Supporting Information

High quality perovskite films fabricated from Lewis acid-base adduct through molecular exchange

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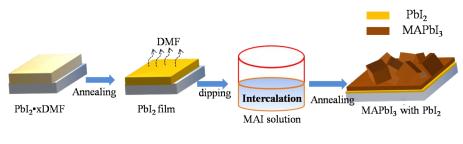




Fig. S1 Schematic illustration of the fabrication of perovskite films from typical twostep method.

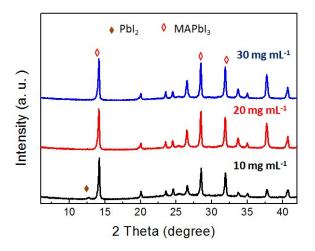


Fig. S2 XRD patterns of the perovskite films fabricated from Lewis adduct by immersing in MAI/2-proponol solutions with different MAI concentrations at a fixed reaction time of 30 s

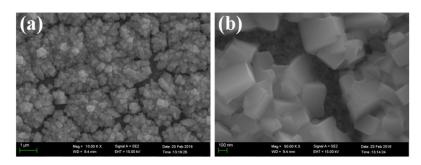


Fig. S3 SEM images of a perovskite film prepared from a solution of MAI/2-propanol with MAI concentration of 7.5 mgmL⁻¹ via Lewis adduct reaction. There are evident gaps among the clusters of perovskite grains.

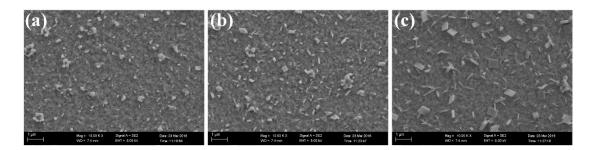


Fig. S4 SEM images of the perovskite films prepared from the concentration of MAI 20 mg mL⁻¹ by the typical TSM for different reaction time. (a) 1 min, (b) 2 min, (c) 5 min.

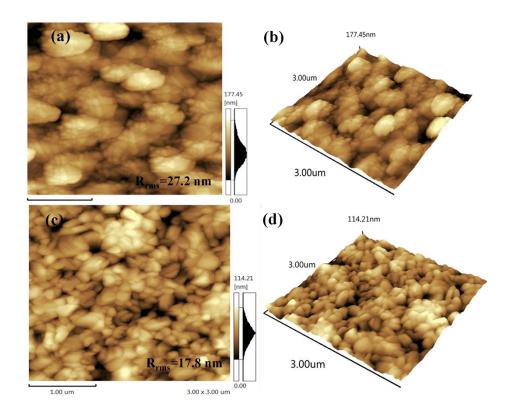


Fig. S5 Planar (Left) and 3D topographic (Right) atomic force microscope images of the perovskite films fabricated from different routes in MAI/2-propanol solution with a concentration of 20 mg mL⁻¹. (a) and (b) typical TSM. (c) and (d) Lewis adduct reaction.

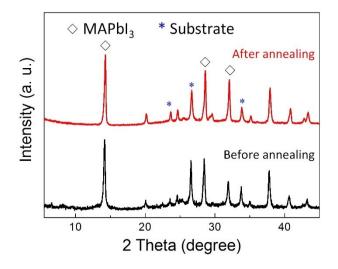


Fig. S6 XRD spectra of the MAPbI₃ films fabricated from the Lewis adduct reaction before and after annealing.

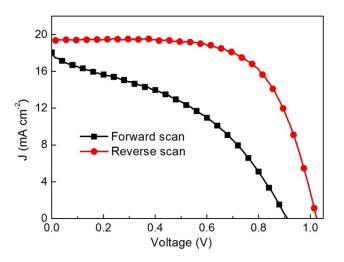


Fig. S7 J-V curves obtained from the forward and reverse scans of a perovskite solar cell fabricated from the Lewis adduct reaction.