Support information

Importance of reagents addition order in contaminant degradation in Fe(II)/PMS system

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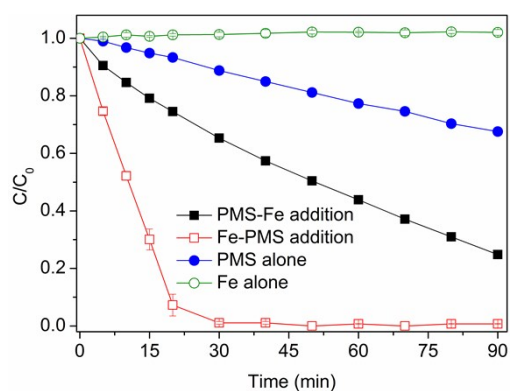*Corresponding author: Email: zhaohuiwang@dhu.edu.cn (Z. Wang).

Fig. S1 The degradation of MB as a function of reaction time in Fe(II)/PMS system.

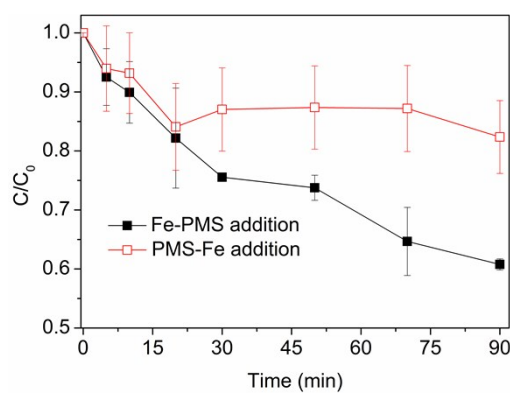Conditions: $[MB]_0$=0.05 mM; $[PMS]_0$=0.3 mM.

Fig. S2 The degradation of 4-Chloro-2-Nitrophenol (4C2NP) in Fe(II)/PMS system.

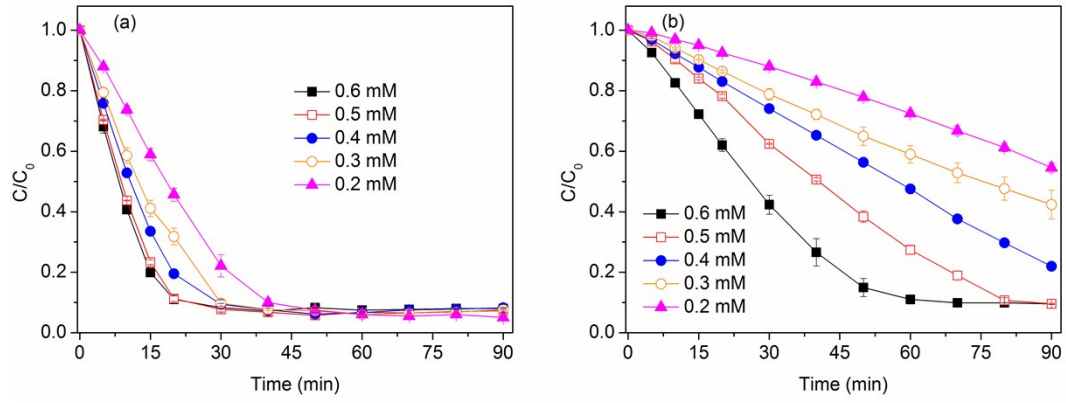Conditions: $[4C2NP]_0$=0.05 mM; $[Fe(II)]_0$=0.3 mM; $[PMS]_0$=0.3 mM

Fig. S3 The degradation of AO7 in Fe(II)/PMS system. (a) Fe(II)-PMS addition; (b) PMS-Fe(II) addition. Conditions: $[AO7]_0=0.05$ mM; $[PMS]_0=0.3$ mM; pH=3.0.
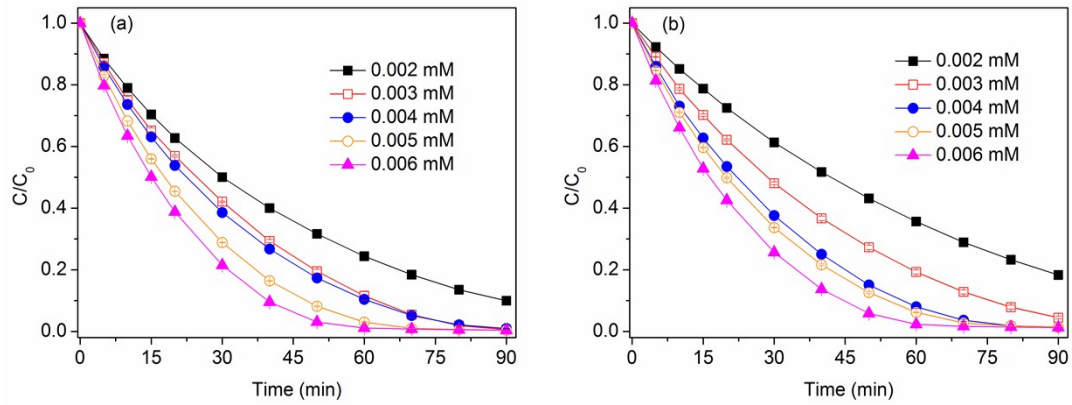

Fig. S4 The degradation of AO7 as a function of reaction time in Co(II)/PMS system. (a) Co(II)-PMS addition; (b) PMS-Co(II) addition. Conditions: $[AO7]_0=0.05$ mM; $[PMS]_0=0.3$ mM; pH=3.0.
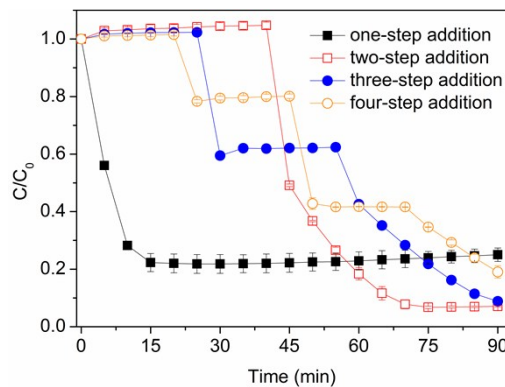
Fig. S5 Stepwise addition of PMS on AO7 degradation process as a function of reaction time in Fe(II)/PMS system. Conditions: $[AO7]_0$=0.1 mM; $[Fe(II)]_0$=0.6 mM; $[PMS]_{total}$=0.3 mM.
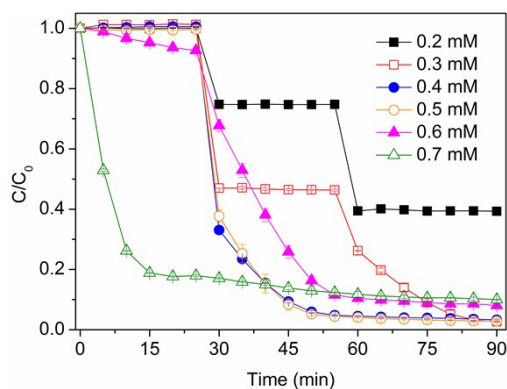

Fig. S6 Effect of PMS concentration on the degradation of AO7 for three-step addition in Fe(II)/PMS system. Conditions: $[AO7]0$=0.05 mM; $[Fe(II)]0$=0.3 mM.