

ELECTRONIC SUPPLEMENTARY INFORMATION

IMPROVED THROMBIN BINDING APTAMER ANALOGUES CONTAINING INVERSION OF POLARITY SITES: STRUCTURAL EFFECTS OF EXTRA-RESIDUES AT THE ENDS

A. Virgilio,^a T. Amato,^a L. Petraccone,^b R. Filosa,^c M. Varra,^a L. Mayol,^a V. Esposito^{a*} and A. Galeone^{a*}

^a Department of Pharmacy, University of Naples Federico II, via D. Montesano, 49, 80131 Naples, Italy. E-mail: galeone@unina.it, Tel: +39 081678542; verespos@unina.it, Tel: +39 081678746.

^b Department of Chemical Sciences, University of Naples Federico II, via Cintia, I-80126 Naples, Italy

^c Department of Experimental Medicine, Second University of Naples, via Costantinopoli 16, 80138 Naples, Italy

*Corresponding Authors

galeone@unina.it

verespos@unina.it

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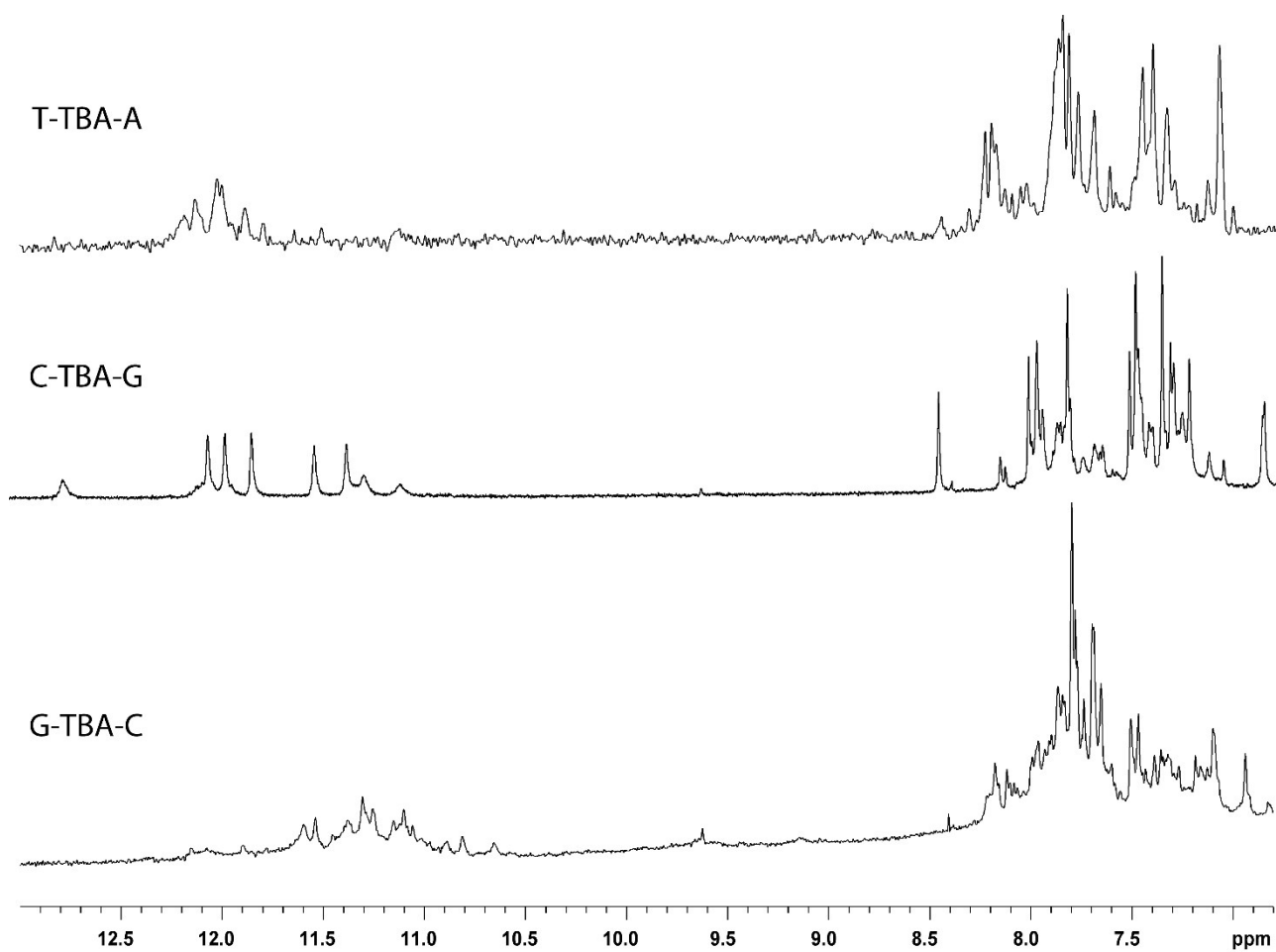
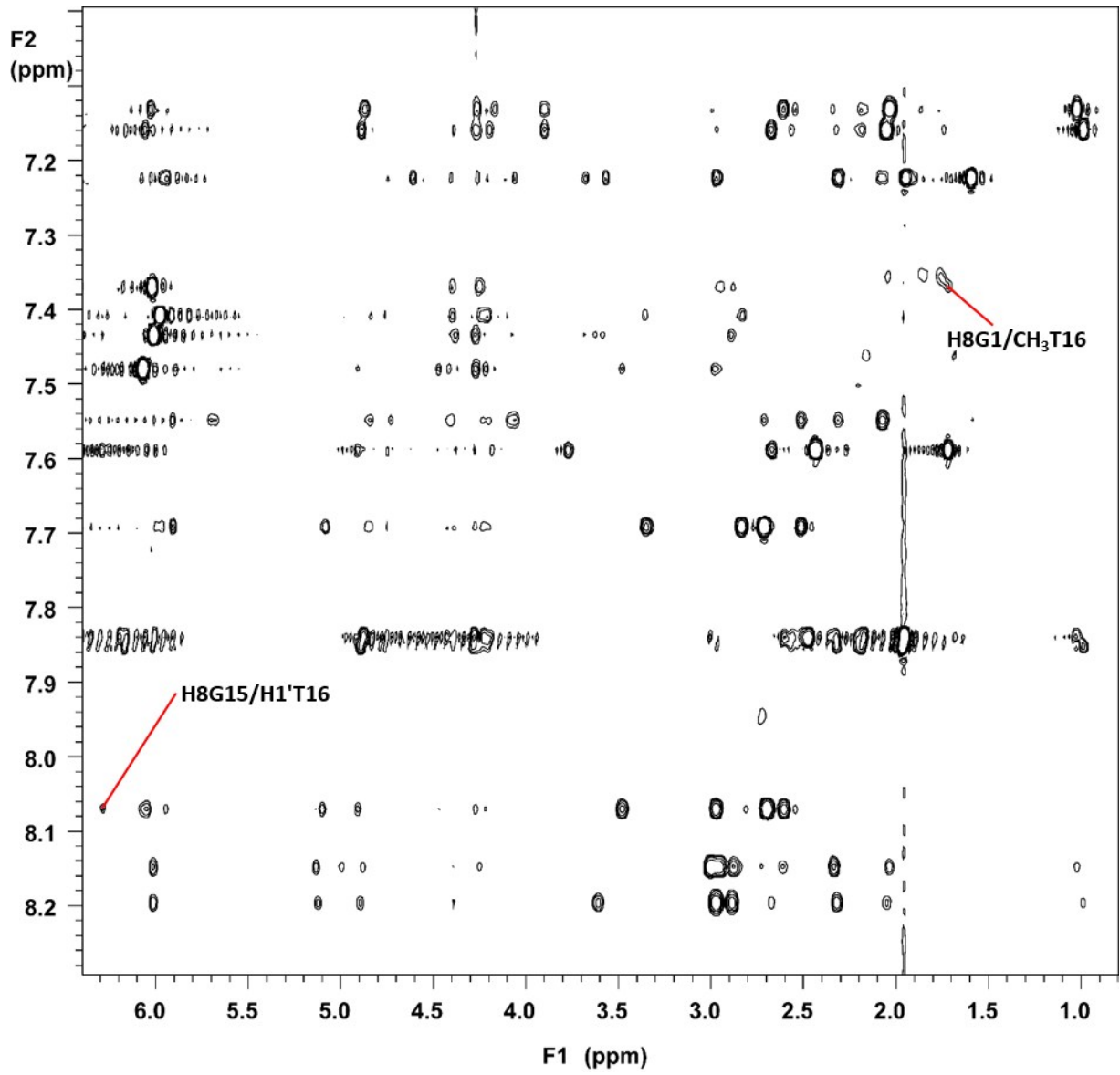
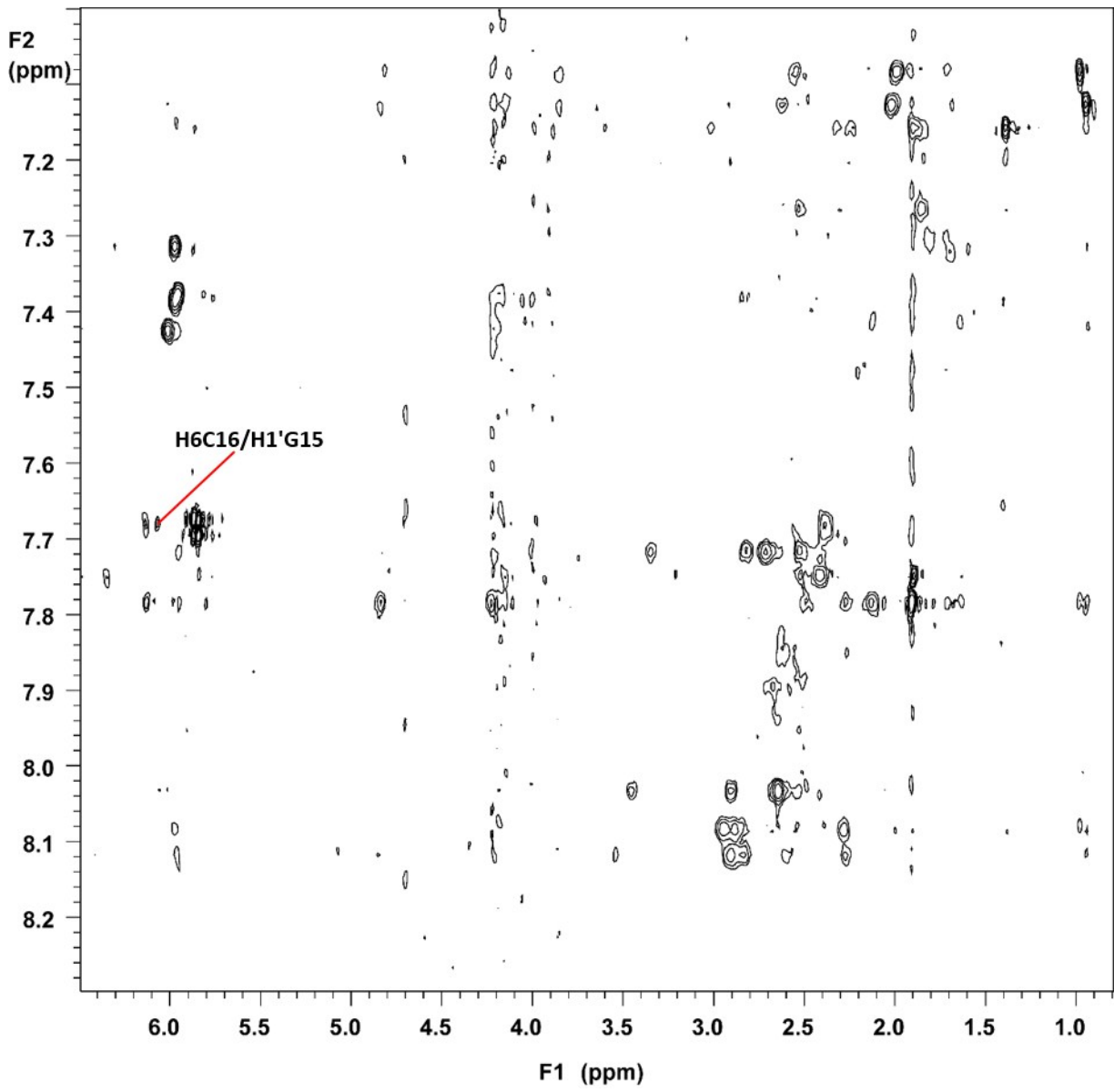


Fig. S1. Aromatic and imino proton regions of the ¹H-NMR spectra (700 MHz, T = 25°C) of the TBA analogues T-TBA-A, C-TBA-G and G-TBA-C in PBS buffer solution and 0.2 mM EDTA (pH 7.0).

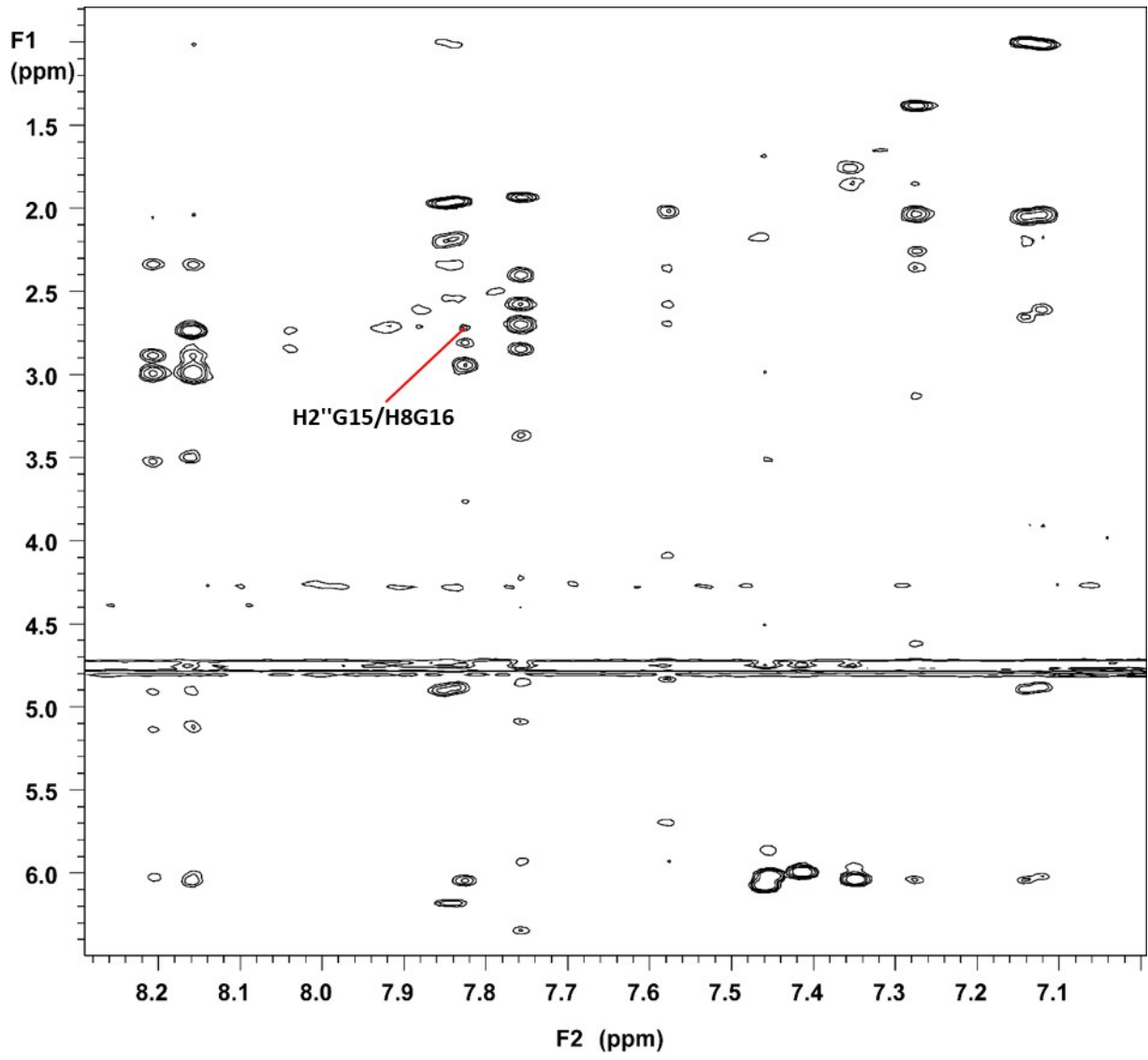
TBA-T



TBA-C



TBA-G



A-TBA-T

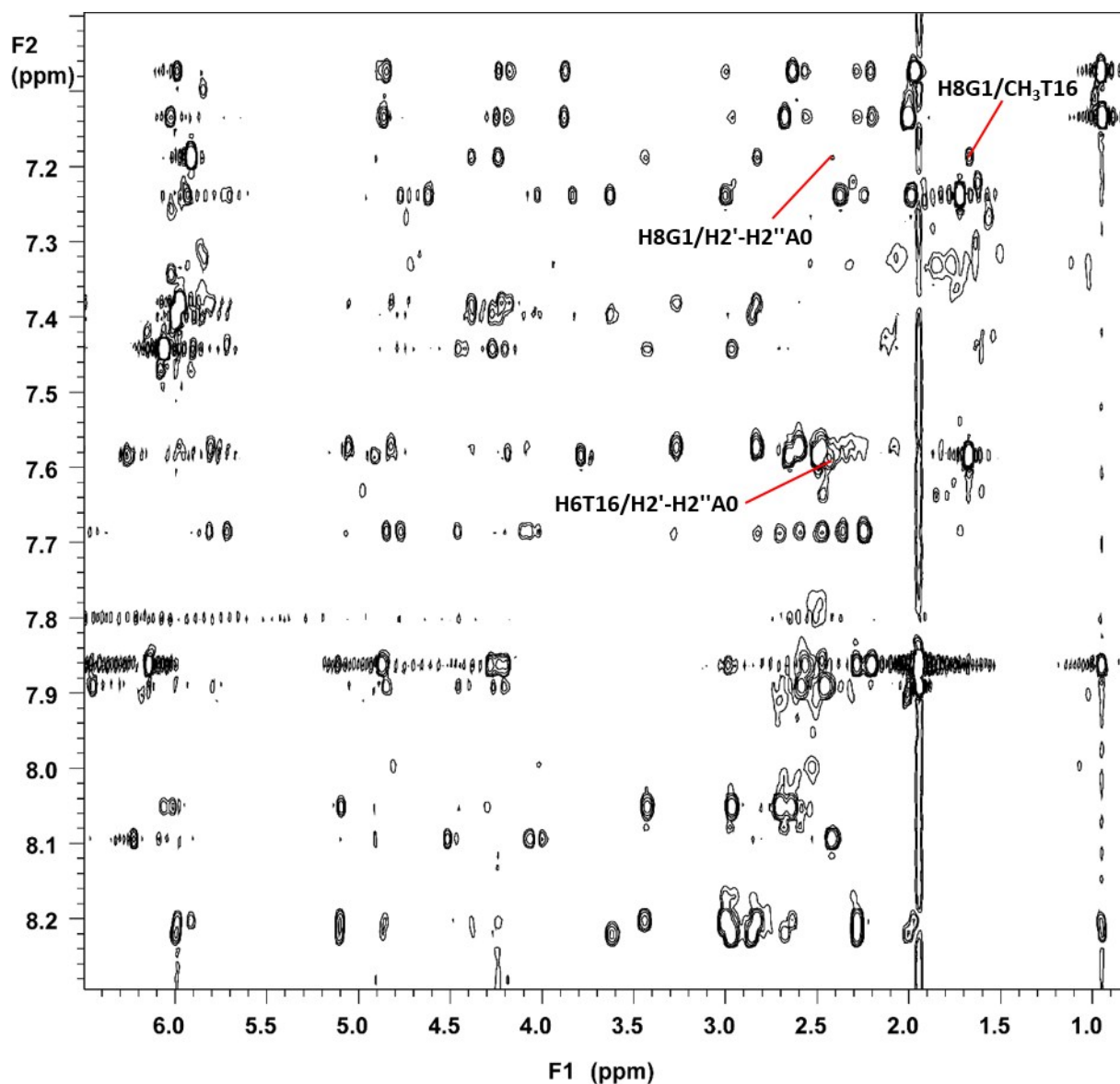


Fig. S2. Expanded region of 2D NOESY spectra of TBA-T, TBA-C, TBA-G and A-TBA-T (700 MHz; 25°C; strand concentration 2 mM in PBS buffer solution (H₂O/D₂O 9:1) and 0.2 mM EDTA (pH 7.0); total volume 0.6 ml; mixing time 180 ms) correlating bases H8/H6 and anomeric with sugar protons. The NOEs involving the extra-residue are labelled.

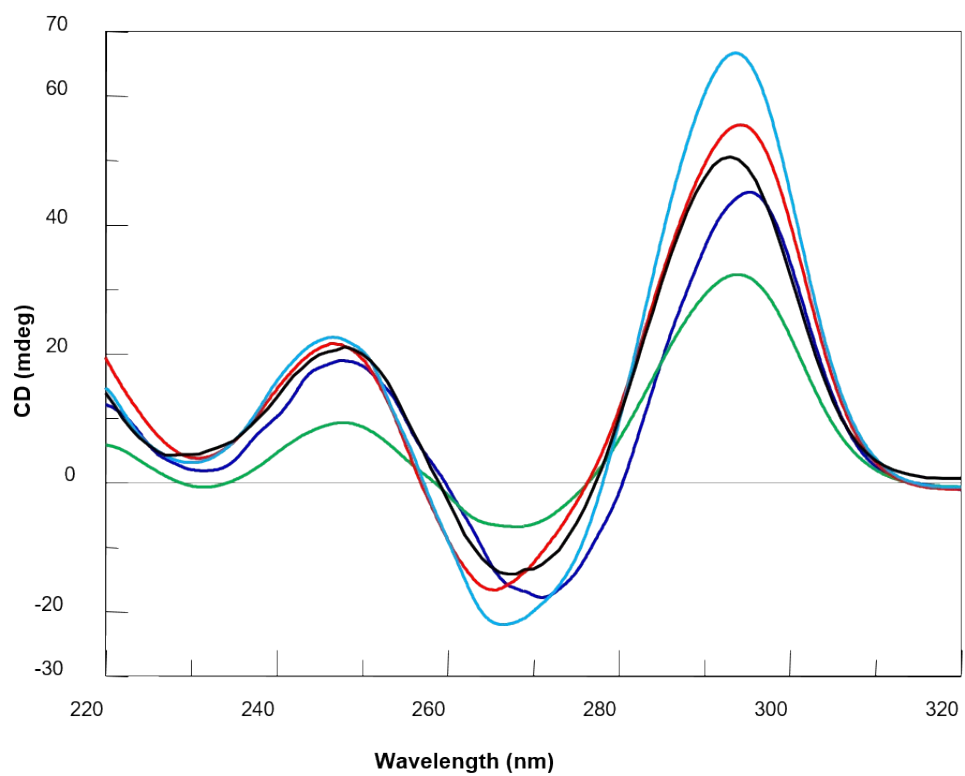


Fig. S3. CD spectra at 20°C (buffer solution 10 mM $\text{KH}_2\text{PO}_4/\text{K}_2\text{HPO}_4$, 70 mM KCl, pH 7.0) of TBA (black), TBA-G (red), TBA-A (dark blue), TBA-C (light blue), TBA-T (green).

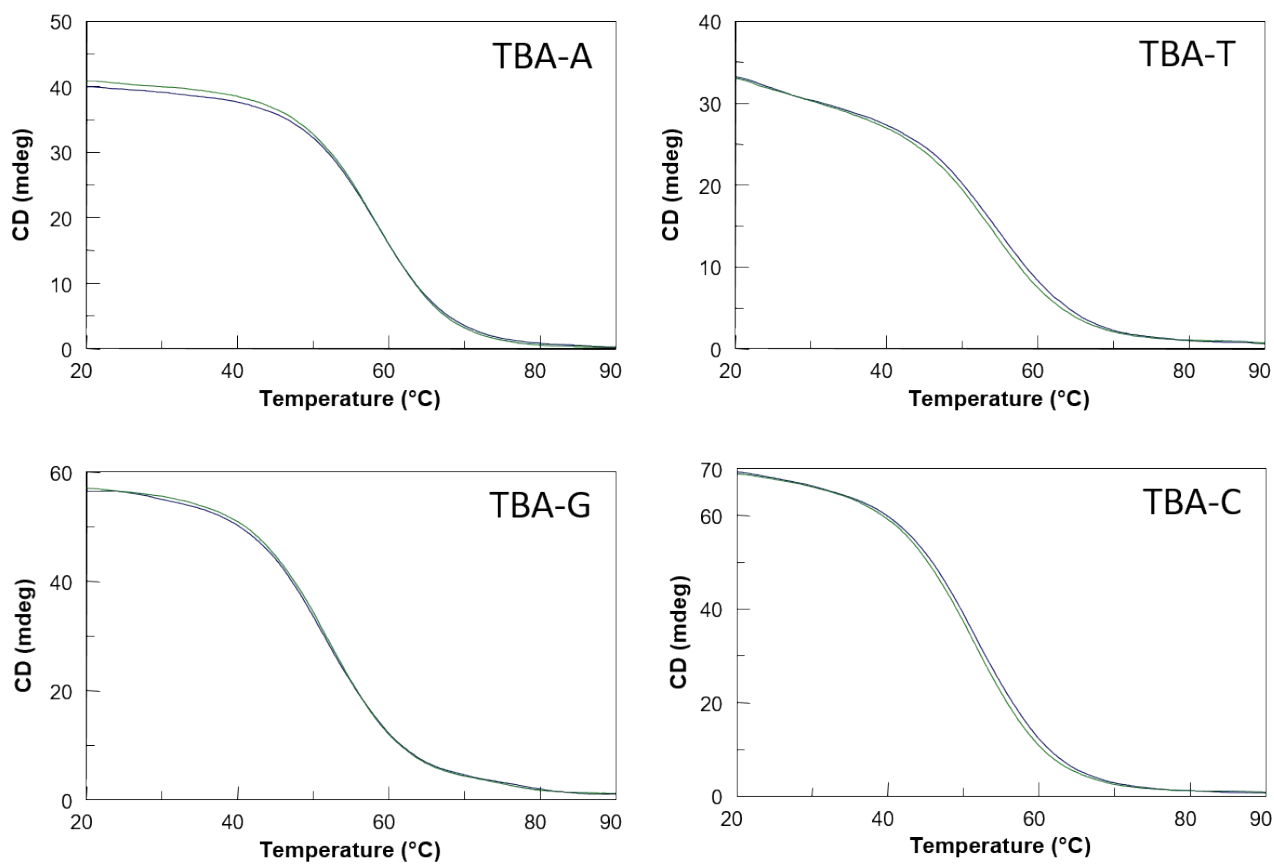


Fig. S4. CD melting (blue) and annealing (green) curves of modified TBAs registered as a function of temperature (range: 20-90°C) for all G-quadruplexes at their maximum Cotton effect wavelengths (294-295 nm). The CD data were recorded in a 0.1 cm pathlength cuvette at 100 μ M ODN strand concentration in a buffer solution 10 mM $\text{KH}_2\text{PO}_4/\text{K}_2\text{HPO}_4$, 70 mM KCl (pH 7.0), with a scan rate of 0.5°C/min.

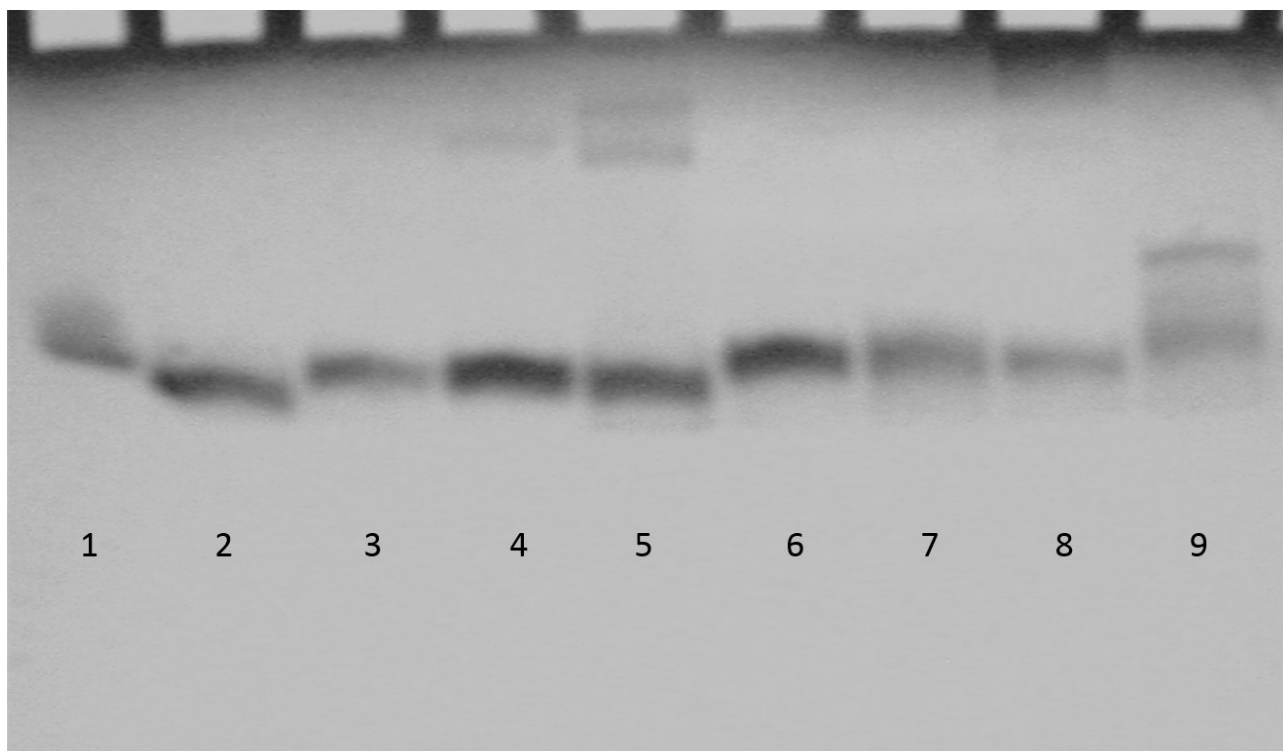


Fig. S5. Polyacrylamide non-denaturing gel electrophoresis of the TBA analogues and their natural counterpart. See Table 1 and the main text for details and experimental procedures. Lane 1: TBA; lane 2: TBA-A; lane 3: TBA-G; lane 4: TBA-T; lane 5: TBA-C; lane 6: A-TBA-T; lane 7: T-TBA-A; lane 8: C-TBA-G; lane 9: G-TBA-C.

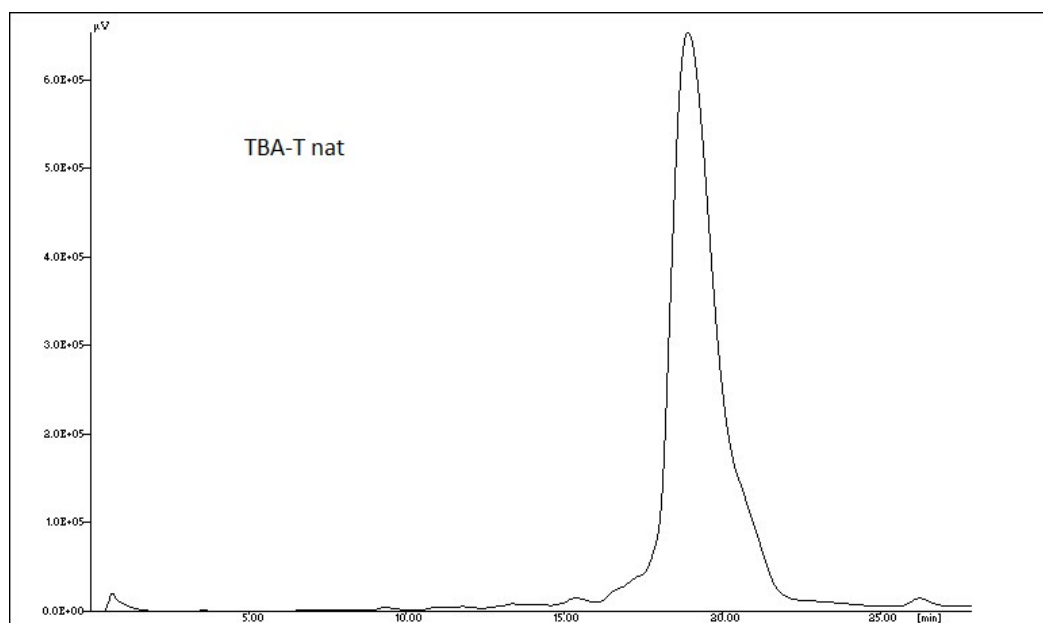
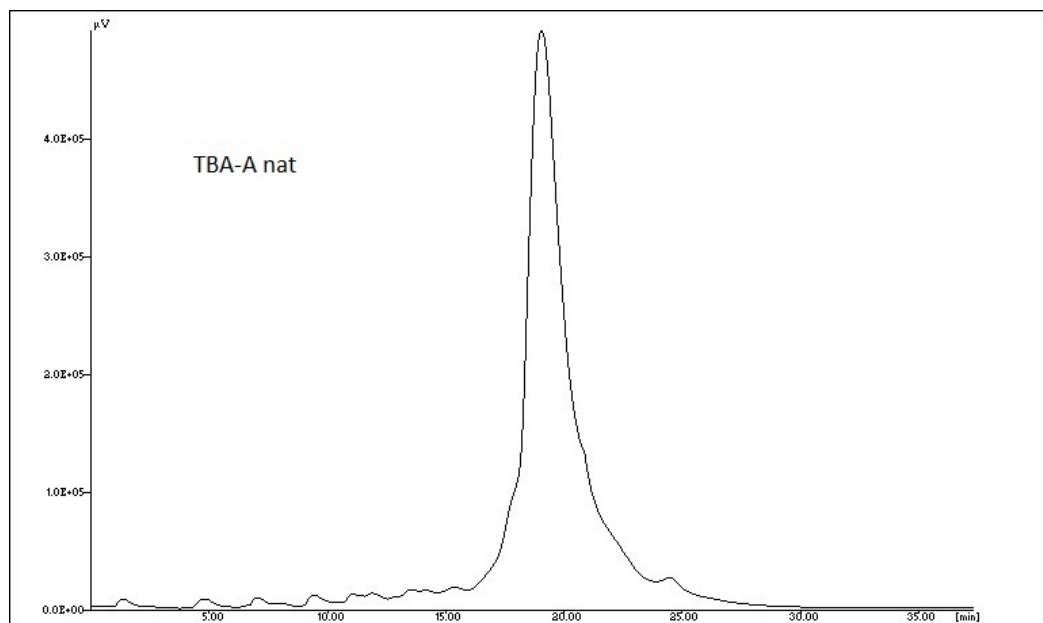


Fig. S6. HPLC chromatograms of purified TBA-A nat and TBA-T nat. See experimental section for details.

A-TBA-T A₀5'-5'G₁G₂T₃T₄G₅G₆T₇G₈T₉G₁₀G₁₁T₁₂T₁₃G₁₄G₁₅3'-3'T₁₆

	H8/H6	H1'	H2'/H2''	H3'	H4'	H5'/H5''	CH ₃ /H2
A ₀	8.10	6.23	2.43	4.52	4.45	4.07/4.00	7.81
<u>G</u> ₁	7.19	5.91	2.83/3.44	4.90	-	4.39/4.24	
G ₂	8.21	5.99	2.28/3.00	5.10	4.86	4.39/4.24	
T ₃	7.86	6.14	2.20/2.56	4.87	-	4.28/4.22	1.95
T ₄	7.08	5.99	1.97/2.64	4.85	4.24	4.17/3.87	0.95
<u>G</u> ₅	7.38	5.98	2.84/3.27	4.82	4.39	4.22/4.18	
G ₆	7.58	5.81	2.47/2.60	5.05	4.38	4.08	
T ₇	7.89	6.45	2.46/2.58	4.85	4.45	4.26/4.20	1.95
G ₈	7.69	5.71	2.25/2.36	4.77	4.46	4.08/4.02	
T ₉	7.24	5.93	1.99/2.38	4.62	4.02	3.83/3.63	1.72
<u>G</u> ₁₀	7.40	6.00	2.86/3.62	-	4.39	4.28/4.24	
G ₁₁	8.22	6.00	2.28/2.97	5.10	4.38	-	
T ₁₂	7.86	6.14	2.20/2.56	4.87	-	4.28/4.22	1.95
T ₁₃	7.14	6.03	2.01/2.68	4.86	4.25	4.19/3.88	0.95
<u>G</u> ₁₄	7.44	6.06	2.97/3.42	-	4.44	4.27/4.20	
G ₁₅	8.05	6.01	2.64/2.70	5.09	4.30	-	
T ₁₆	7.59	6.27	2.50/2.65	4.92	4.18	3.79	1.67

TBA-G 5'G ₁ G ₂ T ₃ T ₄ G ₅ G ₆ T ₇ G ₈ T ₉ G ₁₀ G ₁₁ T ₁₂ T ₁₃ G ₁₄ G ₁₅ 3'-3'G ₁₆							
	H8/H6	H1'	H2'/H2''	H3'	H4'	H5'/H5''	CH ₃
<u>G</u> ₁	7.35	6.04	2.90	4.92	4.40	4.25	
G ₂	8.16	6.05	2.36	5.15	-	-	
T ₃	7.83	6.18	2.19/2.56	4.90	4.29	4.22	1.95
T ₄	7.12	6.03	2.06/2.61	4.88	-	4.17/3.91	1.01
<u>G</u> ₅	7.41	6.00	2.85/3.39	4.87	4.27	-	
G ₆	7.76	5.93	2.58/2.70	5.08	4.51	4.21	
T ₇	7.76	6.34	2.43/2.58	4.84	4.42	4.21	1.94
G ₈	7.58	5.70	2.04/2.37	4.73	-	4.10	
T ₉	7.27	6.04	2.05/2.26	4.63	3.62	3.52/3.13	1.38
<u>G</u> ₁₀	7.45	6.02	2.91/3.53	4.92	4.40	4.27	
G ₁₁	8.21	6.03	2.35/3.01	5.14	4.40	-	
T ₁₂	7.85	6.18	2.20/2.56	4.89	4.29	4.22	1.97
T ₁₃	7.14	6.05	2.07/2.66	4.89	-	4.20/3.91	1.00
<u>G</u> ₁₄	7.46	6.06	2.99/3.51	4.92	4.27	-	
G ₁₅	8.16	6.05	2.63/2.73	4.98	4.36	-	
G ₁₆	7.82	6.05	2.82/2.95	5.04	4.29	3.77	

TBA-C 5'G ₁ G ₂ T ₃ T ₄ G ₅ G ₆ T ₇ G ₈ T ₉ G ₁₀ G ₁₁ T ₁₂ T ₁₃ G ₁₄ G ₁₅ 3'-3'C ₁₆							
	H8/H6	H1'	H2'/H2''	H3'	H4'	H5'/H5''	CH ₃ /H5
<u>G</u> ₁	7.31	5.97	2.88	4.94	-	-	
G ₂	8.09	5.96	2.29/2.94	5.08	-	-	
T ₃	7.79	6.13	2.14/2.49	-	-	-	1.91
T ₄	7.09	5.96	1.99/2.55	-	-	-	0.98
<u>G</u> ₅	7.38	5.96	2.83/3.35	4.82	-	-	
G ₆	7.72	5.95	2.53/2.72	5.06	-	-	
T ₇	7.75	6.35	2.42/2.52	-	-	-	1.90
G ₈	7.27	5.59	1.86/2.32	4.67	-	-	
T ₉	7.16	5.86	2.25/2.32	-	-	-	1.40
<u>G</u> ₁₀	7.39	5.97	2.84/3.55	4.86	-	-	
G ₁₁	8.12	5.96	2.27/2.90	5.08	-	-	
T ₁₂	7.79	6.13	2.14/2.49	-	-	-	1.91
T ₁₃	7.13	5.96	2.02/2.62	4.84	-	-	0.95
<u>G</u> ₁₄	7.43	6.01	2.91/3.45	4.87	-	-	
G ₁₅	8.04	6.06	2.41/2.65	5.01	-	-	
C ₁₆	7.68	6.13	2.38/2.56	-	-	-	5.85

TBA-A 5' <u>G</u> ₁ G ₂ T ₃ T ₄ G ₅ G ₆ T ₇ G ₈ T ₉ G ₁₀ G ₁₁ T ₁₂ T ₁₃ G ₁₄ G ₁₅ 3'-3'A ₁₆							
	H8/H6	H1'	H2'/H2''	H3'	H4'	H5'/H5''	CH ₃ /H2
<u>G</u> ₁	7.20	6.05	2.80/2.94	5.01	4.24	4.00	
G ₂	8.12	5.96	2.01/2.31	5.12	-	-	
T ₃	7.82	6.17	2.17/2.31	4.87	-	4.26/4.21	1.94
T ₄	7.09	6.00	2.00/2.57	4.86	4.25	4.16/3.89	0.98
<u>G</u> ₅	7.44	5.98	2.79/3.31	4.84	4.48	4.22	
G ₆	7.84	6.10	2.52/2.79	5.12	-	-	
T ₇	7.62	6.32	2.52/2.56	4.85	4.26	4.20/4.14	1.92
G ₈	7.51	6.11	2.52	5.23	4.44	4.15	
T ₉	7.01	6.07	1.87/2.10	5.24	4.47	3.45/3.35	0.83
<u>G</u> ₁₀	7.48	6.05	2.91/3.71	5.11	4.92	4.41	
G ₁₁	8.26	6.07	2.35/3.04	5.14	-	-	
T ₁₂	7.87	6.18	2.23/2.59	4.90	4.29	4.23	1.97
T ₁₃	7.07	6.00	2.02/2.64	4.89	4.27	4.19/3.91	1.00
<u>G</u> ₁₄	7.50	6.13	3.04/3.51	4.91	4.30	-	
G ₁₅	8.21	6.16	2.67/2.86	5.05	-	4.30/4.21	
A ₁₆	8.33	6.21	2.91/2.97	5.11	-	-	7.78

TBA-T	5'G ₁ G ₂ T ₃ T ₄ G ₅ G ₆ T ₇ G ₈ T ₉ G ₁₀ G ₁₁ T ₁₂ T ₁₃ G ₁₄ G ₁₅ 3'-3'T ₁₆						
	H8/H6	H1'	H2'/H2''	H3'	H4'	H5'/H5''	CH ₃
<u>G₁</u>	7.37	6.02	2.88/2.95	5.00	4.40	4.25/4.03	
G ₂	8.15	6.02	2.33/2.99	5.13	-	4.25	
T ₃	7.84	6.20	2.18/2.55	4.89	-	4.27/4.19	1.95
T ₄	7.13	6.03	2.04/2.61	4.87	4.47	4.17/3.90	1.02
<u>G₅</u>	7.41	5.98	2.83/3.35	4.85	4.40	4.22	
G ₆	7.69	5.91	2.52/2.71	5.09	-	4.22	
T ₇	7.84	6.43	2.48/2.60	4.86	-	-	1.96
G ₈	7.55	5.70	2.07/2.31	4.73	4.41	4.21/4.07	
T ₉	7.22	5.96	1.94/2.31	4.61	-	3.68/3.57	1.59
<u>G₁₀</u>	7.44	6.01	2.88/3.60	4.90	4.39	4.27/4.11	
G ₁₁	8.20	6.02	2.32/2.97	5.12	-	4.28	
T ₁₂	7.85	6.17	2.19/2.56	4.89	4.47	4.27/4.19	1.96
T ₁₃	7.16	6.06	2.05/2.67	4.88	4.47	4.19/3.90	0.98
<u>G₁₄</u>	7.48	6.07	2.97/3.48	4.91	4.48	4.27/4.21	
G ₁₅	8.07	6.06	2.61/2.70	5.10	4.48	4.27/4.22	
T ₁₆	7.59	6.28	2.44/2.67	4.90	4.19	3.77	1.72

Table S1. Proton chemical shifts assignment for G-quadruplex structures formed by ODNs A-TBA-T, TBA-G, TBA-C, TBA-A, TBA-T (700 MHz; 25°C) in PBS buffer solution (H₂O/D₂O 9:1) and 0.2 mM EDTA (pH 7.0). Deoxyguanosines adopting *syn* glycosidic conformation are underlined. The hypens indicate values not determined.

Names	T_m (°C) (± 1)	ΔT_m (°C)	$\Delta H_{v.H.}$ (kJ/mol) ^a	$\Delta S_{v.H.}$ (kJ/mol K) ^a
TBA	50	-	154	0.48
TBA-A	59	+9	194	0.58
TBA-T	56	+6	174	0.53
TBA-G	52	+2	175	0.54
TBA-C	53	+3	172	0.53

^a The errors on $\Delta H_{v.H.}$ and $\Delta S_{v.H.}$ are within the 10 %.

Table S2. Thermodynamic parameters of some modified aptamers and their natural counterpart (TBA) in potassium buffer obtained from the van't Hoff analysis of the CD melting curves (see text for details).