## SUPPLEMENTARY

TABLE S. The calculated O-O, C-O, C\*-O, N-O, C-C\*, and C\*-N bond lengths of the N-SWNTs<sup>Q</sup> for all three adsorption states in Å. Where C\* is one of the carbon atoms adjacent to N atom on which oxygen adsorption is considered.

N-SWNT	Physisorpt	ion state	Chemisor	ption state	Dissociativ	e adsorption		
Charge (Q)	0	-1	0	-1	0	-1		
(15,0)MZ	Diameter = 11.74 Å							
0-0	1.24	1.24	1.50	1.50	2.80	2.80		
C-0	3.67	3.58	1.52	1.52	1.45	1.46		
C*-0	3.45	3.94 (N-O 3.49)	1.47	1.47	1.43	1.43		
C-C*	1.41	1.41	1.53	1.53	1.49	1.49		
C*-N	1.41	1.41	1.48	1.48	1.465	1.465		
(18,0)MZ	Diameter	r = 14.09  Å						
0-0	1.24	1.24	1.50	1.50	2.67	2.67		
C-0	3.61	3.62	1.52	1.52	1.43	1.43		
C*-0	3.33	3.65 (N-O 3.45)	1.47	1.47	1.24	1.24		
C-C*	1.42	1.42	1.53	1.53	1.53	1.53		
C*-N	1.41	1.41	1.48	1.48	2.10	2.07		
(21,0)MZ	Diame	eter = 16.4406 Å						
0-0	1.24	1.24	1.50	1.50	2.79	2.79		
C-0	3.56	3.71	1.52	1.53	1.45	1.46		
<u>C*-O</u>	3.23	3.28	1.47	1.48	1.43	1.43		
C-C*	1.42	1.42	1.53	1.53	1.49	1.49		
C*-N	1.41	1.41	1.48	1.48	1.47	1.47		
(16,0)SZ	Diamet	ter = 12.53 Å						
0-0	1.25	1.25	1.50	1.50	2.73	2.68		
<u>C-O</u>	3.20	3.61	1.53	1.53	1.42	1.42		
<u>C*-O</u>	3.07	3.23	1.48	1.47	1.35	1.24		

C-C*	1.41	1.41	1.54	1.53	1.53	1.52				
C*-N	1.42	1.42	1.48	1.48	1.56	2.12				
(17,0)SZ	Diar	meter = 13.25 Å								
0-0	1.24	1.25	1.50	1.50	2.68	2.68				
C-0	3.76	3.61	1.52	1.53	1.42	1.43				
C*-0	3.76	4.14 (N-O 3.71)	1.47	1.48	1.24	1.24				
C-C*	1.41	1,42	1.53	1.53	1.53	1.53				
C*-N	1.40	1.41	1.48	1.48	2.11	2.10				
(19,0)SZ	Diameter = 14.91 Å									
0-0	1.24	1.24	1.50	1.50	2.80	2.80				
C-0	3.69	3.84	1.52	1.55	1.45	1.46				
C*-0	3.64	3.51	1.47	1.47	1.42	1.42				
C-C*	1.42	1.42	1.53	1.53	1.49	1.49				
C*-N	1.41	1.41	1.48	1.48	1.47	1.47				
(20,0)SZ	Diamet	ter = 15.66 Å								
0-0	1.24	1.25	1.50	1.50	2.79	2.79				
C-0	3.72	3.75	1.53	1.54	1.46	1.45				
<u>C*-0</u>	3.28	3.30	1.47	1.48	1.42	1.42				
C-C*	1.42	1.42	1.53	1.53	1.49	1.49				
C*-N	1.41	1.41	1.48	1.48	1.47	1.47				
(9,9)MA	Diamet	ter = 12.204 Å								
0-0	1.25	1.26	1.50	1.49	2.81	2.84				
C-0	3.32	3.07	1.52	1.59	1.45	1.45				
C*-0	3.23	2.96	1.47	1.50	1.25	1.28				
C-C*	1.41	1.42	1.57	1.56	1.56	1.57				
C*-N	1.41	1.41	1.46	1.45	1.88	1.72				
(10,10)MA	Diameter = $13.56$ Å									

0-0	1.25	1.26	1.50	1.49	3.02	2.83	
<u>C-O</u>	3.30	3.89	1.53	1.59	1.46	1.45	
<u>C*-O</u>	3.06	3.67	1.47	1.50	1.42	1.27	
C-C*	1.41	1.42	1.57	1.56	1.46	1.57	
C*-N	1.41	1.41	1.46	1.45	1.46	1.73	
(11,11)MA	Dian	neter = 14.916 Å					
0-0	1.25	1.28	1.50	1.49	3.04	2.81	
C-O	3.31	3.12	1.53	1.59	1.46	1.45	
<u>C*-0</u>	3.23	3.09 (N-O 2.92)	1.47	1.50	1.42	1.28	
C-C*	1.41	1.42	1.57	1.56	1.52	1.57	
C*-N	1.41	1.41	1.46	1.45	1.46	1.72	
(12,12)MA	Diam	neter = 16.272 Å					
0-0	1.25	1.26	1.50	1.49	2.76	2.80	
<u>C-0</u>	3.36	3.18	1.53	1.59	1.44	1.45	
C*-0	3.06	2.99	1.47	1.50	1.26	1.27	
C-C*	1.41	1.42	1.57	1.56	1.56	1.57	
C*-N	1.41	1.41	1.46	1.45	1.86	1.73	



FIG. S1. Local density of states of zigzag N-SWNTs<sup>(Q = -1 and 0)</sup> (a) all C atoms, and (b) a N atom. E<sub>F</sub> represents the Fermi level.



FIG. S2. Local density of states of armchair N-SWNTs<sup>(Q=-1 and 0)</sup> (a) all C atoms, and (b) a N atom. E<sub>F</sub> represents the Fermi