

## Supporting Information

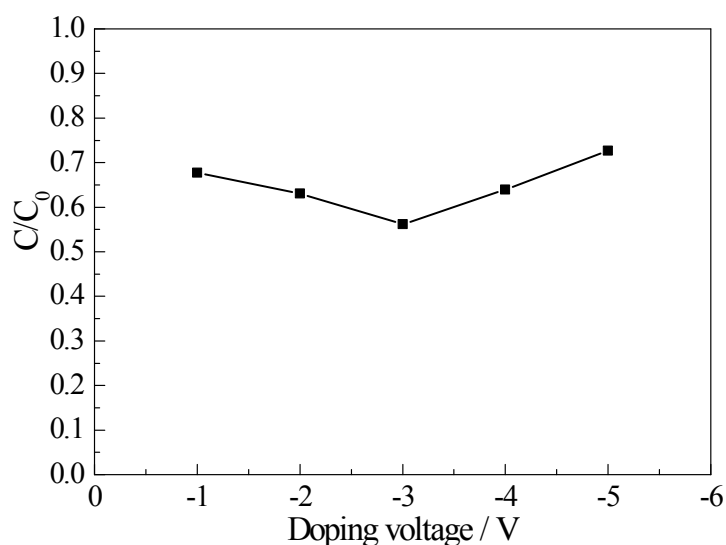
### UV-light Aided Photoelectrochemical Synthesis of Au/TiO<sub>2</sub> NTs for

#### Photoelectrocatalytic Degradation of HPAM

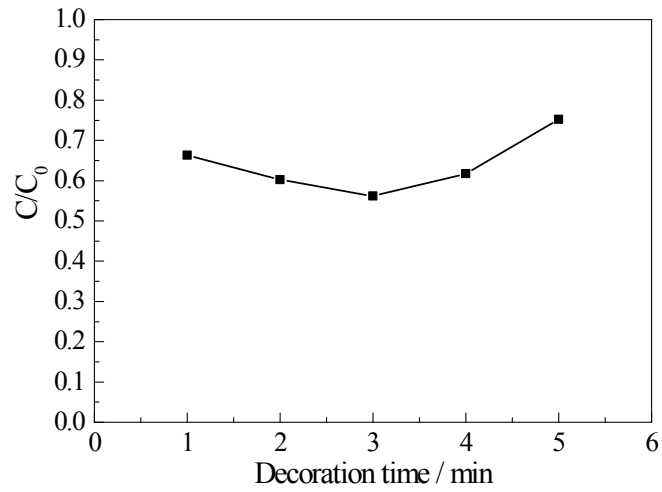
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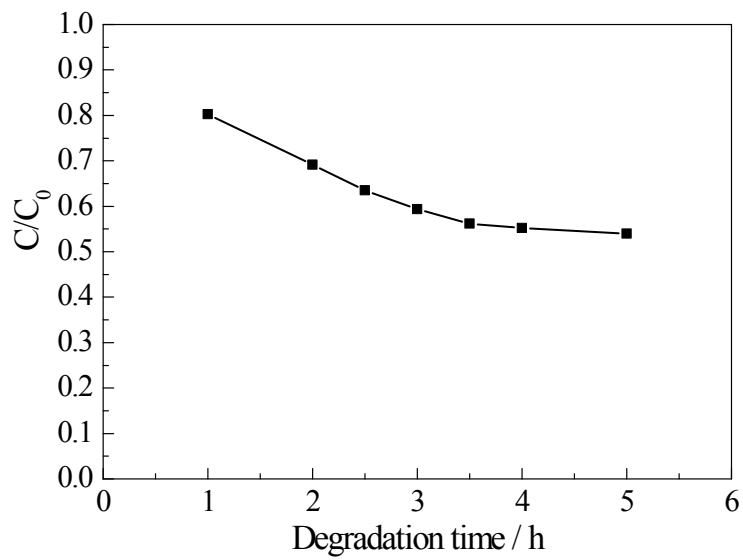
#### 1. Degradation rate of HPAM using Au/TiO<sub>2</sub> NTs with different conditions



**Fig. S1** Degradation rate of HPAM using Au/TiO<sub>2</sub> NTs with different  
doping voltage

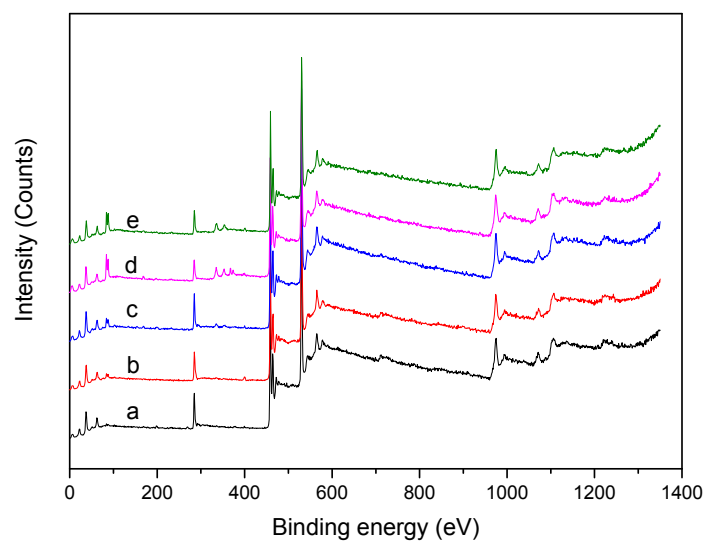


**Fig. S2** Degradation rate of HPAM using Au/TiO<sub>2</sub> NTs with different decoration time



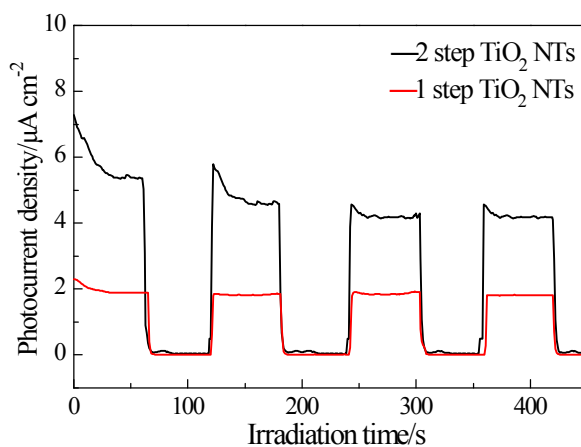
**Fig. S3** Degradation rate of HPAM using Au/TiO<sub>2</sub> NTs with different degradation time

**2. XPS of Au/TiO<sub>2</sub> NTs immersed with different concentrations of H<sub>2</sub>AuCl<sub>4</sub>**



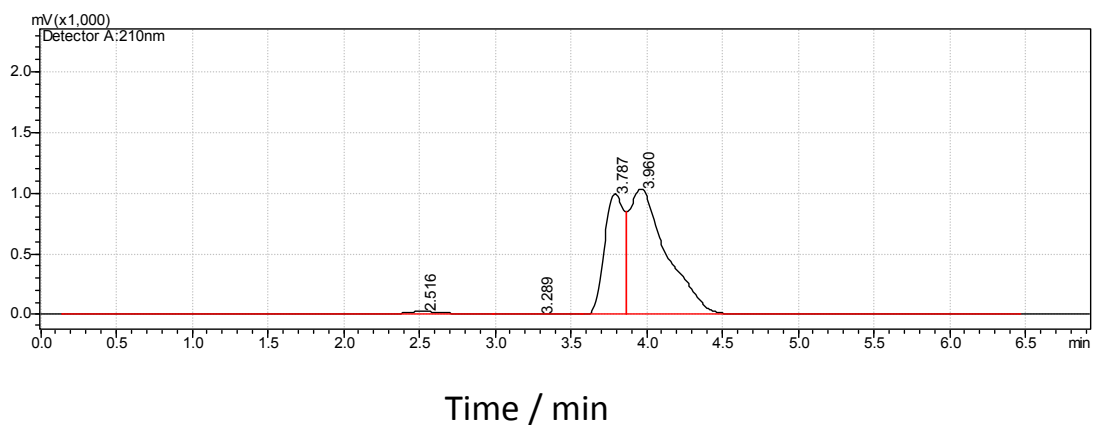
**Fig. S4** XPS of Au/TiO<sub>2</sub> NTs immersed with different concentrations of HAuCl<sub>4</sub>: curve a to e response to the concentrations of HAuCl<sub>4</sub> 0.005 g L<sup>-1</sup> to 0.025 g L<sup>-1</sup>.

### 3. The photocurrent response of TiO<sub>2</sub> NTs

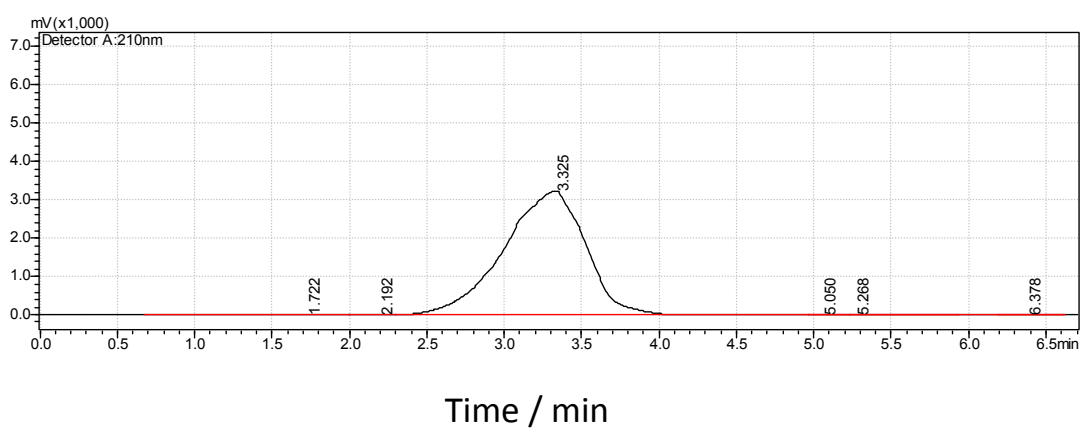


**Fig. S5** Amperometric I-t curves at an applied potential of 0.5V under illumination of 300 W high pressure mercury lamp with 60 s light on/off cycles.

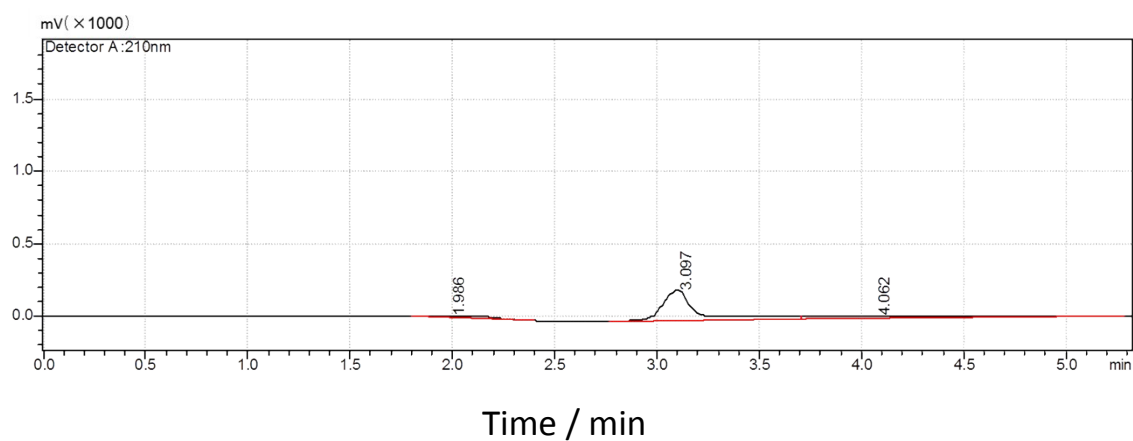
### 4. Product analysis of the HPAM degradation



**Fig. S6** HPLC of acrylamide standard sample



**Fig. S7** HPLC of acrylic acid standard sample



**Fig. S8** HPLC of HPAM degradation after 3.5h irradiation