

## Supporting Information

### Palladium-Catalyzed Direct C-H Arylation of Ferrocenecarboxamides with Aryl Halides

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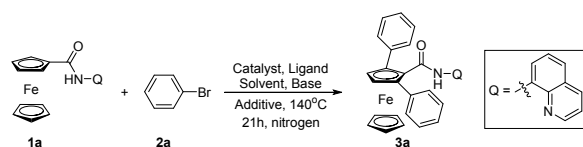
## 1. Preparation of Substrates

All amides were prepared from the corresponding ferrocenecarboxylic acid and amines according to the reported procedure.<sup>[1]</sup>

## 2. Optimization of Reaction Conditions

A 25 mL schlenk tube was equipped with a magnetic stir bar and charged with **1a** (71.2 mg, 0.2 mmol), **2a** (63  $\mu$ L, 0.6 mmol, 3 equiv), base (0.4 mmol, 2 equiv), catalyst (0.02 mmol, 10 mol %), ligand (0.02 mmol, 10 mol %), PivOH (6.2mg, 0.06 mmol, 30 mol %) in solvent (1.0 mL). The resulting mixture was heated under nitrogen at 140 °C for 21 h, and cooled to room temperature. Upon completion, CH<sub>2</sub>Cl<sub>2</sub> (20 mL) was added to the reaction system, and the resulting mixture was filtered through a pad of Celite. The filtrate was extracted with H<sub>2</sub>O (20 mL), and the aqueous layer was extracted with CH<sub>2</sub>Cl<sub>2</sub> (2  $\times$  10 mL). The combined organic layer was dried over anhydrous Na<sub>2</sub>SO<sub>4</sub> and filtered. After evaporation of the solvent under vacuum, the residue was purified by column chromatography on silica gel (100–200 mesh) using hexane-EtOAc as an eluent to afford the pure product **3a**.

**Table S1 Screening of Reaction Conditions<sup>a</sup>**



Entry	Solvent	Base	Catalyst	Ligand	Additive	Yield(%) <sup>[b]</sup>
1	toluene	K <sub>2</sub> CO <sub>3</sub>	Pd(OAc) <sub>2</sub>	PPh <sub>3</sub>	PivOH	8
2	DCE	K <sub>2</sub> CO <sub>3</sub>	Pd(OAc) <sub>2</sub>	PPh <sub>3</sub>	PivOH	Trace
3	dioxane	K <sub>2</sub> CO <sub>3</sub>	Pd(OAc) <sub>2</sub>	PPh <sub>3</sub>	PivOH	Trace
4	CH <sub>3</sub> CN	K <sub>2</sub> CO <sub>3</sub>	Pd(OAc) <sub>2</sub>	PPh <sub>3</sub>	PivOH	Trace
5	DMF	K <sub>2</sub> CO <sub>3</sub>	Pd(OAc) <sub>2</sub>	PPh <sub>3</sub>	PivOH	Trace
6	<i>o</i> -xylene	K <sub>2</sub> CO <sub>3</sub>	Pd(OAc) <sub>2</sub>	PPh <sub>3</sub>	PivOH	30
7	<i>o</i> -xylene	Na <sub>2</sub> CO <sub>3</sub>	Pd(OAc) <sub>2</sub>	PPh <sub>3</sub>	PivOH	20
8	<i>o</i> -xylene	Cs <sub>2</sub> CO <sub>3</sub>	Pd(OAc) <sub>2</sub>	PPh <sub>3</sub>	PivOH	Trace
9	<i>o</i> -xylene	K <sub>3</sub> PO <sub>4</sub>	Pd(OAc) <sub>2</sub>	PPh <sub>3</sub>	PivOH	18
10	<i>o</i> -xylene	KHCO <sub>3</sub>	Pd(OAc) <sub>2</sub>	PPh <sub>3</sub>	PivOH	Trace
11	<i>o</i> -xylene	K <sub>2</sub> CO <sub>3</sub>	-	PPh <sub>3</sub>	PivOH	Trace
12	<i>o</i> -xylene	K <sub>2</sub> CO <sub>3</sub>	PdCl <sub>2</sub>	PPh <sub>3</sub>	PivOH	11
13	<i>o</i> -xylene	K <sub>2</sub> CO <sub>3</sub>	Pd <sub>2</sub> dba <sub>3</sub>	PPh <sub>3</sub>	PivOH	17
14	<i>o</i> -xylene	K <sub>2</sub> CO <sub>3</sub>	Pd(CF <sub>3</sub> COO) <sub>2</sub>	PPh <sub>3</sub>	PivOH	10
15	<i>o</i> -xylene	K <sub>2</sub> CO <sub>3</sub>	Ni(OAc) <sub>2</sub>	PPh <sub>3</sub>	PivOH	Trace
16	<i>o</i> -xylene	K <sub>2</sub> CO <sub>3</sub>	[RuCl <sub>2</sub> (cymene)] <sub>2</sub>	PPh <sub>3</sub>	PivOH	Trace
17	<i>o</i> -xylene	K <sub>2</sub> CO <sub>3</sub>	Pd(OAc) <sub>2</sub>	PPy <sub>3</sub>	PivOH	23
18	<i>o</i> -xylene	K <sub>2</sub> CO <sub>3</sub>	Pd(OAc) <sub>2</sub>	DPPF	PivOH	49
19	<i>o</i> -xylene	K <sub>2</sub> CO <sub>3</sub>	Pd(OAc) <sub>2</sub>	RuPhos	PivOH	61

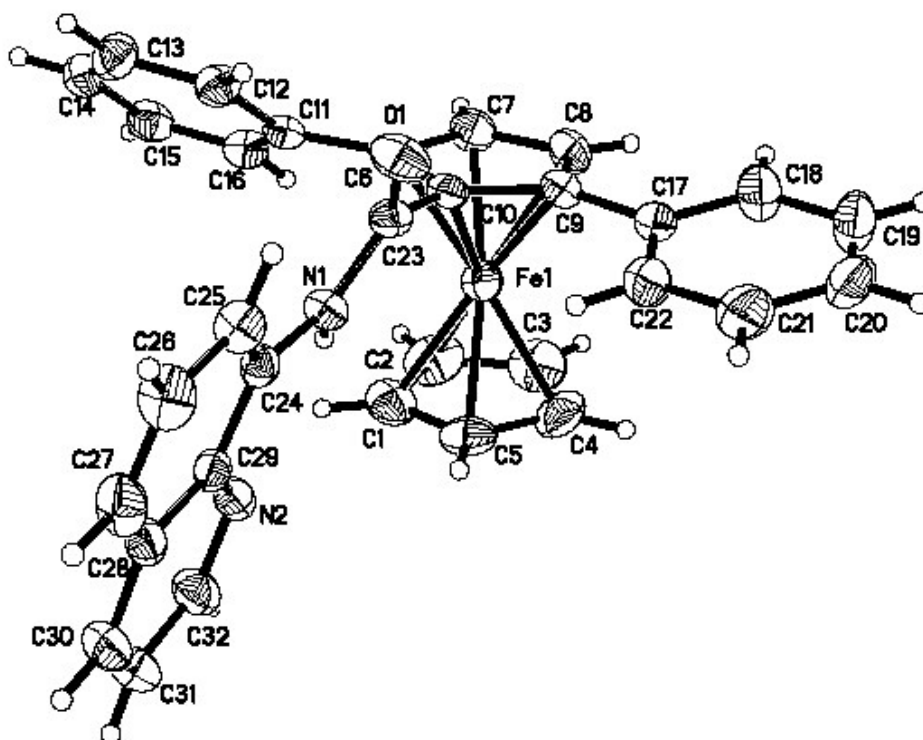
20	<i>o</i> -xylene	K <sub>2</sub> CO <sub>3</sub>	Pd(OAc) <sub>2</sub>	XPhos	PivOH	83
21	<i>o</i> -xylene	K <sub>2</sub> CO <sub>3</sub>	Pd(OAc) <sub>2</sub>	XantPhos	PivOH	42
22	<i>o</i> -xylene	K <sub>2</sub> CO <sub>3</sub>	Pd(OAc) <sub>2</sub>	<sup>t</sup> BuXPhos	PivOH	60
23	<i>o</i> -xylene	K <sub>2</sub> CO <sub>3</sub>	Pd(OAc) <sub>2</sub>	XPhos	AcOH	61
24	<i>o</i> -xylene	K <sub>2</sub> CO <sub>3</sub>	Pd(OAc) <sub>2</sub>	XPhos	AcOK	75
25	<i>o</i> -xylene	K <sub>2</sub> CO <sub>3</sub>	Pd(OAc) <sub>2</sub>	XPhos	PhCOOH	74
26	<i>o</i> -xylene	K <sub>2</sub> CO <sub>3</sub>	Pd(OAc) <sub>2</sub>	XPhos	-	72
27 <sup>[c]</sup>	<i>o</i> -xylene	K <sub>2</sub> CO <sub>3</sub>	Pd(OAc) <sub>2</sub>	XPhos	PivOH	82
28 <sup>[d]</sup>	<i>o</i> -xylene	K <sub>2</sub> CO <sub>3</sub>	Pd(OAc) <sub>2</sub>	XPhos	PivOH	51

<sup>a</sup> Reaction conditions: **1a** (0.2 mmol), **2a** (0.6 mmol), Catalyst (0.02 mmol), Ligand (0.02 mmol), PivOH (0.06 mmol), Base (0.4 mmol) and Solvent (1.0 mL) under nitrogen at 140 °C for 21 h unless otherwise noted. <sup>b</sup> Isolated yield based on **1a**. <sup>c</sup> Without PivOH. <sup>d</sup> At 150 °C. <sup>e</sup> At 130 °C.

### 3. References

1. (a) L. D. Tran, J. Roane and O. Daugulis, *Angew. Chem., Int. Ed.* 2013, **52**, 6043; (b) T. Truong, K. Klimovica and O. Daugulis, *J. Am. Chem. Soc.* 2013, **135**, 9342.

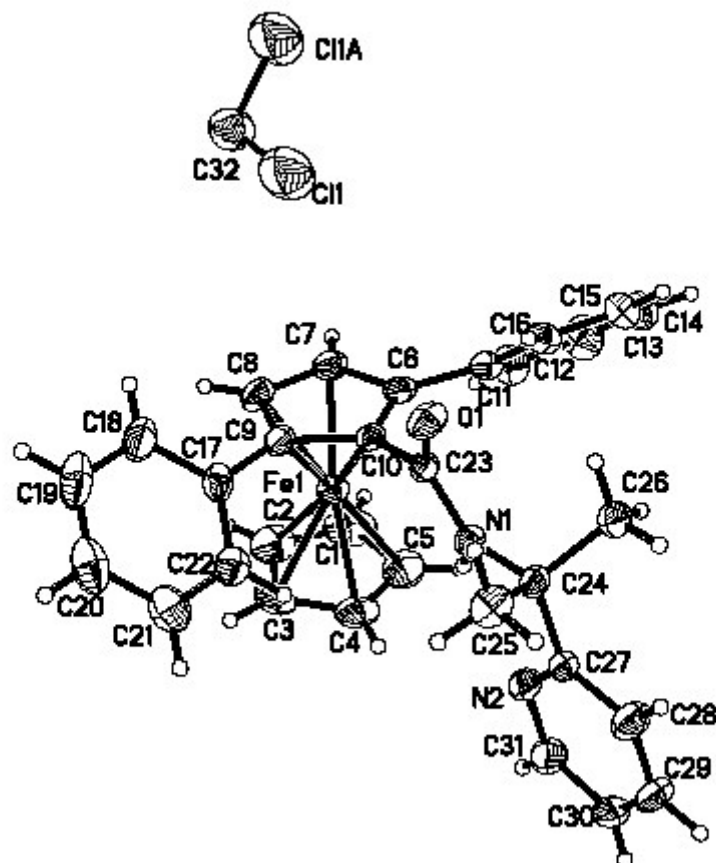
### 4. The Single Crystal X-ray Diffraction Study of **3a** and **5a**



CCDC 1446343 (**3a**) contains the supplementary crystallographic data for this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre *via* [www.ccdc.cam.ac.uk/data\\_request/cif](http://www.ccdc.cam.ac.uk/data_request/cif).

**Table S2 Crystal dData and Structure Refinement for 3a.**

Identification code	201506168
Empirical formula	C <sub>32</sub> H <sub>24</sub> FeN <sub>2</sub> O
Formula weight	508.38
Temperature/K	291.15
Crystal system	monoclinic
Space group	P2 <sub>1</sub> /c
a/Å	11.3423(6)
b/Å	14.2822(4)
c/Å	16.0589(7)
α/°	90
β/°	107.980(5)
γ/°	90
Volume/Å <sup>3</sup>	2474.4(2)
Z	4
ρ <sub>calc</sub> /cm <sup>3</sup>	1.365
μ/mm <sup>-1</sup>	5.099
F(000)	1056.0
Crystal size/mm <sup>3</sup>	0.2 × 0.2 × 0.16
Radiation	CuKα (λ = 1.54184)
2θ range for data collection/°	8.196 to 134.122
Index ranges	-13 ≤ h ≤ 13, -17 ≤ k ≤ 14, -19 ≤ l ≤ 13
Reflections collected	9066
Independent reflections	4408 [R <sub>int</sub> = 0.0267, R <sub>sigma</sub> = 0.0364]
Data/restraints/parameters	4408/13/325
Goodness-of-fit on F <sup>2</sup>	1.018
Final R indexes [I >= 2σ (I)]	R <sub>1</sub> = 0.0401, wR <sub>2</sub> = 0.0962
Final R indexes [all data]	R <sub>1</sub> = 0.0531, wR <sub>2</sub> = 0.1034
Largest diff. peak/hole / e Å <sup>-3</sup>	0.32/-0.21



CCDC 1446344 (**5a**) contains the supplementary crystallographic data for this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre *via* [www.ccdc.cam.ac.uk/data\\_request/cif](http://www.ccdc.cam.ac.uk/data_request/cif).

**Table S3 Crystal Data and Structure Refinement for 5a.**

Identification code	201506167
Empirical formula	C <sub>63</sub> H <sub>58</sub> Cl <sub>2</sub> Fe <sub>2</sub> N <sub>4</sub> O <sub>2</sub>
Formula weight	1085.73
Temperature/K	291.15
Crystal system	monoclinic
Space group	P2 <sub>1</sub> /c
a/Å	14.7314(2)
b/Å	10.30632(19)
c/Å	17.8542(4)
α/°	90
β/°	95.6050(17)
γ/°	90
Volume/Å <sup>3</sup>	2697.78(9)
Z	2
ρ <sub>calc</sub> /cm <sup>3</sup>	1.337

$\mu/\text{mm}^{-1}$	5.596
F(000)	1132.0
Crystal size/ $\text{mm}^3$	$0.22 \times 0.2 \times 0.17$
Radiation	CuK $\alpha$ ( $\lambda = 1.54184$ )
$2\Theta$ range for data collection/ $^\circ$	6.028 to 134.156
Index ranges	$-17 \leq h \leq 17, -12 \leq k \leq 6, -21 \leq l \leq 21$
Reflections collected	9723
Independent reflections	4818 [ $R_{\text{int}} = 0.0292, R_{\text{sigma}} = 0.0363$ ]
Data/restraints/parameters	4818/4/340
Goodness-of-fit on $F^2$	1.053
Final R indexes [ $I \geq 2\sigma(I)$ ]	$R_1 = 0.0497, wR_2 = 0.1388$
Final R indexes [all data]	$R_1 = 0.0625, wR_2 = 0.1497$
Largest diff. peak/hole / $e \text{ \AA}^{-3}$	0.68/-0.64

## 5. Copies of NMR Spectra for the Products

