

## Supplementary data

**Table A.** Th electrophoretic mobility variation as function of DTPA, conditions of separation  $[H^+] = 0.040$  M,  $I = 0.1$  M  $NaNO_3$ ,  $T = 25-26$  °C,  $V = +10$  kV. DMF is used as electroosmotic flow marker.

$C_{DTPA}$ (M) + $C_{NTA} = 10^{-6}$ M	$\mu$ ( $\times 10^4$ cm <sup>2</sup> V <sup>-1</sup> s <sup>-1</sup> )	$C_{DTPA}$ (M) + $C_{NTA} = 0$	$\mu$ ( $\times 10^4$ cm <sup>2</sup> V <sup>-1</sup> s <sup>-1</sup> )
0	+1.105*	0	+2.561*
$10^{-8}$	+1.103	0	+2.352
$10^{-5}$	+0.225	$10^{-8}$	+2.546
$5 \times 10^{-5}$	-0.416	$10^{-7}$	+2.444
$10^{-4}$	-0.662	$10^{-6}$	-0.808
$5 \times 10^{-4}$	-0.976	$10^{-6}$	-1.021
$10^{-3}$	-0.882	$5 \times 10^{-4}$	-0.939
$5 \times 10^{-3}$	-0.975	$10^{-4}$	-0.834
$10^{-2}$	-0.965	$10^{-3}$	-0.936
		$8 \times 10^{-3}$	-0.948

\* Remark: at  $C_{DTPA} = 0$  and  $C_{NTA} = 10^{-6}$  M, the main Th species is  $ThNTA^+$  ( $\mu = +1.105 \times 10^4$  cm<sup>2</sup>V<sup>-1</sup>s<sup>-1</sup>) whereas for  $C_{DTPA} = C_{NTA} = 0$ , the main species is  $Th^{4+}$  ( $\mu = +2.561 \times 10^4$  cm<sup>2</sup>V<sup>-1</sup>s<sup>-1</sup>). The limiting  $\mu$  value at high DTPA concentration ( $\approx -0.96 \times 10^4$  cm<sup>2</sup>V<sup>-1</sup>s<sup>-1</sup>) corresponds to the complex  $ThDTPA^-$ .

**Table B.** Pu relative area variation as function of of DTPA, conditions of separation  $[H^+] = 0.040$  M,  $I = 0.1$  M  $NaNO_3$ ,  $T = 25-26$  °C,  $V = +10$  kV.

pH with $C_{DTPA} = 10^{-4}$ M	Relative area PuDTPA <sup>-</sup> (%)	Relative area Pu(NTA) <sub>2</sub> <sup>2-</sup> (%)
$10^{-7}$	14.42	85.58
$5 \times 10^{-7}$	17.35	82.65
$10^{-6}$	26.09	73.91
$5 \times 10^{-5}$	55.22	44.78
$10^{-5}$	38.23	61.77
$10^{-4}$	60.02	39.98
$5 \times 10^{-4}$	79.13	20.87

**Table C.** Th electrophoretic mobility variation as function of pH, conditions of separation  $C_{DTPA} = 10^{-4}$  and  $10^{-2}$  M,  $I = 0.1$  M TMAX (X = MES or HEPES),  $T = 25-26$  °C,  $V = +10$  kV. DMF is used as electroosmotic flow marker.

pH with $C_{DTPA} = 10^{-4}$ M	$\mu$ ( $\times 10^4$ cm <sup>2</sup> V <sup>-1</sup> s <sup>-1</sup> )	pH with $C_{DTPA} = 10^{-2}$ M	$\mu$ ( $\times 10^4$ cm <sup>2</sup> V <sup>-1</sup> s <sup>-1</sup> )
5.978	-2.365	6.045	-1.559
6.455	-2.452	6.503	-1.581
6.995	-2.387	7.114	-1.514
7.565	-2.477	7.488	-1.514
8.451	-3.169	8.091	-1.568
9.377	-3.452	8.548	-2.1817
		9.612	-2.6952

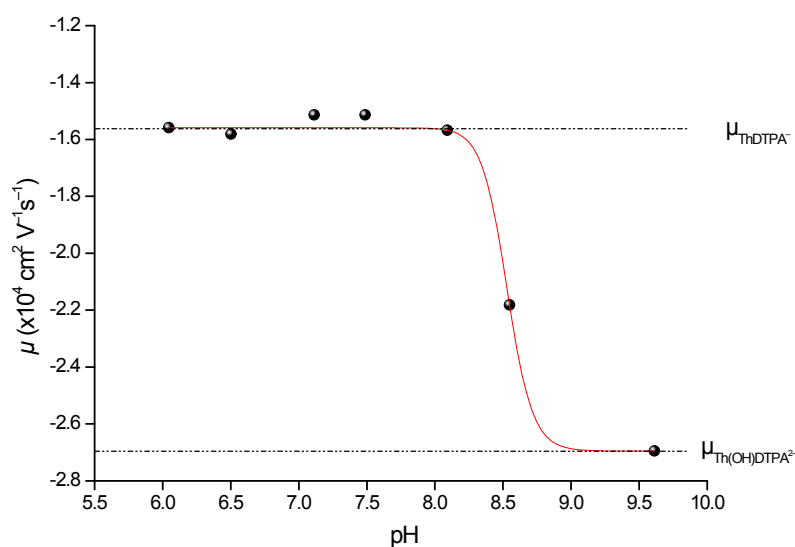


Figure C. example of the variation of the electrophoretic mobility of Th species as function of pH at  $C_{DTPA} = 10^{-2}$  M, 0.1 M TMAX (X = MES or HEPES), T = 25 °C.

Table D. Np electrophoretic mobility variation as function of pH, conditions of separation  $C_{DTPA} = 10^{-4}$  and  $10^{-2}$  M, I = 0.1 M TMA/(MES or HEPES), T = 25-26 °C, V = +10 kV. DMF is used as electroosmotic flow marker.

pH with $C_{DTPA} = 10^{-4}$ M	$\mu$ ( $\times 10^4$ cm $^2$ V $^{-1}$ s $^{-1}$ )	pH with $C_{DTPA} = 10^{-2}$ M	$\mu$ ( $\times 10^4$ cm $^2$ V $^{-1}$ s $^{-1}$ )
5.978	-2.459	6.045	-1.383
6.455	-2.529	6.503	-1.477
6.995	-2.477	7.114	-1.577
7.565	-2.545	7.488	-1.711
7.995	-3.046	8.091	-1.862
8.451	-3.221	8.548	-2.051
9.377	-3.327	9.612	-2.167

Table E. Pu relative area variation as function of pH, conditions of separation  $C_{DTPA} = 10^{-4}$  and  $10^{-2}$  M, I = 0.1 M TMA/(MES or HEPES), T = 25-26 °C, V = +10 kV. DMF is used as electroosmotic flow marker.

pH with $C_{DTPA} = 10^{-4}$ M	Relative area Pu(OH)DTPA $^{2-}$ (%)	Relative area PuDTPA $^{-}$ (%)	pH with $C_{DTPA} = 10^{-2}$ M	Relative area Pu(OH)DTPA $^{2-}$ (%)	Relative area PuDTPA $^{-}$ (%)
5.047	1.97	98.03	6.045	14.87	85.13
6.070	20.63	79.37	6.503	29.75	70.25
7.466	53.57	46.43	7.114	53.05	46.95
7.806	56.65	43.35	7.488	63.51	36.49
8.021	56.60	43.40	8.091	65.39	34.61
8.261	98.88	98.02	8.548	63.63	36.37
8.674	75.23	24.77	9.612	77.67	22.33

**Table F. variations of  $\mu_{Th}$  and  $\mu_{Pu}$  as function of the total concentration of nitrilotriacetic acid (NTA) at  $I = 0.1 \text{ M}$  ( $\text{NaNO}_3$ ),  $T = 25 \text{ }^\circ\text{C}$ , and  $[\text{H}^+] = 0.040 \text{ M}$ ,  $V = +10 \text{ kV}$ . DMF is used as electroosmotic flow marker.**

$C_{NTA} \text{ (M)}$	$\mu_{Th} (\times 10^4 \text{ cm}^2\text{V}^{-1}\text{s}^{-1})$	$C_{NTA} \text{ (M)}$	$\mu_{Pu} (\times 10^4 \text{ cm}^2\text{V}^{-1}\text{s}^{-1})$
$10^{-10}$	2.565	$10^{-8}$	2.742
$10^{-10}$	3.074	$10^{-8}$	2.754
$10^{-10}$	3.146	$10^{-7}$	2.559
$10^{-8}$	2.742	$10^{-7}$	2.999
$10^{-8}$	2.832	$5 \times 10^{-7}$	2.618
$10^{-8}$	2.964	$10^{-6}$	0.922
$10^{-7}$	2.756	$10^{-6}$	0.331
$10^{-7}$	3.094	$5 \times 10^{-6}$	-0.373
$5 \times 10^{-7}$	2.442	$10^{-5}$	-0.426
$10^{-6}$	2.655	$5 \times 10^{-5}$	-1.726
$10^{-6}$	2.906	$10^{-4}$	-1.991
$10^{-6}$	2.404	$10^{-4}$	-2.345
$5 \times 10^{-6}$	2.765	$10^{-3}$	-2.153
$10^{-5}$	2.615	$10^{-3}$	-2.418
$10^{-5}$	2.391	$5 \times 10^{-3}$	-2.417
$5 \times 10^{-5}$	1.943	$8 \times 10^{-3}$	-2.125
$10^{-4}$	1.712		
$10^{-4}$	1.732		
$10^{-4}$	1.449		
$2 \times 10^{-4}$	1.105		
$5 \times 10^{-4}$	-0.477		
$10^{-3}$	-0.898		
$10^{-3}$	0.048		
$10^{-3}$	-1.360		
$5 \times 10^{-3}$	-2.410		
$8 \times 10^{-3}$	-2.371		