Supporting Information

Core/shell Ag@silicate nanoplatelets and poly(vinyl alcohol)

spherical nanohybrids fabricated by coaxial electrospraying as

highly sensitive SERS substrates

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FIGURE CAPTIONS

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Fig. S2 UV-Vis absorption spectra of the colloidal AgNPs in AgNO₃/copolymer/silicate reductions with a weight ratio of 1:1:0. Inset: 0.005 wt% yellow-gold solution confirming AgNP formation following treatment at 80 °C for 9 h.

Fig. S3 UV-Vis absorption spectra of the colloidal AgNPs in AgNO₃/copolymer/silicate reductions with a weight ratio of 1:1:1. Inset: 0.005 wt% yellow-gold solution confirming AgNP formation following treatment at 80 °C for 10 h.

Fig. S4 UV-Vis absorption spectra of the colloidal AgNPs in AgNO₃/copolymer/silicate reductions with a weight ratio of 10:10:1. Inset: 0.005 wt% yellow-gold solution confirming AgNP formation following treatment at 80 °C for 8 h.

Fig. S5 UV-Vis absorption spectra of the colloidal AgNPs in AgNO₃/copolymer/silicate reductions with a weight ratio of 20:20:1. Inset: 0.005 wt% yellow-gold solution confirming AgNP formation following treatment at 80 °C for 8 h.

TABLE LIST

Table S1. Solubility of PIB-SA, POE-2000, and tri-block copolymer in water and organic solvents.

VIDEO LIST

Video S1. A movie demonstrating the Taylor cone from hybrid liquid droplets in the electrospray process is provided. The liquid body of the inner needle PVA solution and the outer needle Ag@silicate

hybrids shows a conical shape, referred to as the Taylor cone, at 25 KV threshold voltage, with a half angle of 49.8°.



Scheme S1. Synthesis of copolymer as organic dispersant.



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Sample	H ₂ O	Ethanol	Toluene	Decane
PIB-SA	-	-	+	+
POE-2000	+	+	+-	-
Tri-block copolymer	+	+	+	+

Table S1. Solubility of PIB-SA, POE-2000, and tri-block copolymer in water and organic solvents.

+: soluble; +-: soluble, but sediments form after 2 h settling; -: insoluble.