

Electronic Supplementary Information (ESI)

A new bisglycolamide substituted calix[4]arene-benzo-crown-6 for the selective removal of Cesium ion: Combined experimental and density functional theoretical investigation

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Figure S1: ¹H NMR of 1,3-dipropenyloxy calix-benzo-crown (4):

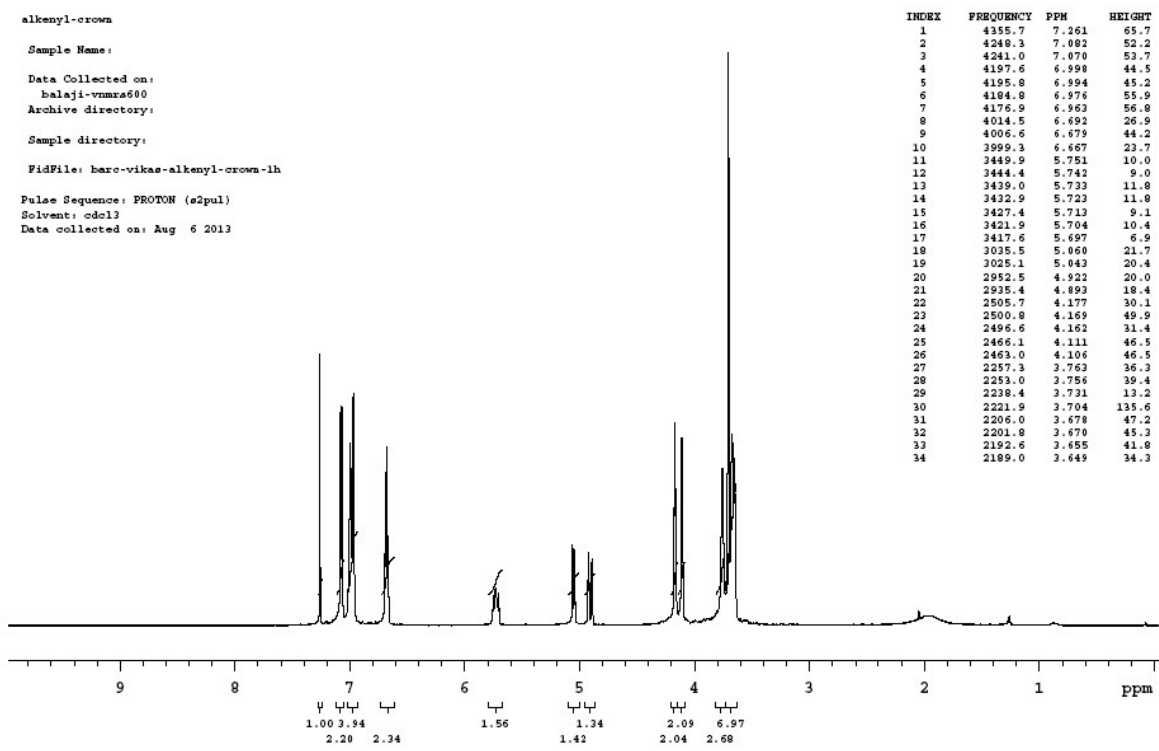


Figure S2: ¹³C NMR of 1,3-dipropenyloxy calix-benzo-crown (4):

calix crown amide

Sample Name:
alkenyl-crown
Data Collected on:
balaji-vmmrs600
Archive directory:

Sample directory:

FidFile: barc-vikas-alkenyl-crown-13c

Pulse Sequence: CARBON (s2pul)
Solvent: cdcl3
Data collected on: Aug 6 2013

INDEX	FREQUENCY	PPM	HEIGHT
1	23539.5	156.058	15.7
2	23497.9	155.782	13.3
3	22462.1	149.916	16.5
4	20256.5	134.294	34.7
5	20140.9	133.527	44.8
6	20127.1	133.435	39.3
7	19761.8	131.013	63.6
8	19627.7	130.125	58.6
9	18457.8	122.369	19.7
10	18434.7	122.216	27.7
11	18370.0	121.786	22.6
12	17473.0	115.839	41.5
13	17343.5	114.981	16.4
14	11642.2	77.184	136.7
15	11609.9	76.969	112.3
16	11592.1	76.785	93.4
17	10699.7	70.869	26.4
18	10675.9	70.777	41.9
19	10625.0	70.440	37.9
20	10583.4	70.164	37.8
21	10472.4	69.428	38.5
22	5658.9	37.517	43.2

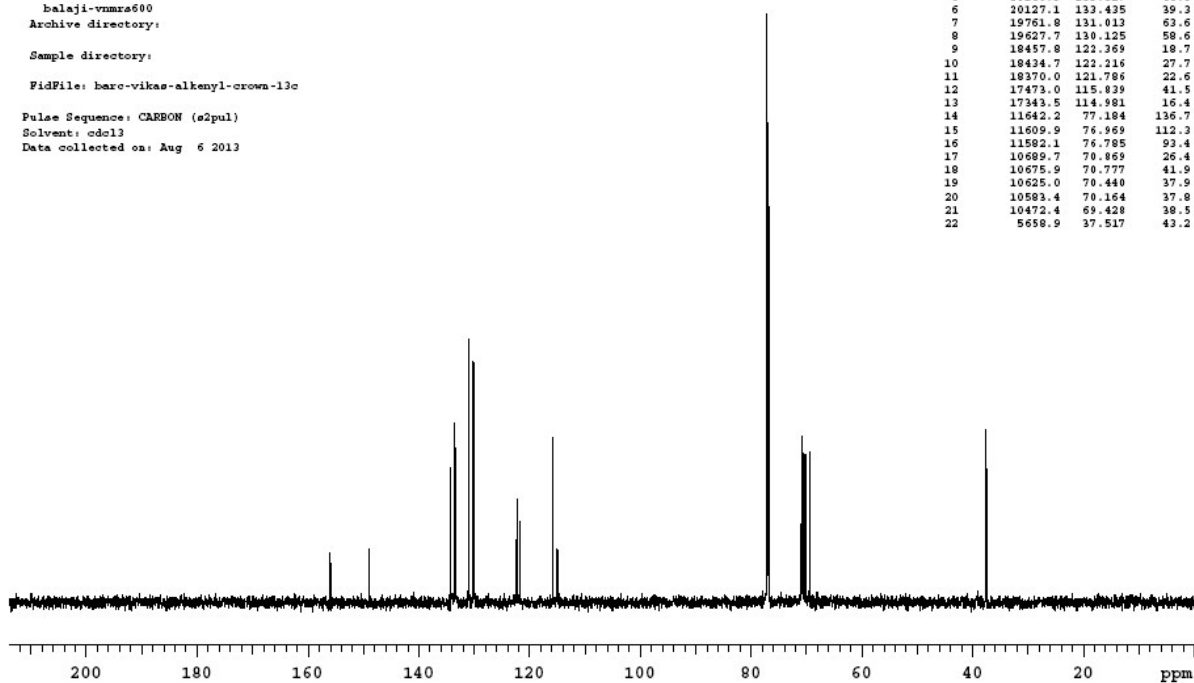


Figure S3: ESI-MS spectra of 1,3-dipropenyloxy calix-benzo-crown (4):

Spectrum Plot - 1/3/2014 3:23 PM

1 A Scan 22 from ...ata\external\barc-vikas\newatbenzyl cc-755.xms

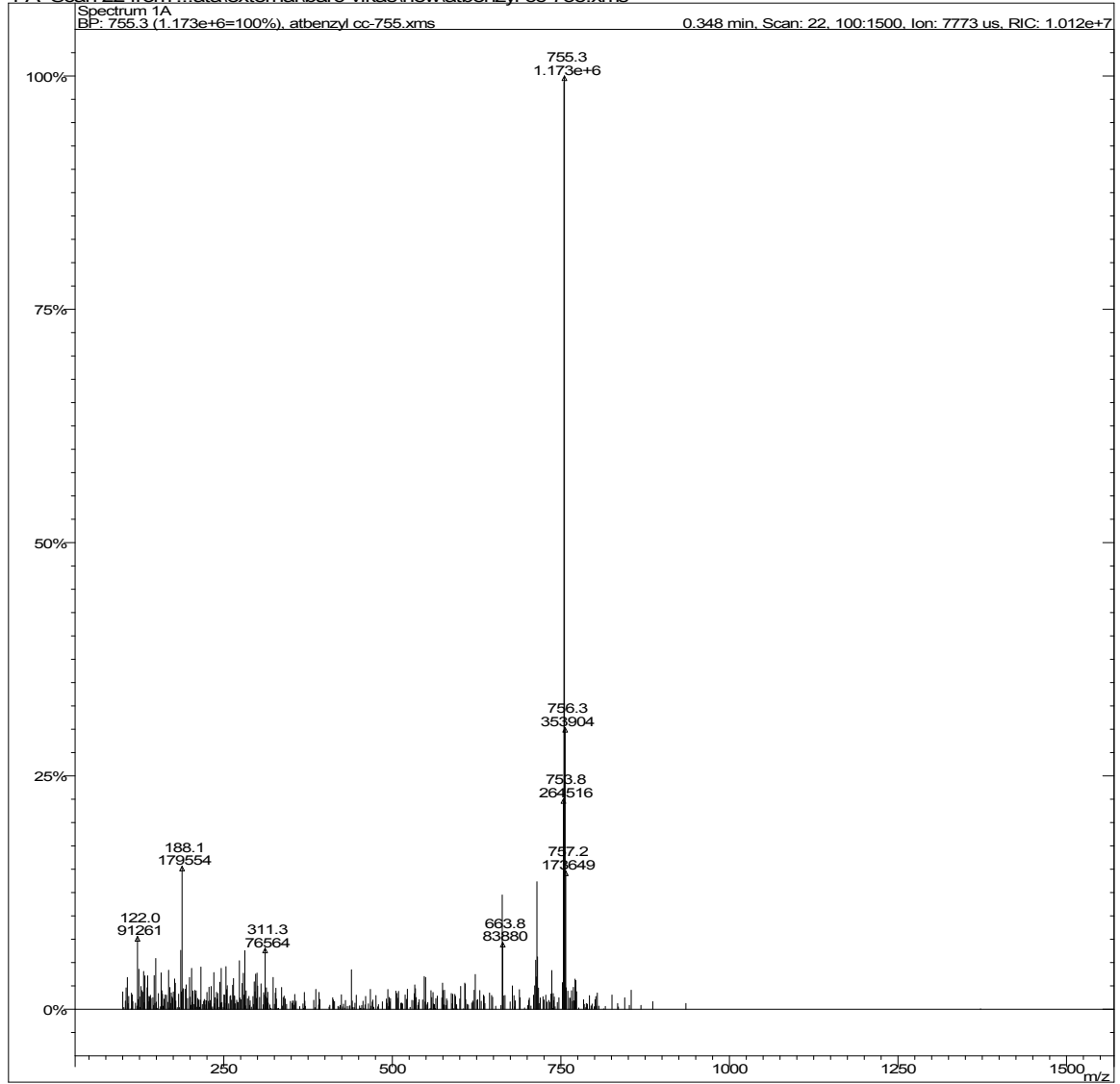


Figure S4: ¹H NMR of 1,3-dipropoxy calix[4]arene-benzocrown diol(5):

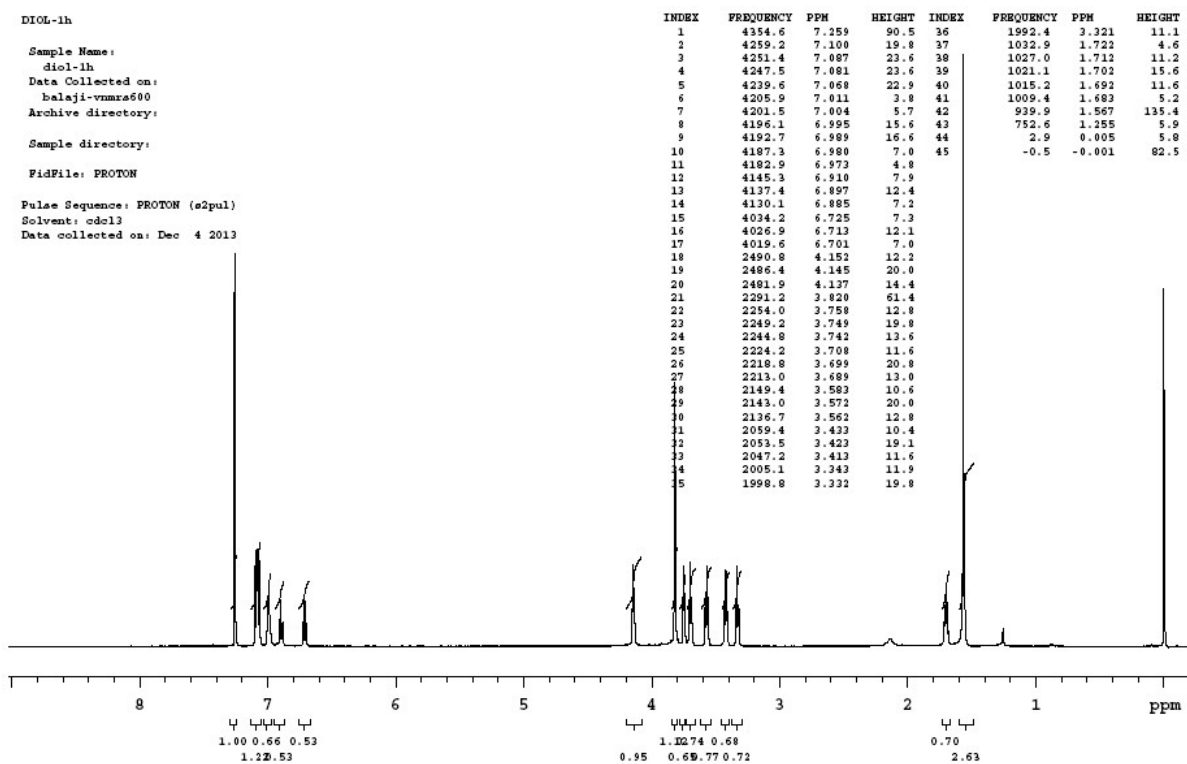


Figure S5: ¹³C NMR of 1,3-dipropoxy calix[4]arene-benzocrown diol(5):

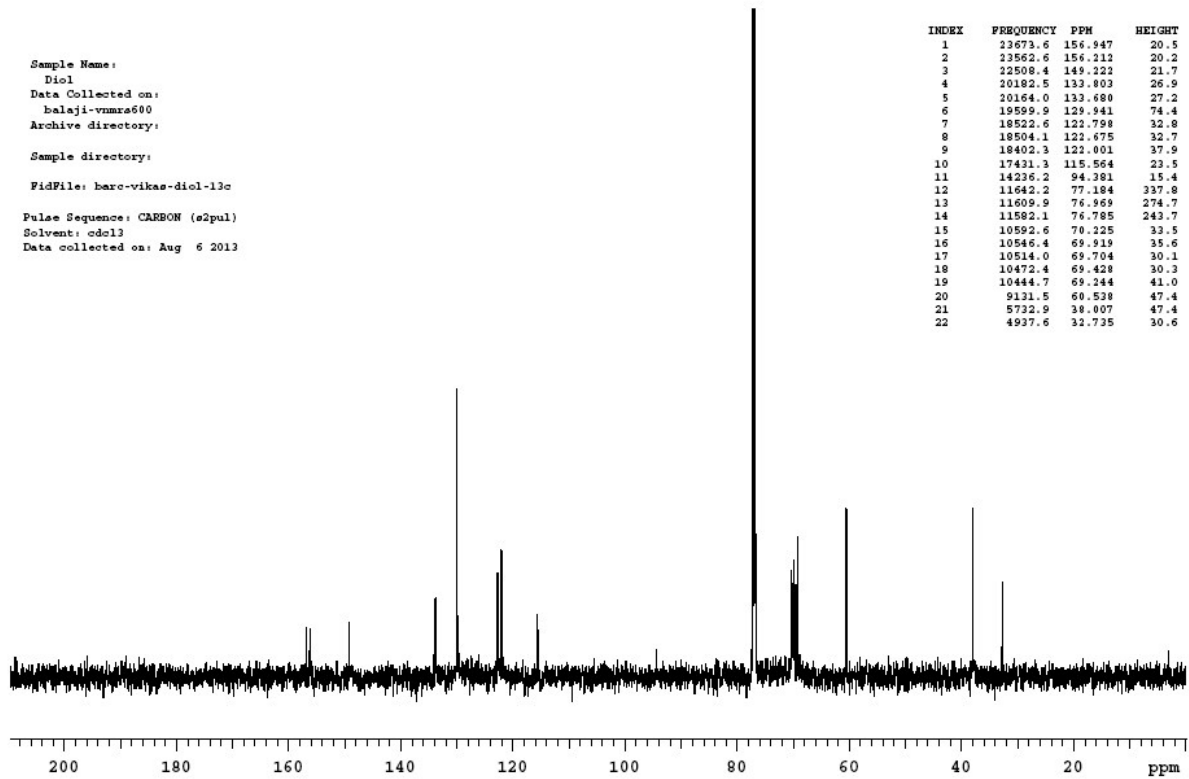


Figure S6: ESI-MS spectra of 1,3-dipropoxy calix[4]arene-benzocrown diol(5):

Spectrum Plot - 1/3/2014 3:39 PM

1 A Scan 55 from c:\data\external\barc-vikas\new\diol-791.xms

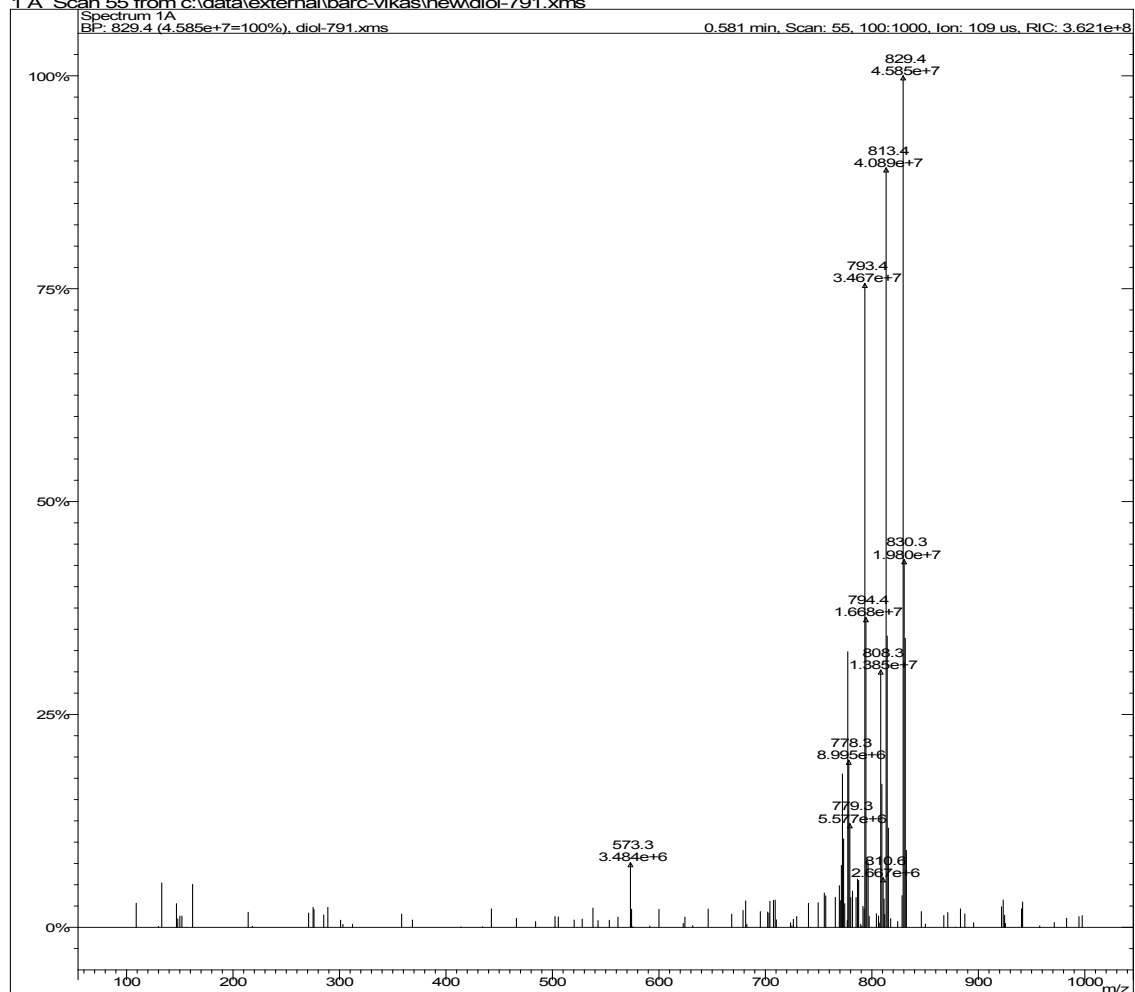


Figure S7: ^1H NMR of 1,3- alternate Calix[4]arene-benzocrown-bisglycolamide (6)

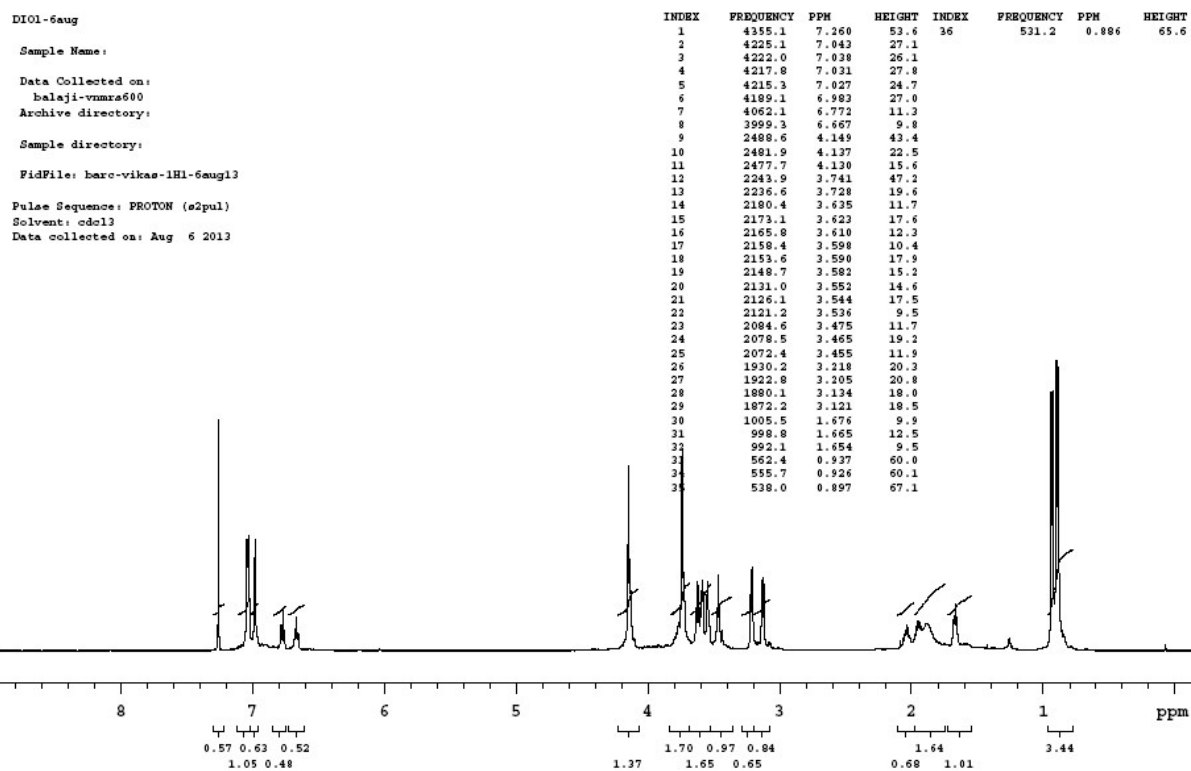


Figure S 8: ^{13}C NMR of 1,3- alternate Calix[4]arene-benzocrown-bisglycolamide (6)

CBCDA-13c

Sample Name:
CBCDA-13c
Data Collected on:
balaji-vnmr600
Archive directory:

Sample directory:

FidFile: CARBON

Pulse Sequence: CARBON (s2pul)
Solvent: cdcl3
Data collected on: Dec 4 2013

INDEX	FREQUENCY	PPM	HEIGHT
1	25571.7	169.531	24.8
2	23606.4	156.502	23.3
3	23565.9	156.234	23.8
4	22482.6	149.051	20.9
5	20201.4	133.928	37.7
6	20186.9	133.832	36.1
7	19597.1	129.922	47.9
8	19578.1	129.796	40.4
9	18423.7	122.143	27.8
10	18412.3	122.067	26.2
11	18377.2	121.834	33.5
12	17395.3	115.325	26.1
13	11634.3	77.132	170.8
14	11603.1	76.924	169.8
15	11571.0	76.712	171.6
16	10591.4	70.217	38.8
17	10575.4	70.111	44.1
18	10565.5	70.045	41.5
19	10499.1	69.605	31.2
20	10361.0	68.690	36.6
21	10265.6	68.057	21.8
22	10259.5	68.017	29.9
23	8155.3	54.067	36.7
24	7883.0	52.261	39.2
25	5826.1	38.625	19.6
26	5672.7	37.608	41.8
27	4480.3	29.702	33.1
28	4461.9	29.581	31.3
29	4346.7	28.817	20.0
30	4111.0	27.254	39.7
31	3937.0	26.101	43.7
32	3449.5	22.869	19.4
33	3025.3	20.057	53.2
34	3013.1	19.976	47.6

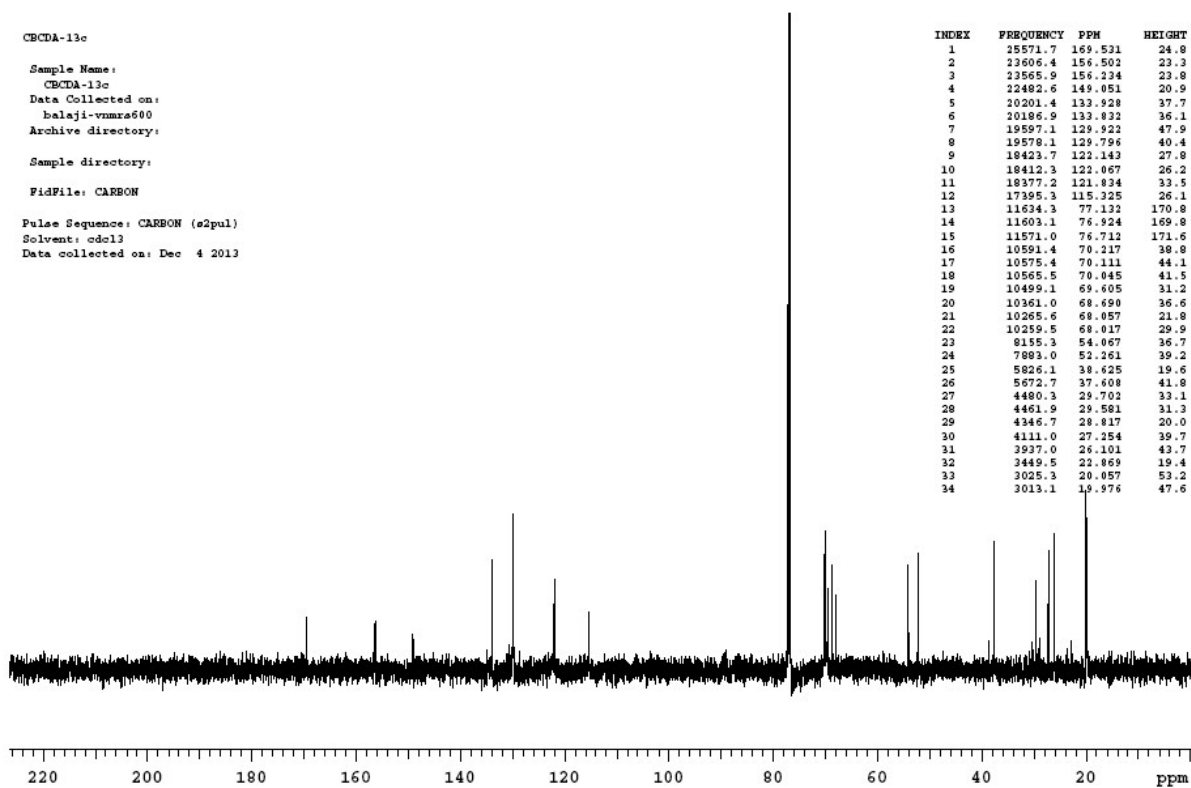


Figure S9: ESI-MS spectra of 1,3- alternate Calix[4]arene-benzocrown-bisglycolamide (6)

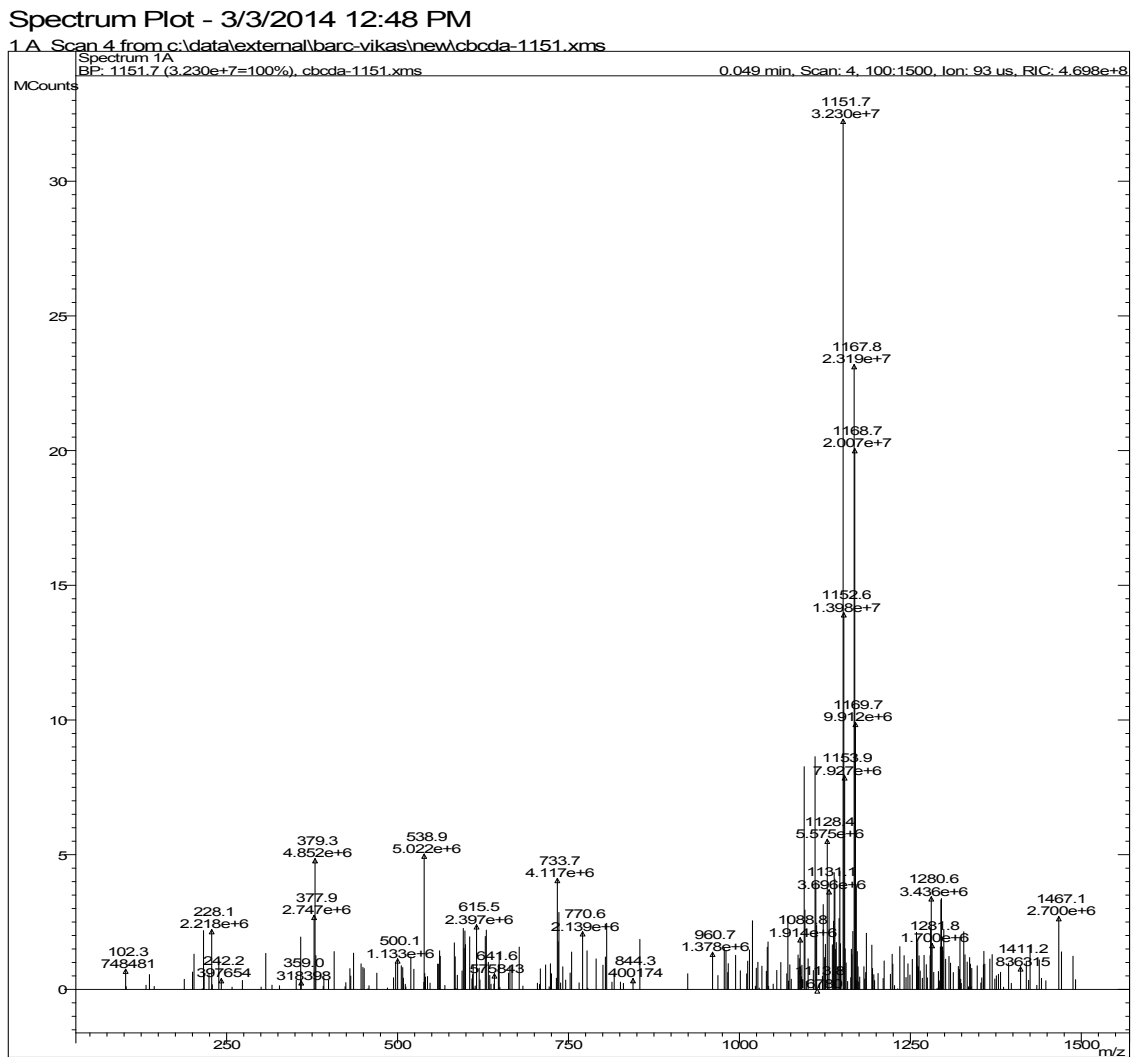


Figure S10: ESI-MS of CBCBGA complexes with Cesium

Spectrum Plot - 1/6/2014 11:41 AM

1 A Scan 37 from c:\data\external\barc-vikas\new\3-complex-1261.xms

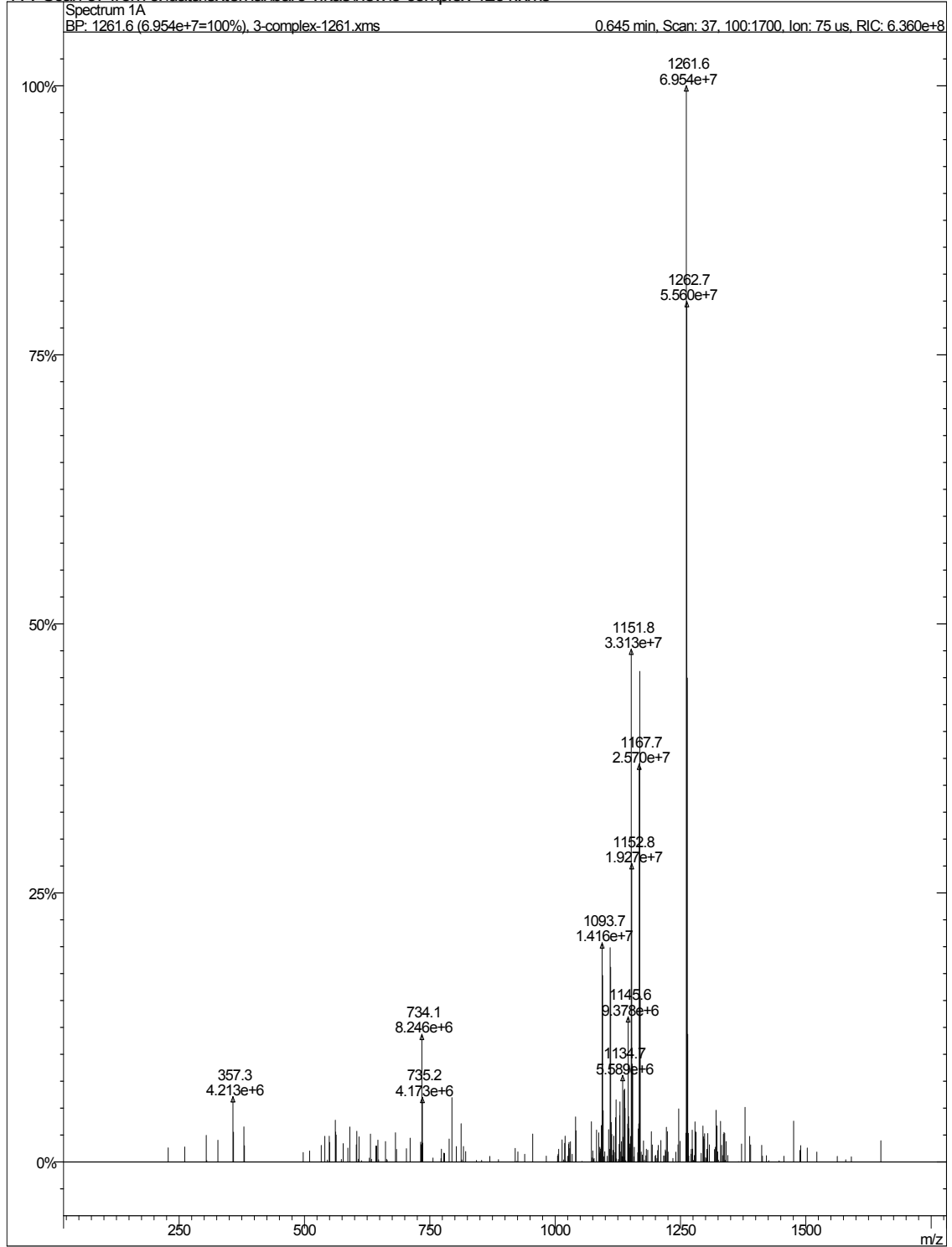


Table S1: Computed values of electron density and Laplacian of electron density and ellipsity of $\text{MNO}_3\text{-CBCBGA}$ and metal ion-hydrated complexes at the B3LYP/DZP level using Bader's AIM calculation.

Metal-ion complex	BCP	ρ	Ave. $\nabla^2\rho$	Ave. ϵ
$\text{Cs}^+\text{-CBCBGA}$	Cs-O1	0.0105	0.039	0.069
	Cs -O2	0.0056		
	Cs -O3	0.0076		
	Cs -O4	0.0103		
	Cs -O5	0.0104		
	Cs -O6	0.0078		
	Cs -O7	0.0111		
	Cs -O8	0.0130		
$\text{Na}^+\text{-CBCBGA}$	Na-O1	0.0105	0.079	0.062
	Na -O2	0.0041		
	Na -O3	0.0091		
	Na -O4	0.0134		
	Na -O5	0.0276		
$\text{Cs}^+\text{-(H}_2\text{O)}_8$	Cs-O1	0.0082	0.031	0.078
	Cs -O2	0.0084		
	Cs -O3	0.0080		
	Cs -O4	0.0084		
	Cs -O5	0.0080		
	Cs -O6	0.0081		
	Cs -O7	0.0076		
	Cs -O8	0.0086		
$\text{Na}^+\text{-(H}_2\text{O)}_6$	Na-O1	0.015	0.096	0.055
	Na -O2	0.015		
	Na -O3	0.015		
	Na -O4	0.015		
	Na -O5	0.015		
	Na -O6	0.015		

Table S2: Calculated values of average second order stabilization energies $E_{ij}^{(2)}$ using NBO analysis at the B3LYP/DZP level of theory.

Donar nbo (i)	Acceptor nbo (j)	$E(2)$ (kcal/mol)
LP(1)O1	LV(1)Cs179	0.68
LP(1)O3	LV(1)Cs179	2.02
LP(2)O3	LV(1)Cs179	0.32
LP(1)O5	LV(1)Cs179	1.02
LP(2)O5	LV(1)Cs179	0.10
LP(1)O6	LV(1)Cs179	2.81
LP(2)O6	LV(1)Cs179	0.22
LP(1)O7	LV(1)Cs179	0.91
LP(2)O7	LV(1)Cs179	0.48
LP(1)O8	LV(1)Cs179	0.94
LP(2)O8	LV(1)Cs179	0.07
LP(1)O13	LV(1)Cs179	1.82
LP(1)O14	LV(1)Cs179	6.49
LP(2)O14	LV(1)Cs179	0.36
LP(3)O14	LV(1)Cs179	0.30
LP(1)O15	LV(1)Cs179	6.03
LP(2)O15	LV(1)Cs179	0.47
LP(3)O15	LV(1)Cs179	0.56
Donar nbo (i)	Acceptor nbo (j)	$E(2)$ (kcal/mol)
LP(1)O69	LV(1)Na87	1.00
LP(2)O69	LV(1)Na87	0.14
LP(1)O70	LV(1)Na87	2.27
LP(1)O71	LV(1)Na87	0.67
LP(2)O71	LV(1)Na87	0.23
LP(1)O74	LV(1)Na87	1.42
LP(2)O74	LV(1)Na87	0.07
LP(1)O75	LV(1)Na87	2.00
LP(1)O76	LV(1)Na87	0.84
LP(1)O81	LV(1)Na87	3.91

LP(2)O81	LV(1)Na87	0.12
LP(1)O82	LV(1)Na87	2.93
LP(2)O82	LV(1)Na87	0.09
LP(1)O83	LV(1)Na87	1.67
