

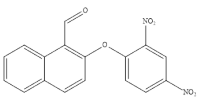
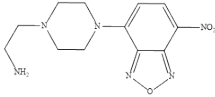
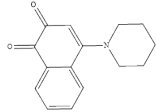
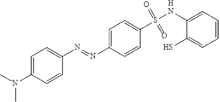
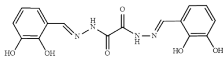
Supporting Information

A colorimetric chemosensor for sulfide in a near-perfect aqueous solution: Practical application using test kit

Seong Youl Lee, Cheal Kim*

Department of Fine Chemistry and Department of Interdisciplinary Bio IT Materials, Seoul National University of Science and Technology, Seoul 139-743, Korea. Fax: +82-2-973-9149; Tel: +82-2-970-6693; E-mail: chealkim@seoultech.ac.kr

Table S1. Examples of various S²⁻ chemosensors with colorimetric test strip

| Sensor | Detection limit (μM) | Binding constant | Interference | Percent of water in solution (%) | Method of detection | Concentration of sensor/S ²⁻ in test strip (μM) | Reference |
|---|-----------------------------------|-------------------|--------------|----------------------------------|----------------------------|---|-----------|
|  | 1.67 | No data | None | 50 | Fluorescence, Colorimetric | 20 / 200 | 1 |
|  | 46.97 | 5.0×10^2 | None | 99.9 | Colorimetric | 3000 / 50 | 2 |
|  | 0.77 | No data | No data | 50 | Fluorescence, Colorimetric | 20 / 100 | 3 |
|  | 28.4 | No data | No data | 99.9 | Colorimetric | 100 / 2000 | 4 |
|  | 28.7 | 1.0×10^2 | None | 99.9 | Colorimetric | 100 / 30 | This work |

References

- 1 A. K. Das, S. Goswami, C. K. Quah and H. Fun, *New J. Chem.*, 2015, **39**, 5669-5675.
- 2 J. J. Lee, Y. S. Kim, E. Nam, S. Y. Lee, M. H. Lim and C. Kim, *Dalton Trans.*, 2016, **45**, 5700-5712.
- 3 A. K. Das, S. Goswami, G. Dutta, S. Maity, T. kanti Mandal, K. Khanra and N. Bhattacharyya, *Org. Biomol. Chem.*, 2016, **14**, 570-576.
- 4 R. Kaushik, A. Singh, A. Ghosh and D. A. Jose, *ChemistrySelect*, 2016, **1**, 1533-1540.

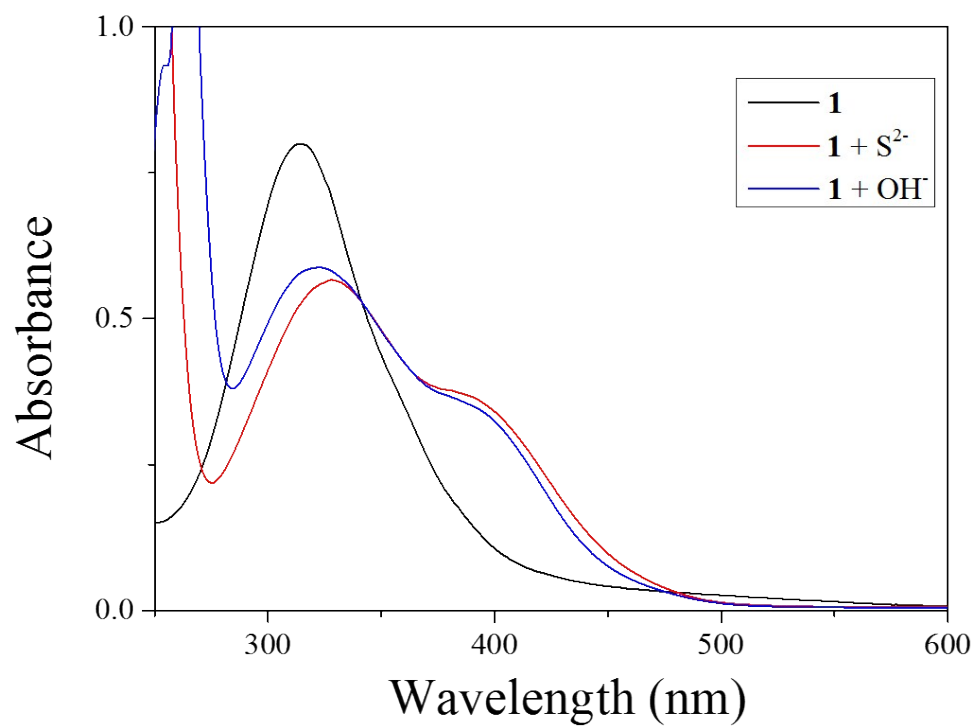


Fig. S1. Absorption changes of **1** (40 μ M) in the presence of NaOH (160 equiv) and Na_2S (100 equiv), respectively, in bis-tris buffer.

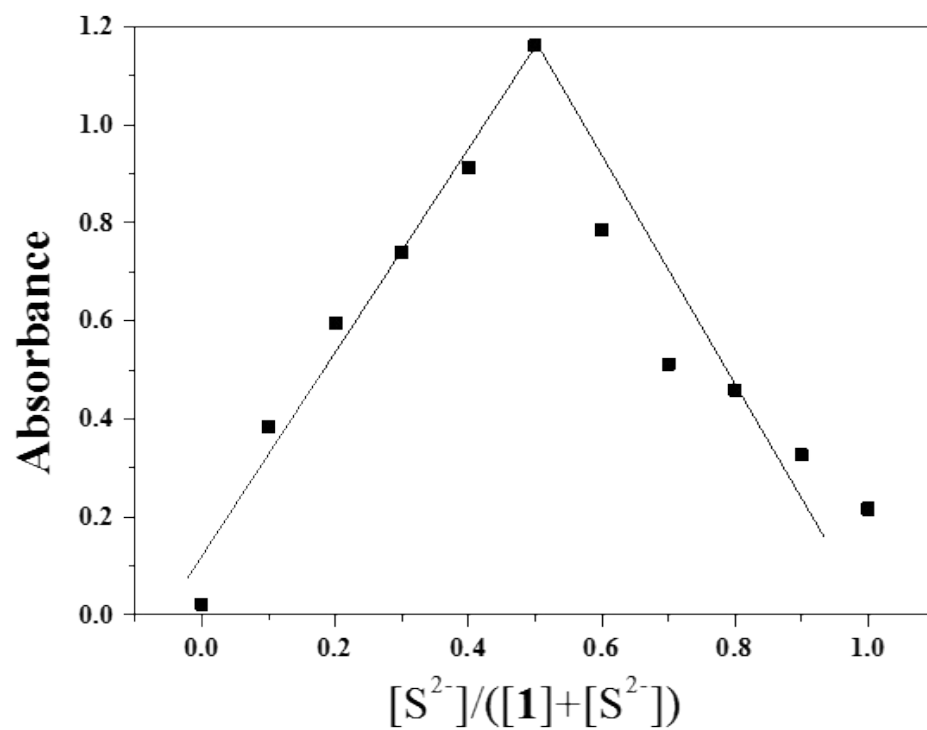


Fig. S2 Job plot of a 1:1 complex of receptor **1** and S²⁻ (absorbance at 400 nm). The total concentration of S²⁻ with receptor **1** was 100 μ M.

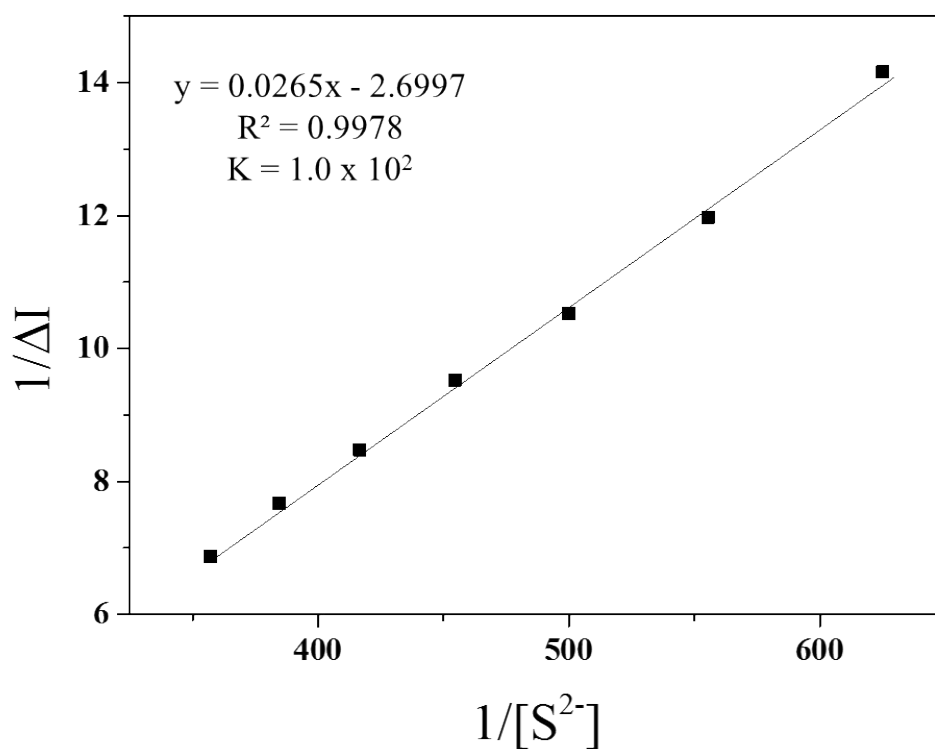


Fig. S3 Benesi-Hildebrand plot (at 400 nm) of **1** based on UV-vis titration, assuming 1:1 stoichiometry for association between **1** and S^{2-} .

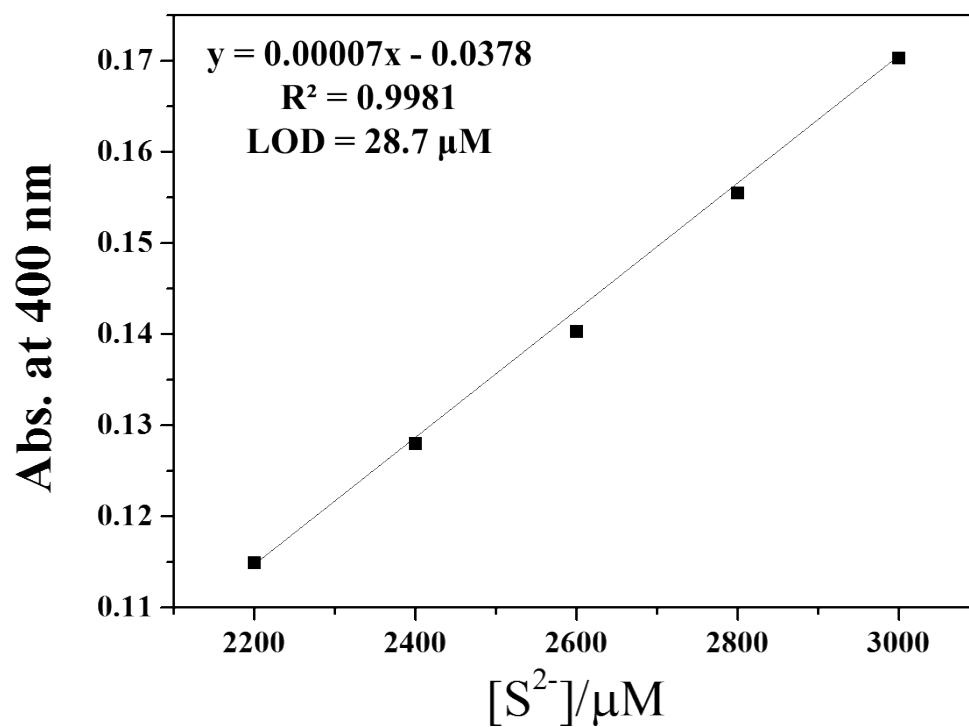


Fig. S4 Determination of the detection limit based on UV-vis titration in the ratio (absorbance intensity at 400 nm) of **1** (40 μM) with S²⁻.

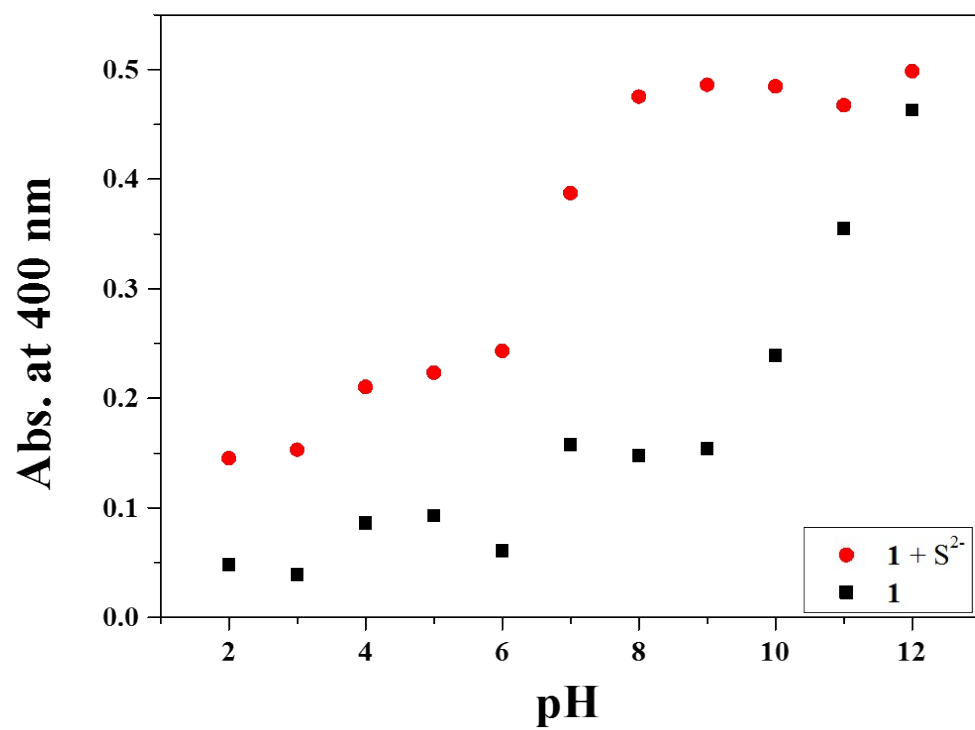
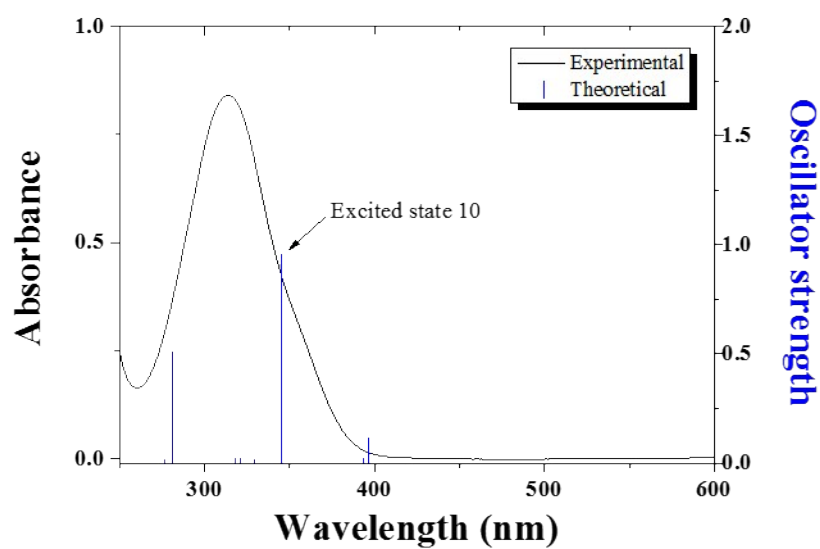


Fig. S5 UV-vis absorbance (400 nm) of **1** and **1-S²⁻** at pH 2-12 in bis-tris buffer at room temperature.

(a)

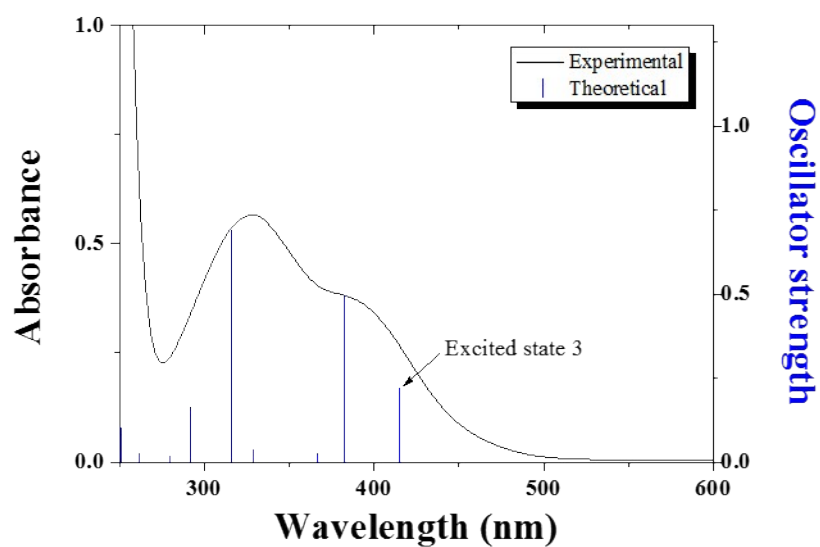


(b)

| Excited state 10 | Wavelength (nm) | Percent (%) | Main character | Oscillator strength |
|------------------|-----------------|-------------|-------------------------|---------------------|
| H → L | 345.6 | 98 | $\pi \rightarrow \pi^*$ | 0.9581 |

Fig. S6 (a) The theoretical excitation energies and the experimental UV-vis spectrum of **1**. (b) The major electronic transition energy and molecular orbital contributions for **1** (H = HOMO and L = LUMO).

(a)



(b)

| Excited state 3 | Wavelength (nm) | Percent (%) | Main character | Oscillator strength |
|-----------------|-----------------|-------------|----------------|---------------------|
| H → L + 1 | 414.88 | 98 | ICT | 0.2214 |

Fig. S7 (a) The theoretical excitation energies and the experimental UV-vis spectrum of **1⁻**. (b) The major electronic transition energy and molecular orbital contributions for **1⁻** (H = HOMO and L = LUMO).