

**Wide blue phase range observed in simple binary mixture
systems containing rodlike racemic biphenyl mesogens with
2-octyloxy tail**

Chiung-Cheng Huang,*^a Chang-Yi Guo,^a Wei-Cheng Hsieh,^a Chih-Yi
Fang,^a Jian-Fu Chiou,^a Mei-Ching Yu,^a Bo-Hao Chen,^b I-Jui Hsu*^b and
Jey-Jau Lee^c

^aDepartment of Chemical Engineering, Tatung University, Taipei 104, Taiwan

E-mail: chchhuang@ttu.edu.tw

^bDepartment of Molecular Science and Engineering, National Taipei University of

Technology, Taipei 106, Taiwan

E-mail: ijuihsu@ntut.edu.tw

Department of Chemical Engineering, Tatung University, Taipei 104, Taiwan.

^cNational Synchrotron Radiation Research Center of Taiwan, Hsinchu 300, Taiwan

Electronic supplementary information (ESI)

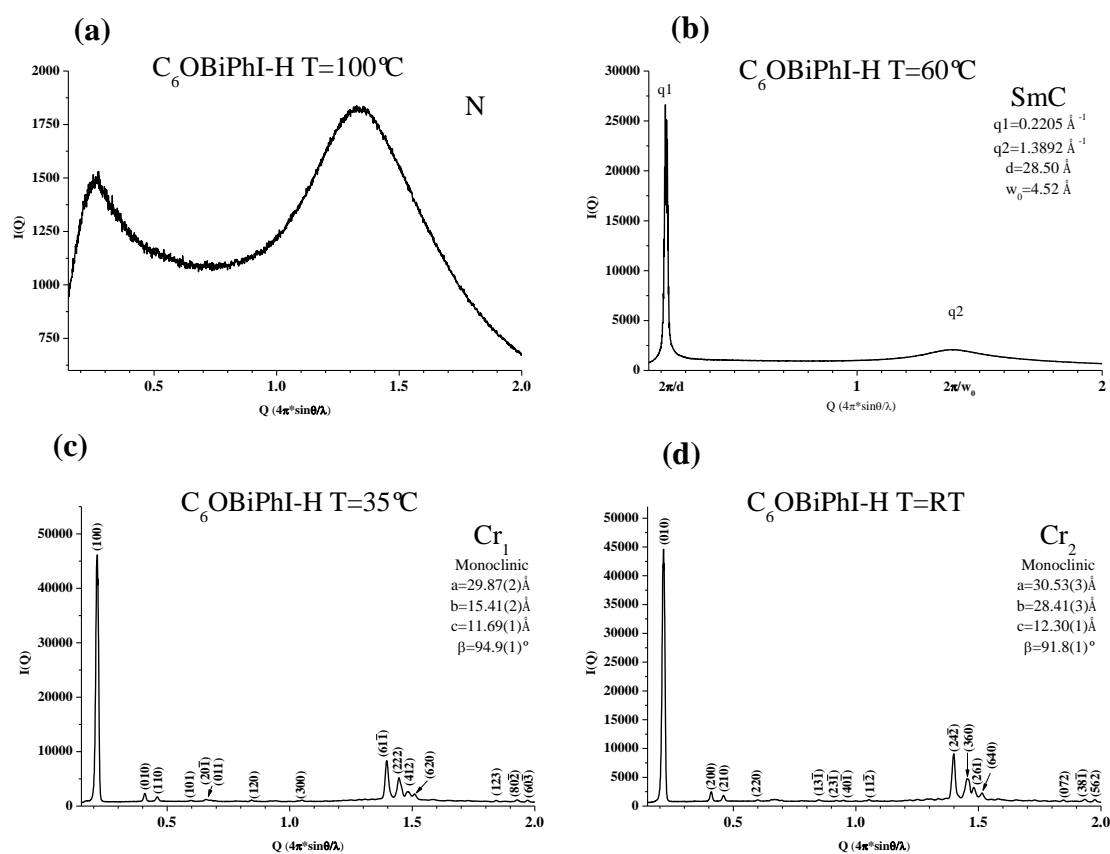


Fig. S1 The variable-temperature XRD measurements of biphenyl mesogen $C_6OBiPhI-H$

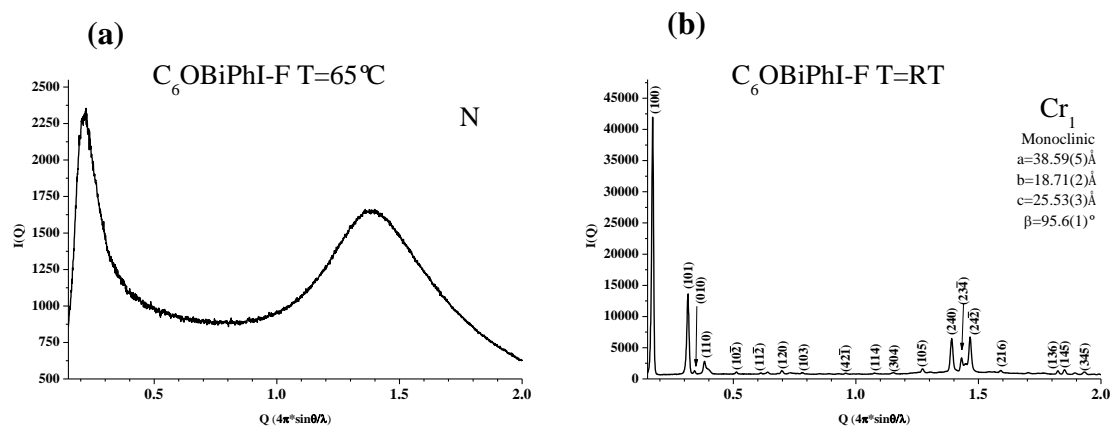


Fig. S2 The variable-temperature XRD measurements of biphenyl mesogen $C_6OBiPhI-F$

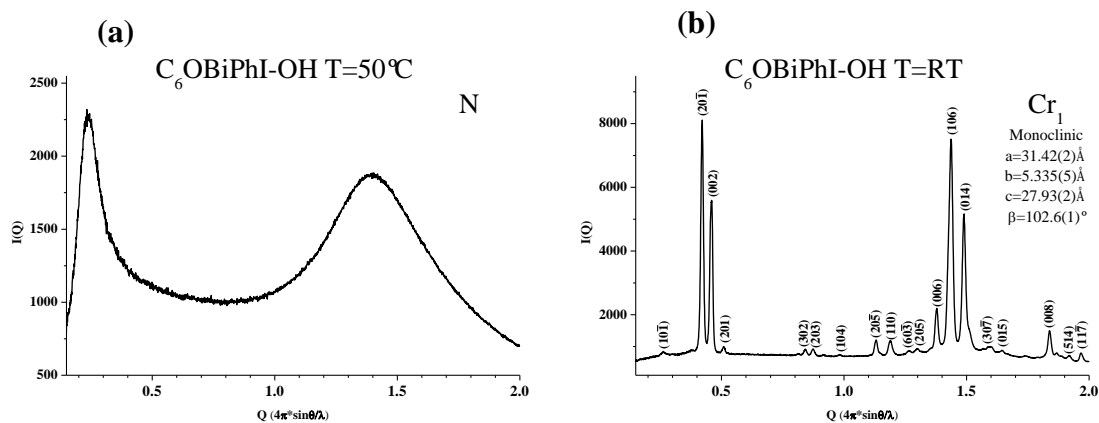


Fig. S3 The variable-temperature XRD measurements of biphenyl mesogen $C_6OBiPhI-OH$

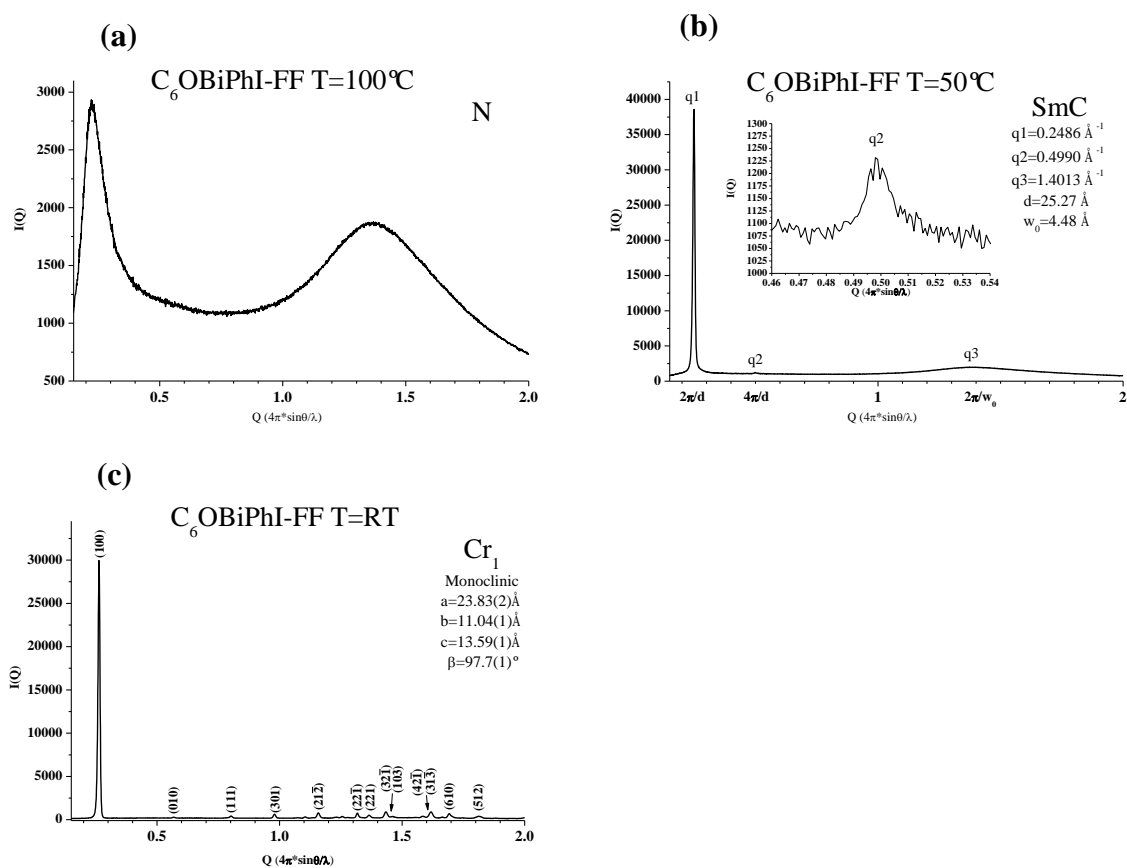


Fig. S4 The variable-temperature XRD measurements of biphenyl mesogen $C_6OBiPhI-FF$

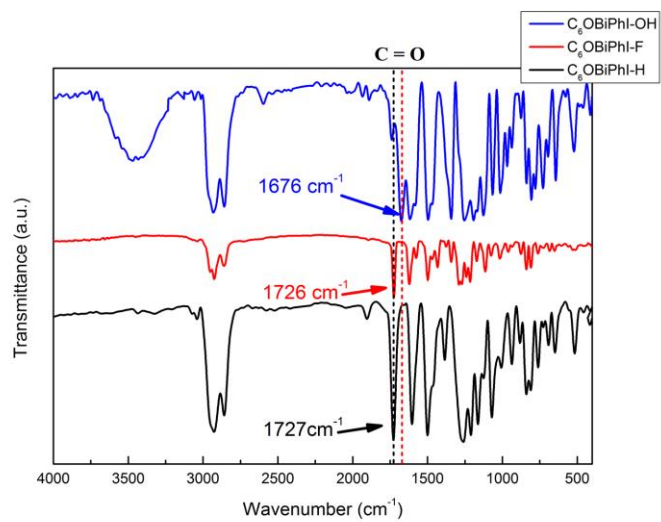


Fig. S5 FT-IR spectra of the biphenyl compounds $C_6OBiPhI-H$, $C_6OBiPhI-F$ and $C_6OBiPhI-OH$.

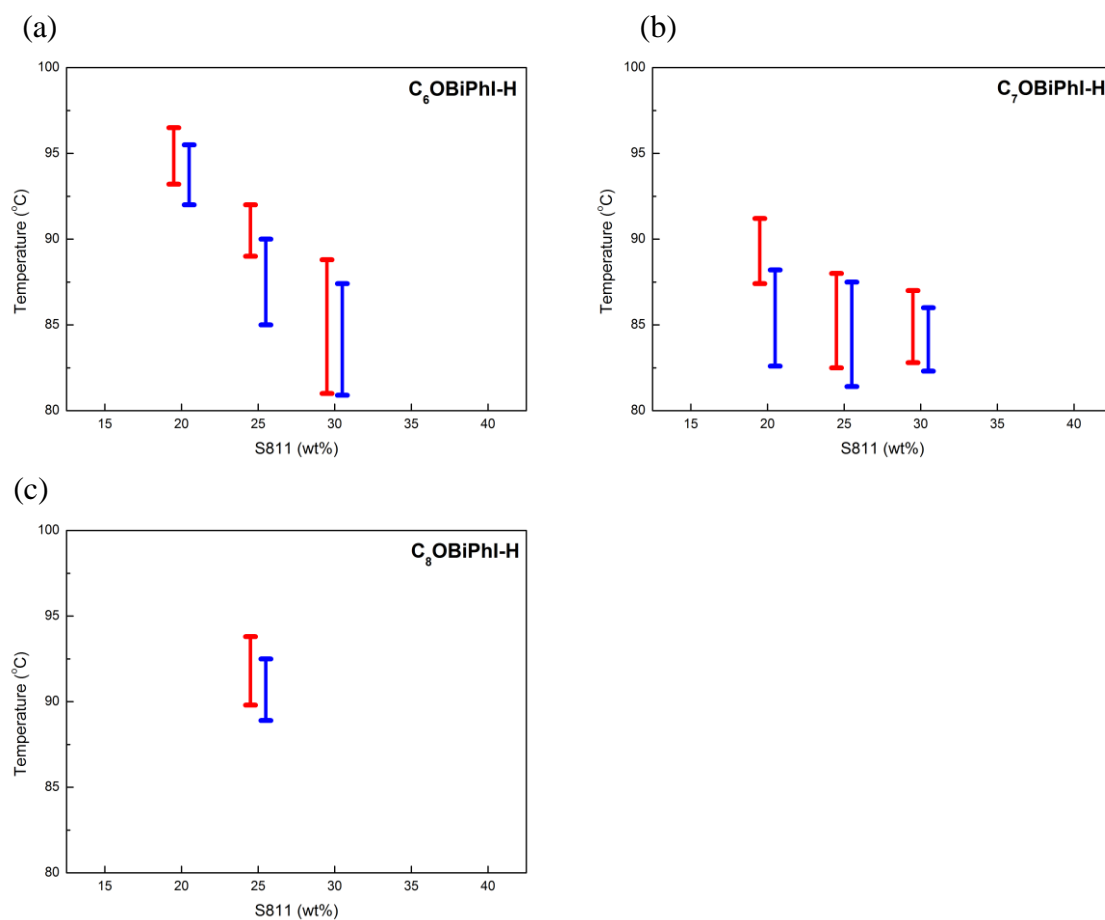


Fig. S6 The comparison of BP temperature range for the blending mixture system composed of **C_nOBiPhI-H** and different amount of chiral dopant **S811** in heating (red line) and cooling processes (blue line).

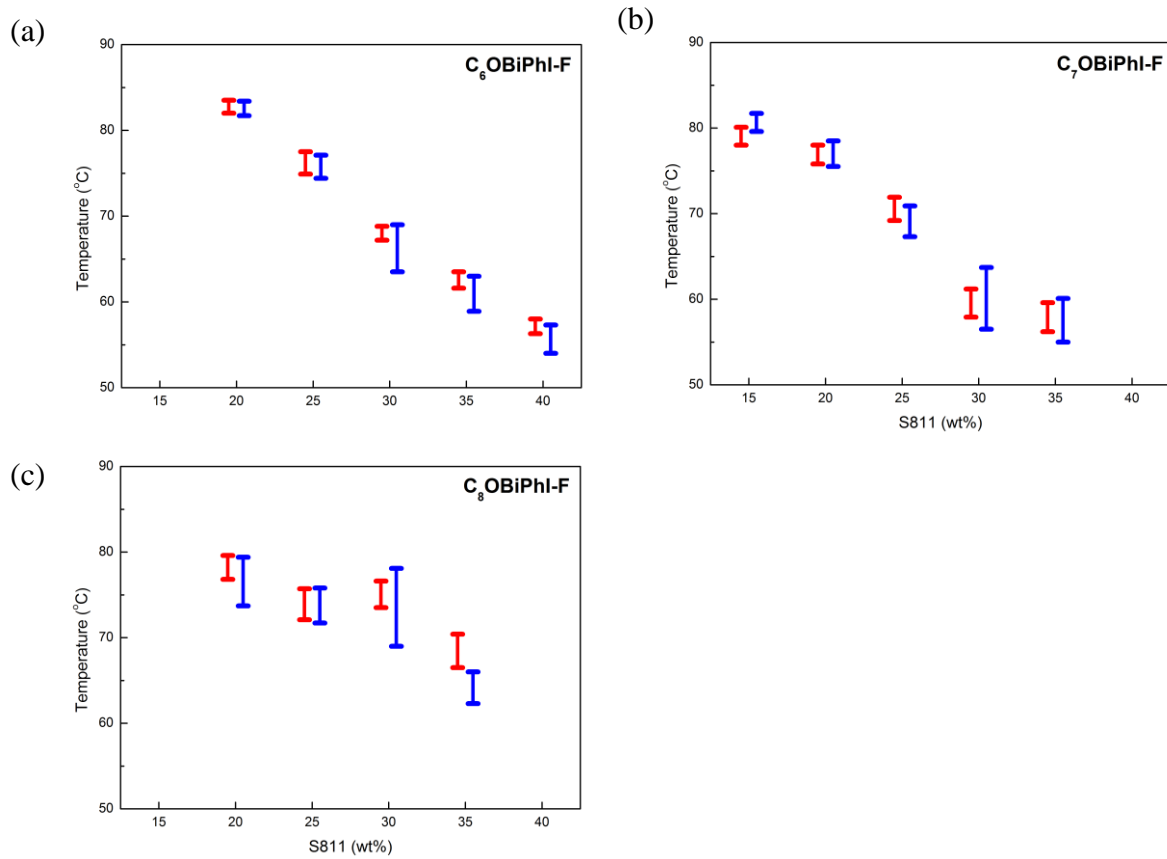


Fig. S7 The comparison of BP temperature range for the blending mixture system composed of $C_nOBiPhI-F$ and different amount of chiral dopant **S811** in heating (red line) and cooling processes (blue line).

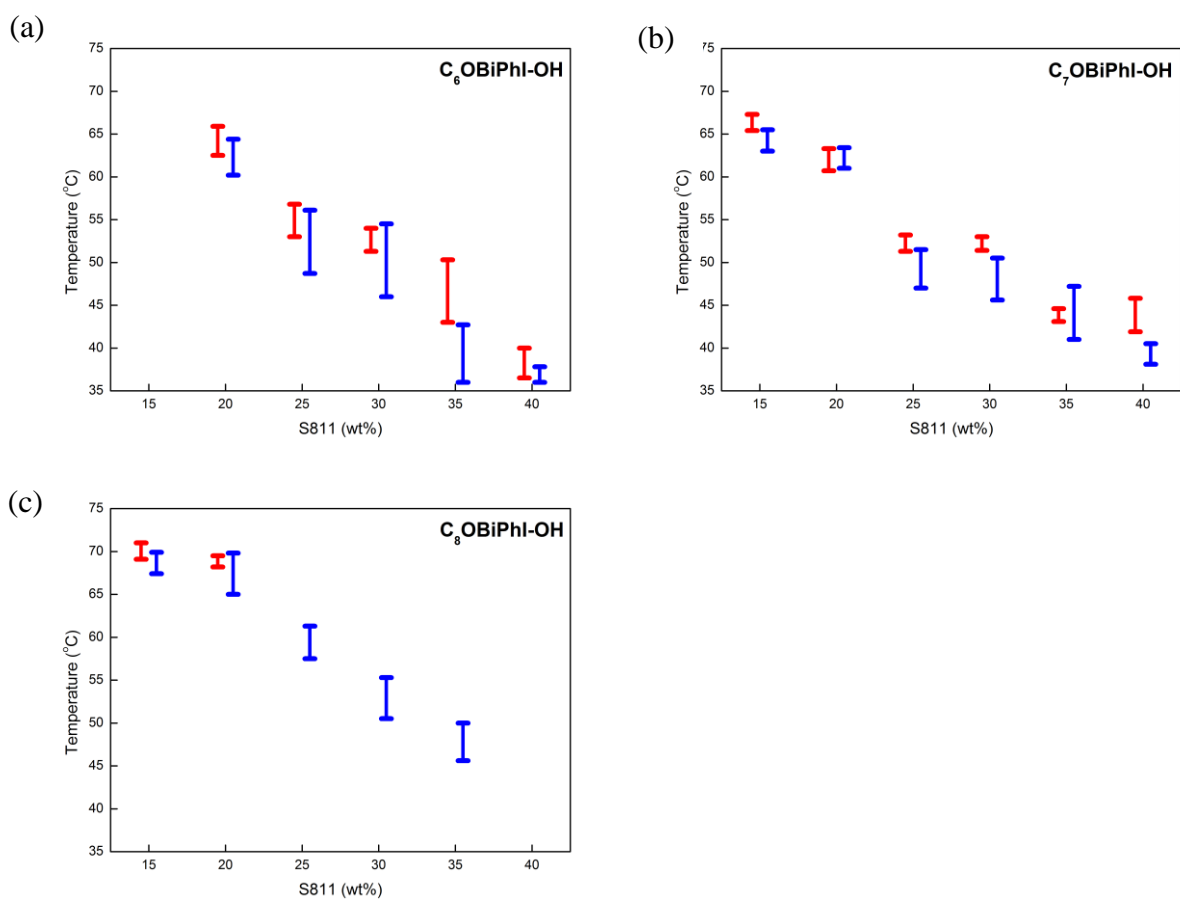


Fig. S8 The comparison of BP temperature range for the blending mixture system composed of C_n OBiPhI-OH and different amount of chiral dopant S811 in heating (red line) and cooling processes (blue line).

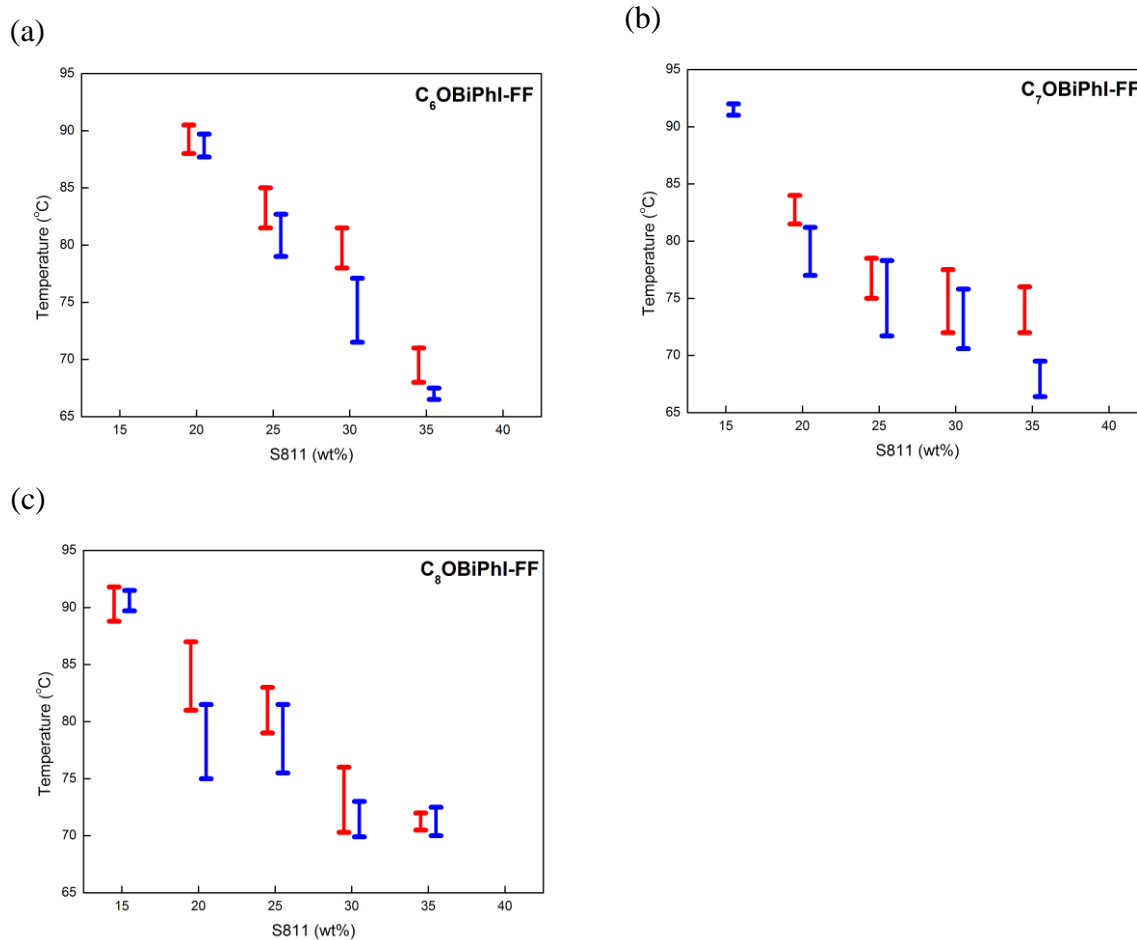


Fig. S9 The comparison of BP temperature range for the blending mixture system composed of $C_n\text{OBiPhI-FF}$ and different amount of chiral dopant **S811** in heating (red line) and cooling processes (blue line).

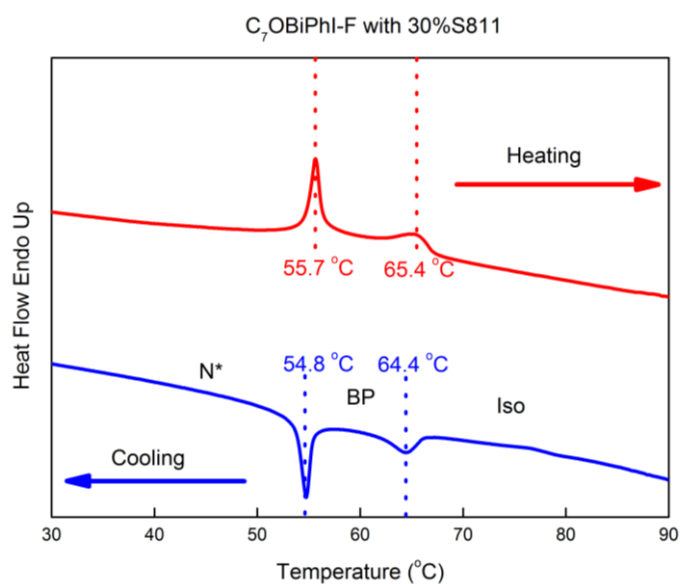


Fig. S10 DSC curves of $C_7\text{OBiPhI-F}$ blended with 30.0 wt% **S811** (with rate of 1 °C/min upon heating and cooling).

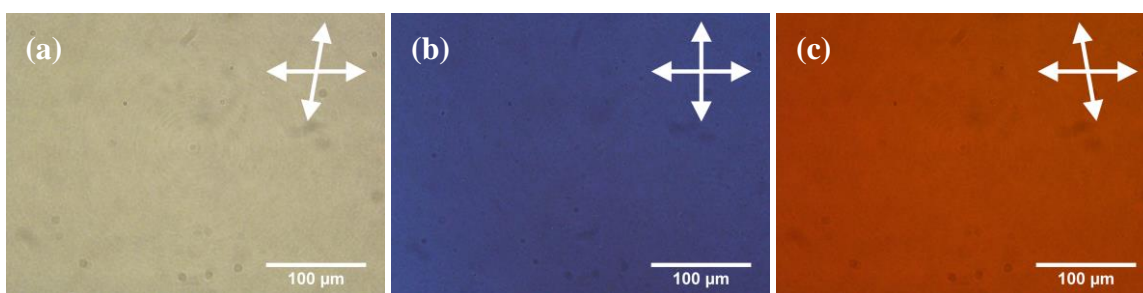


Fig. S11 Optical images of **C₇OBiPhI-OH** blended with 35.0 wt% **S811** were observed by POM at 46.0 °C (a) polarizer was rotated clockwise by a small angle of 10°. (b) Polarizer and analyzer were orthogonal. (c) Polarizer was rotated counterclockwise by a small angle of 10°.

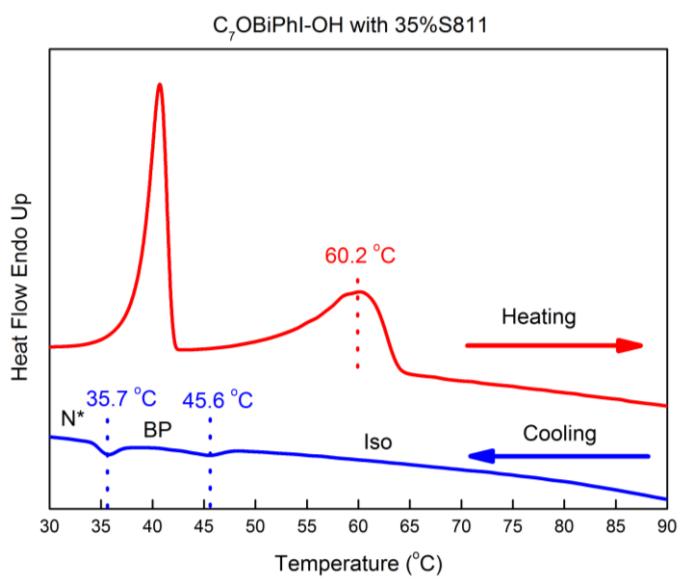


Fig. S12 DSC curves of **C₇OBiPhI-OH** blended with 35.0 wt% **S811** (with rate of 1 °C/min upon heating and cooling).

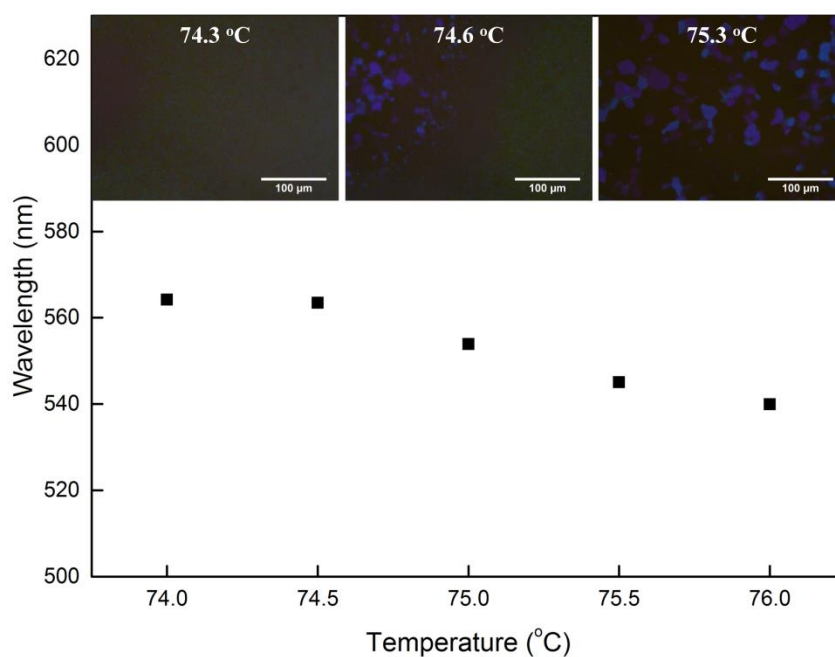


Fig. S13 The temperature dependence of the Bragg reflection wavelength for the blending mixture consisting of **C₈OBiPhI-F** + 30% **S811** during the heating process with a rate of 0.2 °C min⁻¹. The inset shows the POM images of BPs at different temperature.

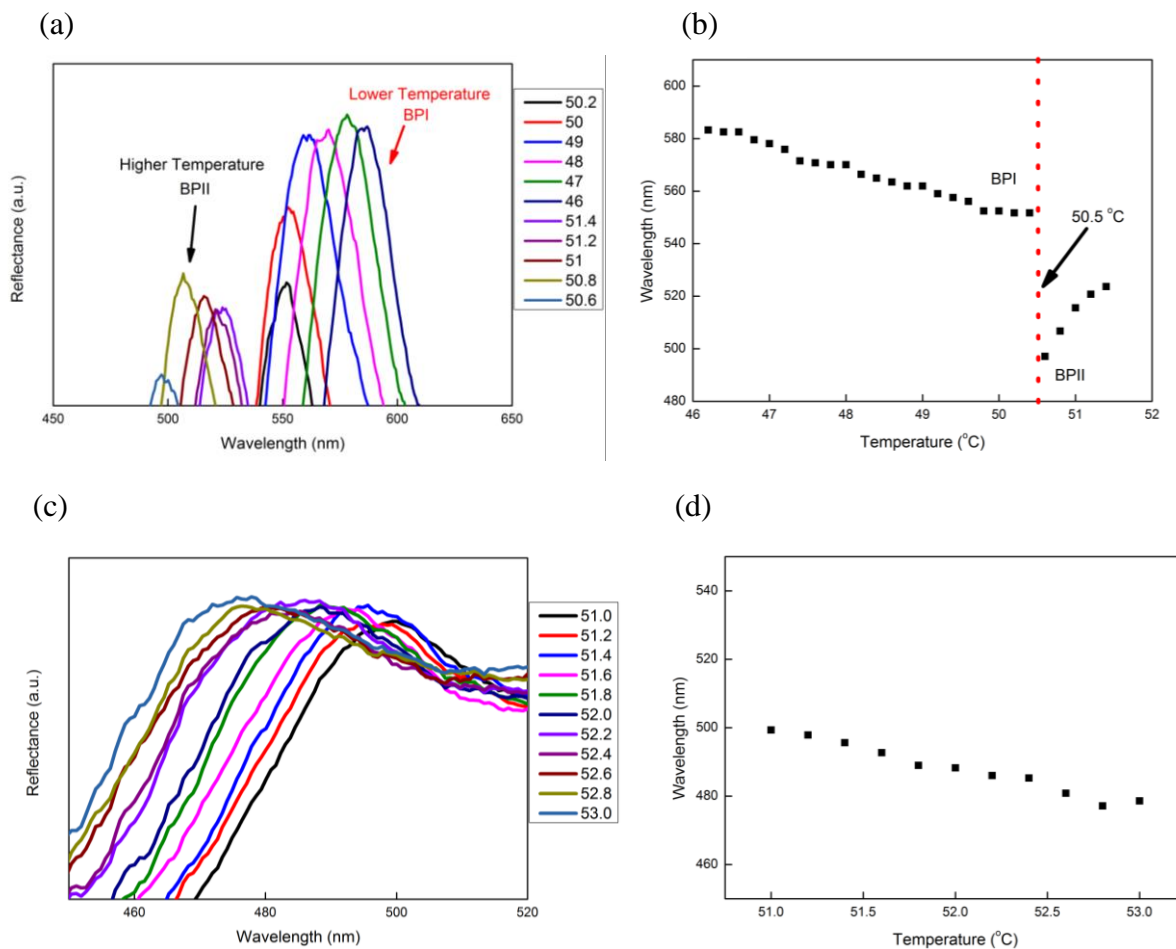


Fig. S14 For the blending mixture consisting of C₇OBiPhI-OH + 25% S811 during the cooling process (a) Temperature dependence of typical reflectance profile, and during the heating process(c); (b) Temperature dependence of the Bragg reflection wavelength during the cooling process and, during the heating process (d) with a rate of 0.2 °C min⁻¹.

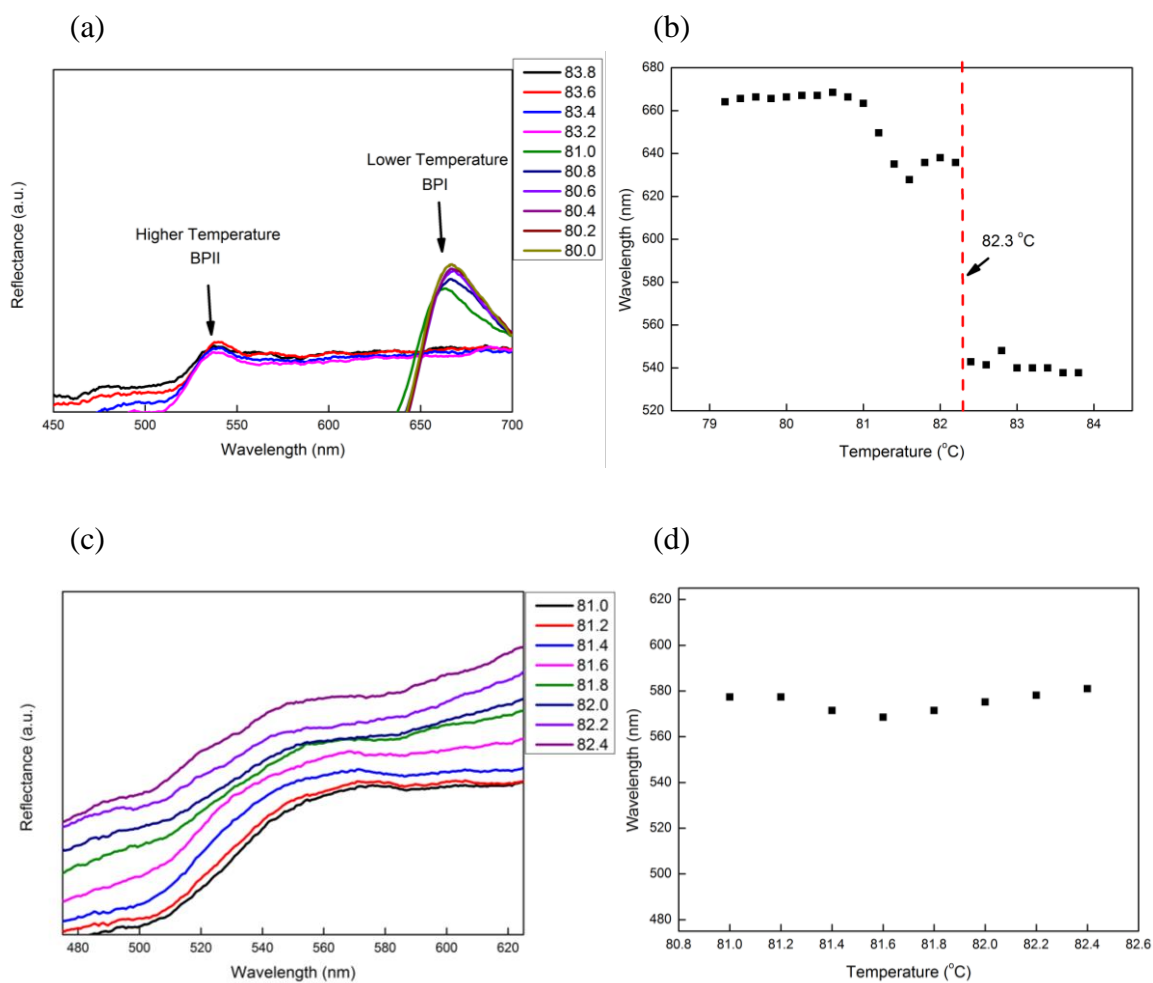


Fig. S15 (a) Temperature dependence of typical reflectance profile during the cooling process and (c) during the heating process; (b) Temperature dependence of the Bragg reflection wavelength during the cooling process and (d) during the heating process for the blending mixture consisting of C₆OBiPhI-FF + 25% S811. (with a rate of 0.2 °C min⁻¹)

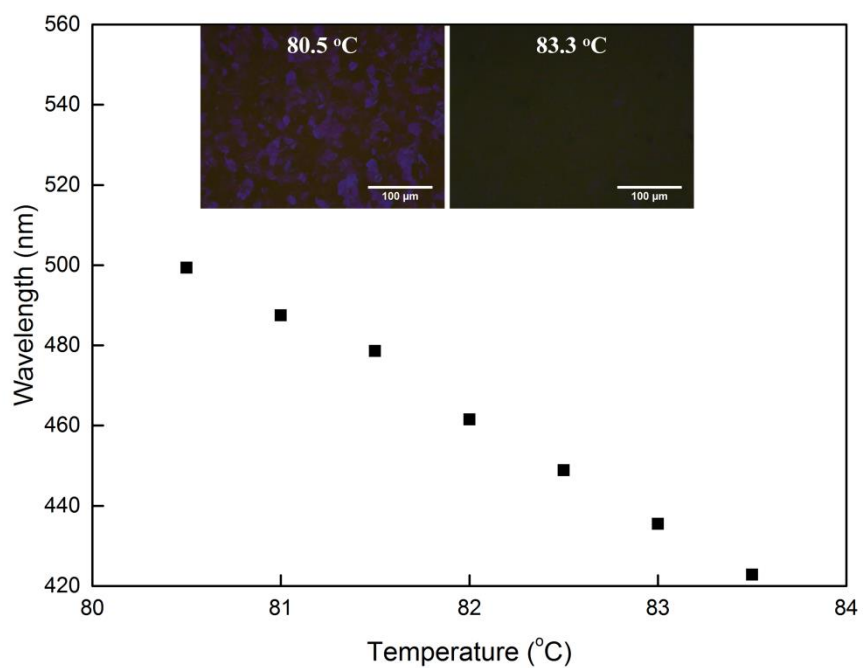


Fig. S16 The temperature dependence of the Bragg reflection wavelength for the blending mixture consisting of **C₈OBiPhI-F** + 10% **ISO(6OBA)₂** during the cooling process with a rate of 0.2 °C min⁻¹. The inset shows the POM images of BPs at different temperature.

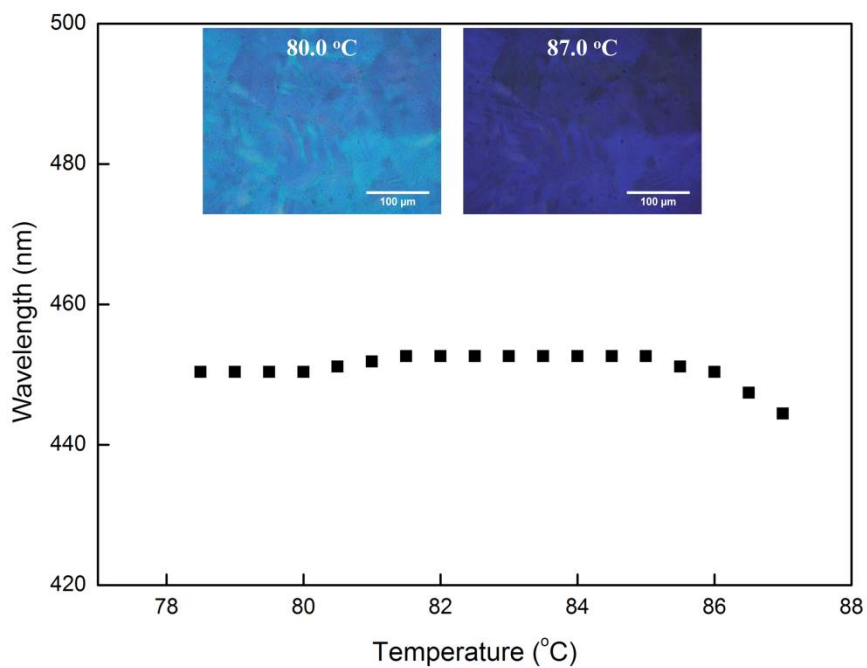


Fig. S17 The temperature dependence of the Bragg reflection wavelength for the blending mixture consisting of $C_8OBiPhI-FF$ + 10% $ISO(6OBA)_2$ during the cooling process with a rate of $0.2 \text{ } ^\circ\text{C min}^{-1}$. The inset shows the POM images of BPs at different temperature.

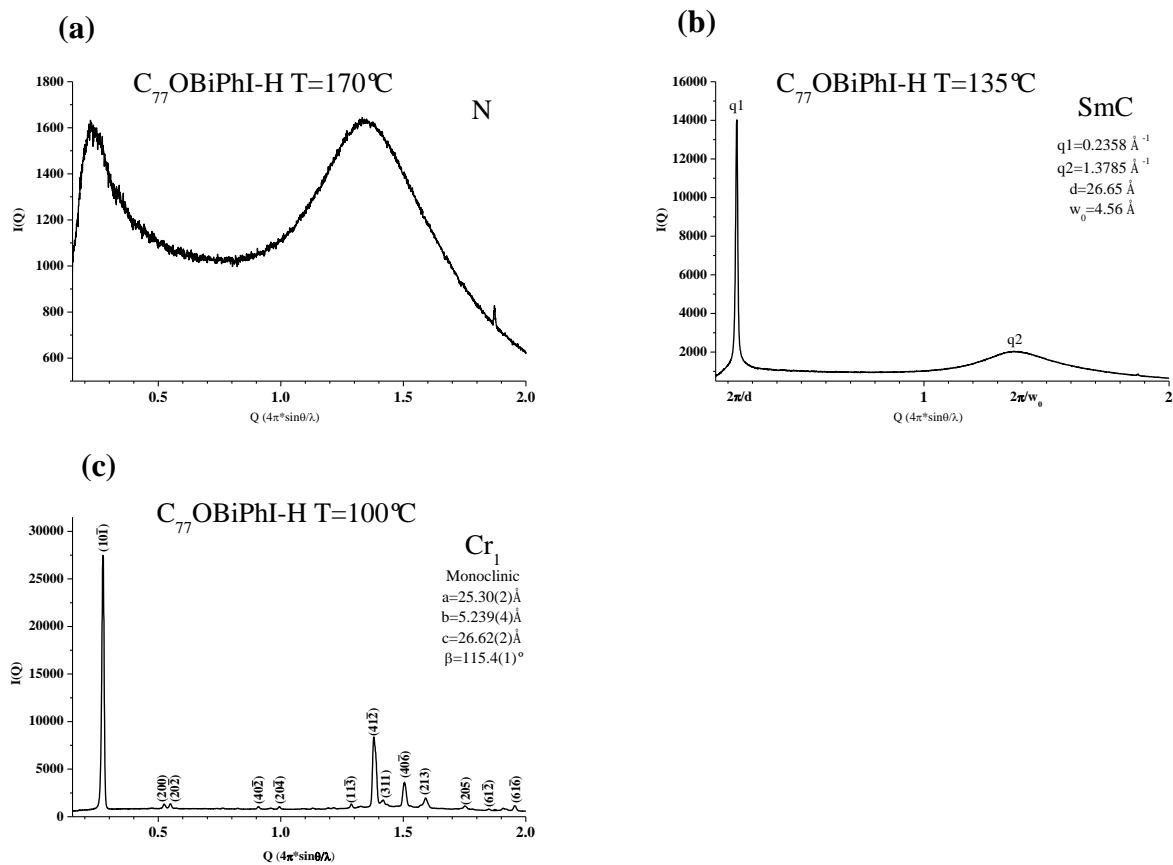
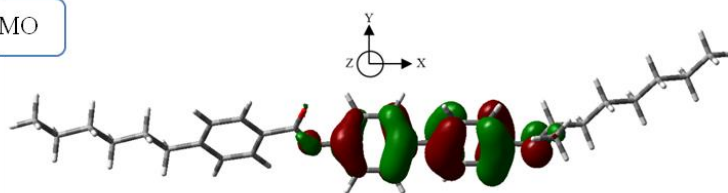


Fig. S18 The variable-temperature XRD measurements of biphenyl mesogen $C_6OBiPhI-FF$

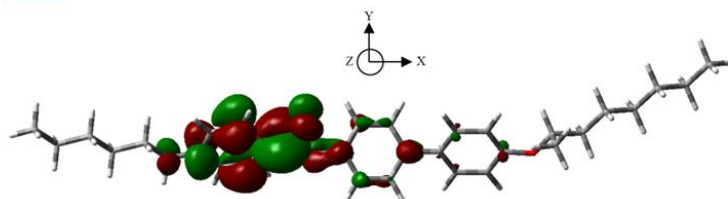
(a)

HOMO



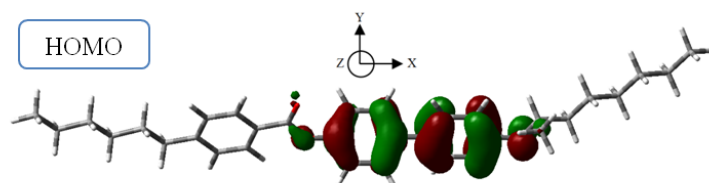
C₆OBiPhI-H

LUMO



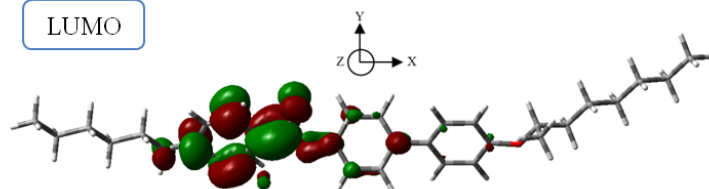
(b)

HOMO



C₆OBiPhI-F

LUMO



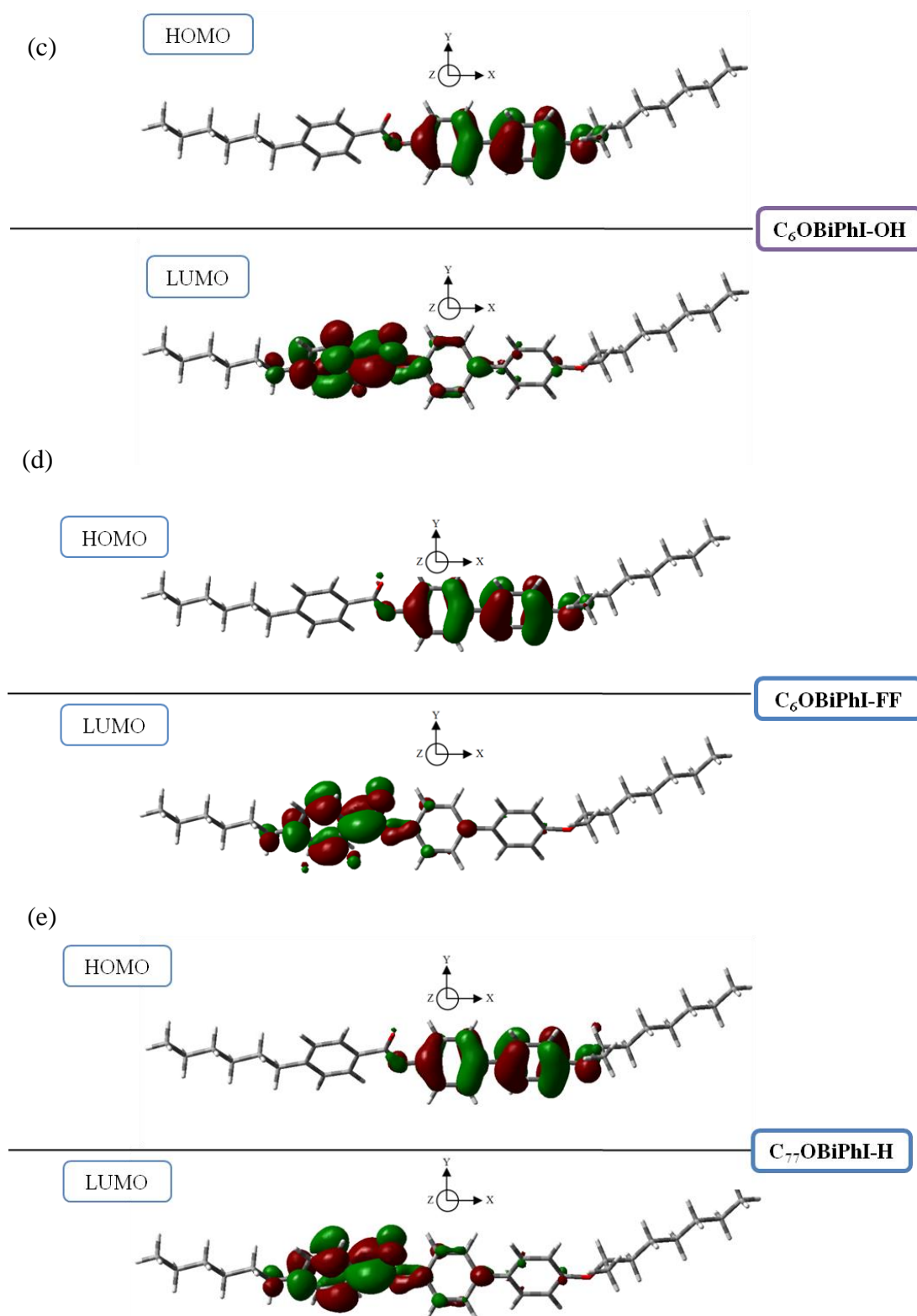


Fig. S19 The HOMO and LUMO of compound (a) C₆OBiPhI-H (b) C₆OBiPhI-F (c) C₆OBiPhI-OH (d) C₆OBiPhI-FF (e) C₇₇OBiPhI-H. The simulation exchange functional and basis set are CAM-B3LYP and 6-311G(d, p), respectively. The isosurface is drawn at value of 0.02

Table S1. DFT calculated HOMO, LUMO, energy gap, dipole moment components, μ_x , μ_y , μ_z and modulus (μ) for the biphenyl compounds.

Compound	Energy (eV)			Dipole moment (m in Debye)			
	HOMO	LUMO	ΔE (eV)	μ_x	μ_y	μ_z	μ_{total}
C₆OBiPhI-H	-7.0074	0.0123	7.0196	-1.4837	0.5665	0.3023	1.6166
C₆OBiPhI-F	-6.9789	-0.1483	6.8306	-2.3668	1.1911	0.9564	2.8170
C₆OBiPhI-OH	-7.1621	-0.0172	7.1449	0.5034	1.2278	1.3010	1.8584
C₆OBiPhI-FF	-7.0240	-0.2903	6.7337	-1.7324	2.2723	1.6355	3.2923
C₇₇OBiPhI-H	-7.0492	0.0023	7.0515	-1.5766	0.4402	0.2524	1.6562