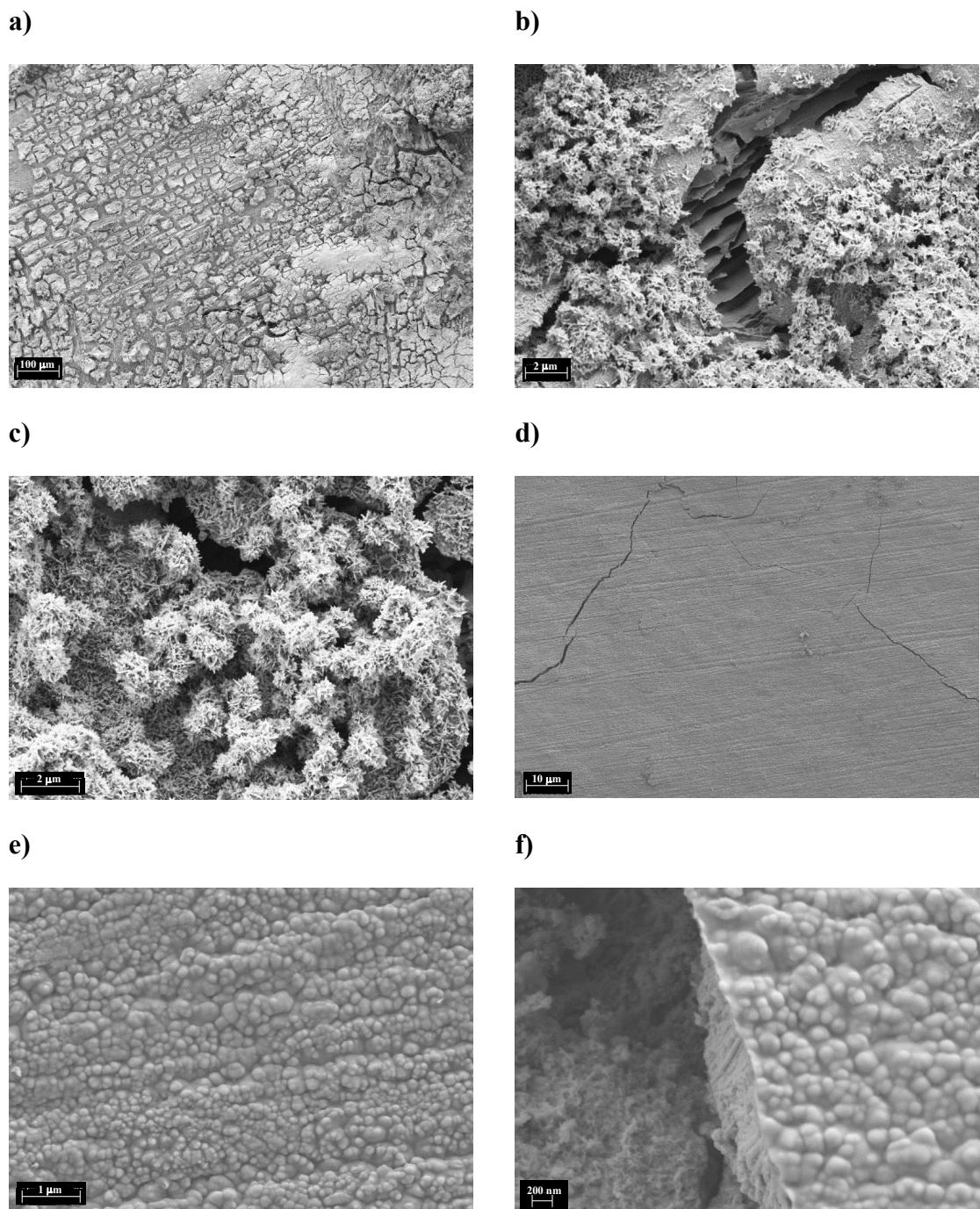


**ZnO/ZnS heterostructures for hydrogen production by photoelectrochemical  
water splitting**

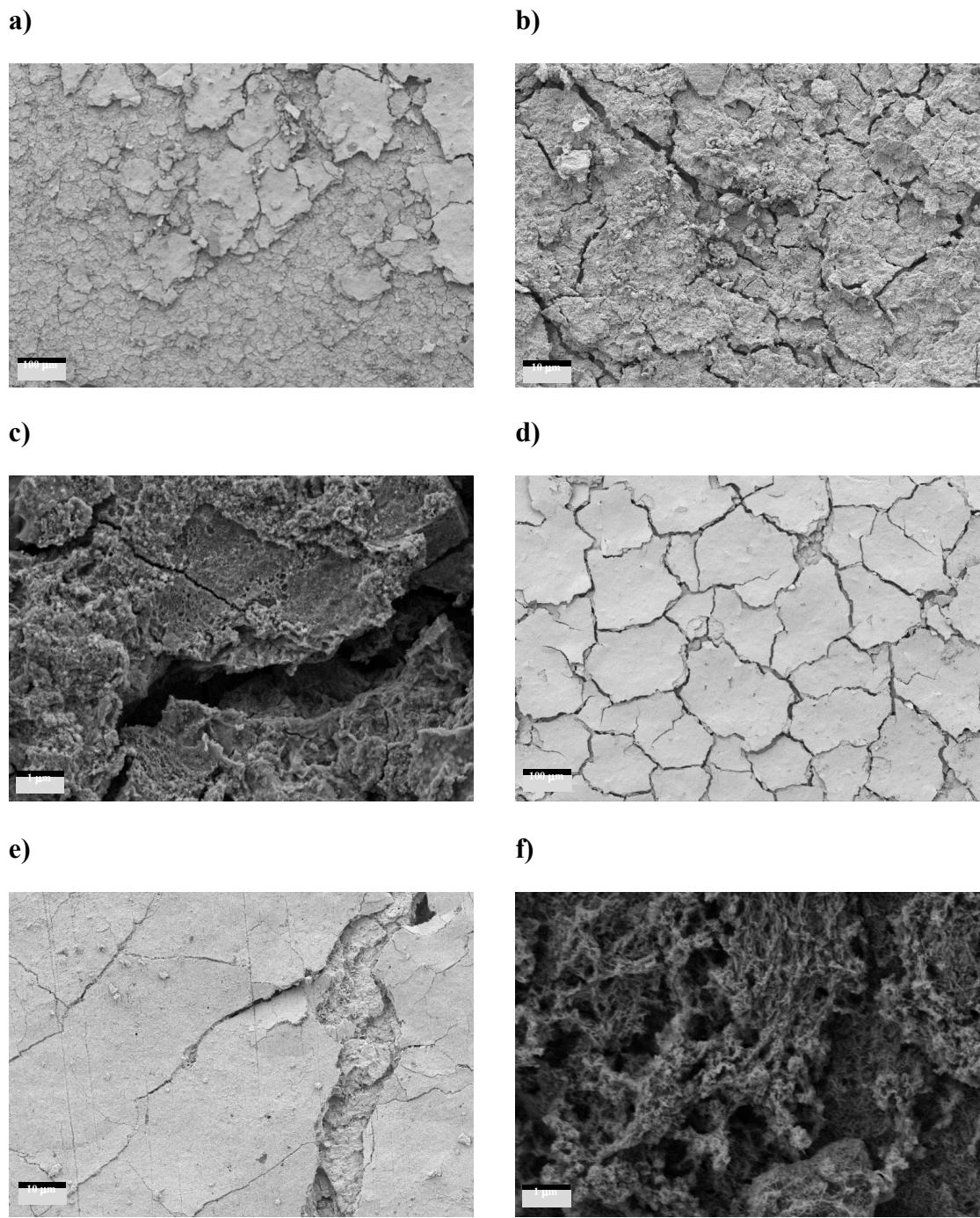
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Nuclear. ETSI Industriales. Universitat Politècnica de València. Camino de Vera s/n,  
46022 Valencia, Spain.*

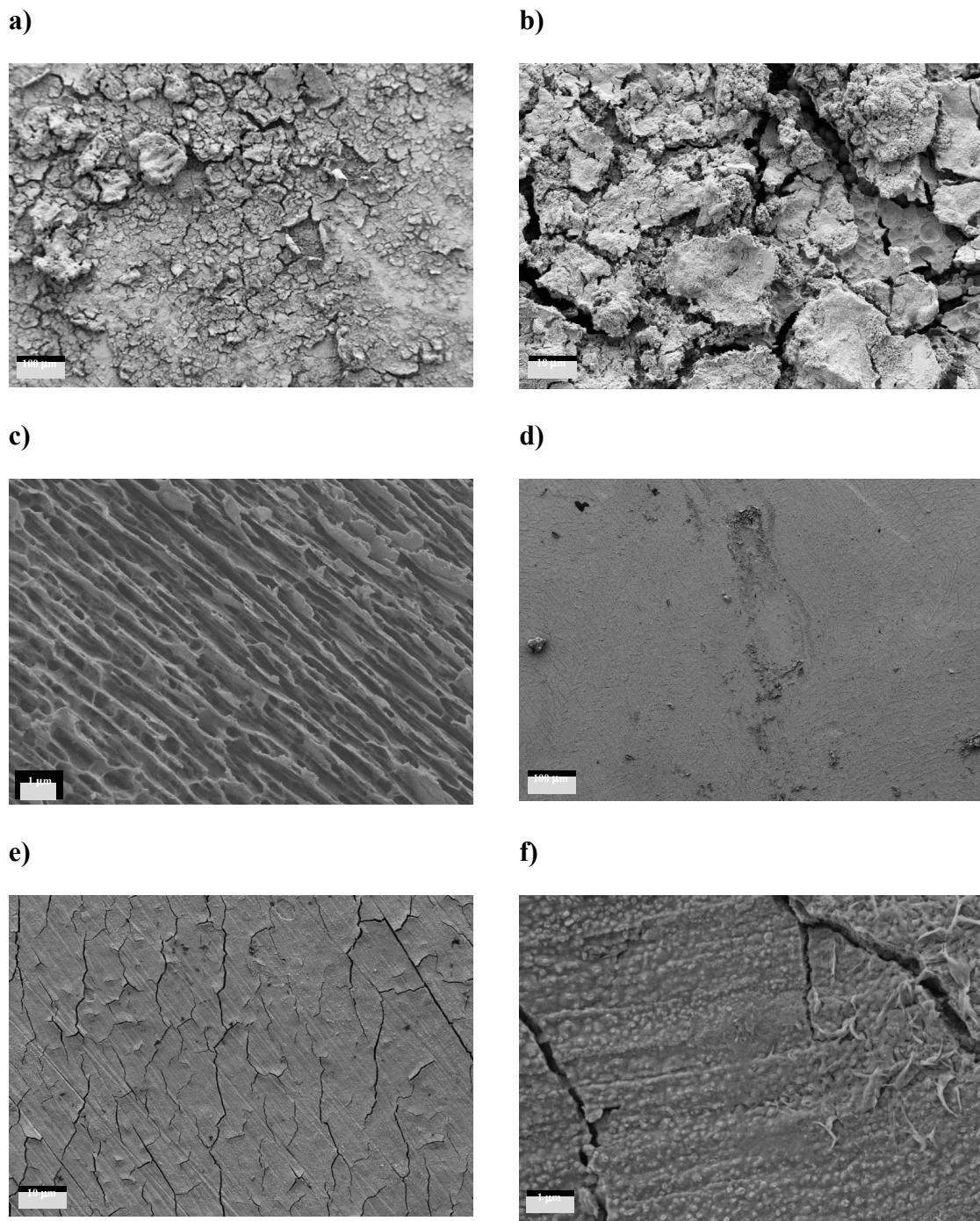
*Tel. 34-96-387 76 32, Fax. 34-96-387 76 39, e-mail: jgarciaa@iqn.upv.es*



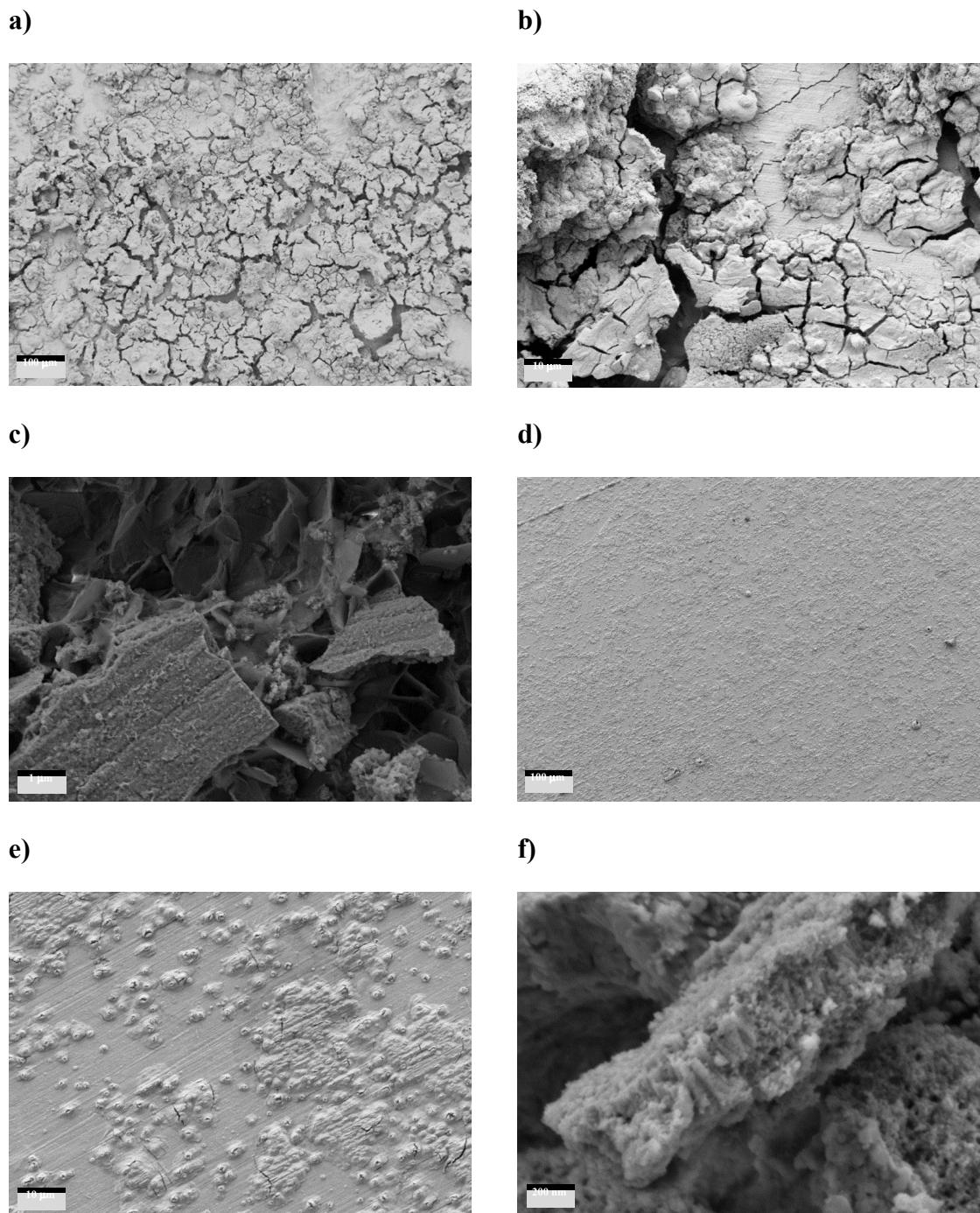
**Figure S1.** FE-SEM images of the ZnO/ZnS heterostructures anodized under static (a-c) and dynamic conditions (d-f) in 0.06M Na<sub>2</sub>S+0.0125M water electrolytes.



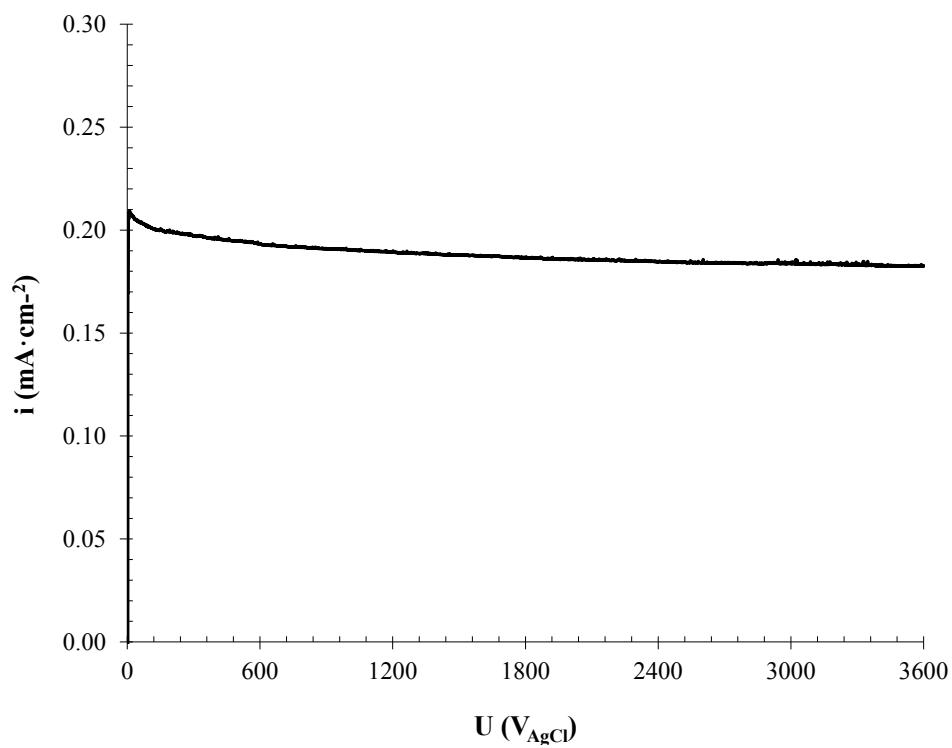
**Figure S2.** FE-SEM images of the ZnO/ZnS heterostructures anodized under static (a-c) and dynamic (d-f) conditions in 0.2 M Na<sub>2</sub>S+0.0125M water electrolytes.



**Figure S3.** FE-SEM images of the ZnO/ZnS heterostructures anodized under static (a-c) and dynamic conditions (d-f) in  $0.06\text{M Na}_2\text{S} + 0.0125\text{M glycerol/water}$  electrolytes.



**Figure S4.** FE-SEM images of the ZnO/ZnS heterostructures anodized under static (a-c) and dynamic conditions (d-f) in 0.2M Na<sub>2</sub>S+0.0125M glycerol/water electrolytes.



**Figure S5.** Photostability in 0.24M  $\text{Na}_2\text{S}$  and 0.35M  $\text{Na}_2\text{SO}_3$  solution during 1 hour of the heterostructure anodized in 0.2M  $\text{Na}_2\text{S} + 0.025\text{M NH}_4\text{F}$  glycerol/water solution stirring the electrolyte at 2000 rpm, holding at 500 mV the potential under AM 1.5 illumination