

Supporting information

Improved activity and stability of dealloyed bimodal nanoporous PtCo catalysts for CO oxidation

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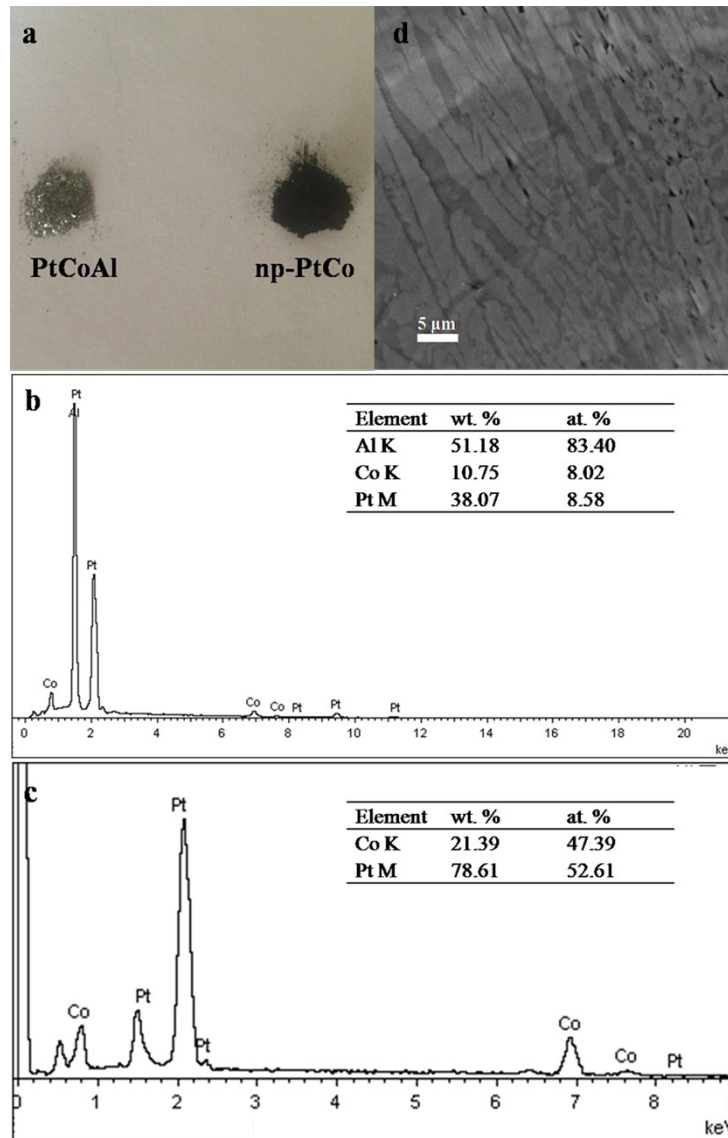


Fig. S1. (a) Macrograph of PtCoAl precursor and as-dealloyed np-PtCo. (b) EDS analysis of PtCoAl precursor. (c) EDS analysis of as-dealloyed PtCo. (d) SEM image of PtCoAl precursor.

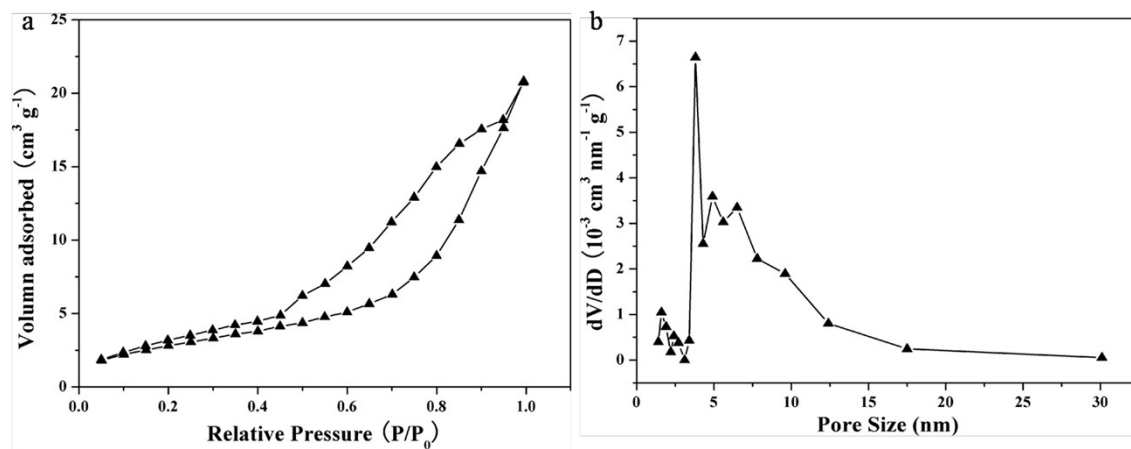


Fig. S2. (a) N₂ adsorption/desorption isotherm of the obtained np-PtCo. (b) Pore-size distribution obtained using the BJH method.

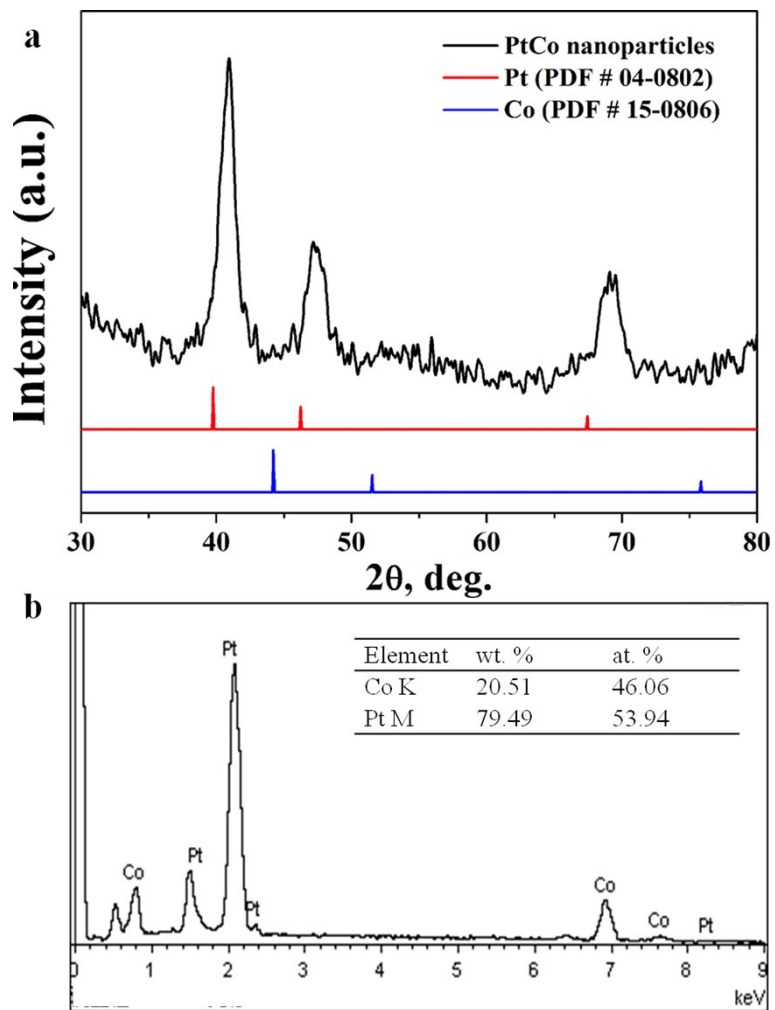


Fig. s3. (a) XRD patterns of as-prepared PtCo nanoparticles. (b) EDS analysis of as-prepared PtCo nanoparticles.

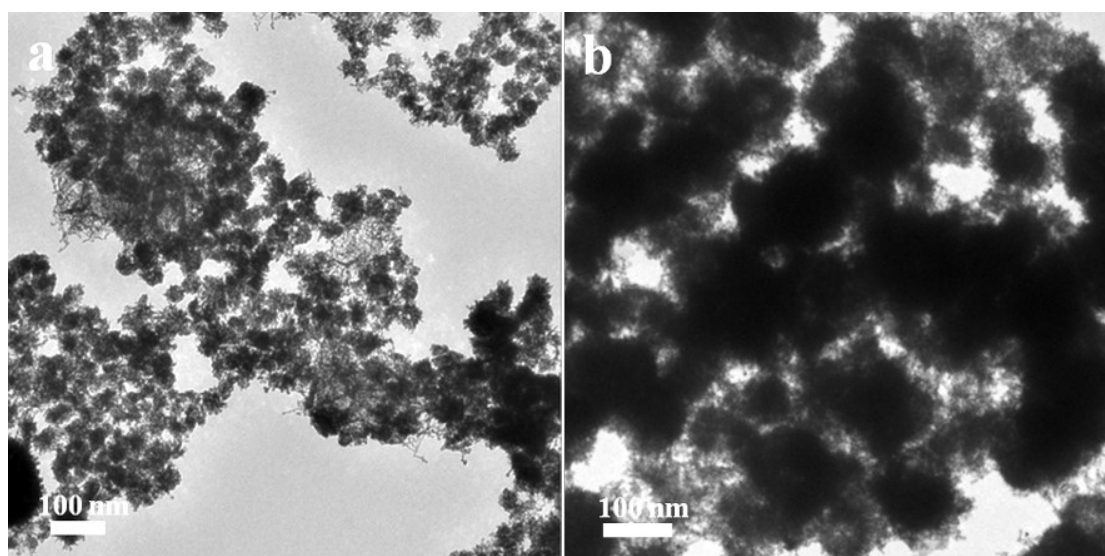


Fig. s4. TEM images of as-prepared PtCo nanoparticles before (a) and after (b) the catalytic reaction.

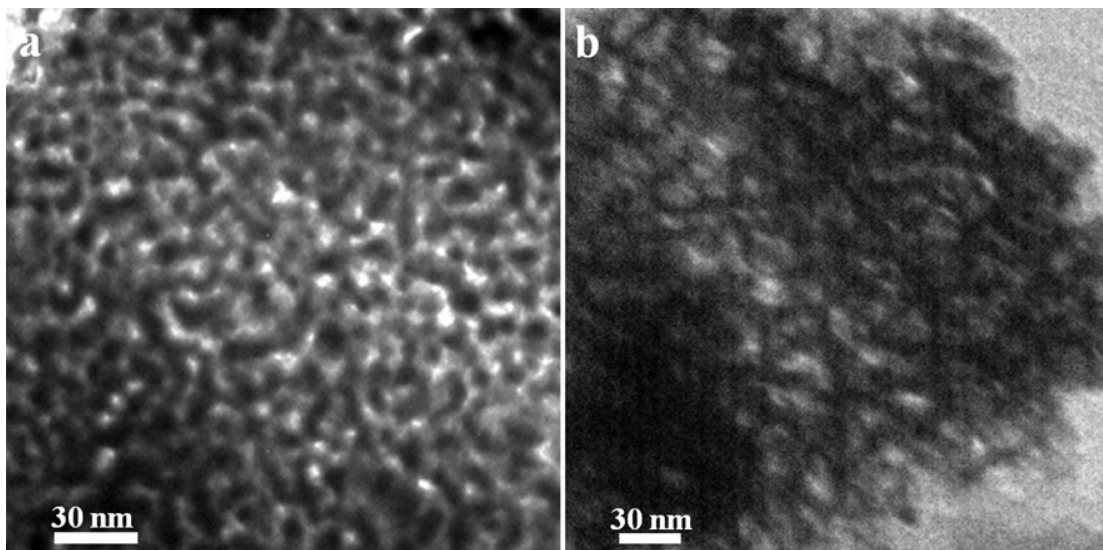


Fig. s5. TEM images of np-Pt before (a) and after (b) the catalytic reaction.

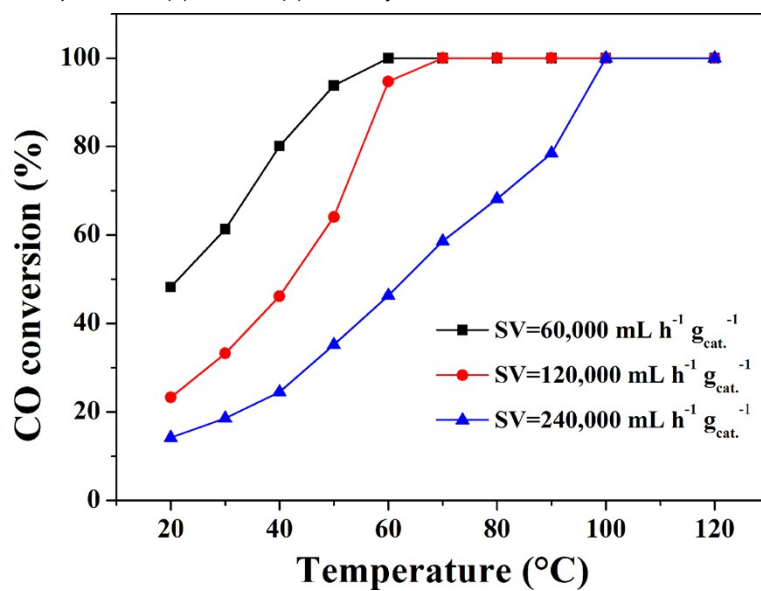


Fig. s6. The influence of space velocity on CO conversion over np-PtCo with 1% CO, 10% O₂, and N₂ balance at space velocity of 60,000 (■), 120,000 (●), and 240,000 mL h⁻¹ g_{cat.}⁻¹ (▲), respectively.