

Content of Electronic Supporting Materials:

S2.2.1. Synthesis of $\text{Er}^{3+}:\text{Y}_3\text{Al}_5\text{O}_{12}$ as up-conversion luminescence agent

Fig. S1. XRD patterns of (a) $\text{Er}^{3+}:\text{Y}_3\text{Al}_5\text{O}_{12}$ (heated-treated at 1100 °C for 120 min), (b) pure NaTaO_3 , (c-g) $\text{Er}^{3+}:\text{Y}_3\text{Al}_5\text{O}_{12}/\text{MoS}_2\text{-NaTaO}_3\text{-PdS}$ nanocomposite photocatalyst (with 0.20 wt% MoS_2 and 0.13 wt% PdS contents and different molar ratios ((c) 0.00:1.00, (d) 0.05:1.00, (e) 0.25:1.00, (f) 0.45:1.00 and (g) 0.65:1.00) of $\text{Er}^{3+}:\text{Y}_3\text{Al}_5\text{O}_{12}$ and NaTaO_3).

Fig. S2. EDX spectra of (a) $\text{Er}^{3+}:\text{Y}_3\text{Al}_5\text{O}_{12}$ powder (heat-treated at 1100 °C for 120 min), (b) pure NaTaO_3 and (c) $\text{Er}^{3+}:\text{Y}_3\text{Al}_5\text{O}_{12}/\text{MoS}_2\text{-NaTaO}_3\text{-PdS}$ nanocomposite photocatalyst (with 0.13 wt% PdS and 0.20 wt% MoS_2 contents and 0.45:1.00 molar ratio of $\text{Er}^{3+}:\text{Y}_3\text{Al}_5\text{O}_{12}$ and NaTaO_3).

Fig. S3. XPS spectra of $\text{Er}^{3+}:\text{Y}_3\text{Al}_5\text{O}_{12}/\text{MoS}_2\text{-NaTaO}_3\text{-PdS}$ nanocomposite photocatalyst (with 0.13 wt% PdS and 0.20 wt% MoS_2 contents and 0.45:1.00 molar ratio of $\text{Er}^{3+}:\text{Y}_3\text{Al}_5\text{O}_{12}$ and NaTaO_3).

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S2.2.1. Synthesis of $\text{Er}^{3+}:\text{Y}_3\text{Al}_5\text{O}_{12}$ as up-conversion luminescence agent

The $\text{Er}^{3+}:\text{Y}_3\text{Al}_5\text{O}_{12}$ was synthesized by the sol-gel method. First, erbium oxide (10 mmol, 2.2715 g) and yttrium oxide (0.334 mmol, 0.0128 g) were dissolved in hot HNO_3 solution (about 60 °C) with magnetic stirring and heated until transparent. Next appropriate amount of aluminum nitrate solution (33.6 mmol, 12.6208 g) and citric acid (168 mmol, 33.9351 g) as chelating agent were added into the mentioned mixture solution (molar ratio of citric acid: metal ion is 3:1), respectively. The solution was continued to be stirred and heated at 50-60 °C until the transparent sol was successfully formed. Then, the transparent sol was heated at 80 °C for 24 h and became the gel, which was ground into good homogeneous powders. For removing residual organic components and nitrate ions the powders were heated at 500 °C for 50 min, and then continued to heat to 1100 °C and kept for 2.0 h and finally permitted to cool down to the room temperature in atmosphere. After fully grinding, the nano-sized $\text{Er}^{3+}:\text{Y}_3\text{Al}_5\text{O}_{12}$ powders were obtained.

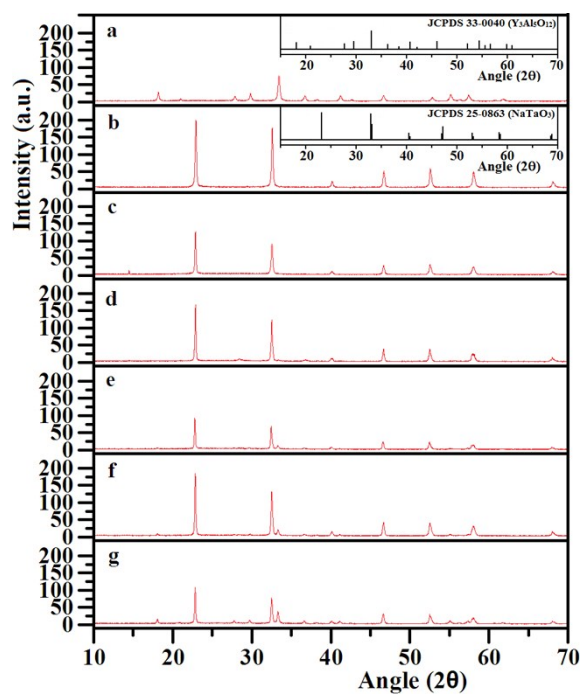


Fig. S1. XRD patterns of (a) $\text{Er}^{3+}:\text{Y}_3\text{Al}_5\text{O}_{12}$ (heated-treated at 1100 °C for 120 min), (b) pure NaTaO_3 , (c-g) $\text{Er}^{3+}:\text{Y}_3\text{Al}_5\text{O}_{12}/\text{MoS}_2\text{-NaTaO}_3\text{-PdS}$ nanocomposite photocatalyst (with 0.20 wt% MoS_2 and 0.13 wt% PdS contents and different molar ratios ((c) 0.00:1.00, (d) 0.05:1.00, (e) 0.25:1.00, (f) 0.45:1.00 and (g) 0.65:1.00) of $\text{Er}^{3+}:\text{Y}_3\text{Al}_5\text{O}_{12}$ and NaTaO_3).

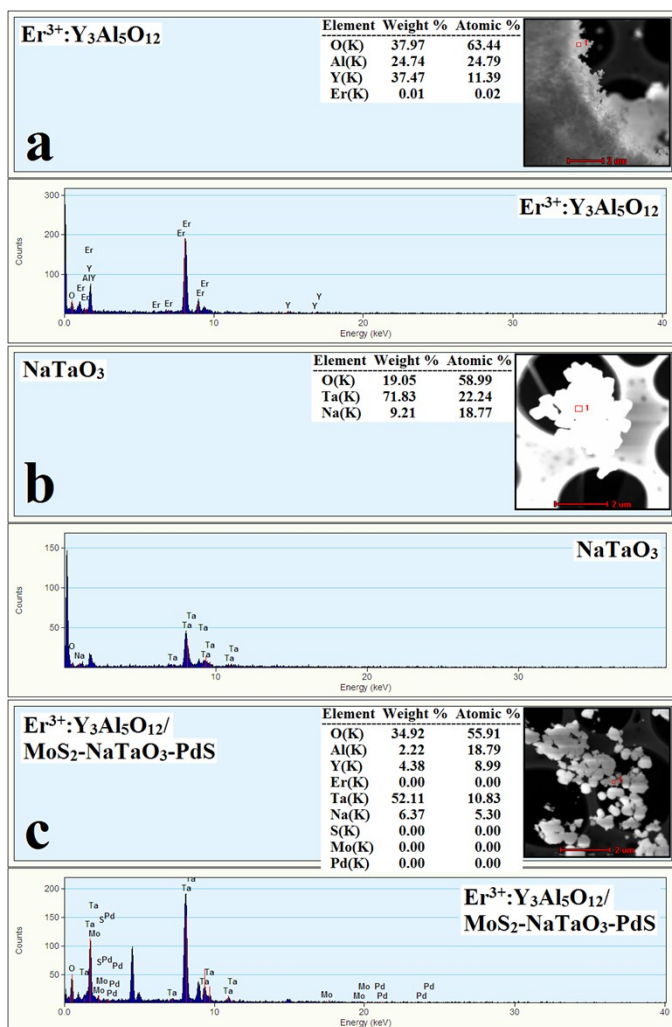


Fig. S2. EDX spectra of (a) Er³⁺:Y₃Al₅O₁₂ powder (heat-treated at 1100 °C for 120 min), (b) pure NaTaO₃ and (c) Er³⁺:Y₃Al₅O₁₂/MoS₂-NaTaO₃-PdS nanocomposite photocatalyst (with 0.13 wt% PdS and 0.20 wt% MoS₂ contents and 0.45:1.00 molar ratio of Er³⁺:Y₃Al₅O₁₂ and NaTaO₃).

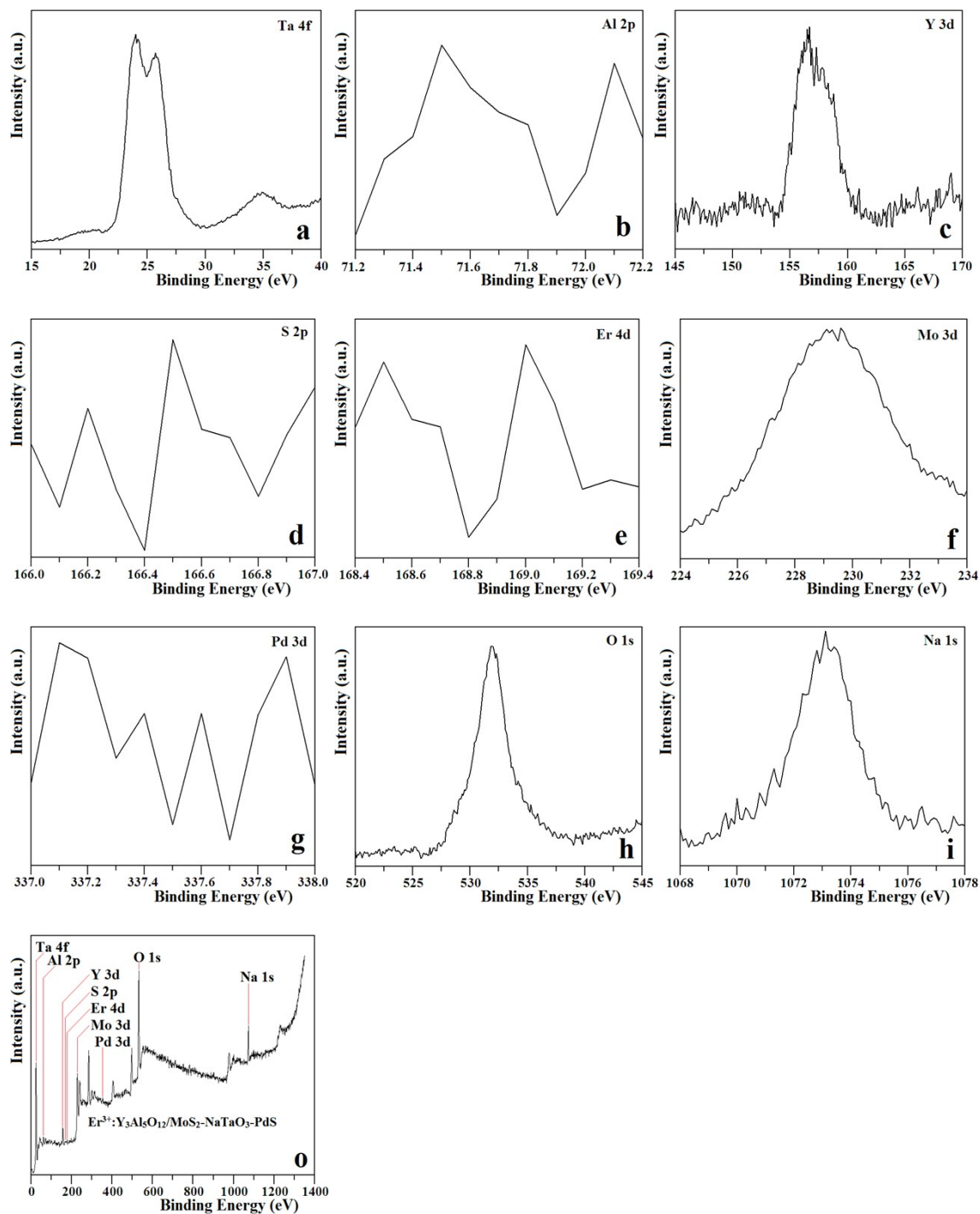


Fig. S3. XPS spectra of $\text{Er}^{3+}:\text{Y}_3\text{Al}_5\text{O}_{12}/\text{MoS}_2\text{-NaTaO}_3\text{-PdS}$ nanocomposite photocatalyst (with 0.13 wt% PdS and 0.20 wt% MoS_2 contents and 0.45:1.00 molar ratio of $\text{Er}^{3+}:\text{Y}_3\text{Al}_5\text{O}_{12}$ and NaTaO_3).