

ELECTRONIC SUPPLEMENTARY INFORMATION

(ESI)

Hybrid Green Nonaqueous Media: Tetraethylene Glycol Modifies Properties of (Choline Chloride + Urea) Deep Eutectic Solvent

Anu Kadyan,^a Kamalakanta Behera,^a and Siddharth Pandey^{*a}

^a Department of Chemistry, Indian Institute of Technology Delhi, Hauz Khas, New Delhi - 110016, India.

*To whom correspondence should be addressed.

E-mail: sipandey@chemistry.iitd.ac.in, Phone: +91-11-26596503, Fax: +91-11-26581102

Table S1: E_{T}^{N} , π^* , α , and β values calculated using absorbance probes Reichardt's Dye 33 (250 μM), nitroaniline (NA, 10 μM), and *N,N*-diethyl-4-nitroaniline (DENA, 25 μM) with equations (1), (2), (3), (4) and (5) for (Reline + TEG) Mixtures at different temperatures as a function of the mole fraction of TEG (x_{TEG}).

T(K)	E_{T}^{N}								
	$x_{\text{TEG}}=0$	$x_{\text{TEG}}=0.05$	$x_{\text{TEG}}=0.10$	$x_{\text{TEG}}=0.15$	$x_{\text{TEG}}=0.2$	$x_{\text{TEG}}=0.4$	$x_{\text{TEG}}=0.6$	$x_{\text{TEG}}=0.8$	$x_{\text{TEG}}=1$
298	0.83	0.82	0.79	0.77	0.77	0.74	0.72	0.70	0.64
313	0.82	0.81	0.77	0.76	0.75	0.72	0.71	0.69	0.62
328	0.80	0.78	0.77	0.75	0.74	0.71	0.68	0.67	0.59
343	0.78	0.78	0.74	0.74	0.72	0.69	0.67	0.66	0.57
358	0.78	0.76	0.74	0.72	0.71	0.67	0.66	0.64	0.54

T(K)	π^*								
	$x_{\text{TEG}}=0$	$x_{\text{TEG}}=0.05$	$x_{\text{TEG}}=0.10$	$x_{\text{TEG}}=0.15$	$x_{\text{TEG}}=0.2$	$x_{\text{TEG}}=0.4$	$x_{\text{TEG}}=0.6$	$x_{\text{TEG}}=0.8$	$x_{\text{TEG}}=1$
298	1.23	1.24	1.21	1.19	1.15	1.08	1.03	0.99	0.90
313	1.23	1.23	1.21	1.19	1.15	1.08	0.99	0.97	0.86
328	1.23	1.23	1.23	1.17	1.15	1.05	0.99	0.93	0.84
343	1.23	1.23	1.21	1.17	1.14	1.05	0.97	0.92	0.82
358	1.23	1.23	1.21	1.17	1.14	1.03	0.95	0.90	0.78

T(K)	α								
	$x_{\text{TEG}}=0$	$x_{\text{TEG}}=0.05$	$x_{\text{TEG}}=0.10$	$x_{\text{TEG}}=0.15$	$x_{\text{TEG}}=0.2$	$x_{\text{TEG}}=0.4$	$x_{\text{TEG}}=0.6$	$x_{\text{TEG}}=0.8$	$x_{\text{TEG}}=1$
298	0.56	0.53	0.50	0.48	0.51	0.51	0.53	0.53	0.49
313	0.55	0.52	0.47	0.47	0.47	0.49	0.54	0.52	0.48
328	0.52	0.48	0.44	0.46	0.45	0.49	0.49	0.52	0.44
343	0.48	0.46	0.41	0.43	0.44	0.45	0.47	0.52	0.42
358	0.46	0.44	0.40	0.40	0.41	0.44	0.47	0.49	0.39

T(K)	β								
	$x_{\text{TEG}}=0$	$x_{\text{TEG}}=0.05$	$x_{\text{TEG}}=0.10$	$x_{\text{TEG}}=0.15$	$x_{\text{TEG}}=0.2$	$x_{\text{TEG}}=0.4$	$x_{\text{TEG}}=0.6$	$x_{\text{TEG}}=0.8$	$x_{\text{TEG}}=1$
298	0.50	0.50	0.55	0.59	0.65	0.69	0.71	0.63	0.67
313	0.50	0.52	0.55	0.59	0.63	0.67	0.73	0.61	0.69
328	0.50	0.50	0.52	0.59	0.61	0.69	0.68	0.63	0.66
343	0.50	0.50	0.55	0.56	0.61	0.66	0.68	0.62	0.61
358	0.50	0.50	0.55	0.56	0.61	0.64	0.65	0.62	0.61

Table S2: Densities ($\rho/\text{g.cm}^{-3}$) of (Reline + TEG) mixtures at pressure $p = 0.1 \text{ MPa}$ and $T = 298 \text{ K}$ to 358 K as a function of mole fraction of TEG (x_{TEG}).

x_{TEG}	Temperature (K)				
	298	313	328	343	358
0.00	1.194	1.186	1.178	1.169	1.157
0.05	1.191	1.182	1.174	1.165	1.154
0.10	1.185	1.176	1.168	1.159	1.148
0.15	1.182	1.172	1.165	1.155	1.144
0.20	1.177	1.167	1.160	1.150	1.139
0.40	1.158	1.147	1.139	1.130	1.119
0.60	1.144	1.133	1.125	1.116	1.105
0.80	1.133	1.122	1.114	1.105	1.095
1.00	1.121	1.109	1.101	1.092	1.082