

Supporting information for:

**Facile high-yield synthesis of unsymmetric
end-off compartmental double Schiff-base
ligands: easy access to mononuclear precursor
and unsymmetric dinuclear complexes**

Markus Schmidt, Helmar Görls, and Winfried Plass*

*Institut für Anorganische und Analytische Chemie, Friedrich-Schiller-Universität Jena,
Humboldtstr. 8, 07743 Jena, Germany.*

E-mail: sekr.plass@uni-jena.de

Phone: +49 3641 948130. Fax: +49 3641 948132

Table S1: Crystallographic data and structure refinement parameters for [Ni(tsc-difo)PPh₃] \cdot 2MeOH and [Zn₂(tsc-hydra)(OAc)₂] \cdot MeCN \cdot H₂O

	[Ni(tsc-difo)PPh ₃] \cdot 2MeOH	[Zn ₂ (tsc-hydra)(OAc) ₂] \cdot MeCN \cdot H ₂ O
formula	C ₃₀ H ₃₂ N ₃ NiO ₄ PS	C ₃₀ H ₃₆ N ₈ O ₆ SZn ₂
formula weight (g mol ⁻¹)	620.33	767.47
color	brown	yellow
crystal size (mm)	0.045 \times 0.034 \times 0.032	0.060 \times 0.060 \times 0.040
crystal system	monoclinic	monoclinic
space group	<i>P</i> 2 ₁ / <i>n</i>	<i>P</i> 2 ₁ / <i>n</i>
<i>a</i> (pm)	1483.10(6)	1347.46(8)
<i>b</i> (pm)	767.72(3)	1502.26(11)
<i>c</i> (pm)	2582.78(9)	1667.45(12)
α (°)	90	90
β (°)	93.346(1)	93.667(4)
γ (°)	90	90
<i>V</i> (10 ⁶ pm ³)	2935.75(19)	3368.4(4)
<i>Z</i>	4	4
<i>T</i> (K)	133(2)	133(2)
ρ_{calc} (g cm ⁻³)	1.403	1.513
<i>F</i> (000)	1296	1584
μ (Mo K α) (mm ⁻¹)	0.826	1.540
Θ range of data collection (°)	2.67–27.45	2.42–27.49
measured reflections	16312	15345
unique reflections (<i>R</i> _{int})	6439 (0.0333)	6769 (0.0447)
no. of parameters	450	457
goodness-of-fit on <i>F</i> ²	1.124	1.177
<i>R</i> 1 (<i>I</i> > 2 σ (<i>I</i>))	0.0478	0.0565
<i>wR</i> 2 (all data)	0.1226	0.1176

Table S2: Selected ^{13}C NMR data for mono and double Schiff bases (δ/ppm)

Ligand	C1	C2	C3	C4	C5	C6	C7	C8	C9	C10
H ₂ sc-difo	156.3	135.5	122.2	134.1	128.8	132.3	121.9	156.4	19.7	194.6
H ₂ tsc-difo	177.9	137.1	122.1	134.2	129.1	134.3	121.8	156.7	19.7	195.7
H ₂ sc-hydra	156.8	134.6	122.3	128.9	126.2	132.7	118.2	158.8	20.0	166.5
H ₂ tsc-hydra	177.7	137.8	121.9	129.7	125.9	133.7	118.0	160.5	19.9	166.5
H ₂ sc-ampy	156.8	134.2	121.0	129.0	127.0	132.9	118.5	157.6	19.9	167.5
H ₂ tsc-ampy	177.8	137.3	121.5	129.4	126.9	133.9	118.5	158.6	19.9	167.5
H ₃ sc-amph	156.8	134.4	122.1	129.3	127.1	133.4	119.3	157.6	20.0	161.4
H ₃ tsc-amph	177.8	137.4	121.6	129.8	127.1	134.3	119.3	158.6	20.0	161.3

Table S3: Bond angles at the nickel(II) ion in the crystal structure of [Ni(tsc-difo)PPh₃] \cdot 2MeOH (in deg)

S-Ni-P	91.46(3)
S-Ni-O1	174.29(7)
P-Ni-O1	86.39(6)
O1-Ni-N3	95.13(10)
S-Ni-N3	87.33(8)
P-Ni-N3	176.35(8)

Table S4: Bond angles at the zinc(II) ions in the crystal structure of $[\text{Zn}_2(\text{tsc-hydra})(\text{OAc})_2]\cdot\text{MeCN}\cdot\text{H}_2\text{O}$ (in deg)

S–Zn1–O1	150.00(9)	O1–Zn2–O4	75.22(11)
S–Zn1–O2	115.13(10)	O1–Zn2–N4	85.40(12)
S–Zn1–O4	100.29(9)	O1–Zn2–N5	161.90(12)
S–Zn1–N3	82.40(10)	O1–Zn2–N6	113.57(13)
O1–Zn1–O2	94.40(13)	O1–Zn2–N7	96.30(13)
O1–Zn1–O4	76.00(11)	O4–Zn2–N4	155.62(13)
O1–Zn1–N3	84.53(12)	O4–Zn2–N5	121.10(12)
O2–Zn1–O4	104.21(13)	O4–Zn2–N6	87.48(13)
O2–Zn1–N3	105.96(13)	O4–Zn2–N7	93.62(13)
O4–Zn1–N3	145.09(13)	N4–Zn2–N5	80.55(13)
		N4–Zn2–N6	86.87(15)
		N4–Zn2–N7	103.22(15)
		N5–Zn2–N6	77.15(14)
		N6–Zn2–N7	76.06(14)
		N6–Zn2–N7	149.27(14)

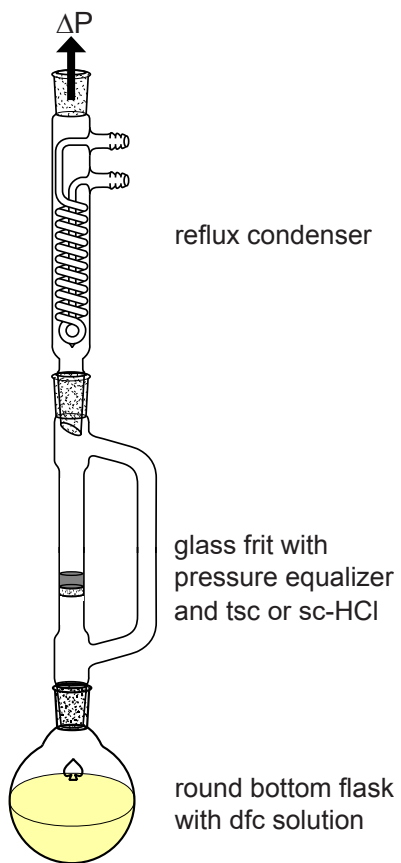


Figure S1: Experimental setup for the synthesis of the proligands H_2sc -difo and H_2tsc -difo in gram scale and high purity.

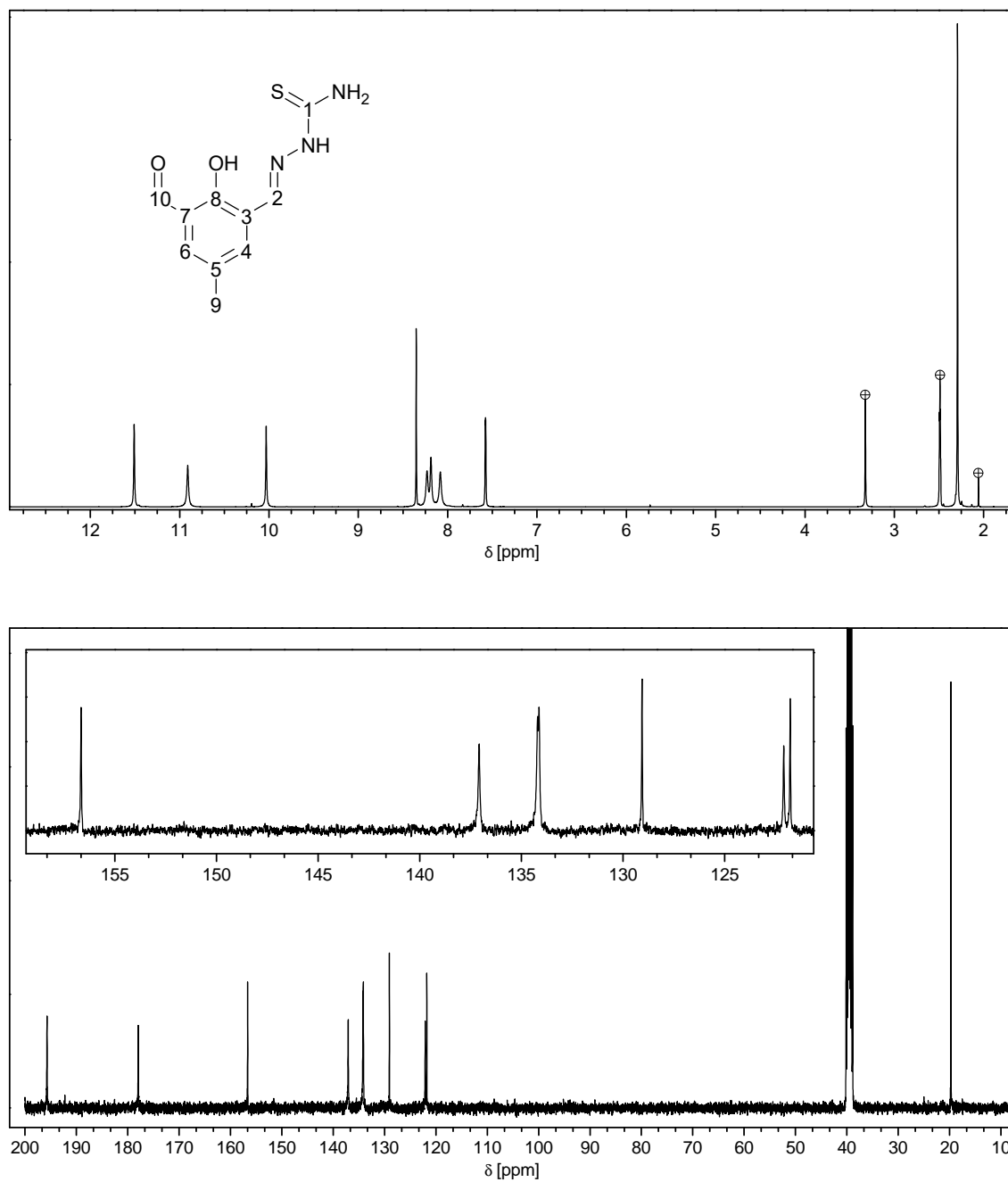


Figure S2: NMR spectra of H₂tsc-difo in DMSO-d₆: (Top) ¹H NMR; marked peaks correspond to water, acetonitrile and DMSO. (Bottom) ¹³C{¹H} NMR.

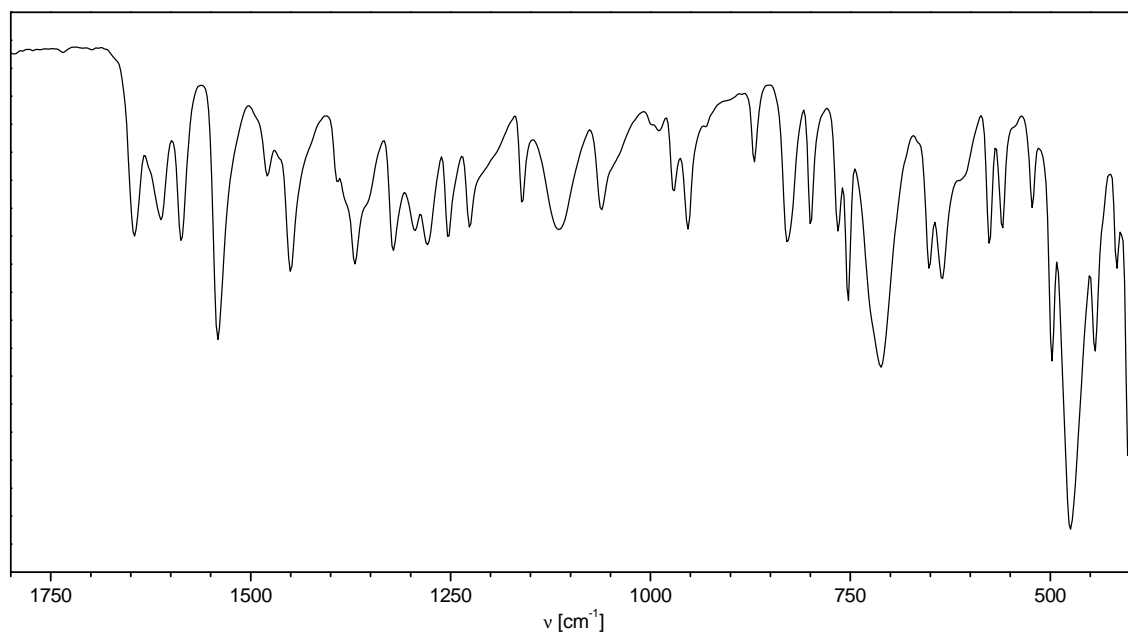
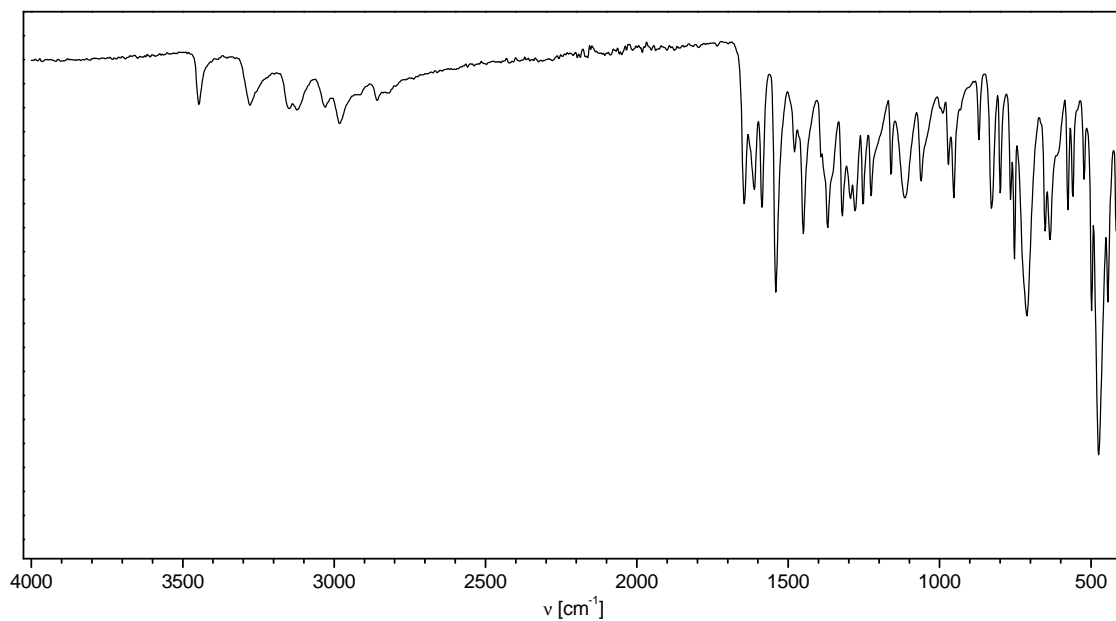


Figure S3: FT-IR spectra of H₂tsc-difo: (Top) Overview range from 4000–400 cm⁻¹. (Bottom) Fingerprint region from 1800–400 cm⁻¹. List of the positions of vibrational bands (in cm⁻¹): 3447, 3278, 3150, 3124, 3034, 2981, 2857, 1644, 1614, 1584, 1540, 1480, 1448, 1370, 1319, 1277, 1251, 1223, 1159, 1112, 1058, 971, 952, 872, 827, 798, 767, 753, 714, 650, 634, 577, 560, 521, 478, 444, 415, 404.

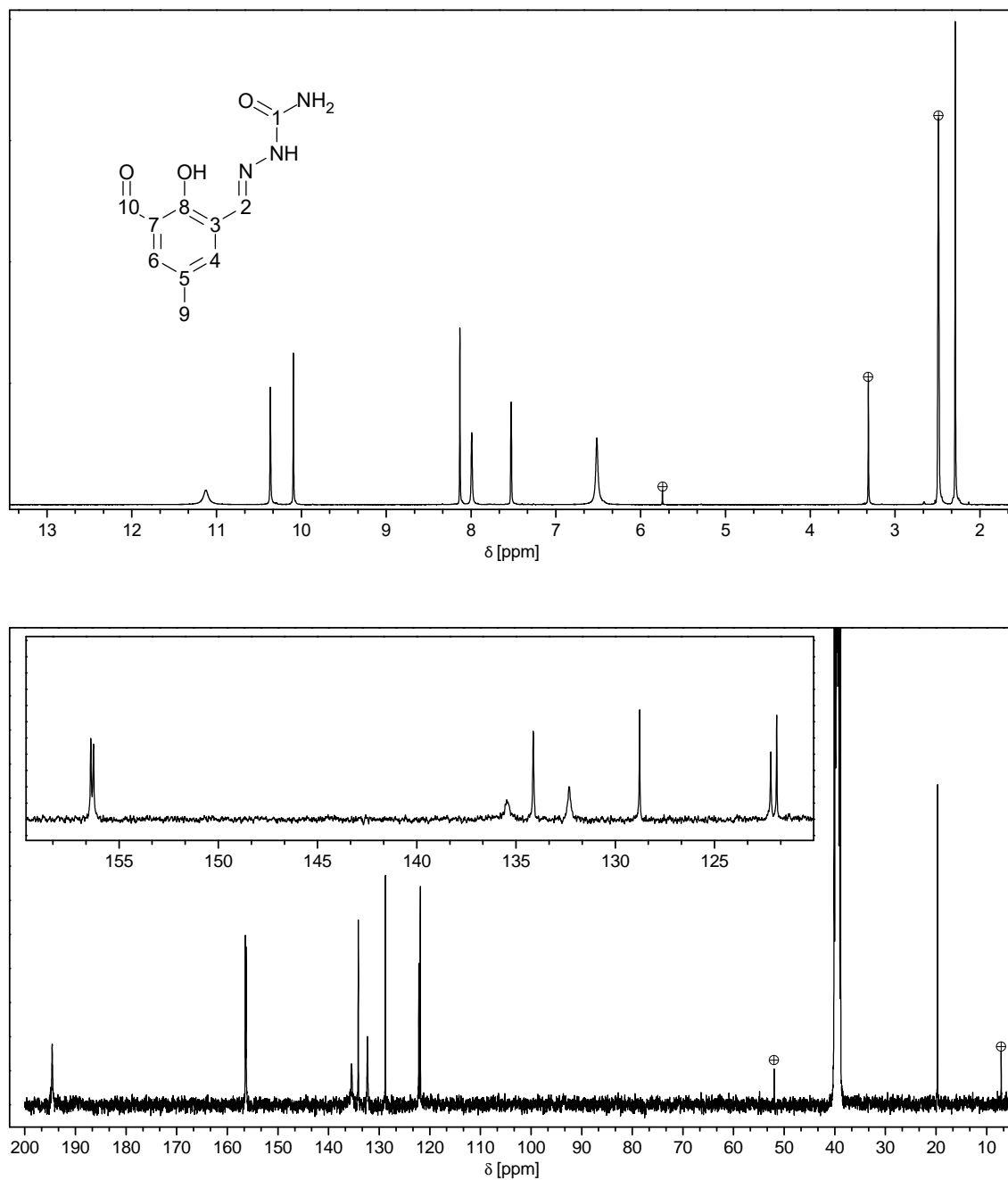


Figure S4: NMR spectra of H₂tsc-difo in DMSO-d₆: (Top) ¹H NMR; marked peaks correspond to water, dichloromethane and DMSO. (Bottom) ¹³C{¹H} NMR; marked peaks correspond diethyl ether.

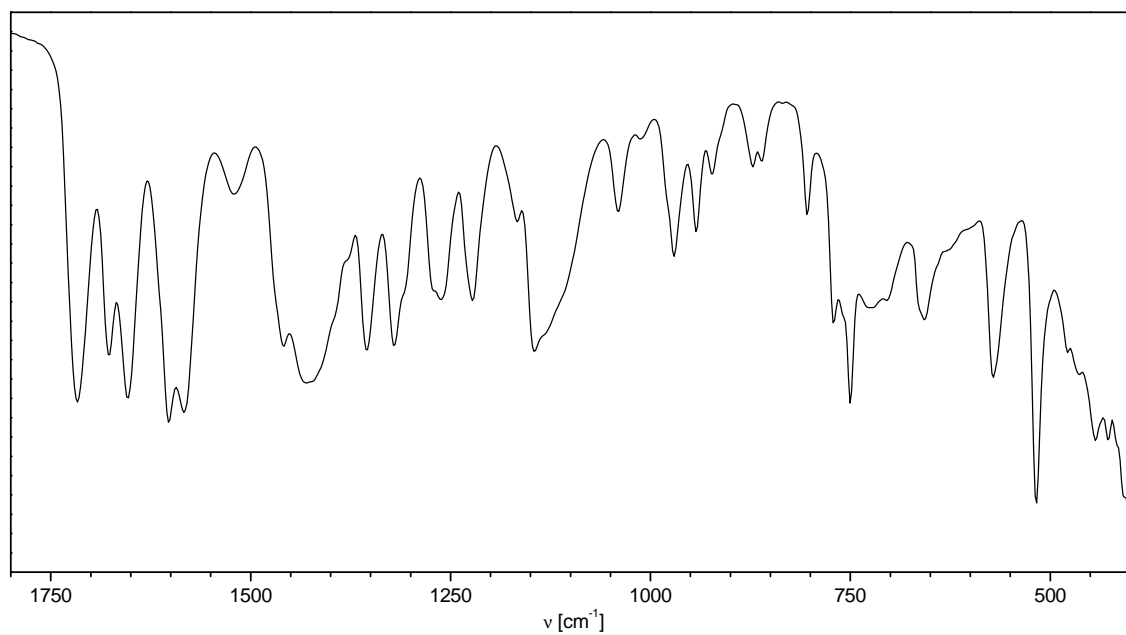
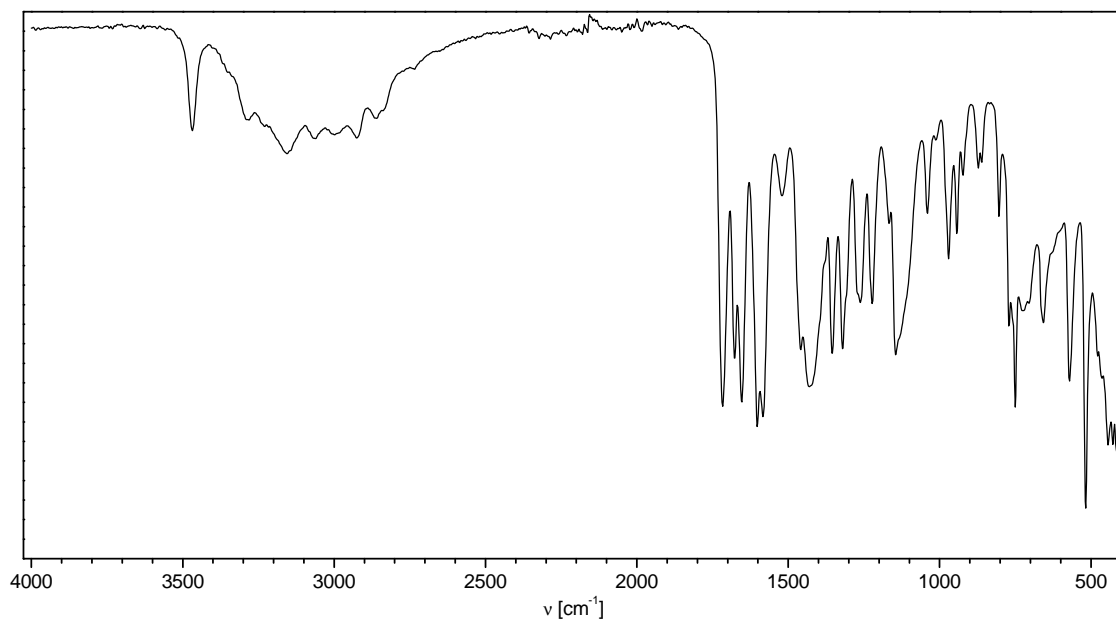


Figure S5: FT-IR spectra of H₂sc-difo: (Top) Overview range from 4000–400 cm⁻¹. (Bottom) Fingerprint region from 1800–400 cm⁻¹. List of the positions of vibrational bands (in cm⁻¹): 3469, 3288, 3158, 3064, 2998, 2922, 2865, 1714, 1675, 1652, 1602, 1582, 1517, 1463, 1428, 1355, 1321, 1261, 1222, 1170, 1145, 1039, 968, 941, 923, 875, 858, 804, 772, 751, 724, 657, 573, 518, 445, 401.

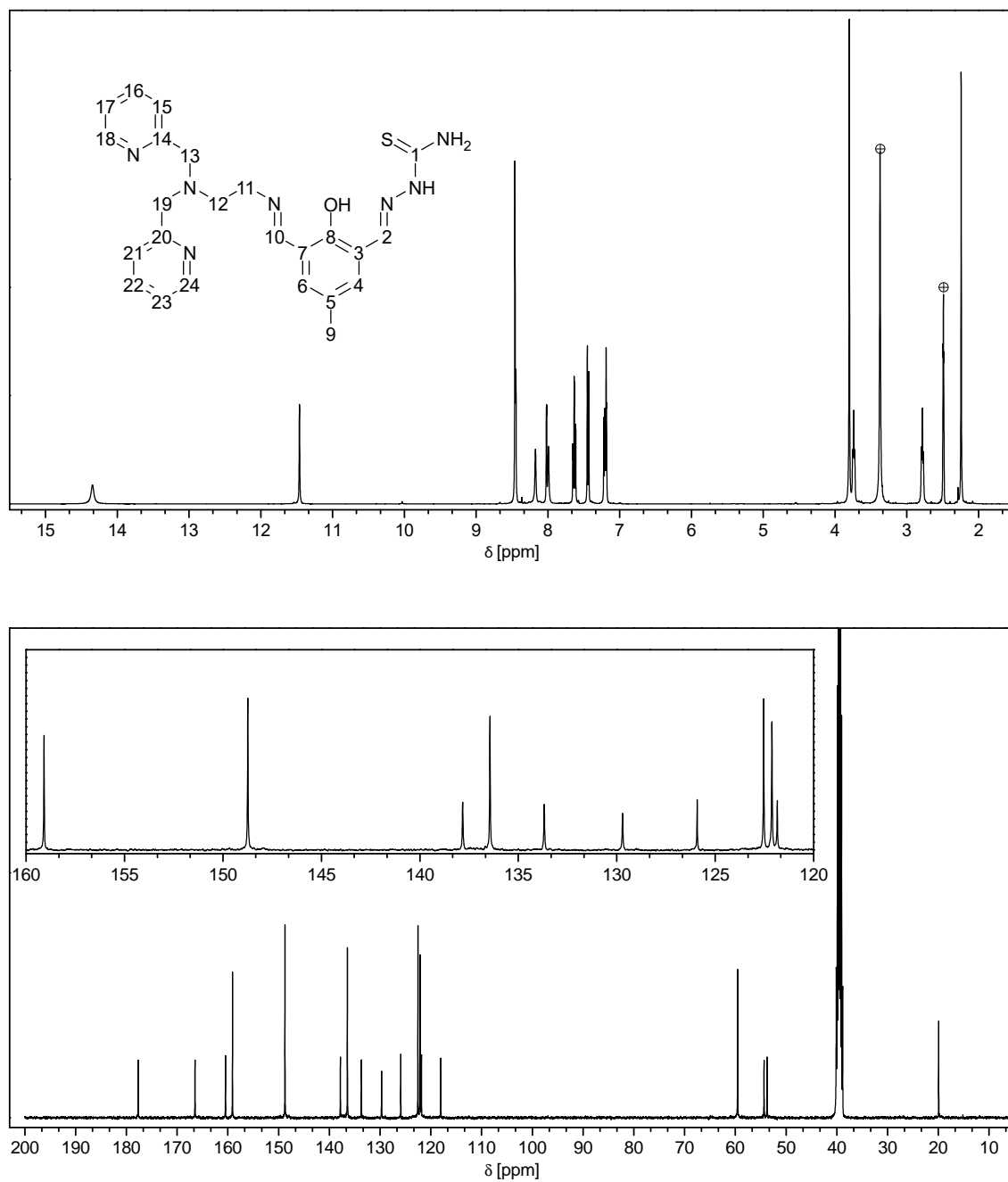


Figure S6: NMR spectra of H₂tsc-hydra in DMSO-d₆: (Top) ¹H NMR; marked peaks correspond to water and DMSO. (Bottom) ¹³C{¹H} NMR.

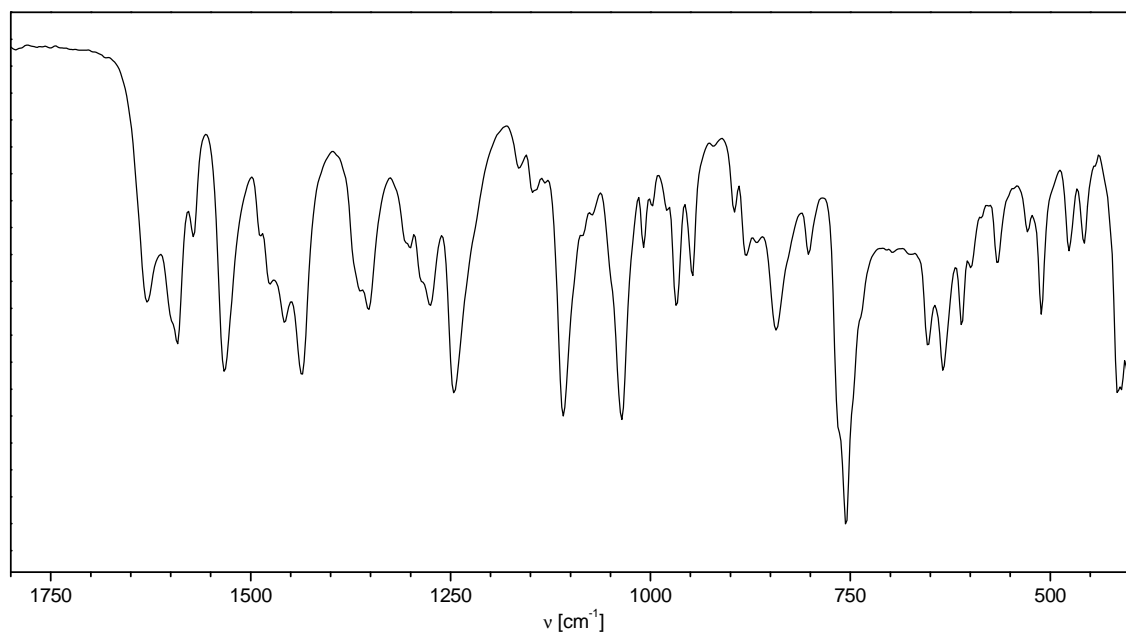
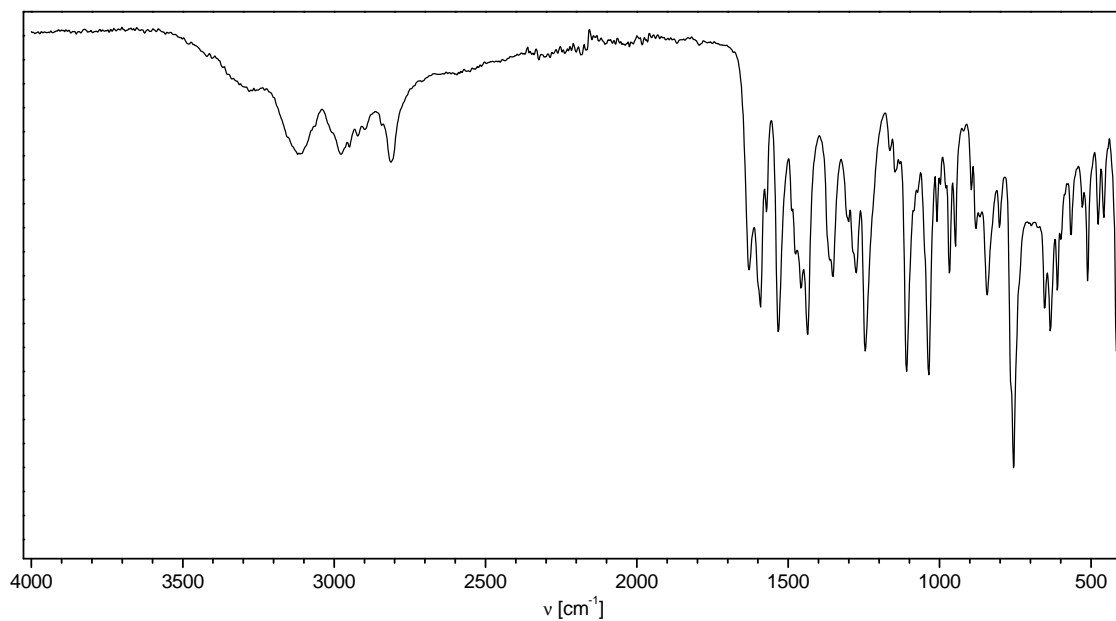


Figure S7: FT-IR spectra of H₂tsc-hydra: (Top) Overview range from 4000–400 cm⁻¹. (Bottom) Fingerprint region from 1800–400 cm⁻¹. List of the positions of vibrational bands (in cm⁻¹): 3117, 2977, 2947, 2922, 2895, 2810, 1630, 1591, 1570, 1531, 1456, 1435, 1364, 1353, 1305, 1275, 1247, 1165, 1145, 1108, 1035, 1010, 968, 948, 898, 879, 843, 801, 753, 653, 637, 611, 566, 529, 513, 477, 458, 420.

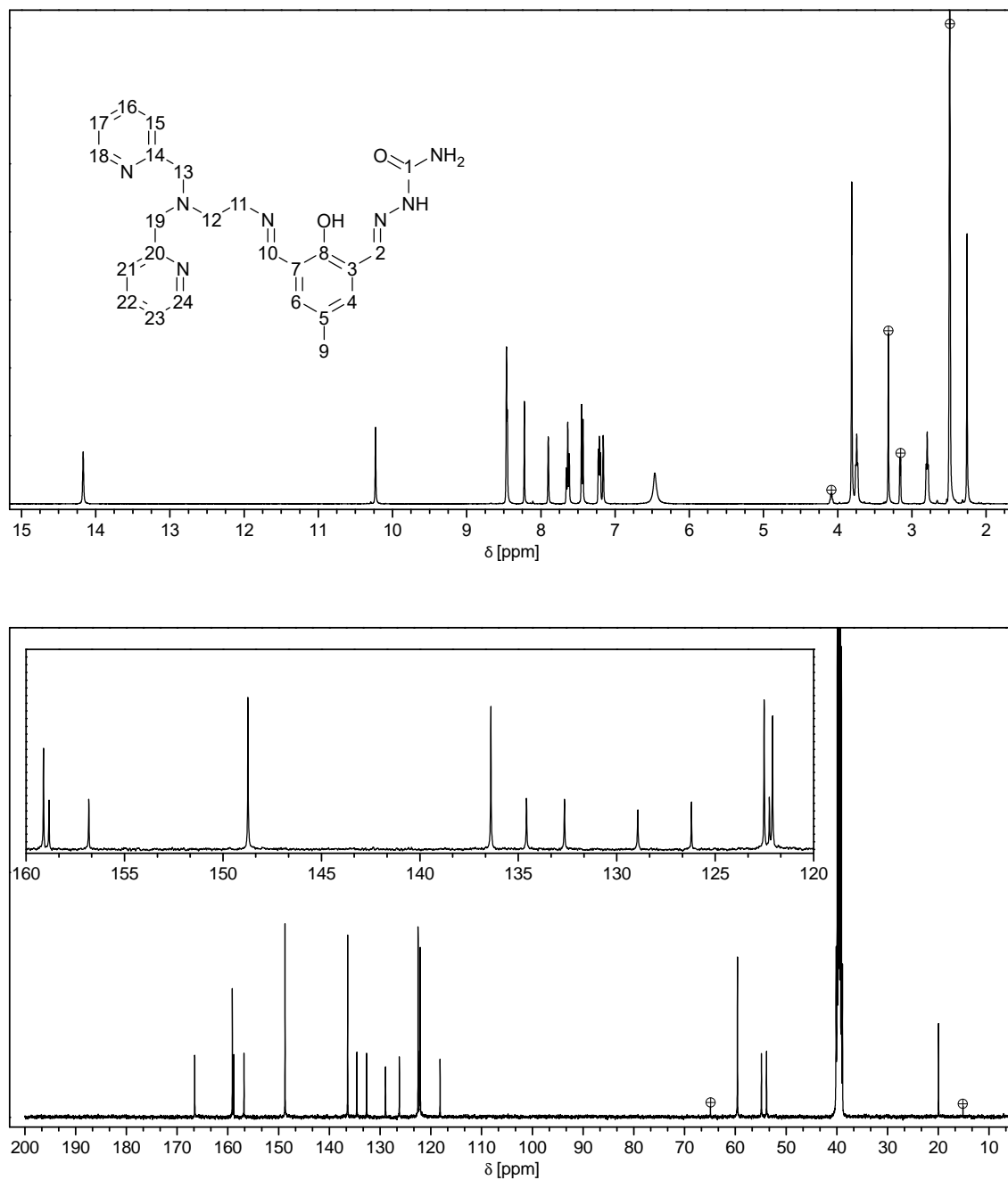


Figure S8: NMR spectra of H₂sc-hydra in DMSO-d₆: (Top) ¹H NMR; marked peaks correspond to water, methanol and DMSO. (Bottom) ¹³C{¹H} NMR; marked peaks are assigned to diethyl ether.

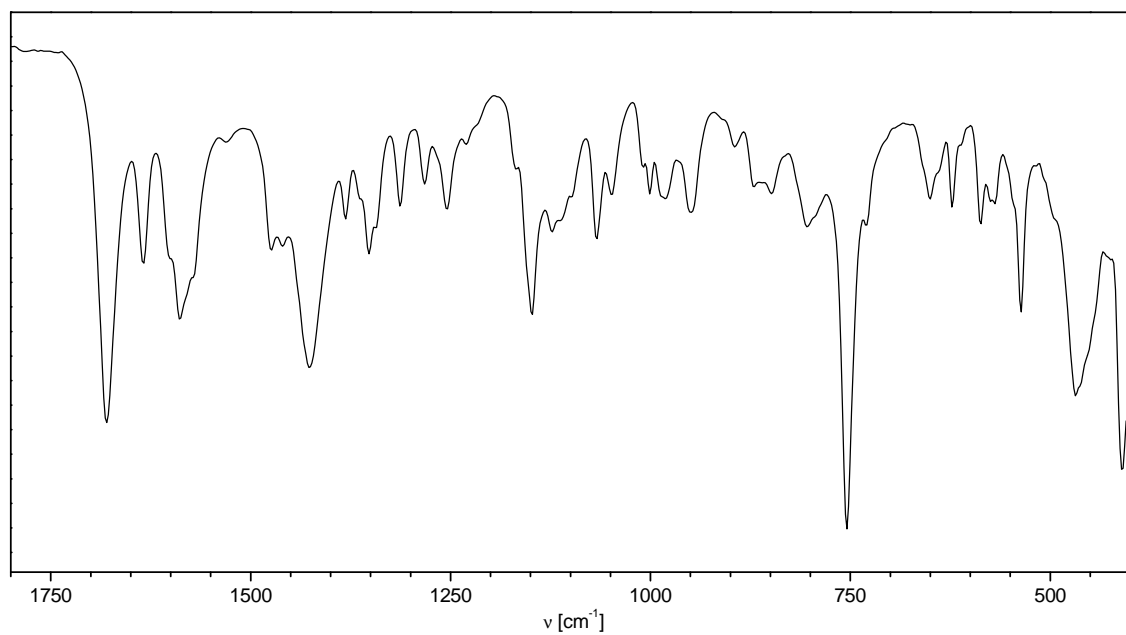
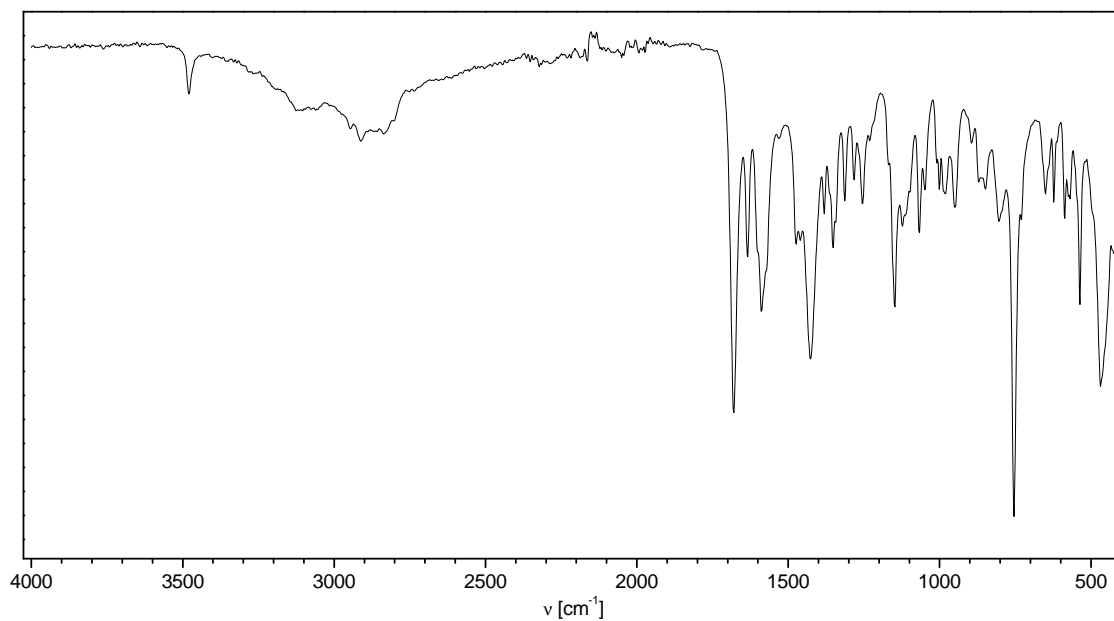


Figure S9: FT-IR spectra of H₂sc-hydra: (Top) Overview range from 4000–400 cm⁻¹. (Bottom) Fingerprint region from 1800–400 cm⁻¹. List of the positions of vibrational bands (in cm⁻¹): 3480, 3124, 3064, 2945, 2918, 2833, 1680, 1634, 1591, 1474, 1460, 1428, 1383, 1353, 1312, 1282, 1252, 1229, 1147, 1122, 1065, 1046, 1005, 984, 948, 898, 872, 849, 808, 756, 730, 650, 625, 589, 570, 538, 470, 412.

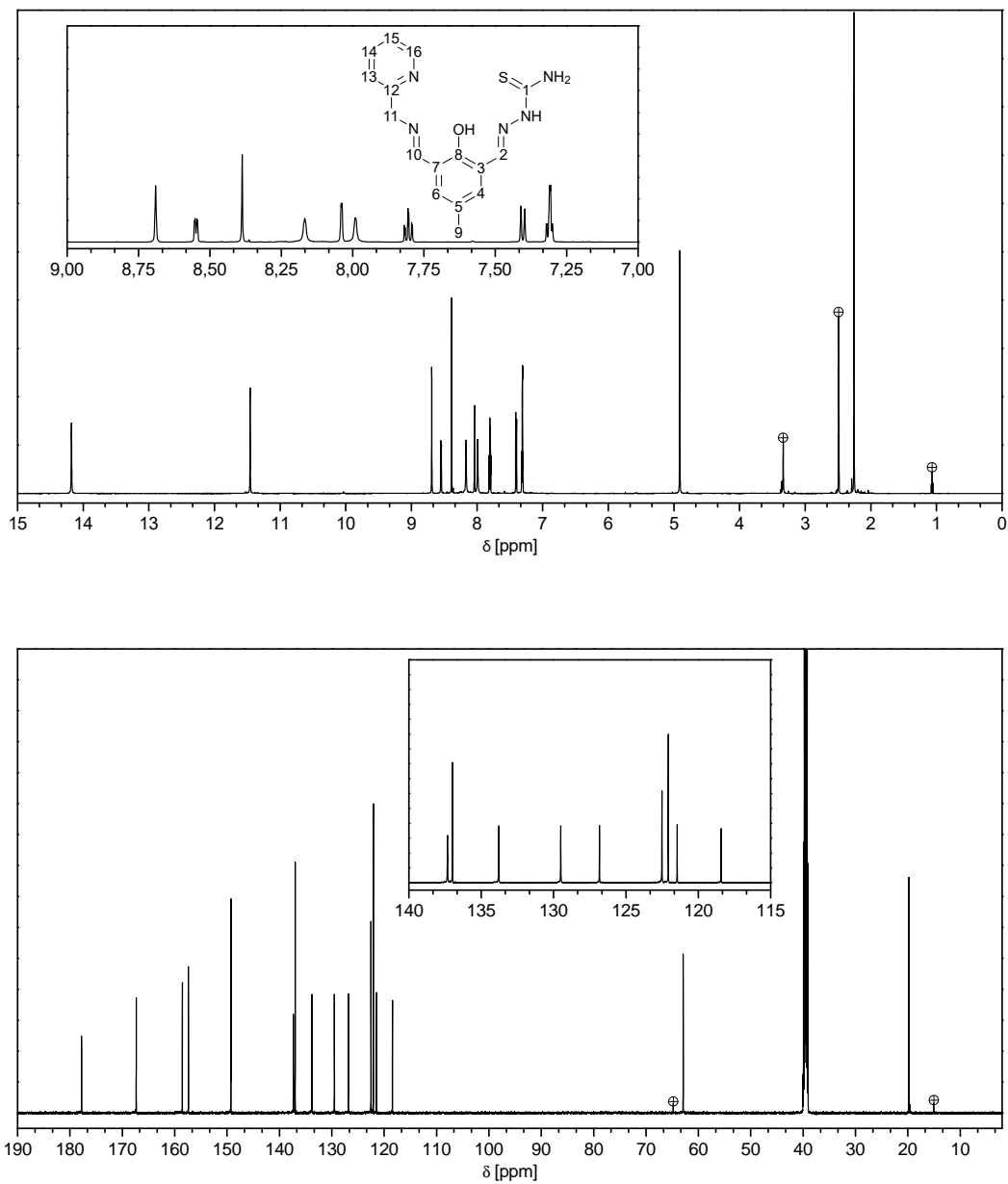


Figure S10: NMR spectra of H₂tsc-ampy in DMSO-d₆: (Top) ¹H NMR; marked peaks correspond to water, DMSO and diethyl ether. (Bottom) ¹³C{¹H} NMR; marked peaks correspond to diethyl ether.

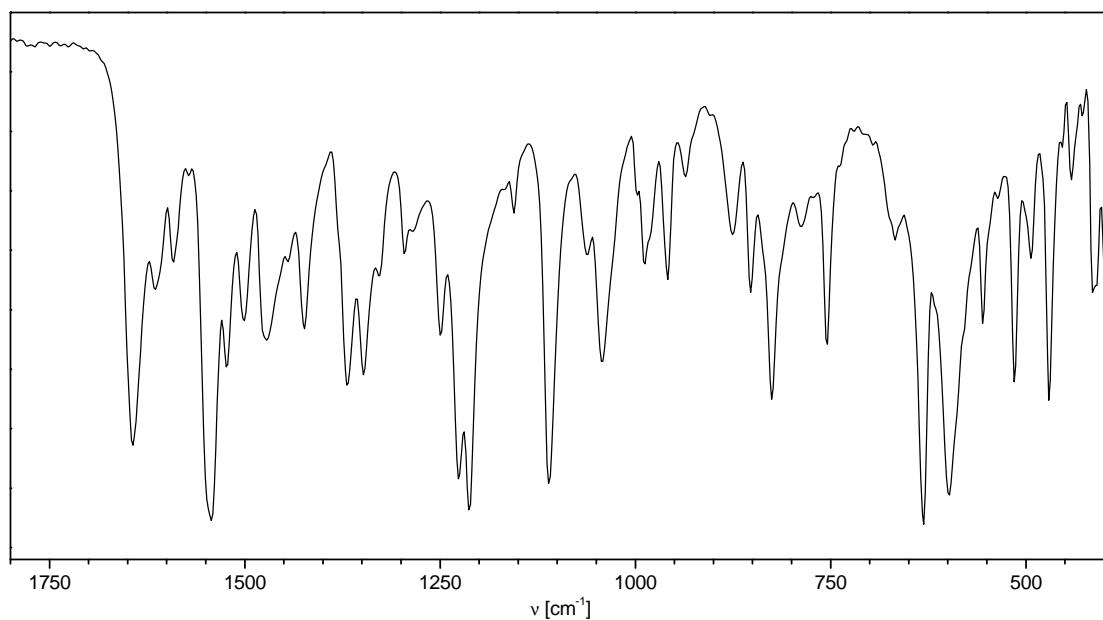
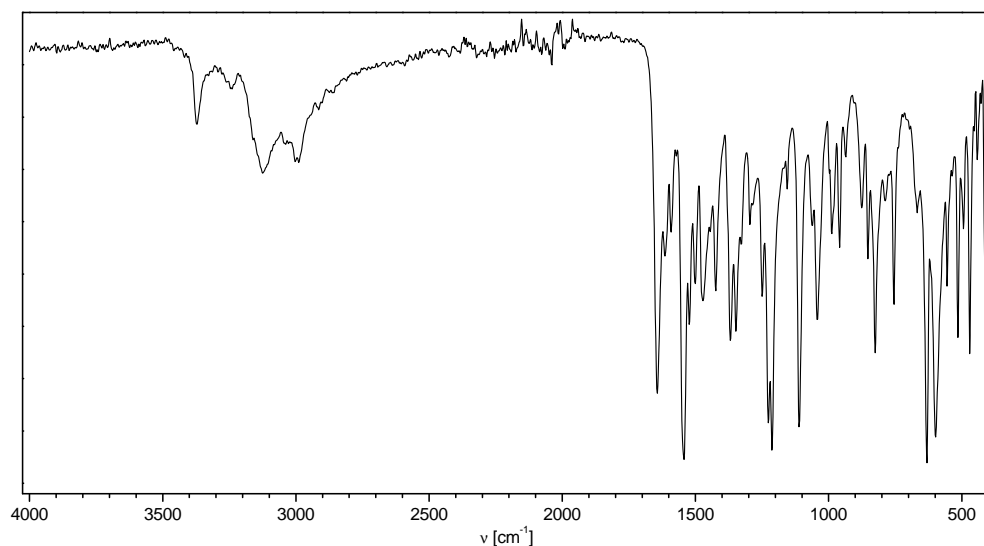


Figure S11: FT-IR spectra of H₂tsc-ampy: (Top) Overview range from 4000–400 cm⁻¹. (Bottom) Fingerprint region from 1800–400 cm⁻¹. List of the positions of vibrational bands (in cm⁻¹): 3374, 3125, 2992, 1640, 1612, 1588, 1544, 1522, 1499, 1471, 1422, 1368, 1347, 1295, 1248, 1227, 1211, 1154, 1108, 1065, 1042, 986, 960, 937, 873, 854, 826, 756, 669, 632, 599, 557, 515, 493, 470, 414.

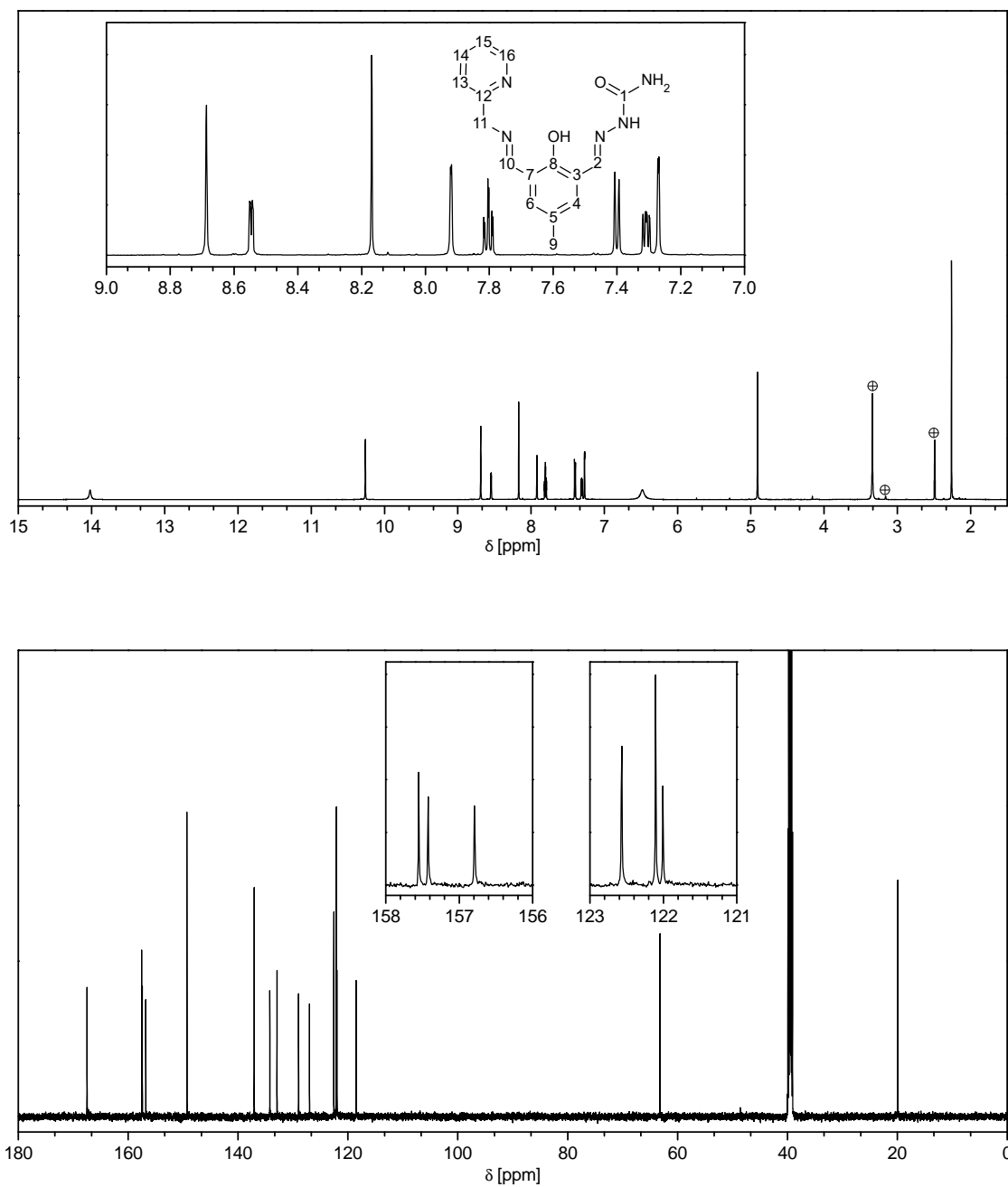


Figure S12: NMR spectra of H₂sc-ampy in DMSO-d₆: (Top) ¹H NMR; marked peaks correspond to water, methanol and DMSO. (Bottom) ¹³C{¹H} NMR.

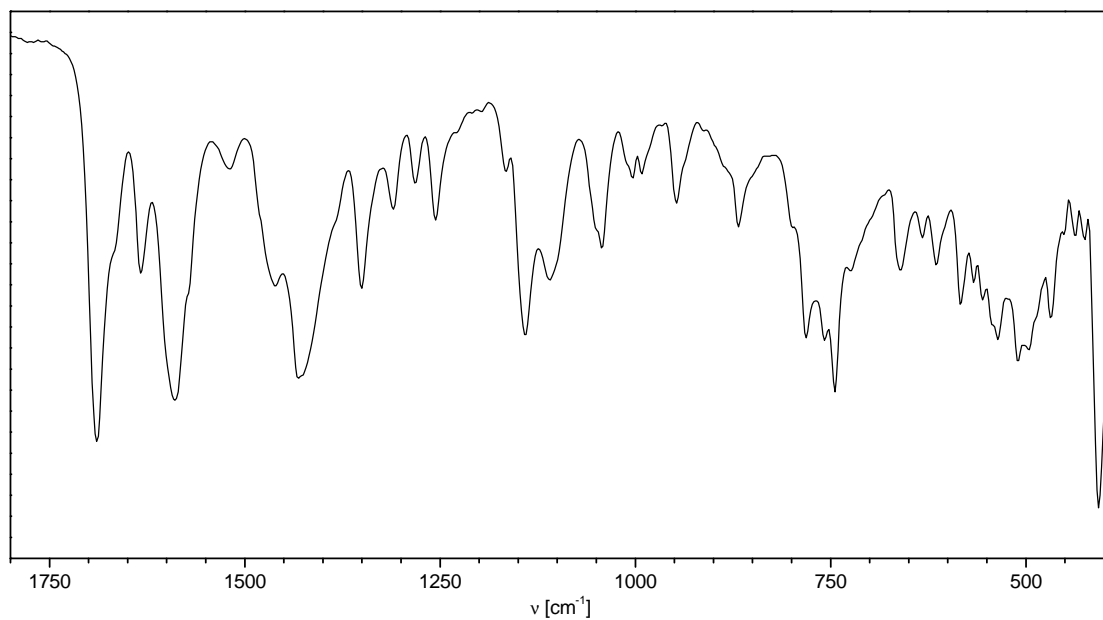
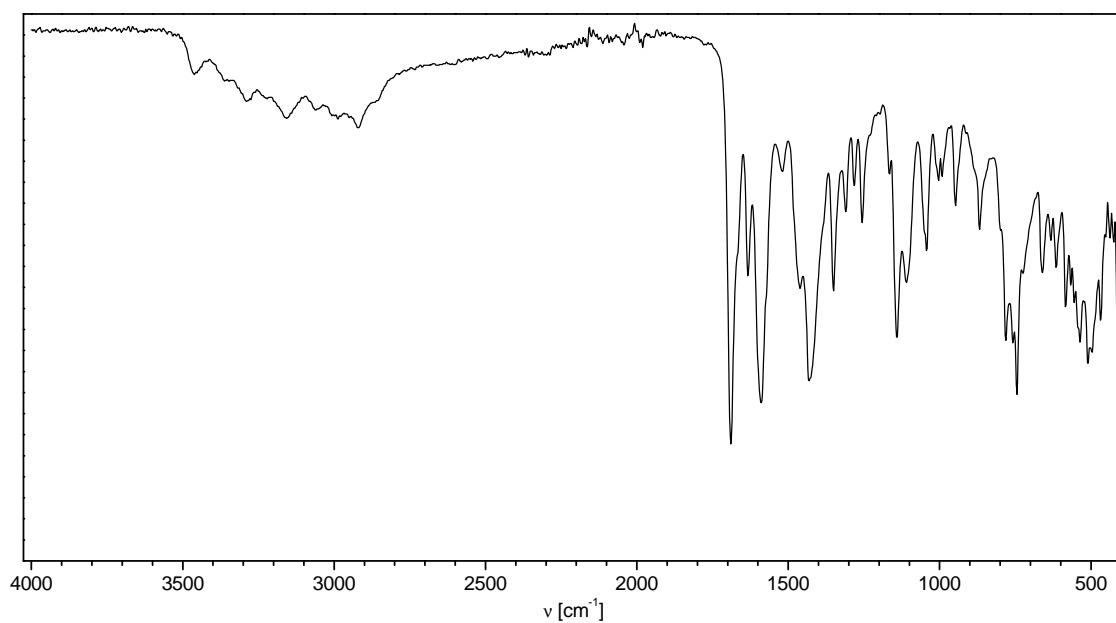


Figure S13: FT-IR spectra of H₂sc-ampy: (Top) Overview range from 4000–400 cm⁻¹. (Bottom) Fingerprint region from 1800–400 cm⁻¹. List of the positions of vibrational bands (in cm⁻¹): 3465, 3286, 3154, 2918, 1688, 1635, 1587, 1521, 1465, 1426, 1350, 1310, 1282, 1261, 1141, 1103, 1040, 1007, 991, 948, 869, 780, 742, 722, 661, 635, 617, 582, 538, 508, 495, 467.

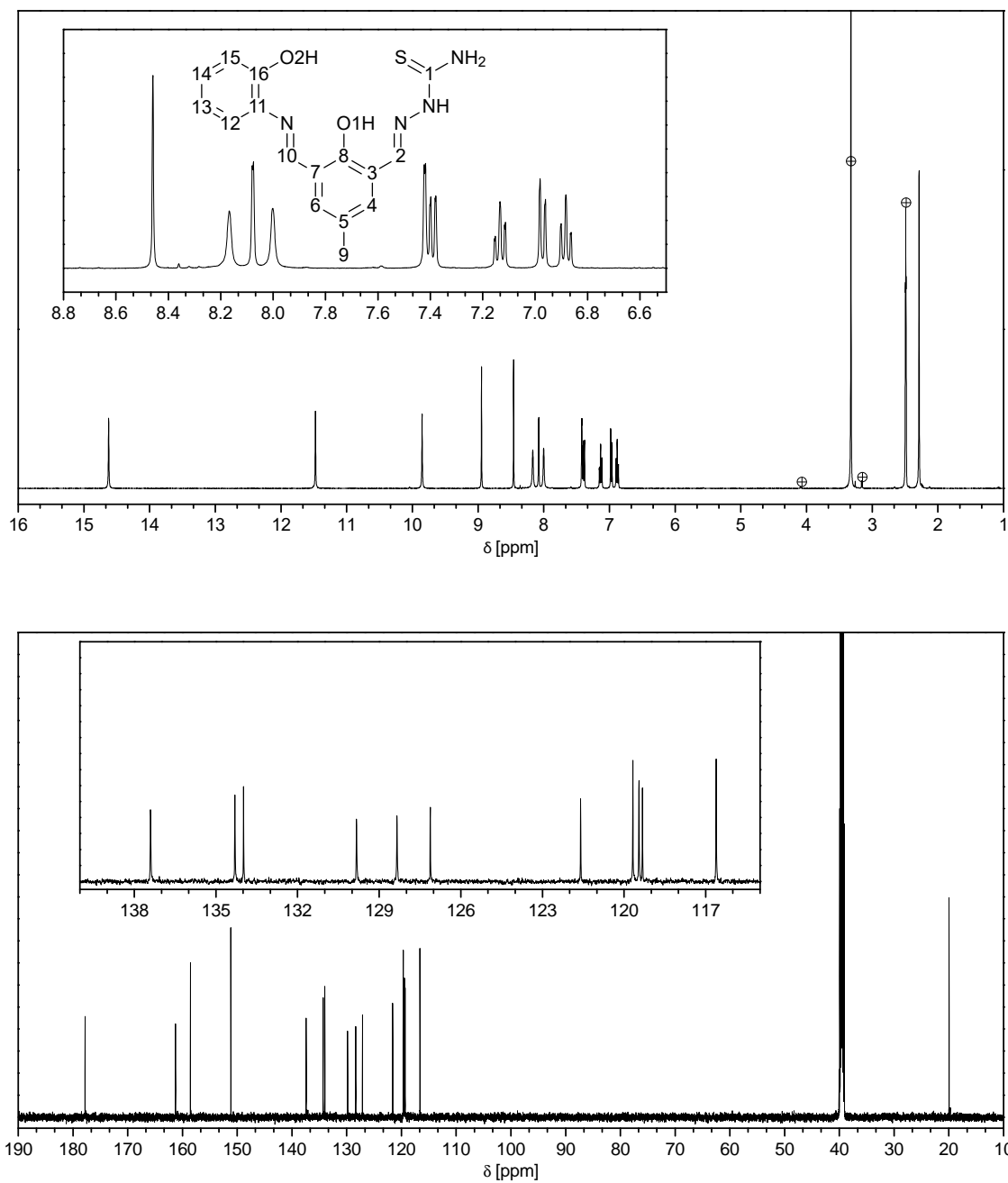


Figure S14: NMR spectra of H₃tsc-amph in DMSO-d₆: (Top) ¹H NMR; marked peaks correspond to water, methanol and DMSO. (Bottom) ¹³C{¹H} NMR.

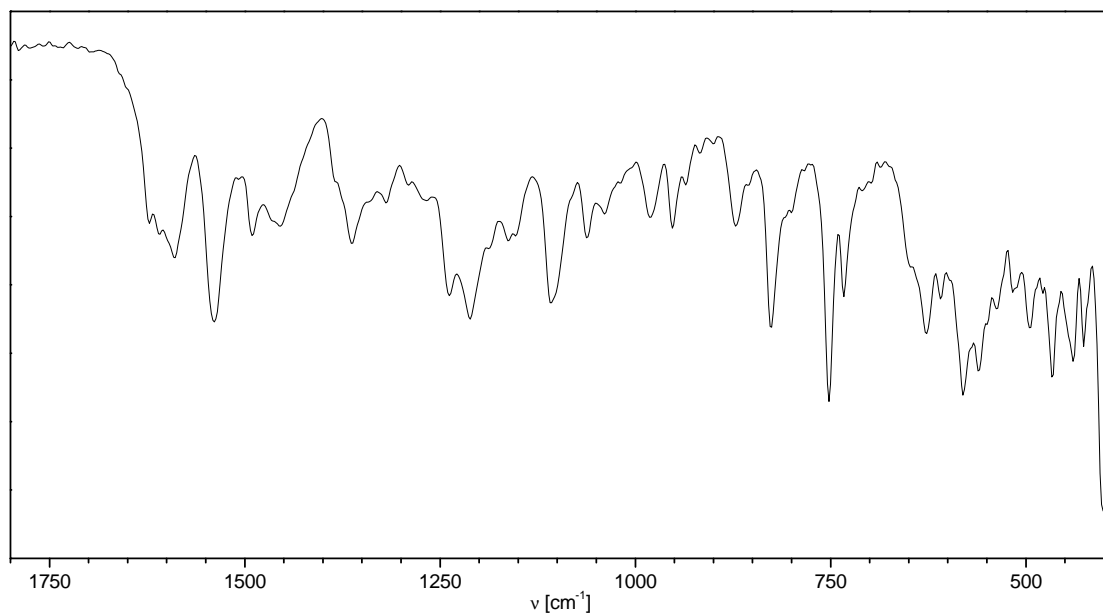
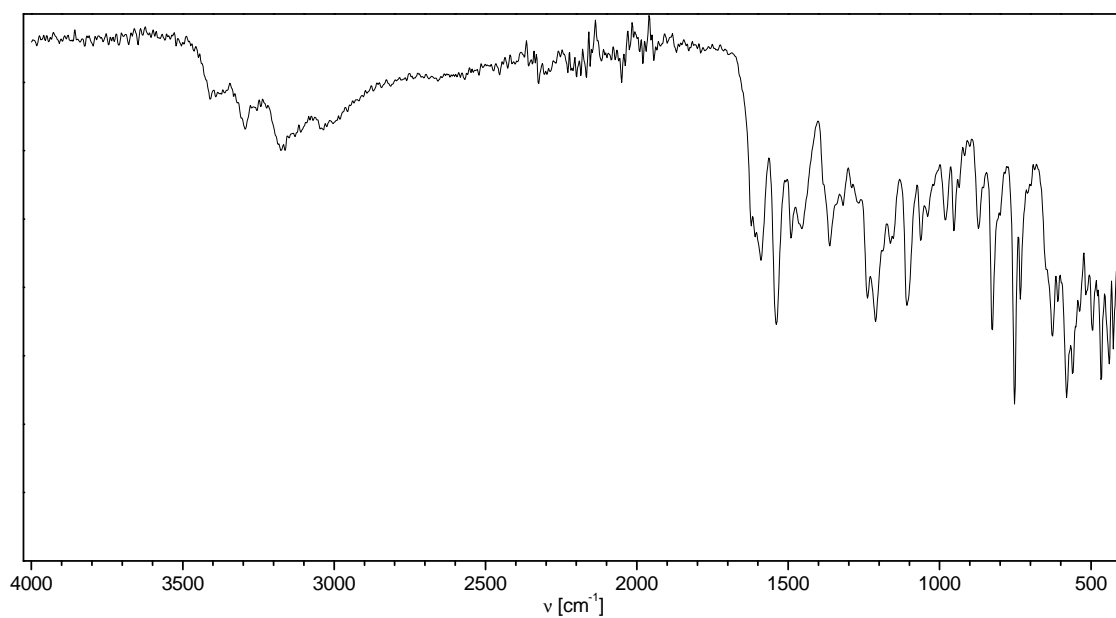


Figure S15: FT-IR spectra of H₃tsc-amph: (Top) Overview range from 4000–400 cm⁻¹. (Bottom) Fingerprint region from 1800–400 cm⁻¹. List of the positions of vibrational bands (in cm⁻¹): 3411, 3294, 3177, 1589, 1541, 1490, 1457, 1363, 1320, 1238, 1213, 1159, 1106, 1065, 1035, 979, 951, 935, 874, 828, 750, 732, 625, 605, 582, 561, 538, 495, 473, 439, 424.

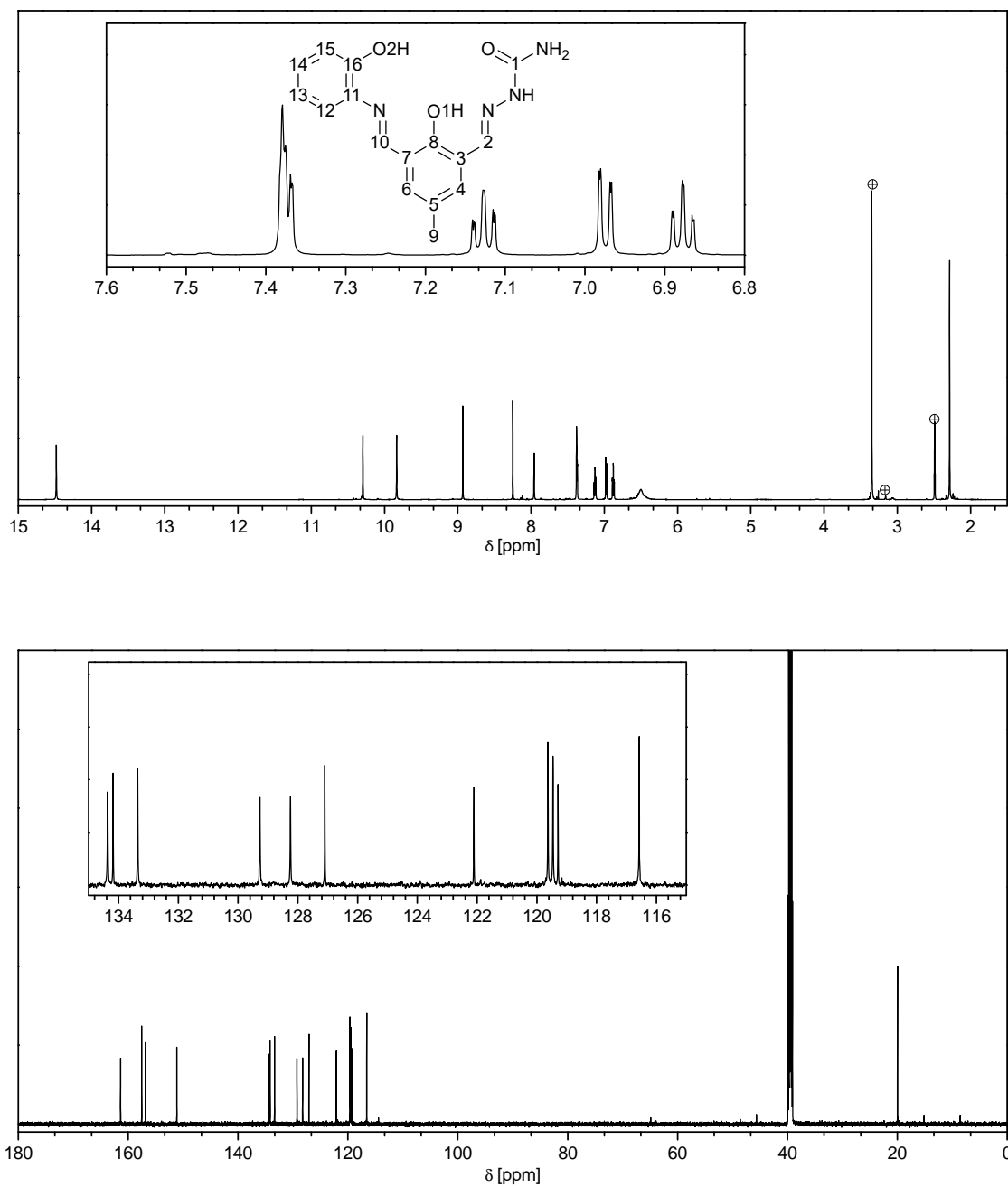


Figure S16: NMR spectra of H₃sc-amph in DMSO-d₆: (Top) ¹H NMR; marked peaks correspond to water, methanol and DMSO. (Bottom) ¹³C{¹H} NMR.

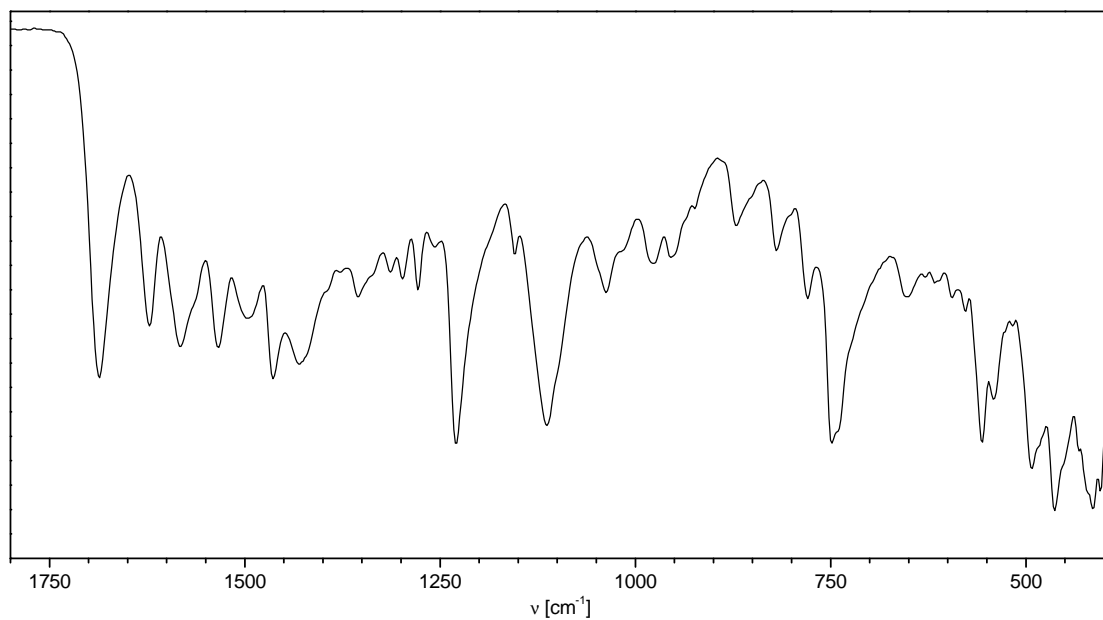
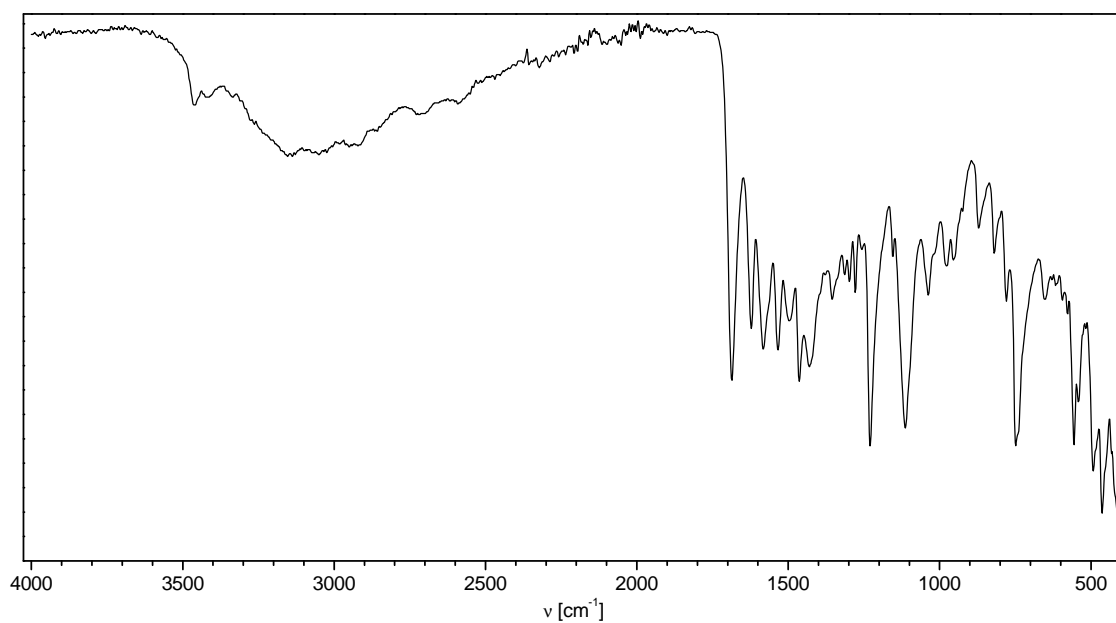


Figure S17: FT-IR spectra of H₃sc-amph: (Top) Overview range from 4000–400 cm⁻¹. (Bottom) Fingerprint region from 1800–400 cm⁻¹. List of the positions of vibrational bands (in cm⁻¹): 3470, 3146, 3047, 2700, 1683, 1620, 1579, 1533, 1500, 1465, 1434, 1355, 1282, 1231, 1157, 1111, 1035, 976, 956, 872, 821, 783, 750, 656, 556, 541, 493, 462, 414.

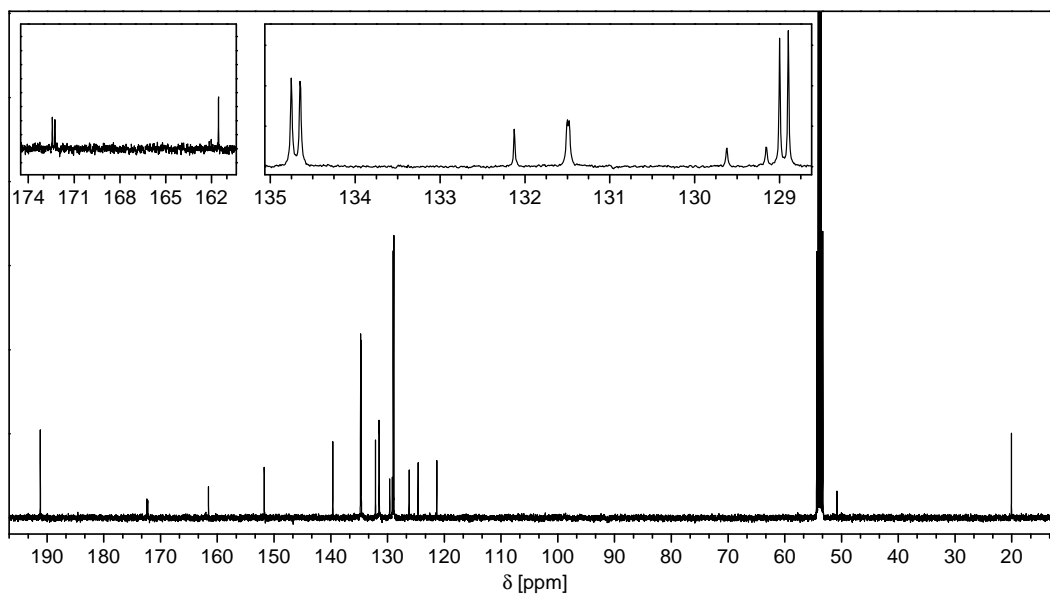
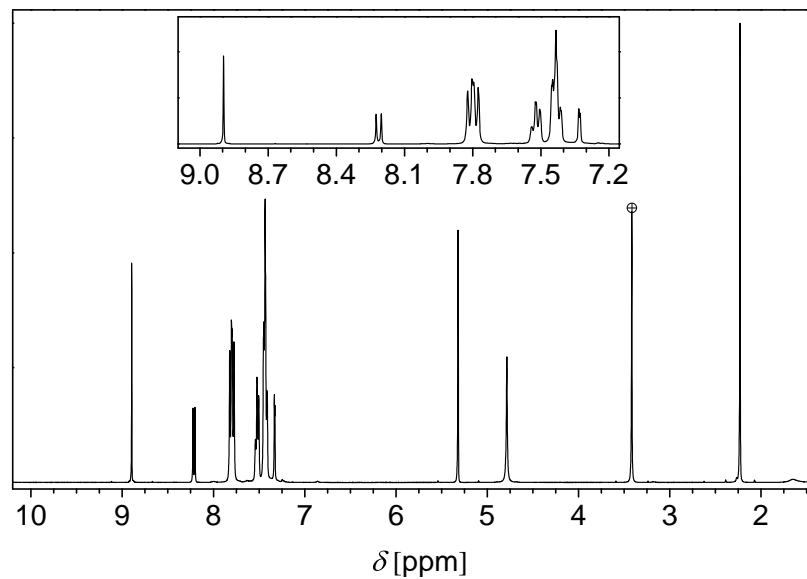


Figure S18: NMR spectra of $[\text{Ni}(\text{tsc-difo})\text{PPh}_3]$ in CD_2Cl_2 : (Top) ^1H NMR; marked peak corresponds to methanol which is included in the crystal packing. (Bottom) $^{13}\text{C}\{^1\text{H}\}$ NMR. For notation of the ligand atoms cf. Fig. S2.

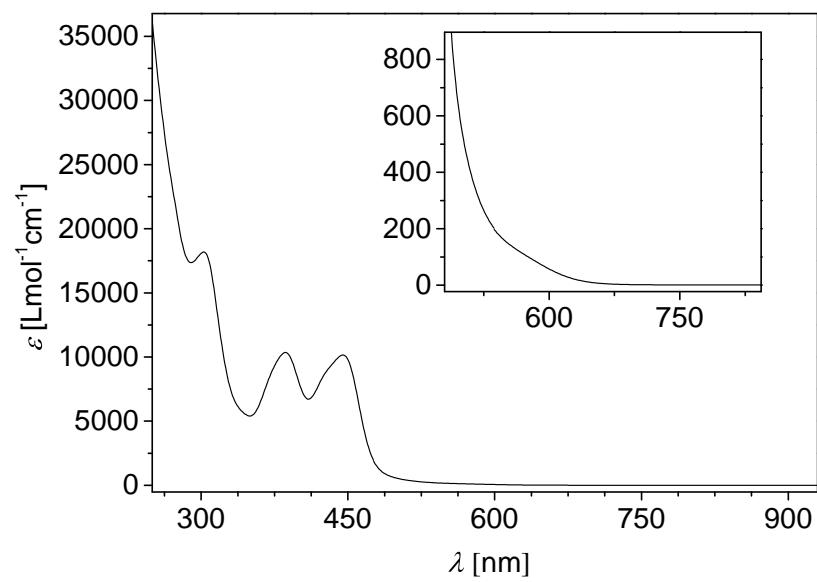


Figure S19: UV/Vis spectrum of $[\text{Ni}(\text{tsc-difo})\text{PPh}_3]$ in dichloromethane.

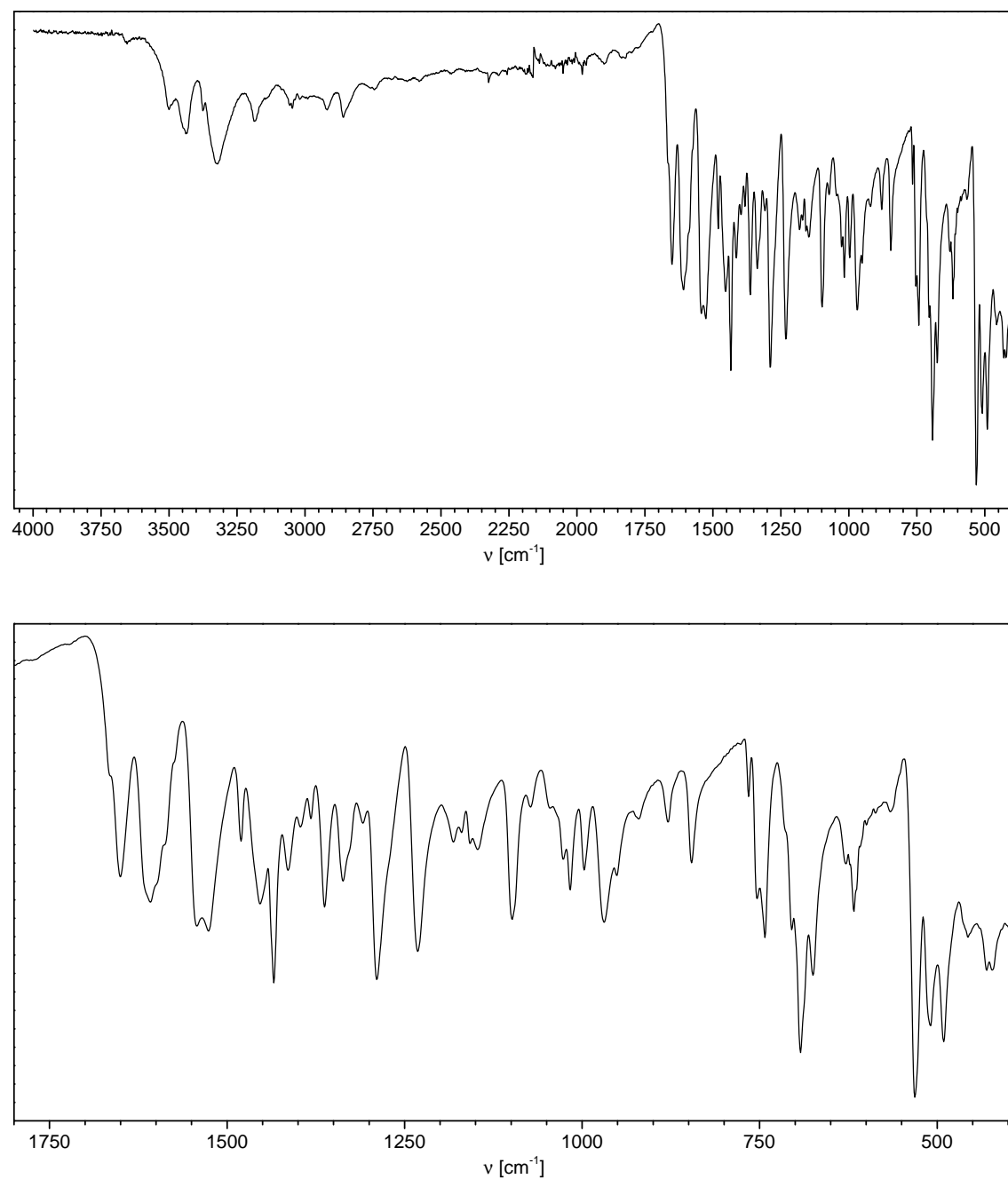


Figure S20: FT-IR spectra of $[\text{Ni}(\text{tsc-difo})\text{PPh}_3]$: (Top) Overview range from 4000–400 cm^{-1} . (Bottom) Fingerprint region from 1800–400 cm^{-1} . List of the positions of vibrational bands (in cm^{-1}): 3503, 3438, 3375, 3324, 3182, 3047, 2916, 2858, 1649, 1603, 1545, 1526, 1480, 1454, 1434, 1413, 1396, 1381, 1362, 1335, 1309, 1287, 1229, 1181, 1145, 1099, 1072, 1026, 1014, 997, 966, 952, 920, 877, 843, 766, 754, 742, 693, 674, 616, 565, 531, 507, 488, 457, 428.

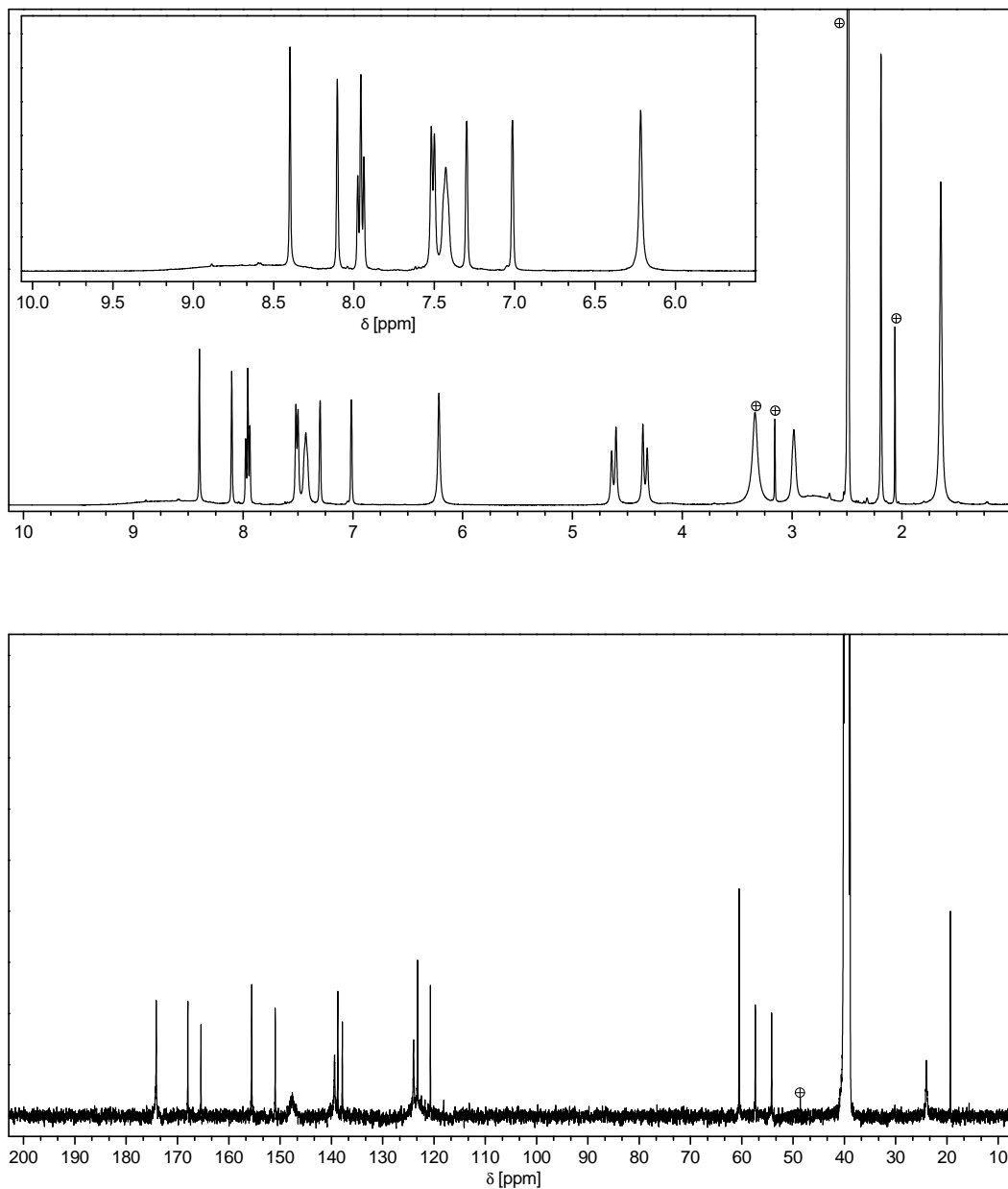


Figure S21: NMR spectra of $[\text{Zn}_2(\text{tsc-hydra})(\text{OAc})_2]$ in DMSO-d_6 : (Top) ^1H NMR; marked peak correspond to DMSO, methanol, acetonitrile and water. (Bottom) $^{13}\text{C}\{^1\text{H}\}$ NMR; marked peak corresponds to methanol. For notation of the ligand atoms cf. Fig. S10.

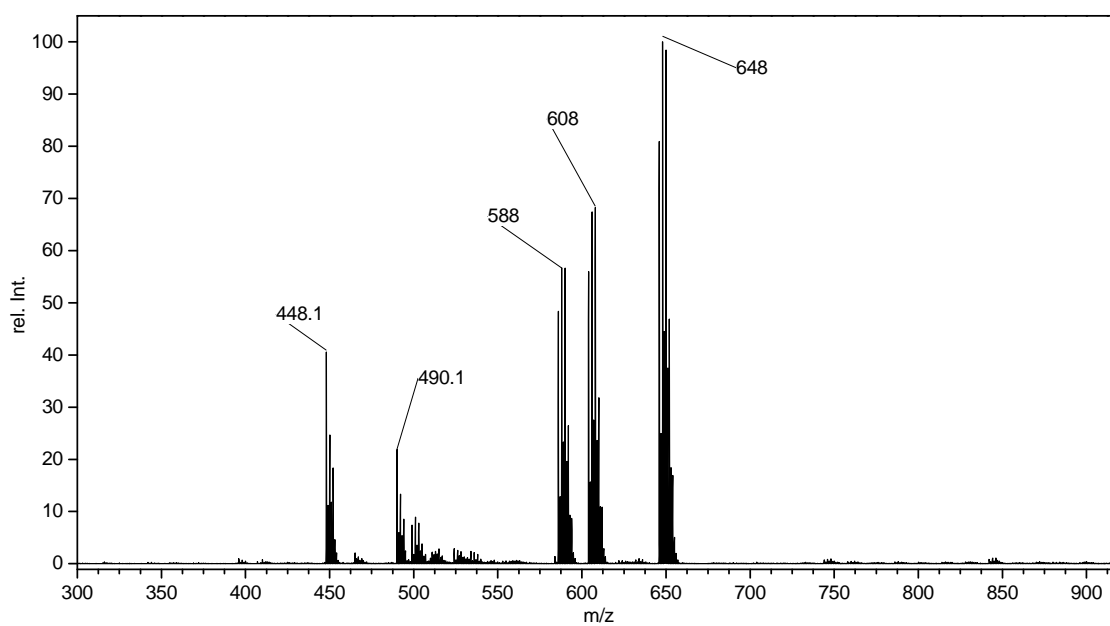


Figure S22: Mass spectrum in ESI positive mode of $[\text{Zn}_2(\text{tsc-hydra})(\text{OAc})_2]$ in acetonitrile.

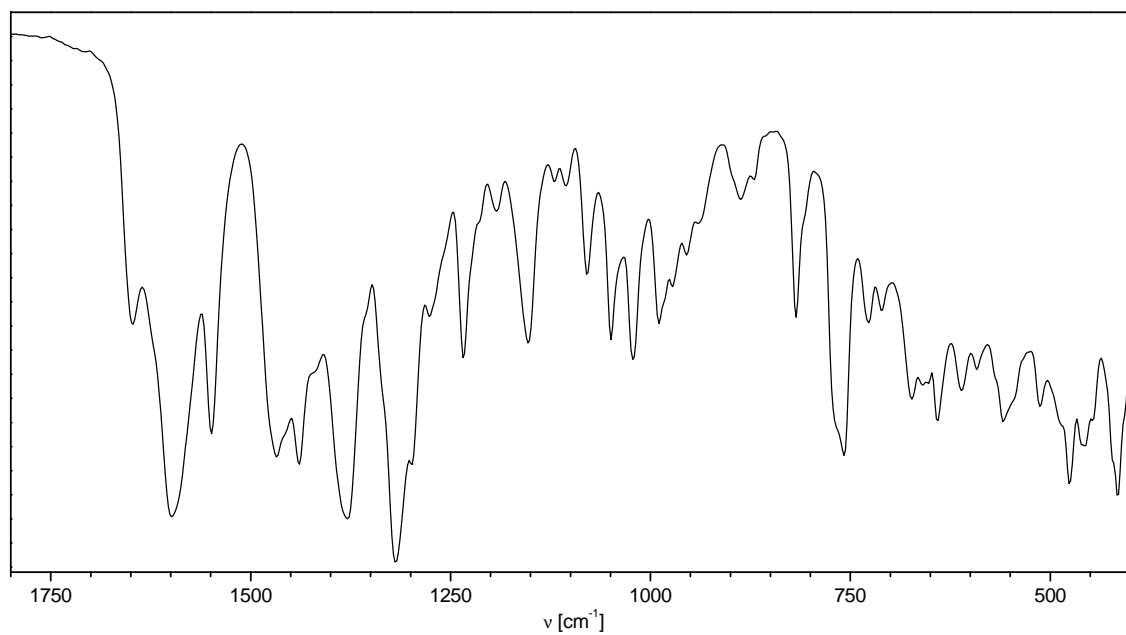
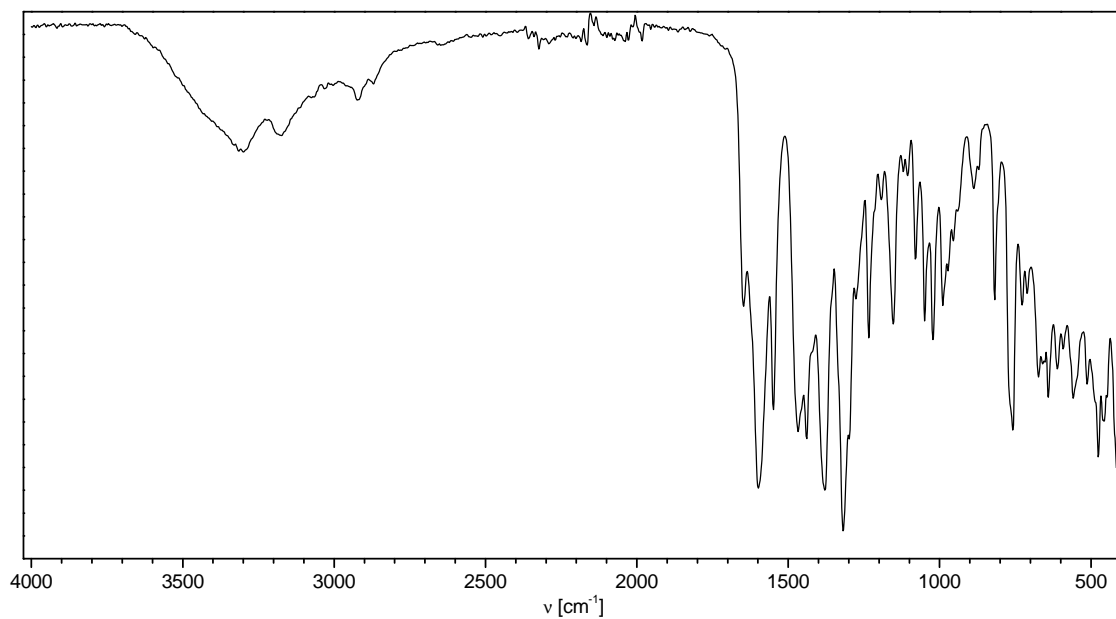


Figure S23: FT-IR spectra of $[\text{Zn}_2(\text{tsc-hydra})(\text{OAc})_2]$: (Top) Overview range from 4000–400 cm^{-1} . (Bottom) Fingerprint region from 1800–400 cm^{-1} . List of the positions of vibrational bands (in cm^{-1}): 3309, 3180, 2921, 1649, 1598, 1550, 1466, 1438, 1378, 1319, 1296, 1276, 1232, 1193, 1152, 1104, 1079, 1051, 1019, 990, 971, 953, 937, 886, 820, 758, 728, 710, 676, 642, 614, 594, 561, 516, 477, 458, 415.

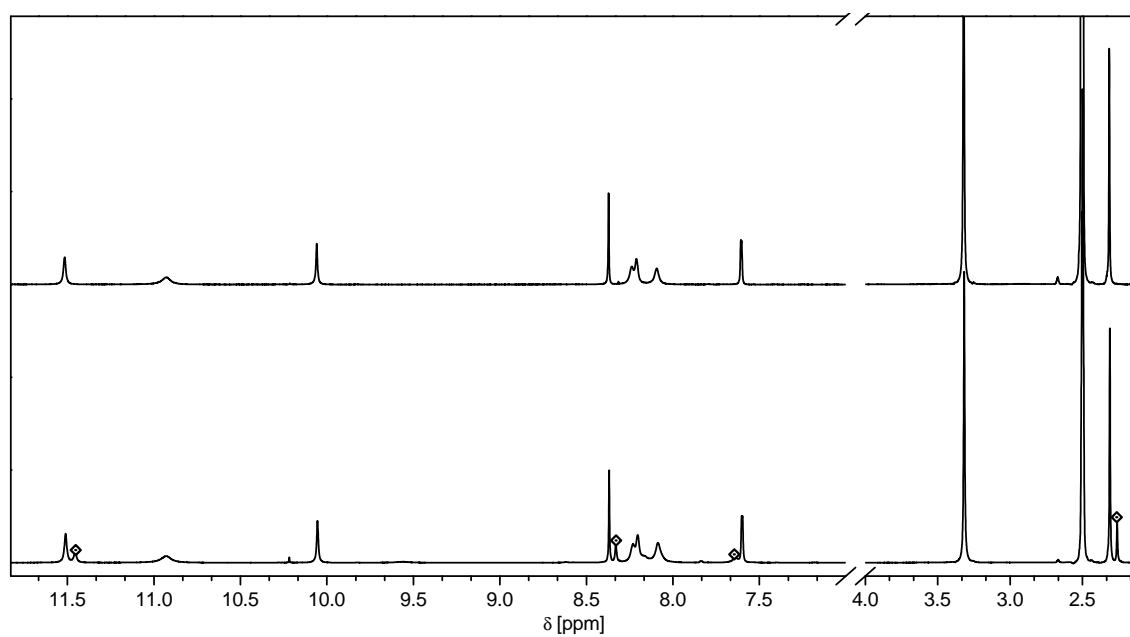


Figure S24: Comparison of the ^1H NMR spectra of $\text{H}_2\text{tsc-difo}$ before (lower line) and after purification by Soxhlet extraction with chloroform; both spectra were measured in DMSO-d_6 ; marked signals correspond to the symmetric byproduct.

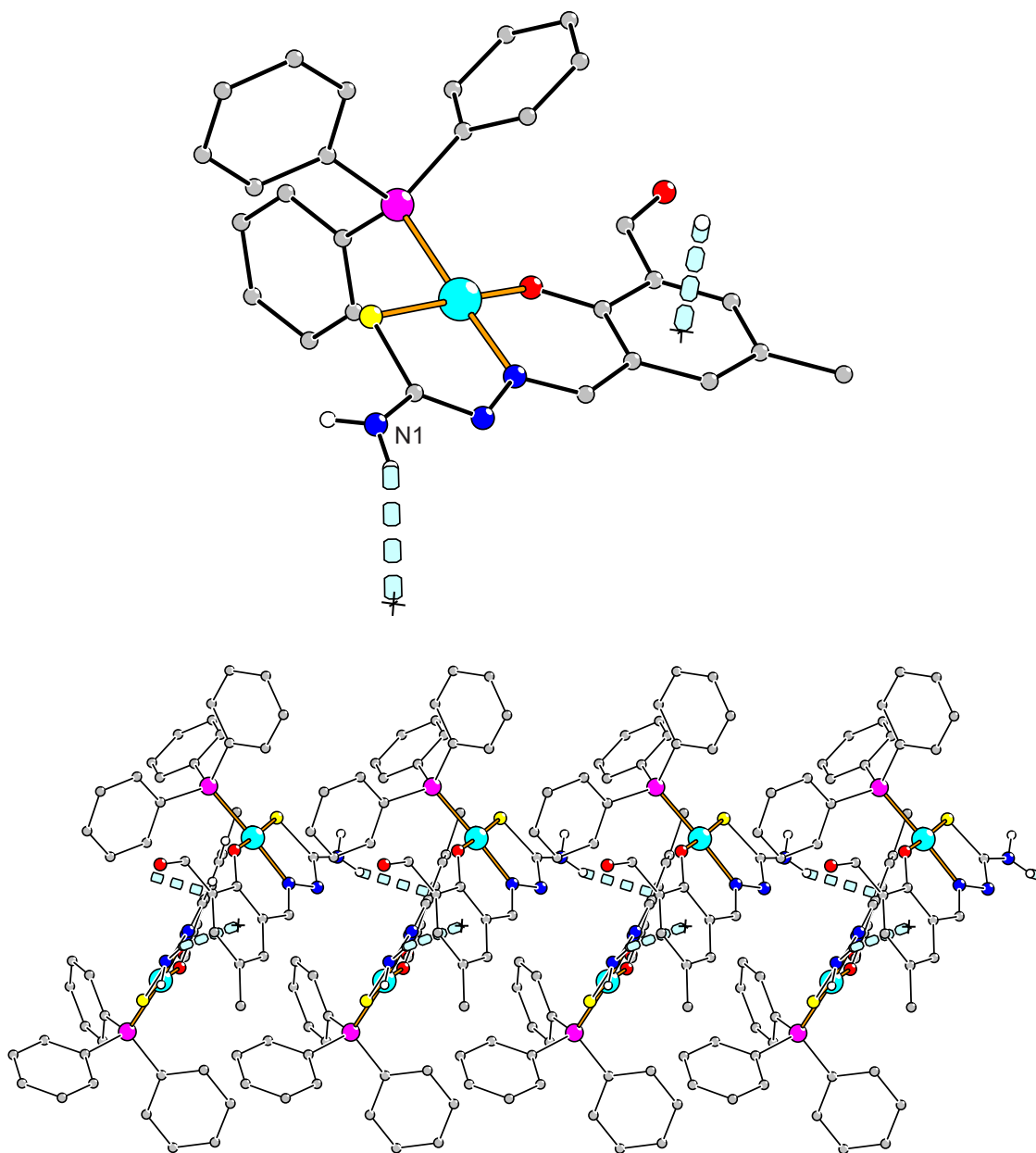


Figure S25: NH/ π interactions in the crystal structure of $[\text{Ni}(\text{tsc-difo})\text{PPh}_3] \cdot 2\text{MeOH}$: (Top) Interactions at one complex molecule. (Bottom) Resulting zig-zag chain like arrangement along the $[010]$ direction. Pertinent distances and angles: $\text{N1} \cdots \text{centroid}$ 326 pm, $\text{H} \cdots \text{centroid}$ 242 pm, $\text{N1-H} \cdots \text{centroid}$ 162° .

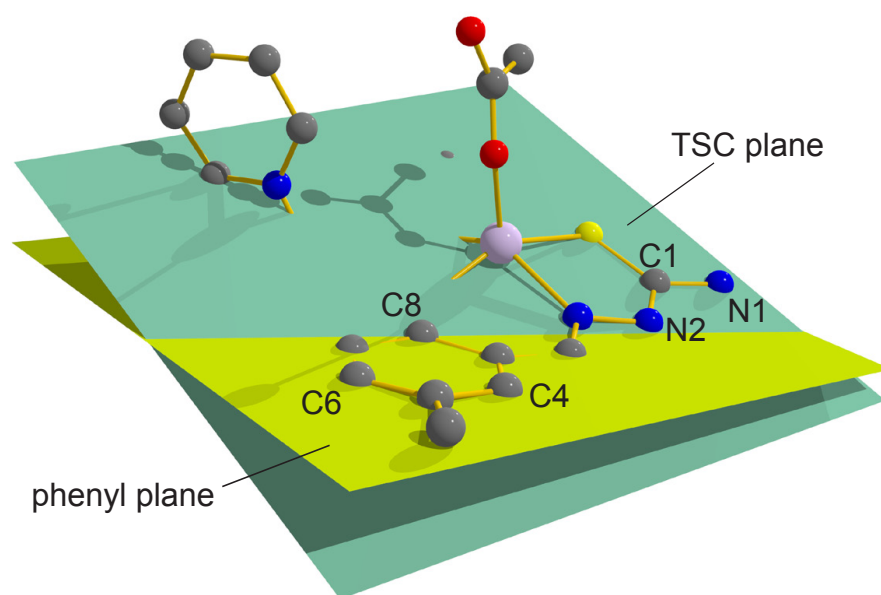


Figure S26: Representation of the mean planes for the thiosemicarbazone (N1, N2, C1) and the phenolate fragment (C4, C6, C8) in the crystal structure of $[\text{Zn}_2(\text{tsc-hydra})(\text{OAc})_2] \cdot \text{MeCN} \cdot \text{H}_2\text{O}$ indicating the rotation about the C2–C3 bond.

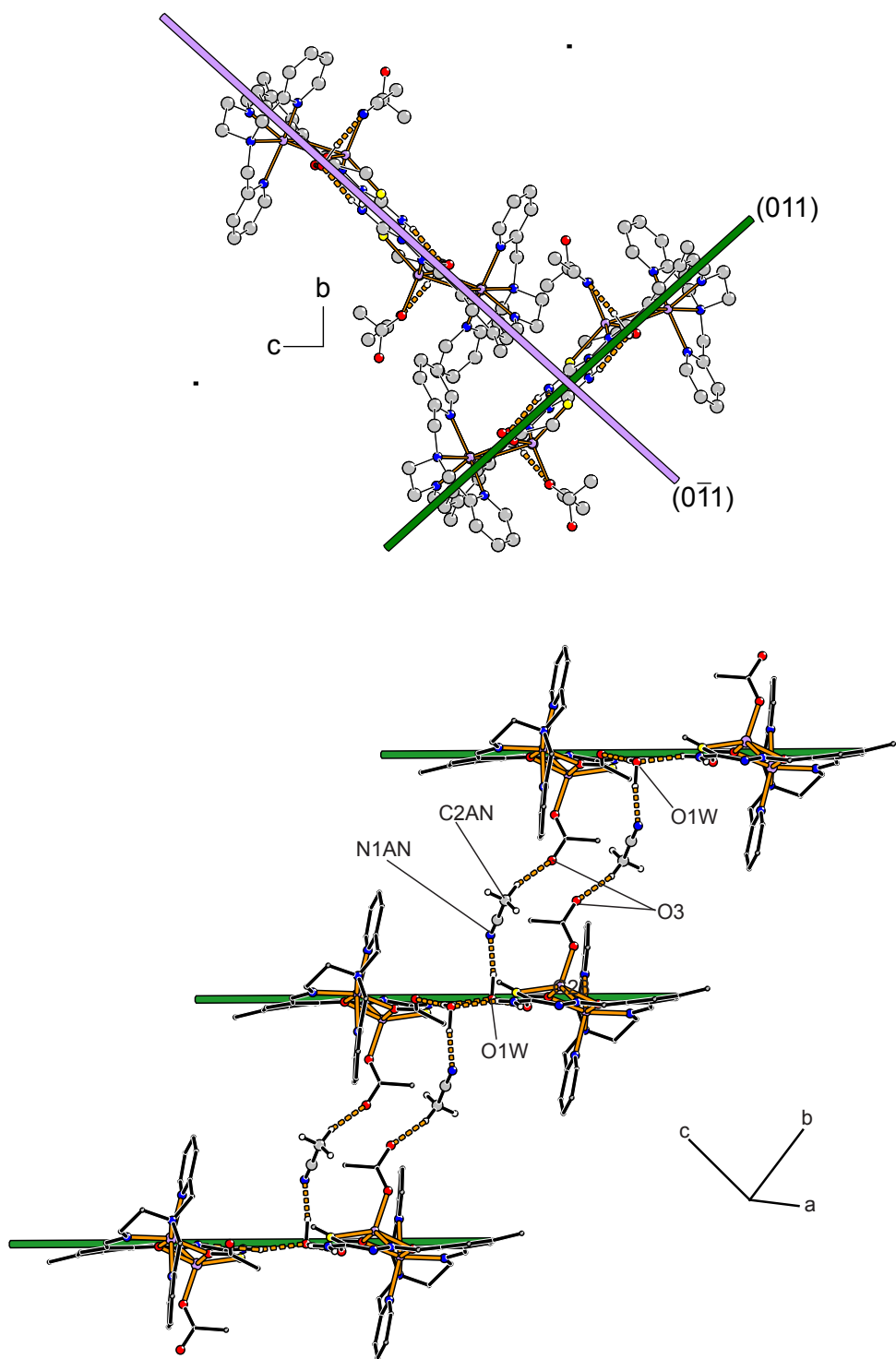


Figure S27: (Top) Representation of the two orientations of the hydrogen bonded chains along the $[100]$ direction in the crystal structure of $[\text{Zn}_2(\text{tsc-hydra})(\text{OAc})_2] \cdot \text{MeCN} \cdot \text{H}_2\text{O}$. (Bottom) Interaction of the acetonitrile molecules with chains of the same orientation. Pertinent distances: $\text{O1W} \cdots \text{N1AN}$ 298.4(9) and $\text{O3} \cdots \text{C2AN}$ 318.2(9) pm.

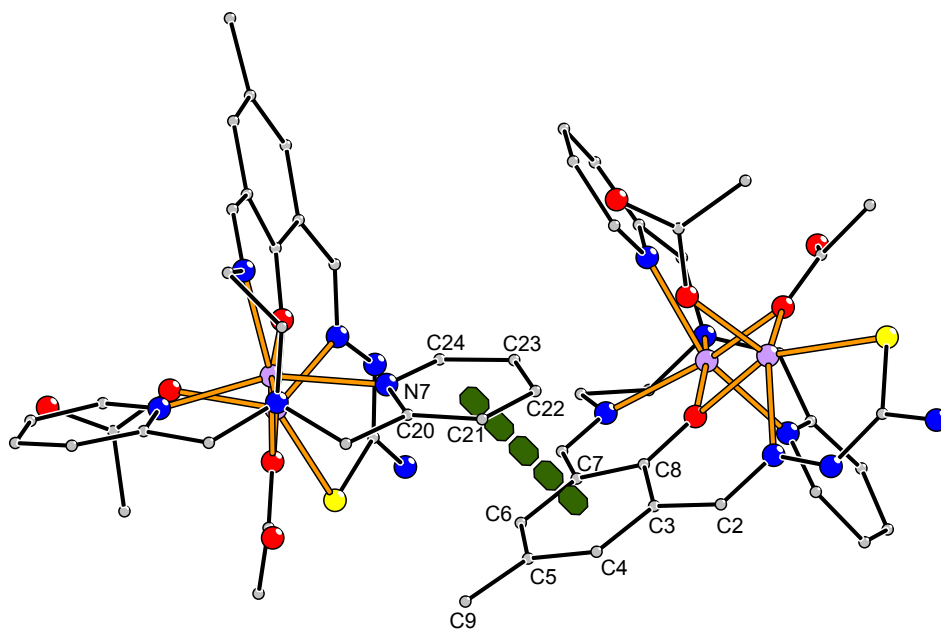


Figure S28: Representation of the π - π interaction in the crystal structure of $[\text{Zn}_2(\text{tsc-hydra})(\text{OAc})_2] \cdot \text{MeCN} \cdot \text{H}_2\text{O}$. Pertinent data: centroid \cdots centroid distance 383.5 pm, shortest C-C contact 346 pm, angle between π planes 19° .