Electronic Supplementary Material (ESI) for RSC Advances. This journal is © The Royal Society of Chemistry 2016

Support Information

^a Oil & Gas Field Applied Chemistry Key Laboratory of Sichuan Province, College of Chemistry and Chemical Engineering, Southwest Petroleum University, 8 Xindu Road, Xindu, Chengdu, 610500, China
^b Key Laboratory of Oil & Gas Fine Chemicals, Ministry of Education, College of Chemistry and Chemical Engineering, Xinjiang University, Urumqi 830046, Xinjiang, China



Figure S1 ¹H NMR chart of M1 (DMSO-d₆)



Figure S2 ¹³C NMR chart of M1 (CF₃COOD)



Figure S3 The IR spectra of M1



Figure S4 HR-MS Spectra of M1, M=307



Figure S5 UV-vis spectral changes of M1 upon addition of Al³⁺ and other metal ions in ethanol at room temperature ([M1] = 10.0 μ M, [Mⁿ⁺] =10.0 equiv.).



Figure S6. Interference from anions in a binary mixture: M1 (10.0 μ M) +Al³⁺ (10.0 equiv.) + Aⁿ⁻ (10.0 equiv.), where Aⁿ⁻=Br, CH₃COO⁻, Cl⁻, ClO₄⁻, CN⁻, CO₃²⁻, F⁻, H₂PO₄⁻, HCO₃⁻, I⁻, NO₂⁻, NO₃⁻, OH⁻, P₂O₇²⁻, PO₄³⁻, S²⁻ (λ_{ex} = 358 nm, ethanol).



Figure S7. Job's plot of M1 and Al^{3+} in ethanol at 442nm



Figure S8. Mass spectrum of complex M1-Al³⁺, M=427