

Composite PS-*b*-P4VP/Ag and PS-*b*-P4VP/Au thin films fabricated via a multilevel self-assembly process

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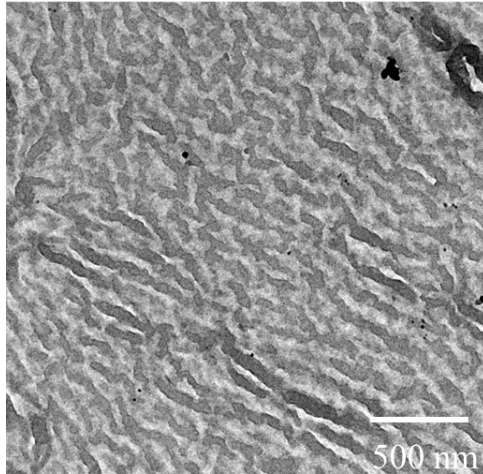


Fig. S1. TEM micrographs of thin films formed at the air/water interface: (a) $V_{\text{DMF}}/V_{\text{CHCl}_3} = 6/4$, the fabrication time is 8h, the concentrations of polymer and AgNO_3 are 0.2 mg mL^{-1} and 0.01 mol L^{-1} , respectively.

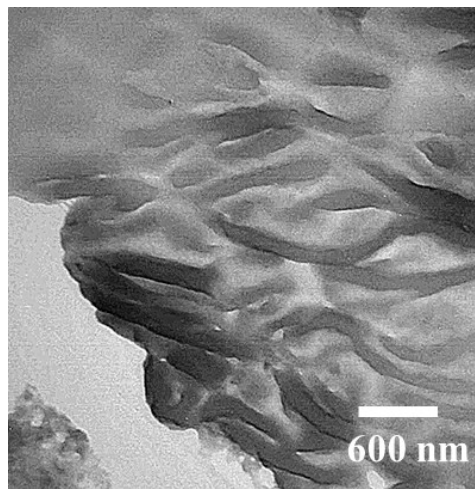


Figure S2. TEM micrographs of the aggregates formed in the upper-phase (the aqueous solution), HAuCl_4 aqueous solution with a concentration of 0.001 mol L^{-1} and $V_{\text{DMF}}/V_{\text{CHCl}_3} = 6/4$.

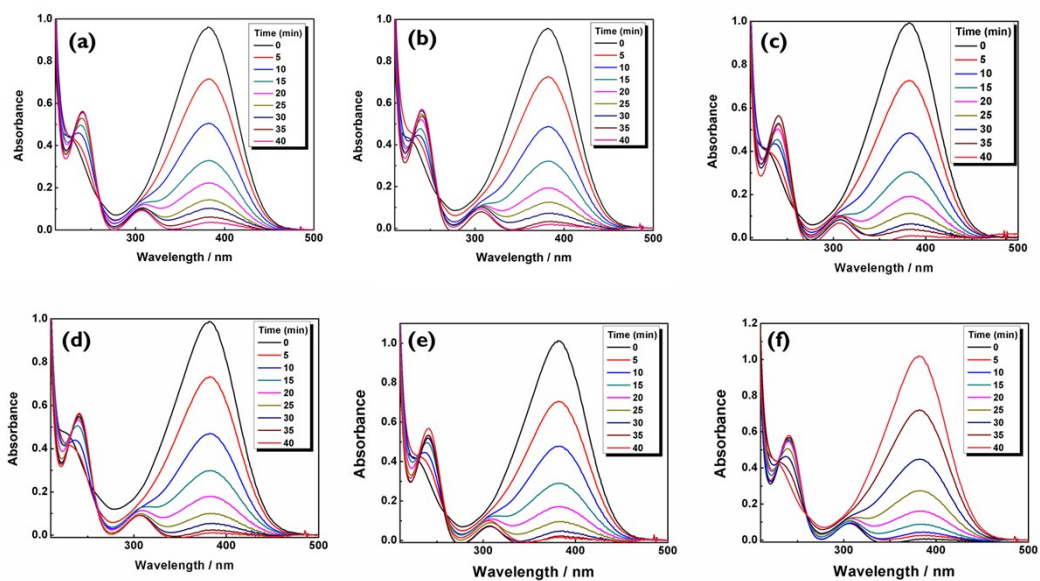


Figure S3. Catalytic reduction of 4-NA. Typical time-dependent UV-vis absorption spectra of the reaction solutions in the presence of composite thin films of PS-b-P4VP/Ag in six successive cycles (a-f).

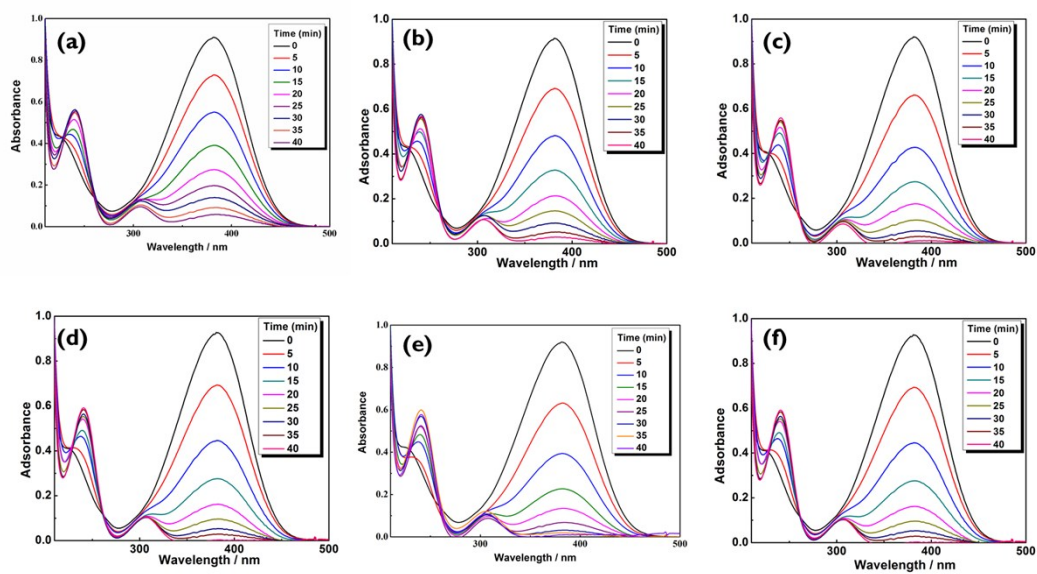


Figure S4. Catalytic reduction of 4-NA. Typical time-dependent UV-vis absorption spectra of the reaction solutions in the presence of composite thin films of PS-b-P4VP/Au in six successive cycles (a-f).