

Selective Fluorescence Detection of Anilines and Fe³⁺ ions by Two Lanthanide Metal – Organic Frameworks

Keheng Xu¹, Fengqin Wang^{*1}, Shuo Huang¹, Zongchao Yu¹, Jianxin Zhang¹, Jianguo Yu¹, Haiyan Gao², Yiyuan Fu¹, Xiuyu Li¹, Yongnan Zhao²

¹College of Environment and Chemical Engineering & State Key Laboratory of Separation Membranes and Membrane Processes, Tianjin Polytechnic University, Tianjin 300387, China

²College of Materials and Engineering & Key Lab of Hollow Fiber Membrane Materials & Membrane Process, Tianjin Polytechnic University, Tianjin 300387, China

E-mail: wangfengqin@tjpu.edu.cn Tel: (+86)-22-83955457

Supplementary materials

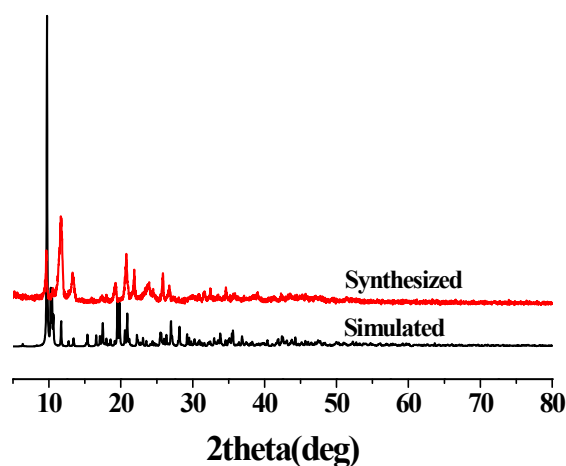


Fig. S1. The XRD patterns of 1 and the simulated.

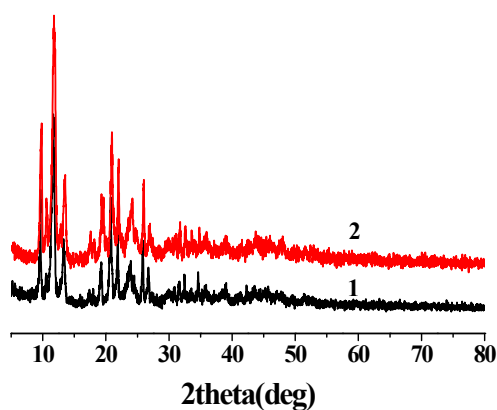


Fig. S2. The XRD patterns of 1 and 2.

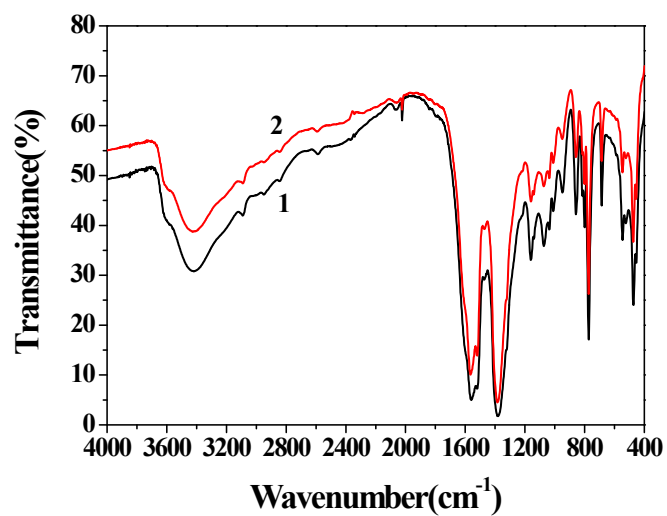


Fig. S3. IR spectra of 1 and 2.

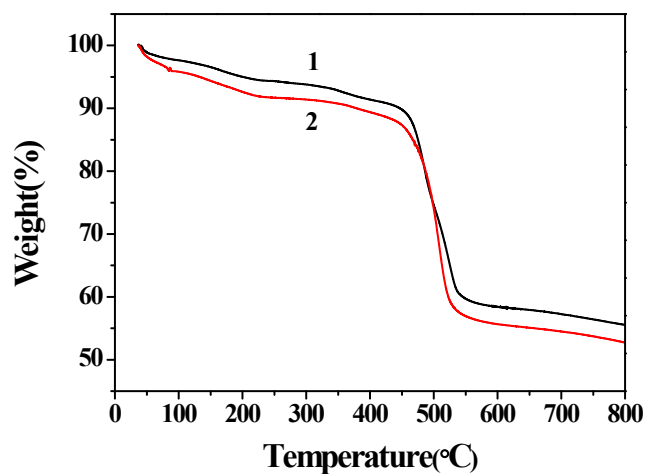


Fig. S4. TGA of 1 and 2.

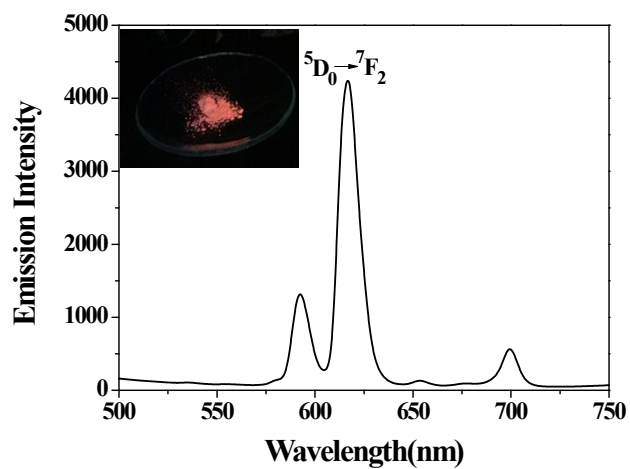


Fig. S5. The fluorescence spectra of 1 in the solid state.

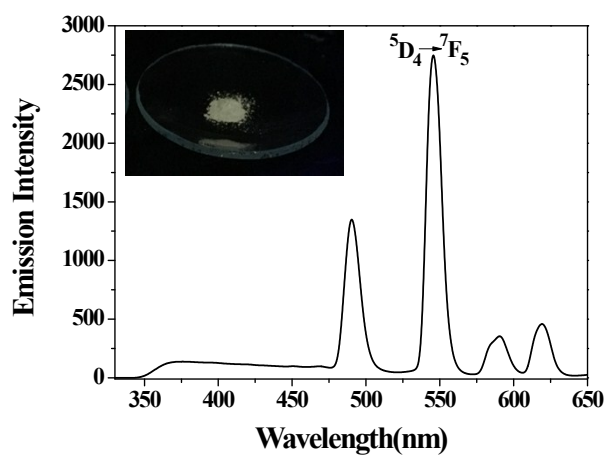


Fig. S6. The fluorescence spectra of **2** in the solid state.

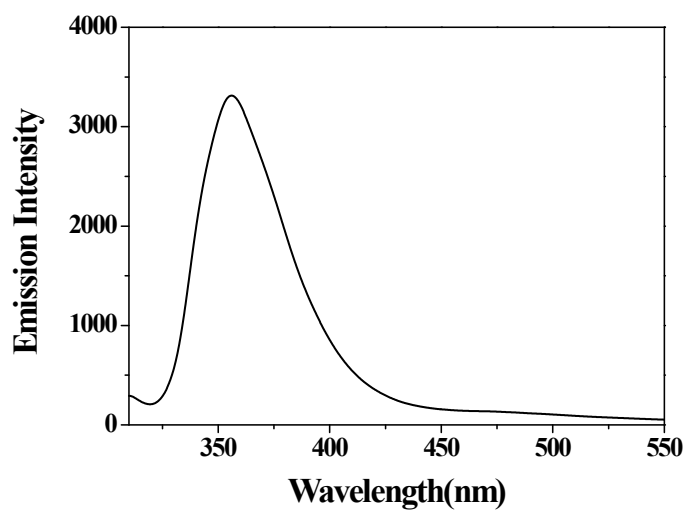


Fig. S7. The fluorescence spectra of L ligand in the solid state.

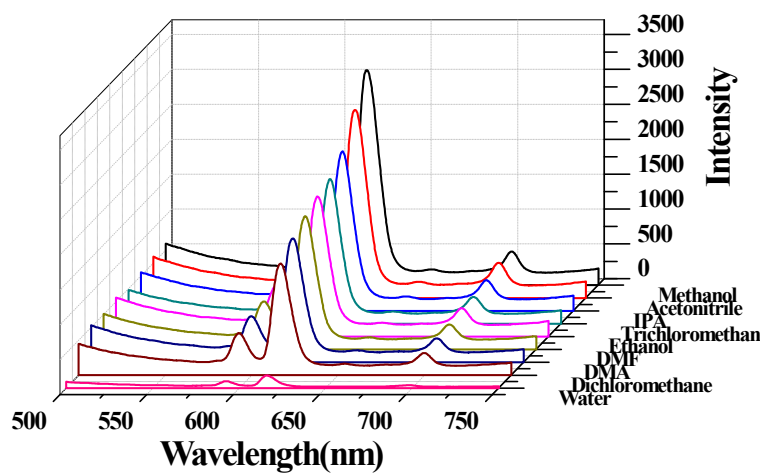


Fig. S8. Emission spectra of **1** in different organic solvent at room temperature.

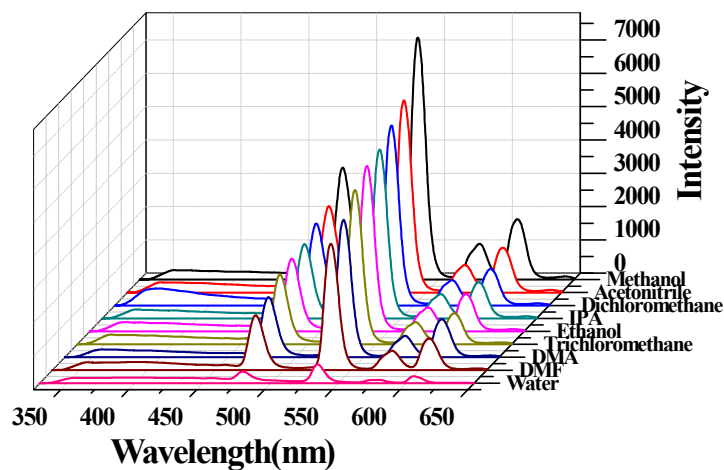


Fig. S9. Emission spectra of **2** in different organic solvent at room temperature.

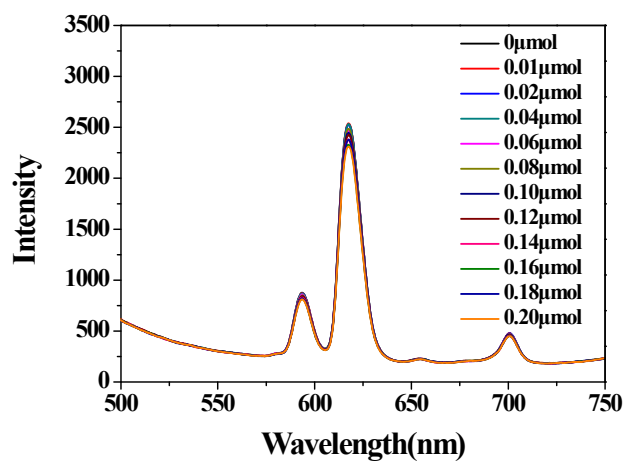


Fig. S10. Fluorescence titration of **1** dispersed in methanol with the addition of different volumes of 10^{-3} M methanol solution of aniline.

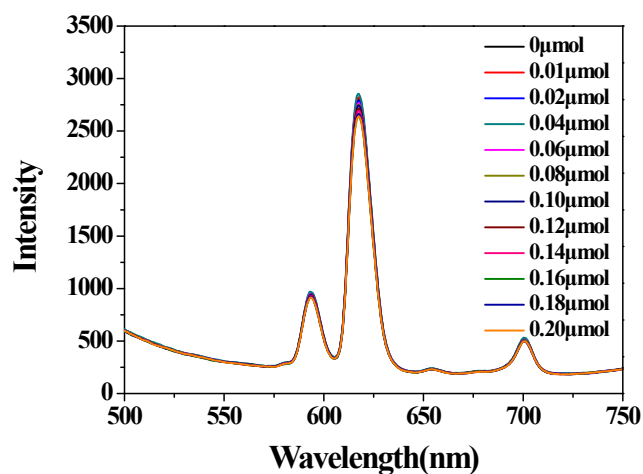


Fig. S11. Fluorescence titration of **1** dispersed in methanol with the addition of different volumes of 10^{-3} M methanol solution of *p*-methylaniline.

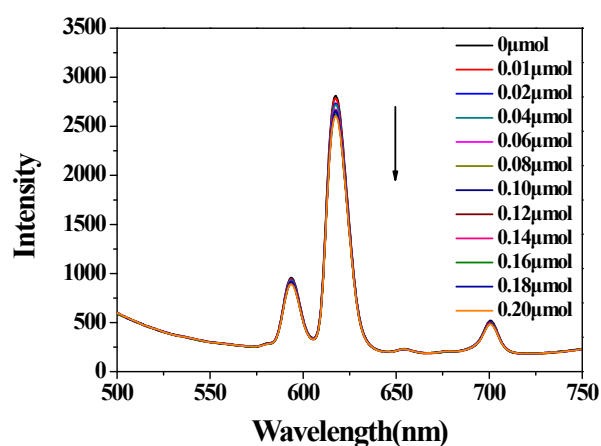


Fig. S12. Fluorescence titration of **1** dispersed in methanol with the addition of different volumes of 10^{-3} M methanol solution of diphenylamine.

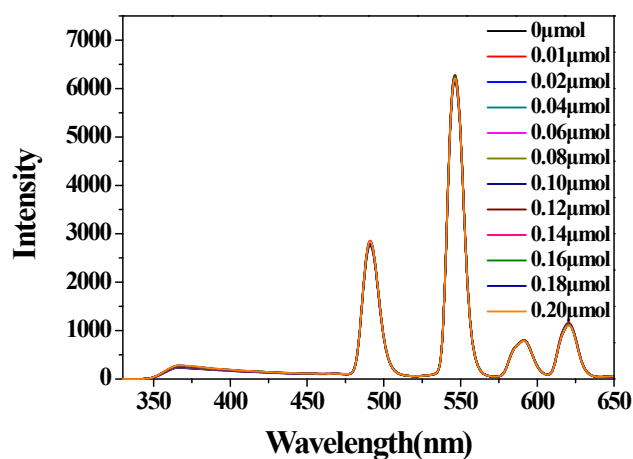


Fig. S13. Fluorescence titration of **2** dispersed in methanol with the addition of different volumes of 10^{-3} M methanol solution of aniline.

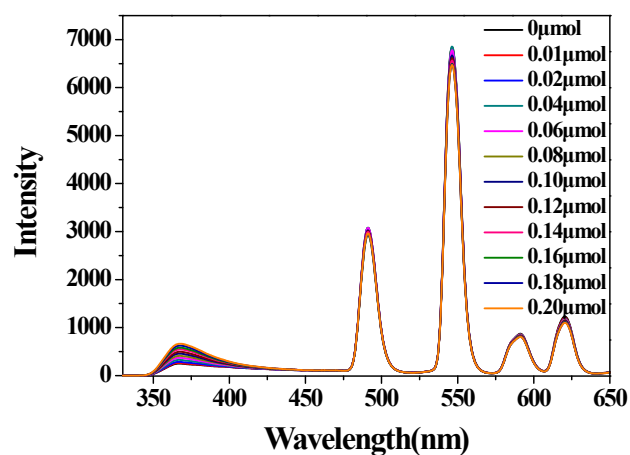


Fig. S14. Fluorescence titration of **2** dispersed in methanol with the addition of different volumes of 10^{-3} M methanol solution of *p*-methylaniline.

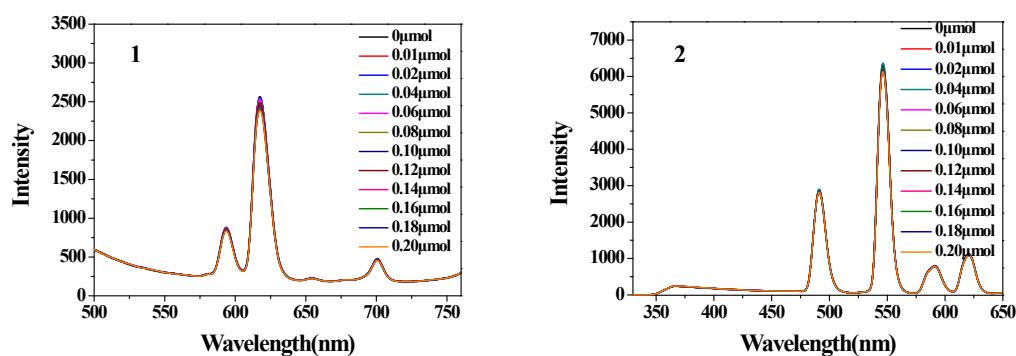


Fig. S15. Fluorescence titration of **1** and **2** dispersed in methanol with the addition of different volumes of 10^{-3} M methanol solution of triethylamine.

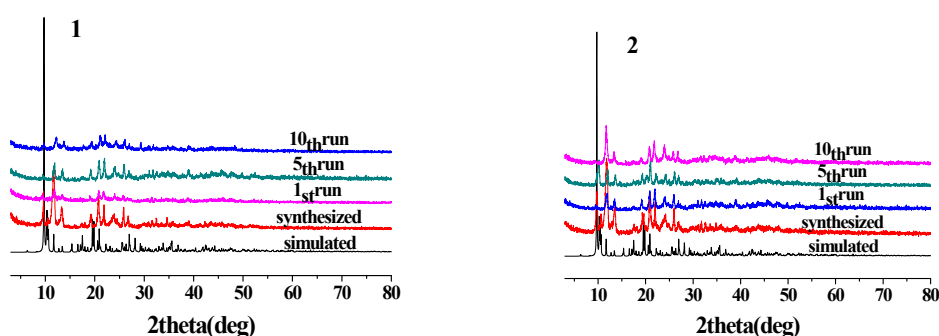


Fig. S16. The powder X-ray diffraction of **1** and **2** before and after the fluorescence titration.

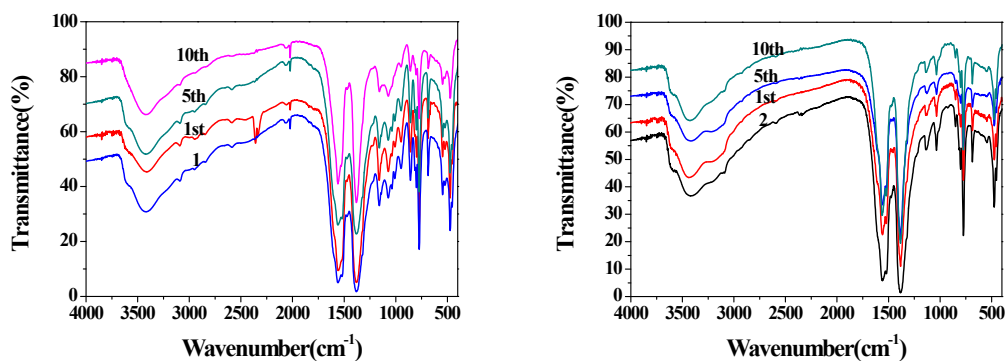


Fig. S17. IR spectra of **1** and **2** before and after the fluorescence titration.

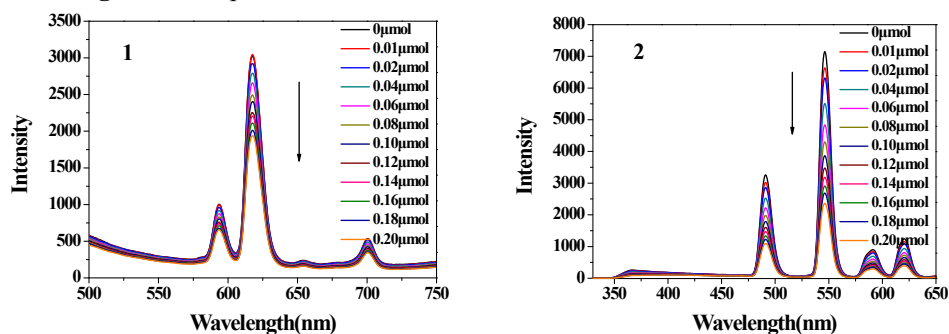


Fig. S18. Fluorescence titration of **1** and **2** dispersed in methanol with the addition of different volumes of 10^{-3} M methanol solution containing Fe^{3+} . The slit widths for excitation and emission were both 5 nm.

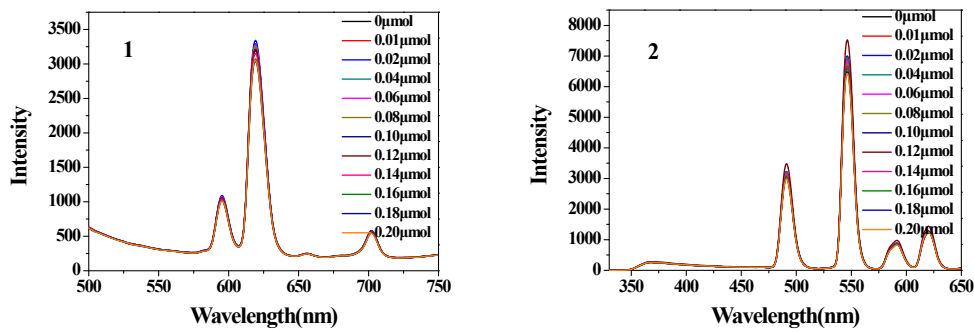


Fig. S19. Fluorescence titration of **1** and **2** dispersed in methanol with the addition of different volumes of 10^{-3} M methanol solution containing Ag^+ .

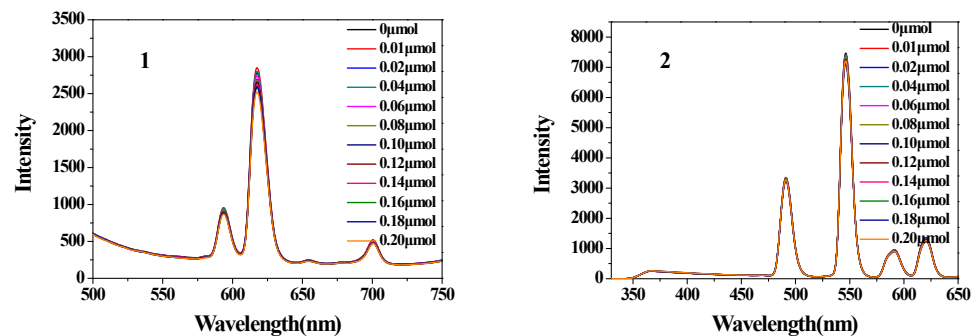


Fig. S20. Fluorescence titration of **1** and **2** dispersed in methanol with the addition of different volumes of 10^{-3} M methanol solution containing Al^{3+} .

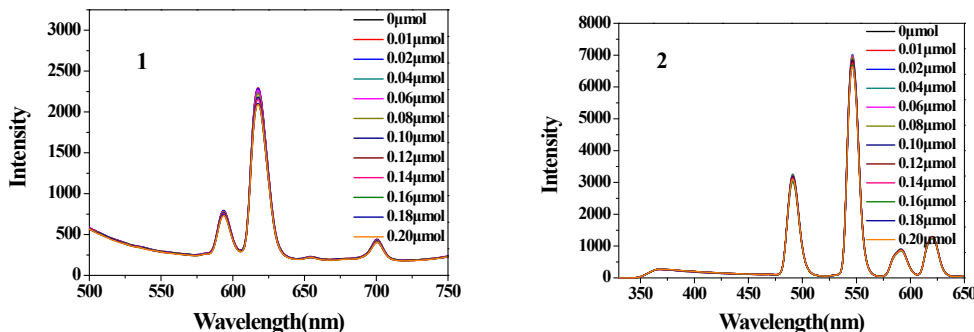


Fig. S21. Fluorescence titration of **1** and **2** dispersed in methanol with the addition of different volumes of 10^{-3} M methanol solution containing Ba^{2+} .

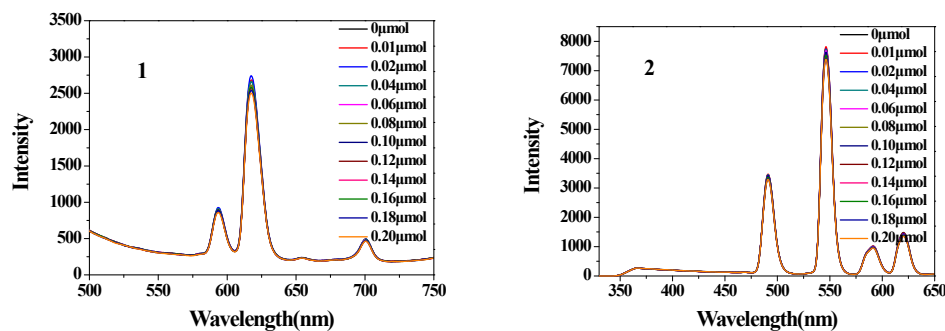


Fig. S22. Fluorescence titration of **1** and **2** dispersed in methanol with the addition of different volumes of 10^{-3} M methanol solution containing Zn^{2+} .

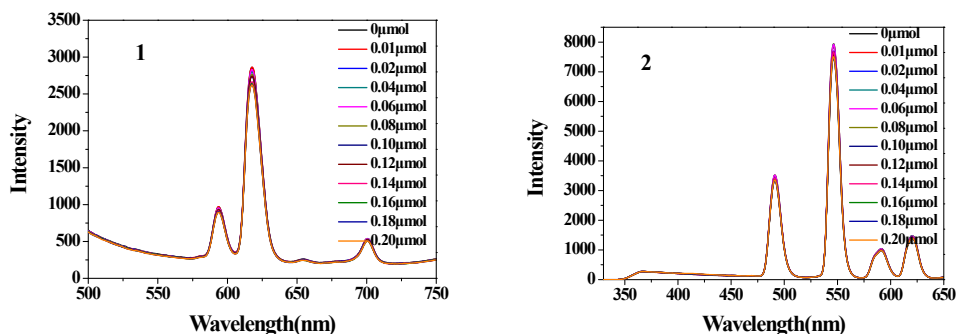


Fig. S23. Fluorescence titration of **1** and **2** dispersed in methanol with the addition of different volumes of 10⁻³ M methanol solution containing Ca²⁺.

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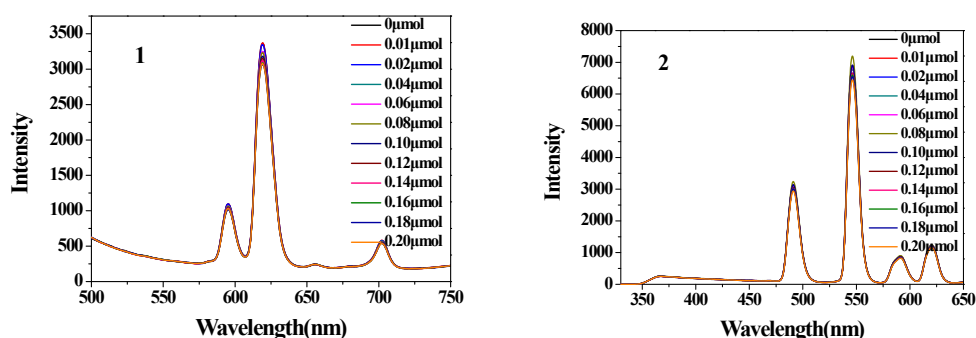


Fig. S24. Fluorescence titration of **1** and **2** dispersed in methanol with the addition of different volumes of 10⁻³ M methanol solution containing Co²⁺.

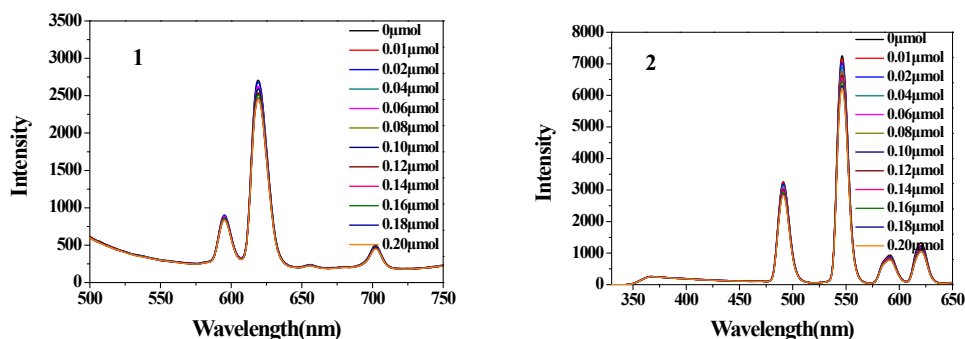


Fig. S25. Fluorescence titration of **1** and **2** dispersed in methanol with the addition of different volumes of 10⁻³ M methanol solution containing Cu²⁺.

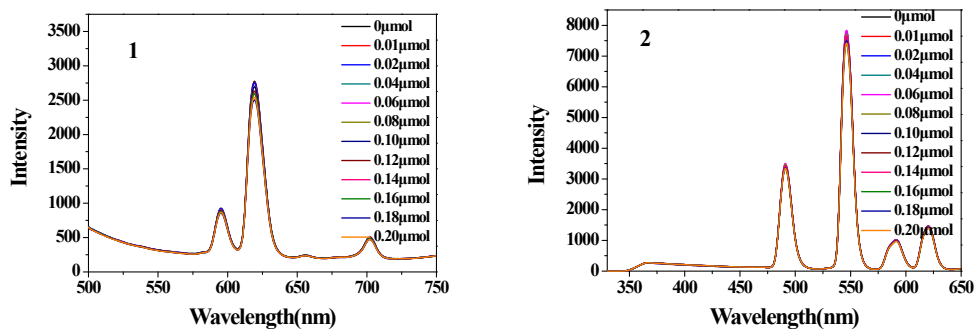


Fig. S26. Fluorescence titration of **1** and **2** dispersed in methanol with the addition of different volumes of 10⁻³ M methanol solution containing Li⁺.

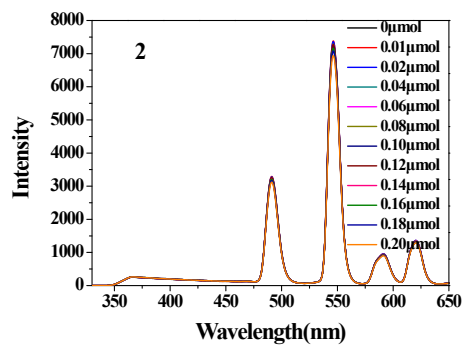
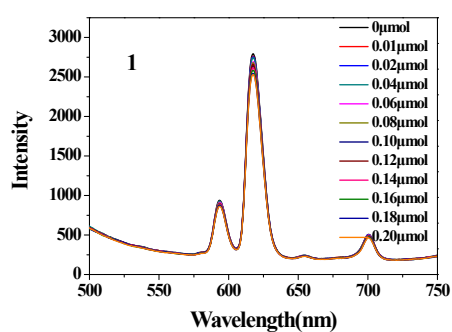


Fig. S27. Fluorescence titration of **1** and **2** dispersed in methanol with the addition of different volumes of 10^{-3} M methanol solution containing Mg^{2+} .

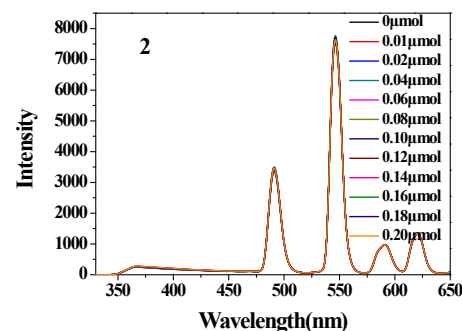
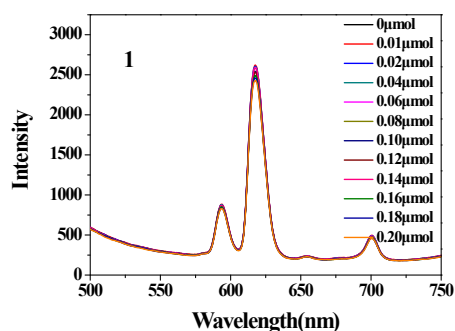


Fig. S28. Fluorescence titration of **1** and **2** dispersed in methanol with the addition of different volumes of 10^{-3} M methanol solution containing Na^{+} .

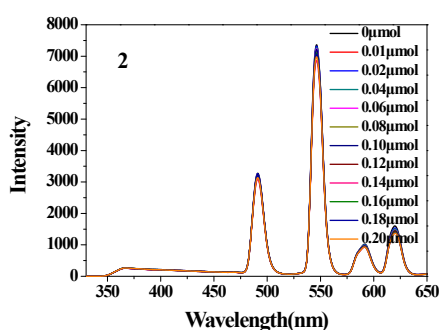
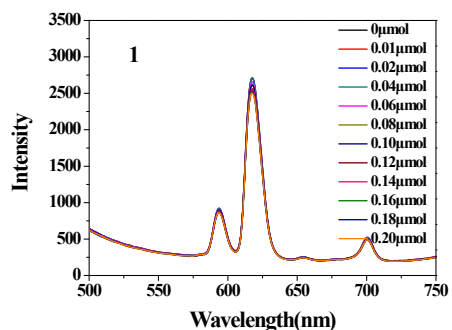


Fig. S29. Fluorescence titration of **1** and **2** dispersed in methanol with the addition of different volumes of 10^{-3} M methanol solution containing Ni^{2+} .

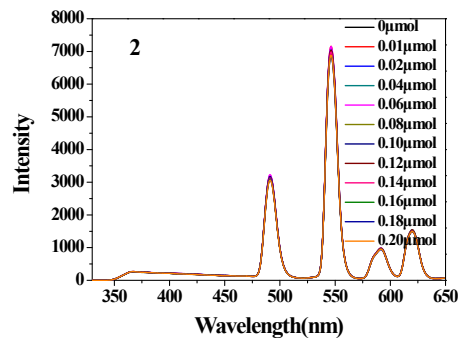
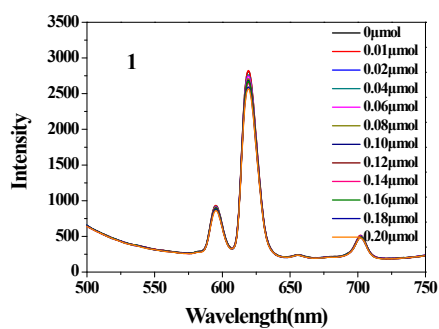


Fig. S30. Fluorescence titration of **1** and **2** dispersed in methanol with the addition of different volumes of 10^{-3} M methanol solution containing Pb^{2+} .