

supporting information

Enhanced Photocatalytic Properties of ZnFe₂O₄ doped ZnIn₂S₄ Heterostructure under Visible Light Irradiation

Wu Yang^a, Dezhi Chen^{*a}, Hongying Quan^b, Shaolin Wu^{*a}, Xubiao Luo^a, Lin Guo^{a,c}

^a Key Laboratory of Jiangxi Province for Persistent Pollutants Control and Resources Recycle, School of Environmental and Chemical Engineering, Nanchang Hangkong University, Nanchang 330063, China.
E-mail: chendz@nchu.edu.cn, 39008@nchu.edu.cn

^b School of Materials Science and Engineering, Nanchang Hangkong University, Nanchang, China.

^c Key Laboratory of Bio-Inspired Smart Interfacial Science and Technology of Ministry of Education, School of Chemistry and Environment, Beihang University, Beijing 100191, China.

Table S1. The detail of nitrogen adsorption/desorption measurements for bare ZnIn₂S₄, ZnFe₂O₄ and the ZFO-ZIS composites

Sample	S _{BET} /(m ² /g)	Average pore size/(nm)	Pore volume/(cm ³ /g)
Pure ZnIn ₂ S ₄	77.5668	5.96610	0.115693
Pure ZnFe ₂ O ₄	138.9283	15.24611	0.529529
1 wt% ZFO-ZIS	170.8030	5.99078	0.255811
2.5 wt% ZFO-ZIS	167.8726	5.55760	0.233242
5 wt% ZFO-ZIS	104.1173	6.28131	0.163498
10 wt% ZFO-ZIS	75.8783	7.53702	0.142974
30 wt% ZFO-ZIS	65.8286	8.7084	0.143315
50 wt% ZFO-ZIS	132.1181	8.23138	0.271879

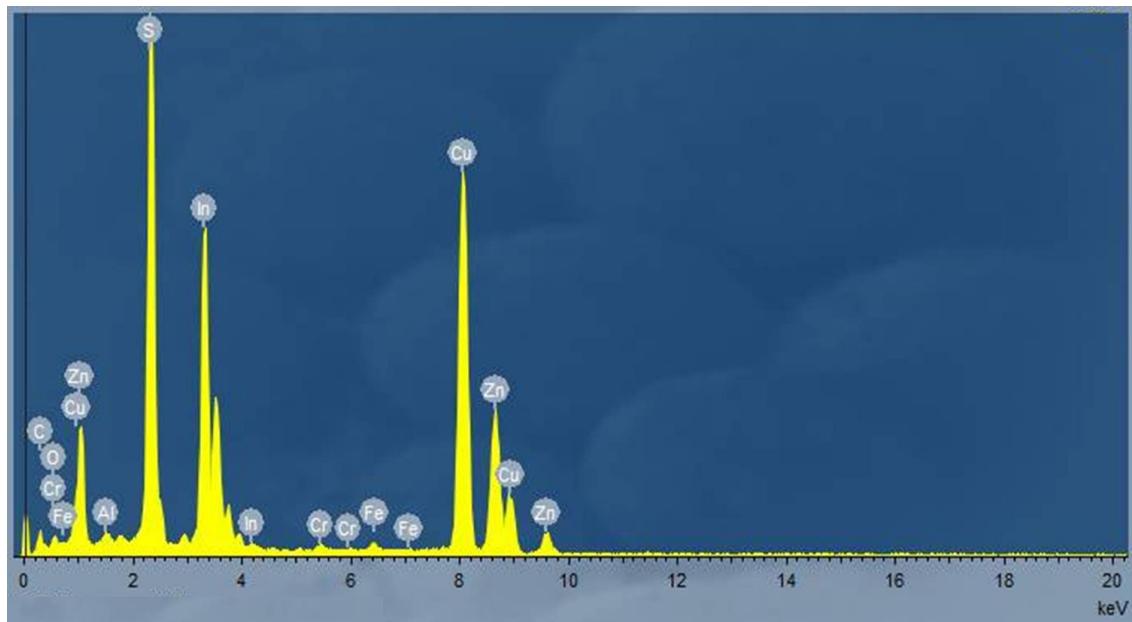


Figure S1. EDS spectrum of 2.5wt% ZFO-ZIS composites

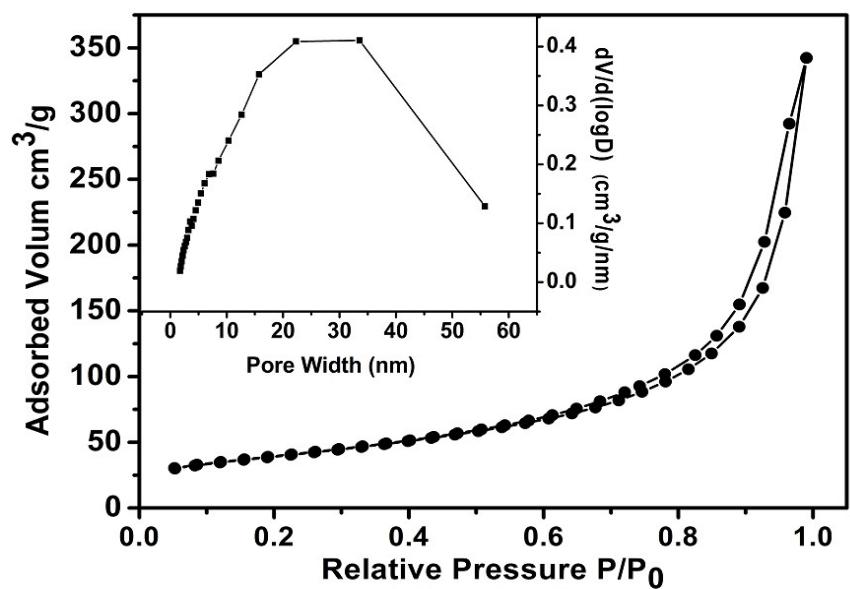


Figure S2. N_2 adsorption–desorption isotherms of pure ZnFe_2O_4 and the pore size distributions of the ZnFe_2O_4 in illustrations.

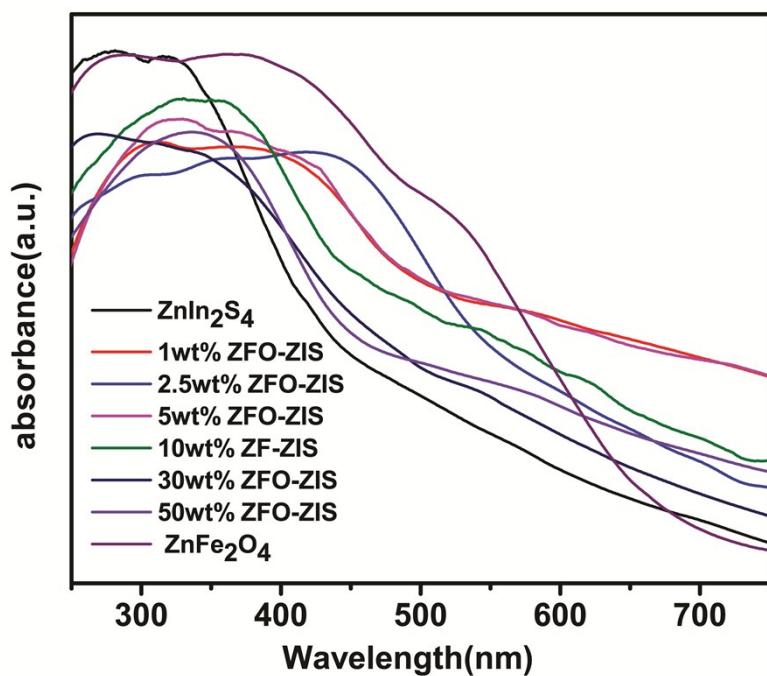


Figure S3. The UV-vis diffuse reflectance spectra of the ZnIn_2S_4 , ZnFe_2O_4 and ZFO-ZIS.

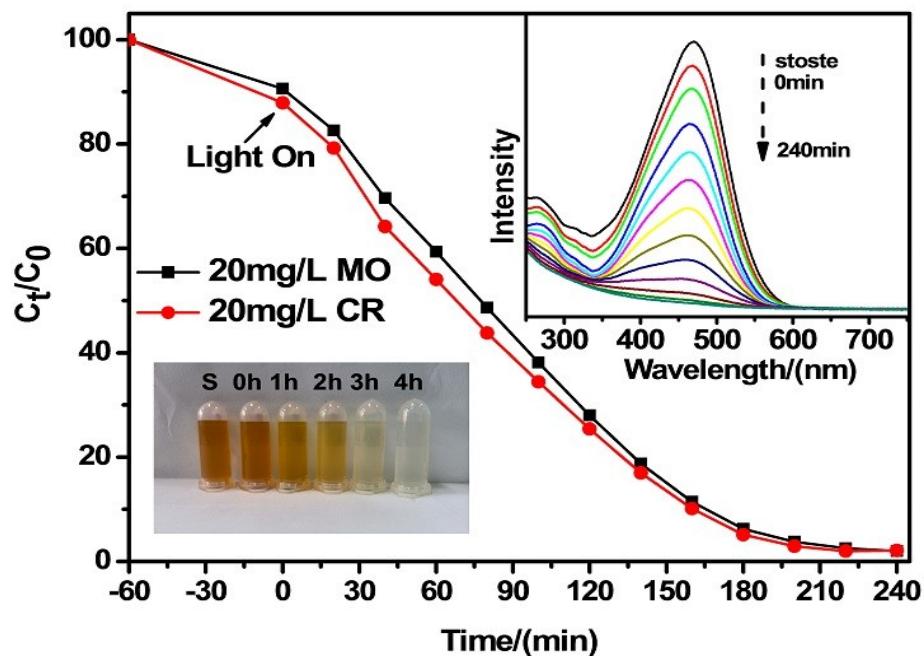


Figure S4. Decolorization of wastewater containing 20 mg/L methyl orange and 20 mg/L Congo red using the 2.5wt% ZFO-ZIS.