5.

4-fluorophenyl 4-methoxybenzoate **3ia** ^[7]



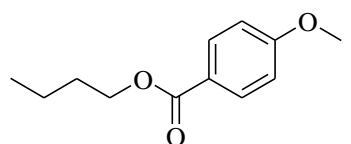
Yield: 44 mg, 72%; ¹H NMR (CDCl₃, 300 MHz): δ (ppm): 8.17-8.13 (m, 2 H), 7.20-7.08 (m, 4 H), 7.01-6.97 (m, 2 H), 3.90 (s, 3H); ¹³C NMR (CDCl₃, 75 MHz): δ (ppm): 164.9, 164.0, 160.1 (d, *J* = 243.0 Hz), 146.8, 132.2, 123.1, 121.4, 115.9 (d, *J* = 23.3 Hz), 113.8, 55.4.

3, 5-dimethylphenyl 4-methoxybenzoate **3ja** ^[8]



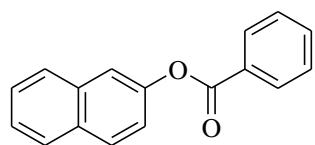
Yield: 34.4 mg, 54%; ¹H NMR (CDCl₃, 300 MHz): δ (ppm): 8.18-8.13 (m, 2 H), 7.01-6.96 (m, 2 H), 6.90 (s, 1H), 6.83 (s, 2H), 3.90 (s, 3H), 2.35 (s, 6H); ¹³C NMR (CDCl₃, 75 MHz): δ (ppm): 165.2, 163.8, 151.0, 139.3, 132.3, 127.5, 122.1, 119.4, 113.8, 55.5, 21.3.

butyl 4-methoxybenzoate **3la** ^[9]



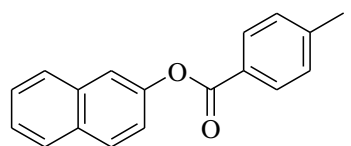
Yield: 38.2 mg, 74%; ¹H NMR (CDCl₃, 300 MHz): δ (ppm): 8.01-7.98 (m, 2 H), 6.92 (d, *J* = 8.7Hz, 2 H), 4.29 (t, *J* = 6.6Hz, 2H), 3.86 (s, 3H), 1.78-1.69 (m, 2 H), 1.54-1.41 (m, 2 H), 0.98 (t, *J* = 7.5Hz, 3H); ¹³C NMR (CDCl₃, 75 MHz): δ (ppm): 166.4, 163.3, 131.5, 123.0, 113.5, 64.5, 55.3, 31.0, 19.3, 13.8.

naphthalen-6-yl benzoate **3ab** ^[10]



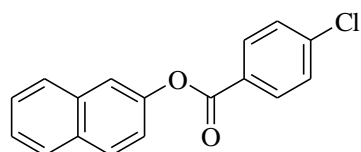
Yield: 50 mg, 81%; ¹H NMR (CDCl₃, 500 MHz): δ (ppm): 8.69 (s, 1H), 8.08 (d, *J* = 8.6 Hz, 1H), 7.89 (d, *J* = 8.1 Hz, 1H), 7.86-7.79 (m, 2H), 7.55-7.46 (m, 2H), 7.39-7.36 (m, 2H), 7.21-7.11 (m, 3H); ¹³C NMR (CDCl₃, 125 MHz): δ (ppm): 165.3, 151.0, 135.8, 132.5, 131.9, 129.6, 129.5, 128.7, 128.4, 127.8, 126.8, 126.0, 125.4, 121.7.

naphthalen-6-yl 4-methylbenzoate **3ac** ^[2]



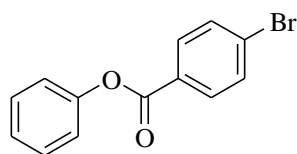
Yield: 54 mg, 82%; ¹H NMR (CDCl₃, 300 MHz): δ (ppm): 8.16 (dd, *J* = 8.1, 1.8 Hz, 2H), 7.93-7.83 (m, 3 H), 7.70 (d, *J* = 2.4 Hz, 1 H), 7.55-7.47 (m, 2 H), 7.38-7.33 (m, 3 H), 2.48 (s, 3H); MS (EI) *m/z* (%) 278.1 (M⁺, 19.9), 119.0 (100), 91.0 (42.8).

naphthalen-3-yl 4-chlorobenzoate **3ad** ^[2]



The general procedure was followed with 2-naphthyl pivalate (0.25 mmol), 4-chlorobenzamide (0.5 mmol) and K₂CO₃ (20 mol%); Yield: 56 mg, 80%; ¹H NMR (CDCl₃, 300 MHz): δ (ppm): 8.20 (d, *J* = 8.7Hz, 2 H), 7.94-7.84 (m, 3 H), 7.71 (d, *J* = 1.8Hz, 1 H), 7.56-7.48 (m, 4 H), 7.37 (dd, *J* = 8.4, 2.4Hz, 1H); ¹³C NMR (CDCl₃, 125 MHz): δ (ppm): 164.6, 148.4, 140.2, 133.8, 131.63, 131.60, 129.6, 129.0, 128.1, 128.0, 127.7, 126.7, 125.9, 121.1, 118.7.

phenyl 4-bromobenzoate **3be** ^[11]



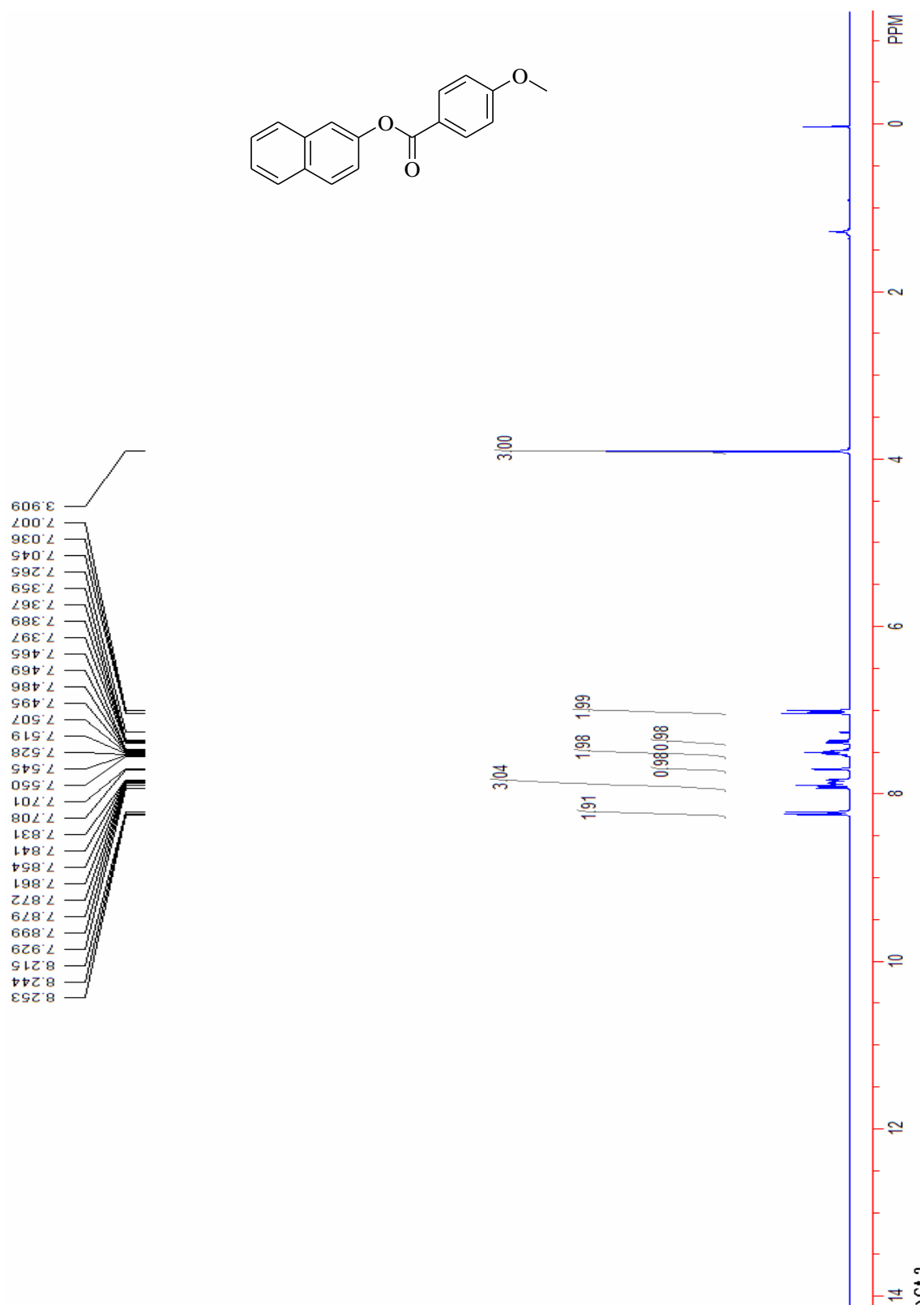
Yield: 49.4 mg, 72%; ¹H NMR (CDCl₃, 300 MHz): δ (ppm): 8.08 (d, *J* = 8.6 Hz, 2H), 7.67 (d, *J* = 8.6 Hz, 2H), 7.48 -7.42 (m, 2H), 7.32-7.28 (m, 1H), 7.23 -7.20 (m, 2H); ¹³C NMR (CDCl₃, 125 MHz): δ (ppm): 164.5, 150.7, 131.9, 131.6, 129.5, 128.8, 128.5, 126.0, 121.6.

Reference

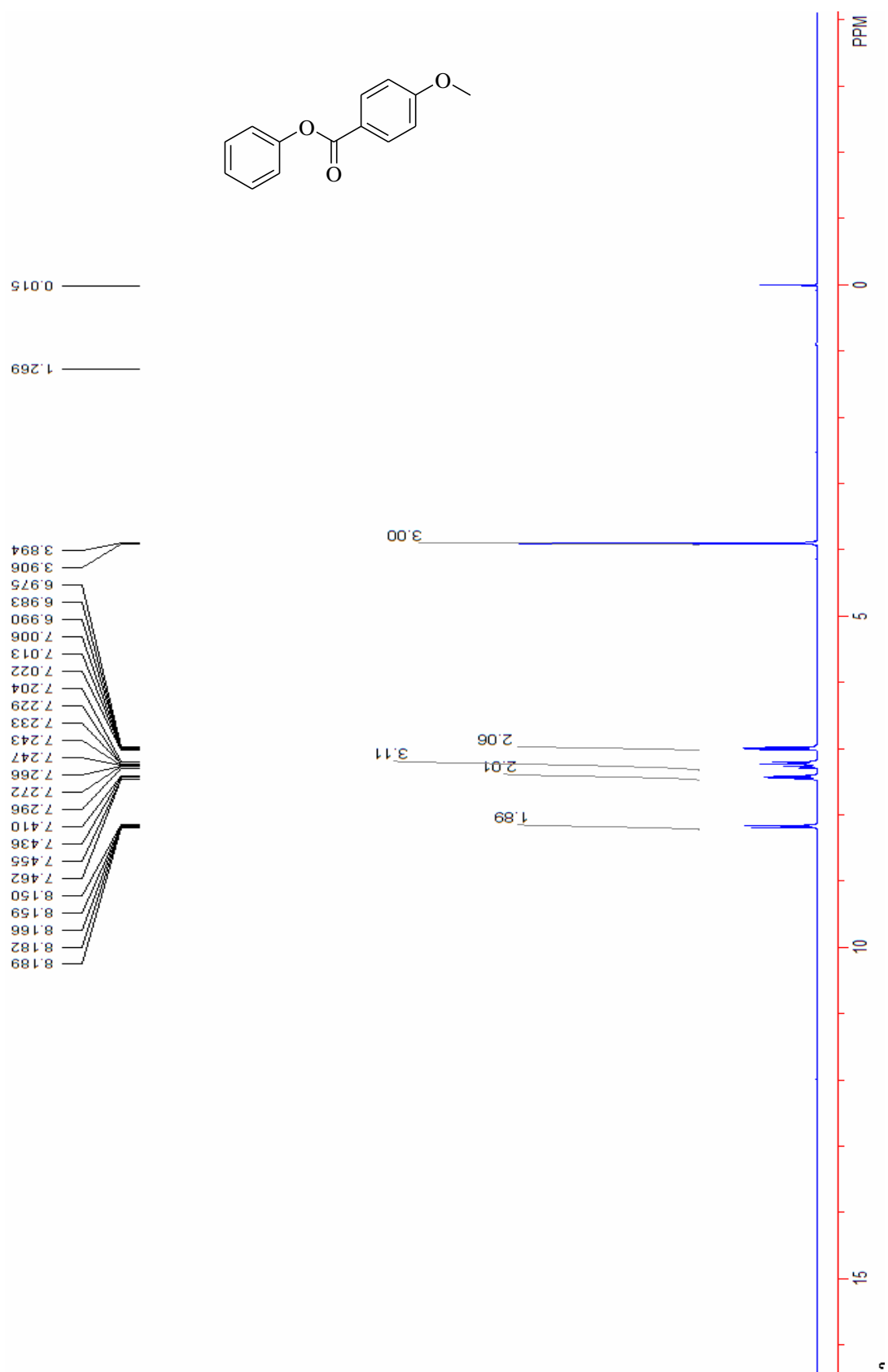
- [1] Rosa, J. N.; Reddy, S.; Candeias, N. R.; Cal, P. M. S. D.; Gois, P. M. P. *Org. Lett.*, 2010, **12**, 2686.
- [2] Arde, P.; Ramanjaneyulu, B. T.; Reddy, V.; Saxena, A.; Anand, R. V. *Org. Biomol. Chem.*, 2012, **10**, 848-851.
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- [9] Ma, G.-Z.; Leng, Y.-T.; Qiao, H.-J.; Yang, F.; Wang, S.-W.; Wu, Y.-J. *Appl. Organomet. Chem.*, 2014, **28**, 44.
- [10] Chen, J. X.; Peng, Y.; Liu, M. C.; Ding, J. C.; Su, W. K.; Wu, H. Y. *Adv. Synth. Catal.*, 2012, **354**, 2117.
- [11] Luo, F.; Pan, C. D.; Qian, P. C.; Cheng, J. *Synthesis.*, 2010, **12**, 2005.

Part II ^1H NMR and ^{13}C NMR

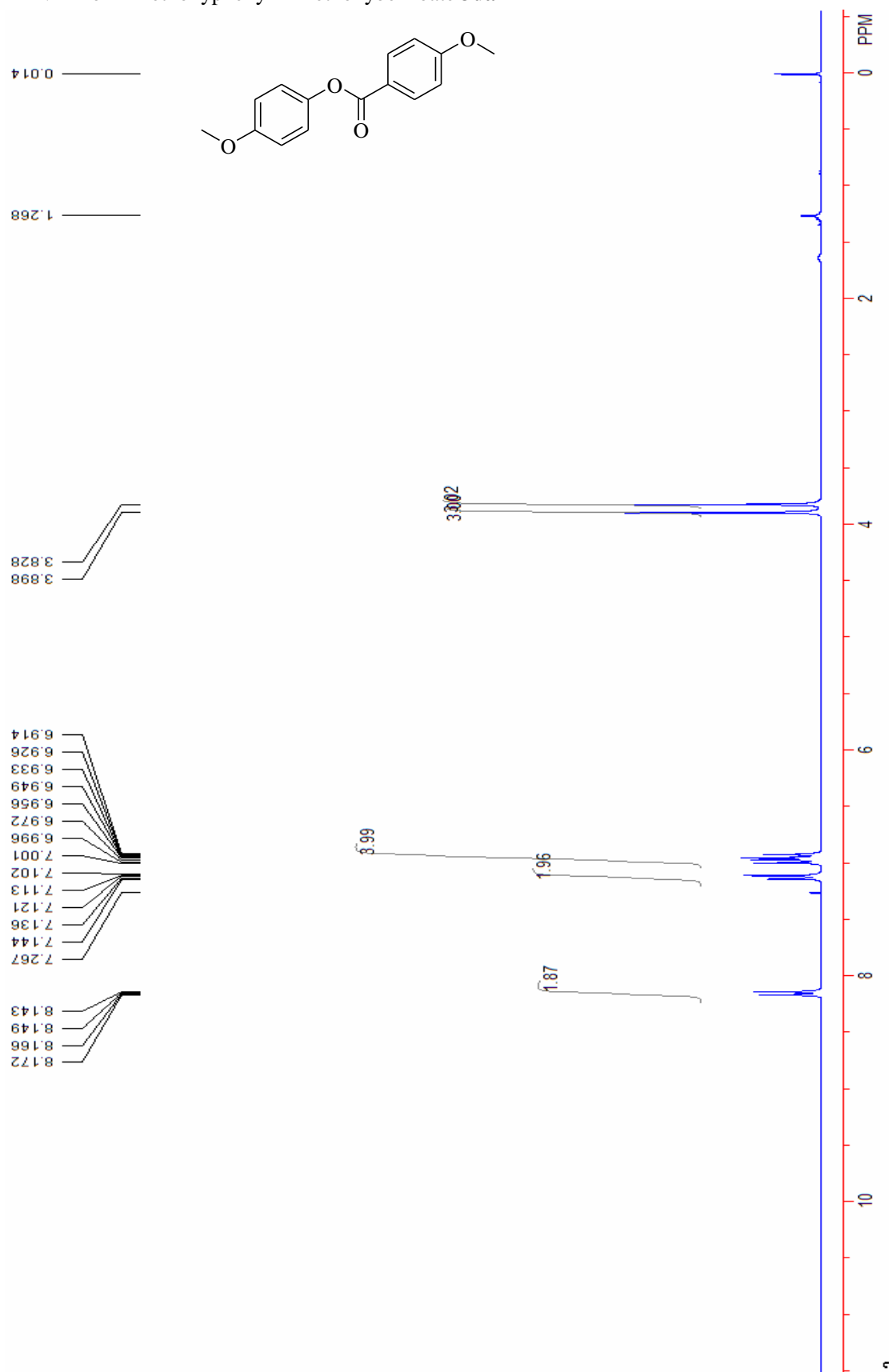
^1H NMR of naphthalen-6-yl 4-methoxybenzoate **3aa**



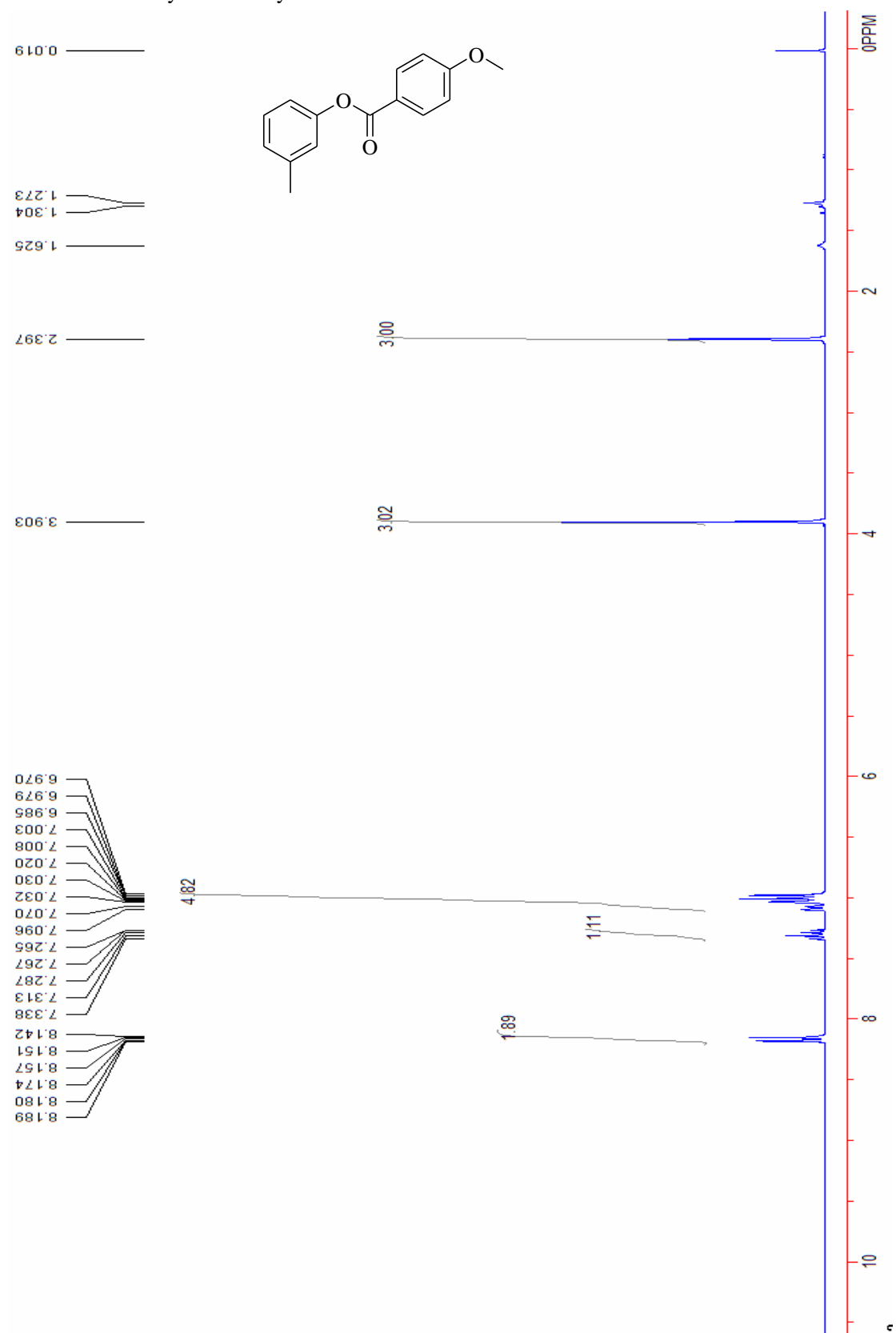
^1H NMR of phenyl 4-methoxybenzoate **3ba**



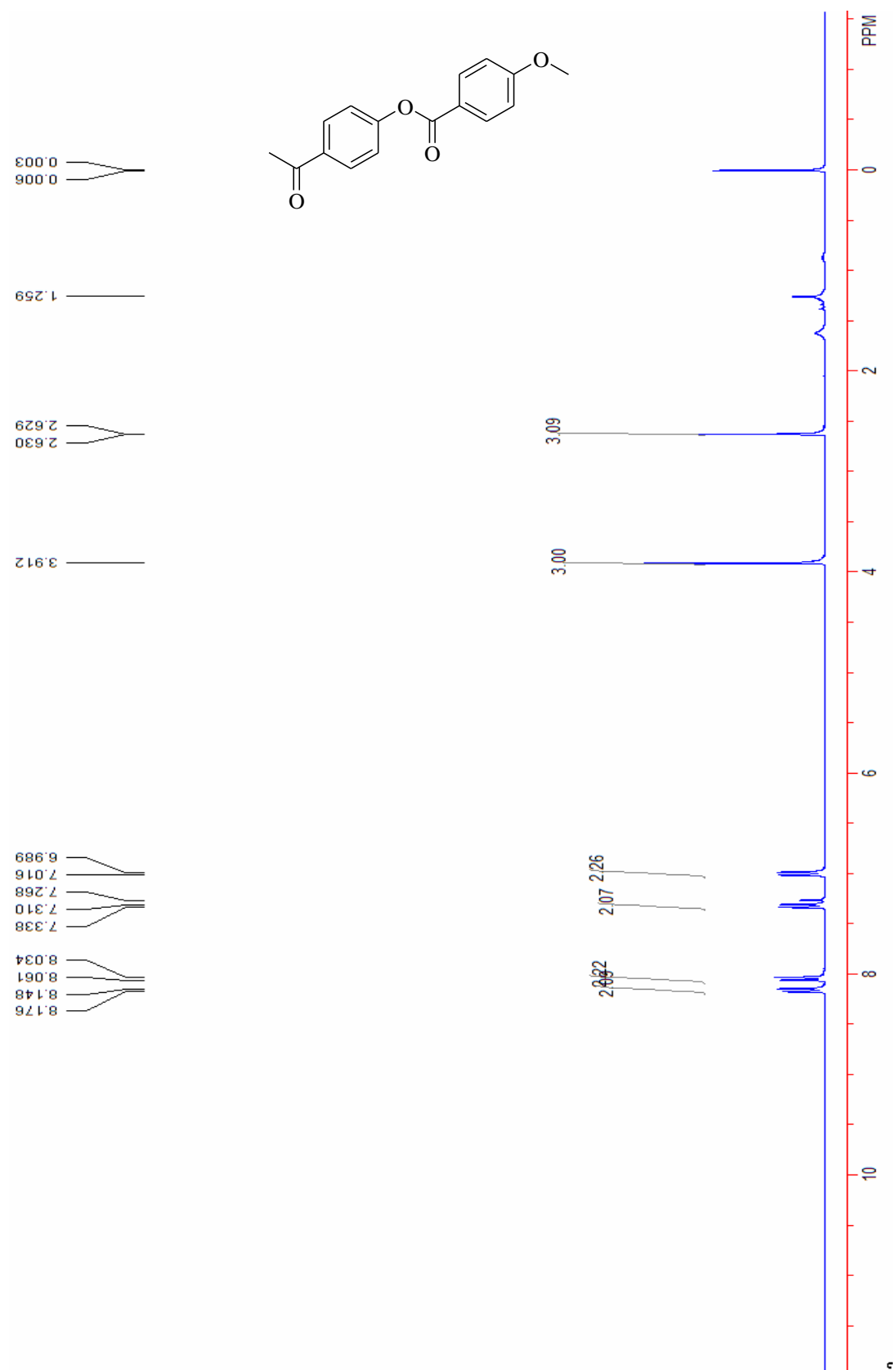
^1H NMR of 4-methoxyphenyl 4-methoxybenzoate **3da**



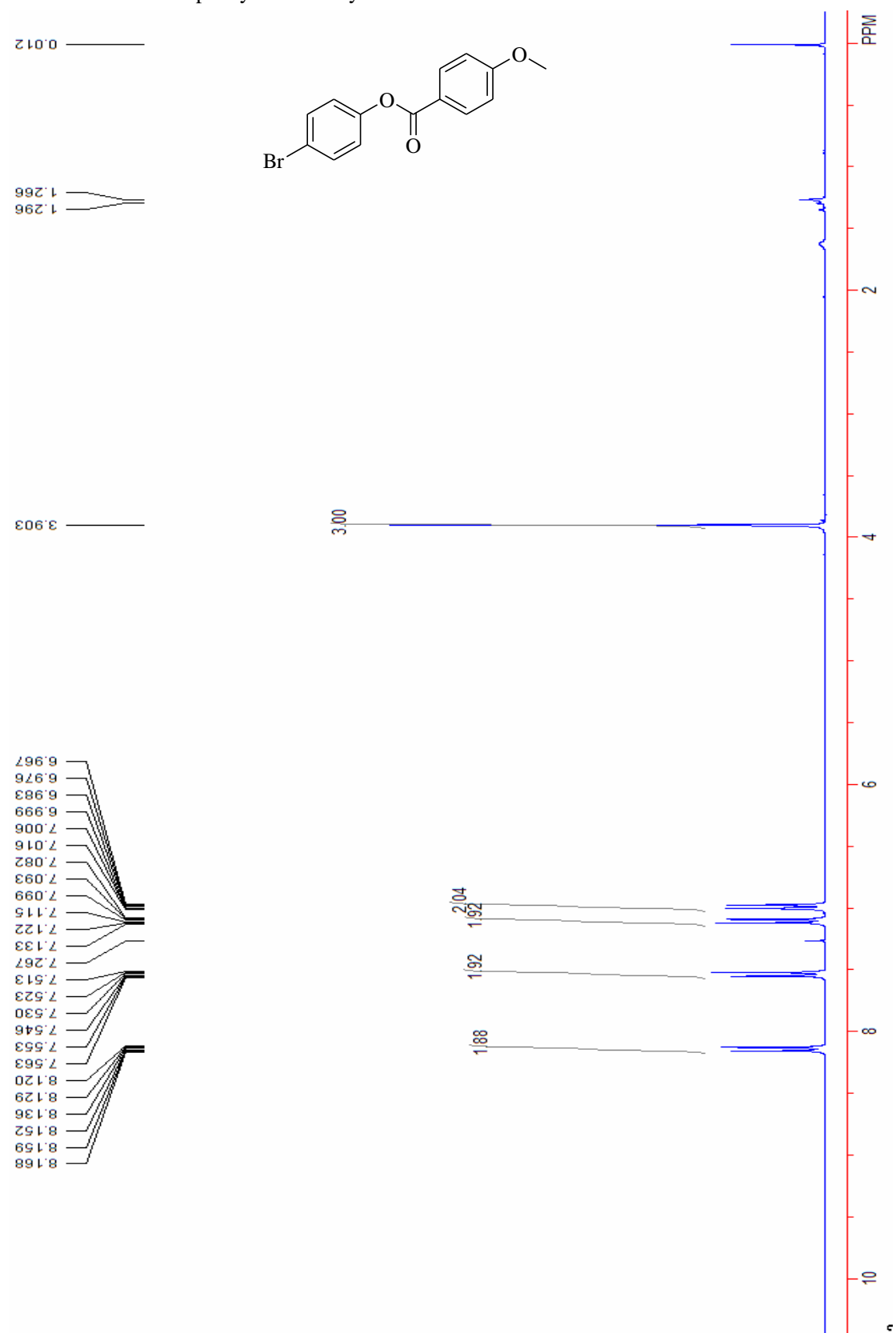
^1H NMR of m-tolyl 4-methoxybenzoate **3ea**



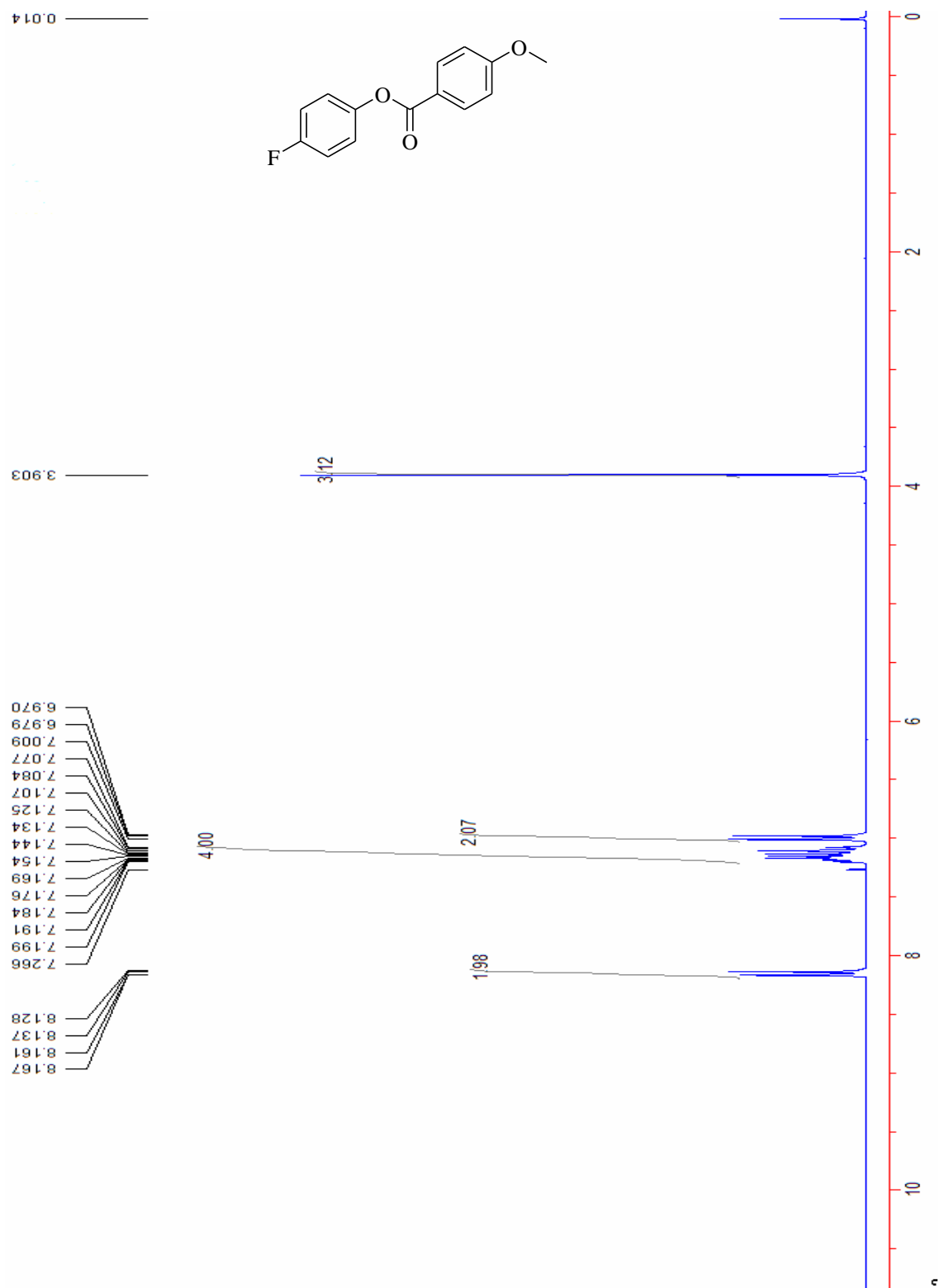
^1H NMR of 4-acetylphenyl 4-methoxybenzoate **3fa**



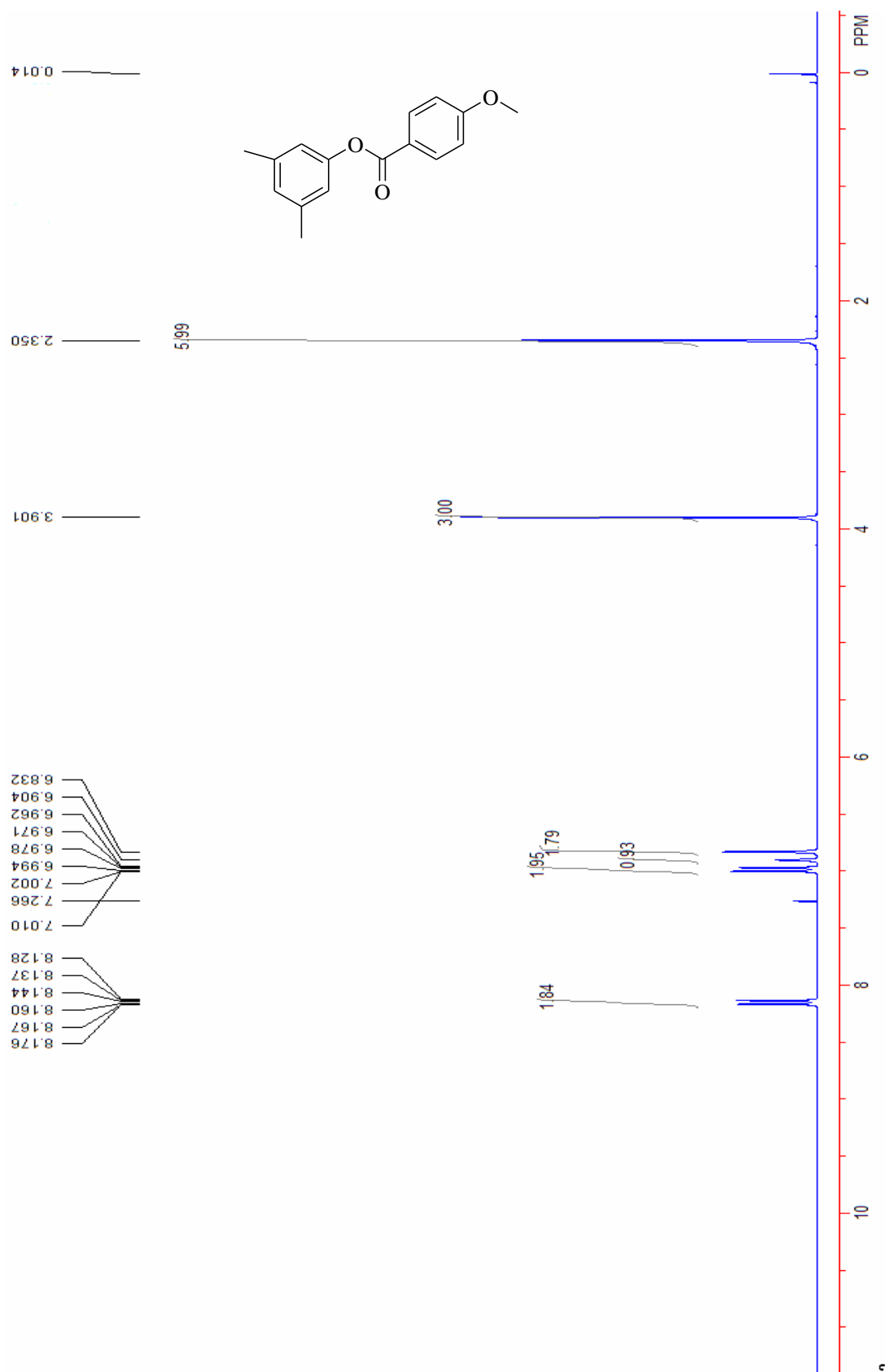
^1H NMR of 4-bromophenyl 4-methoxybenzoate **3ha**



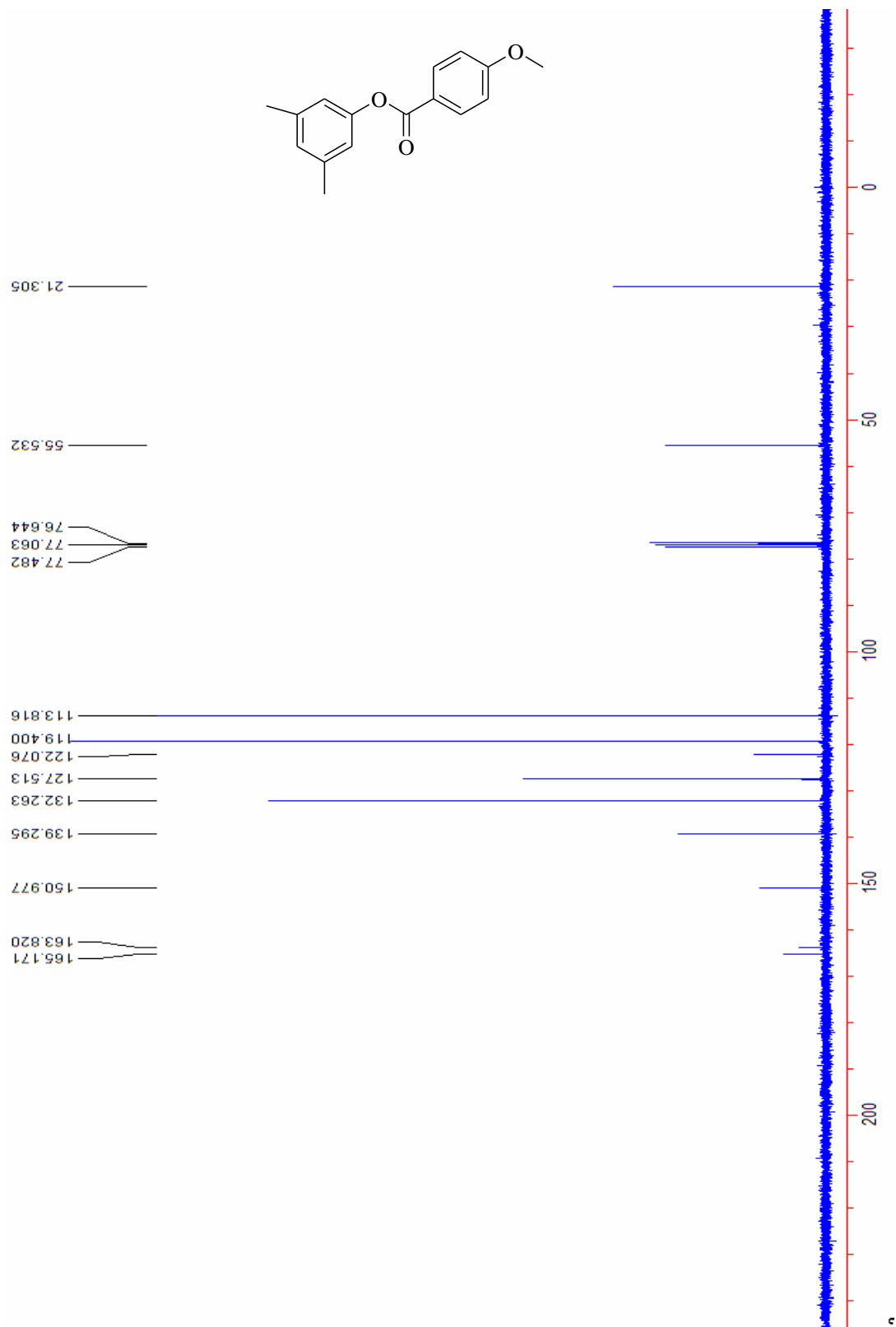
^1H NMR of 4-fluorophenyl 4-methoxybenzoate **3ia**



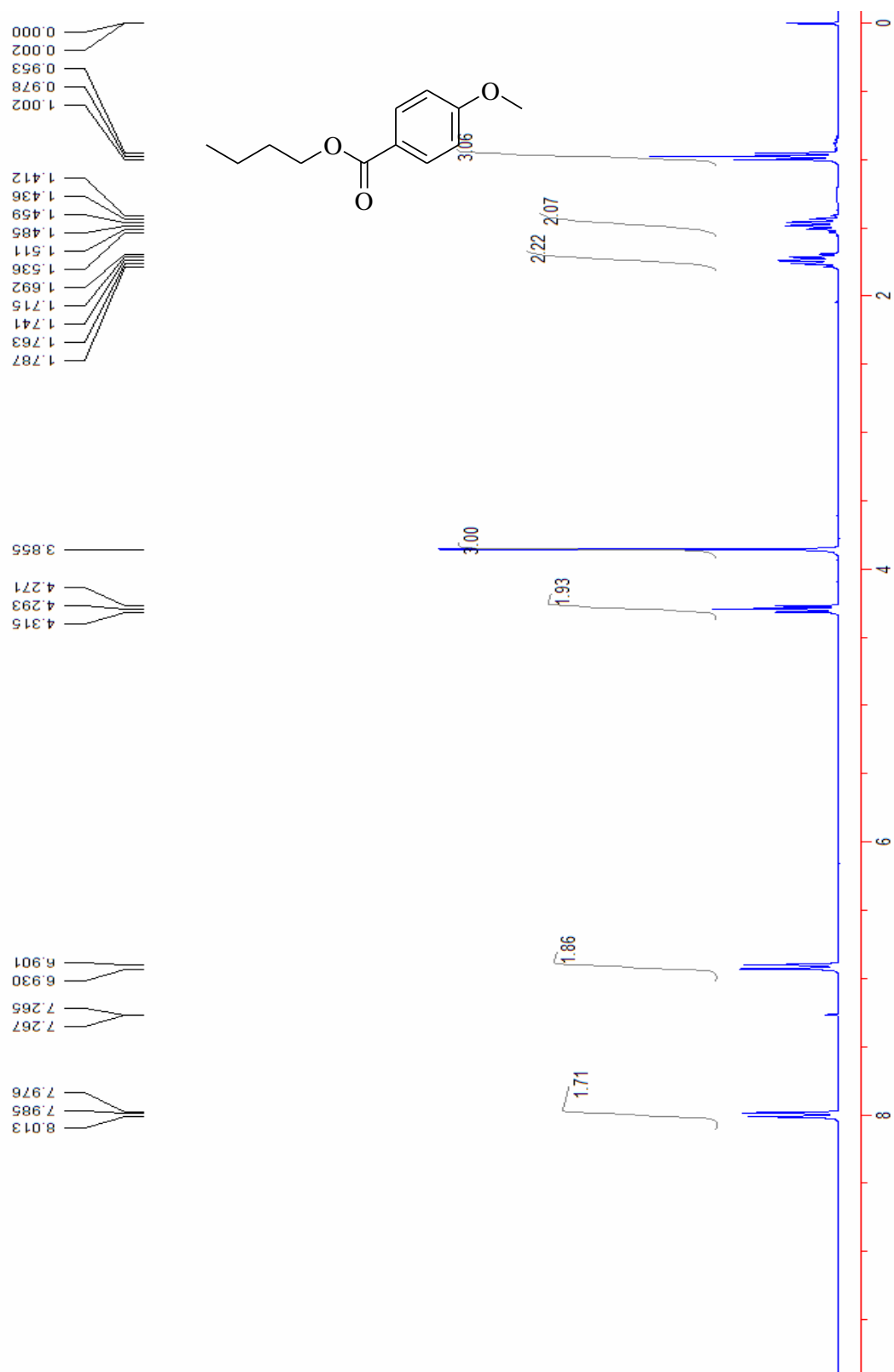
^1H NMR of 3,5-dimethylphenyl 4-methoxybenzoate **3ja**



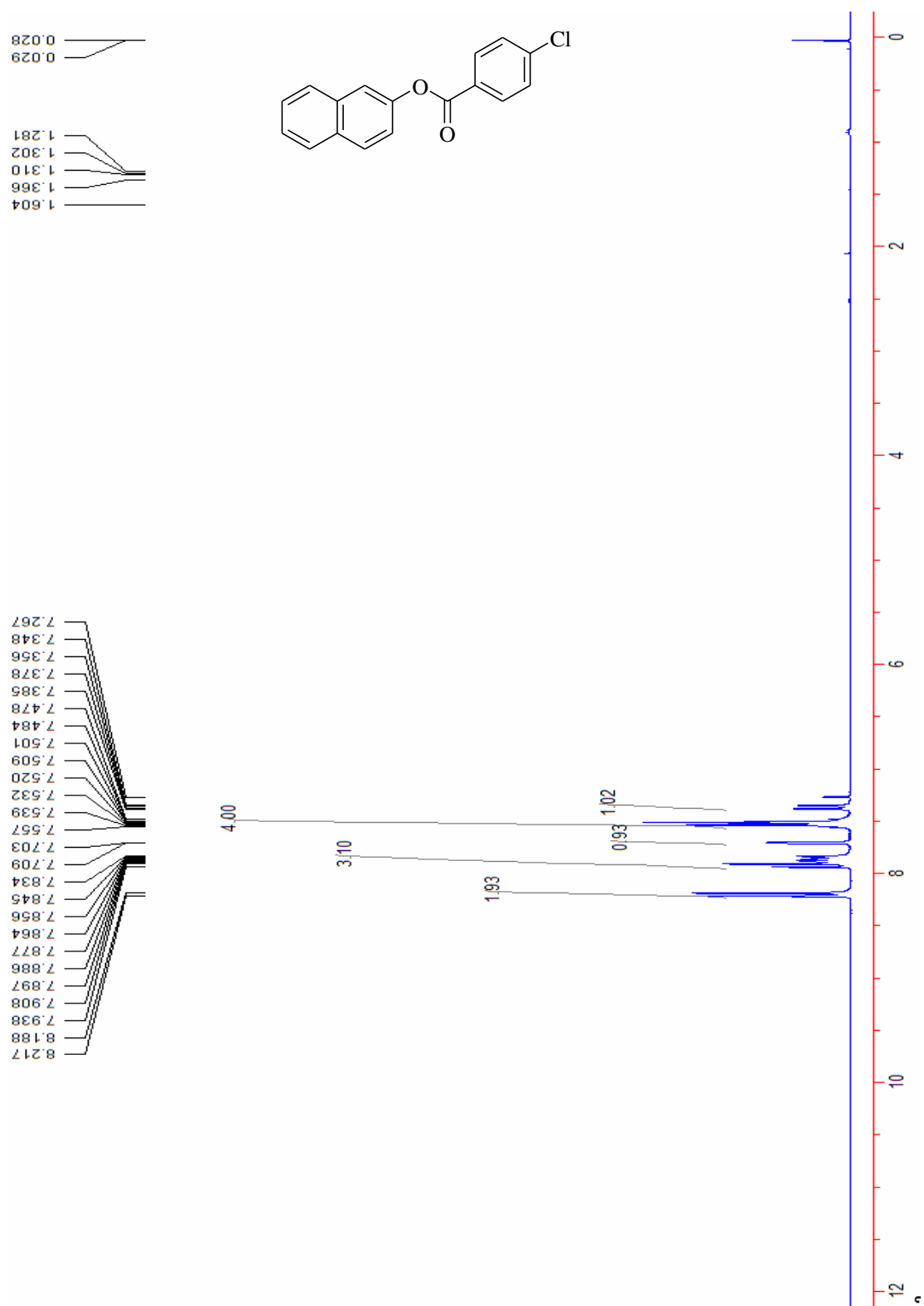
^{13}C NMR of 3,5-dimethylphenyl 4-methoxybenzoate **3ja**



^1H NMR of butyl 4-methoxybenzoate **3la**



^1H NMR of naphthalen-3-yl 4-chlorobenzoate **3ad**



^1H NMR of naphthalen-6-yl 4-methylbenzoate **3ac**

