

Electronic Supplementary Information (ESI)

A new single-component $\text{KCaY}(\text{PO}_4)_2:\text{Dy}^{3+}, \text{Eu}^{3+}$ nanosized phosphor with high color-rendering index and excellent thermal resistance for warm-white NUV-LED

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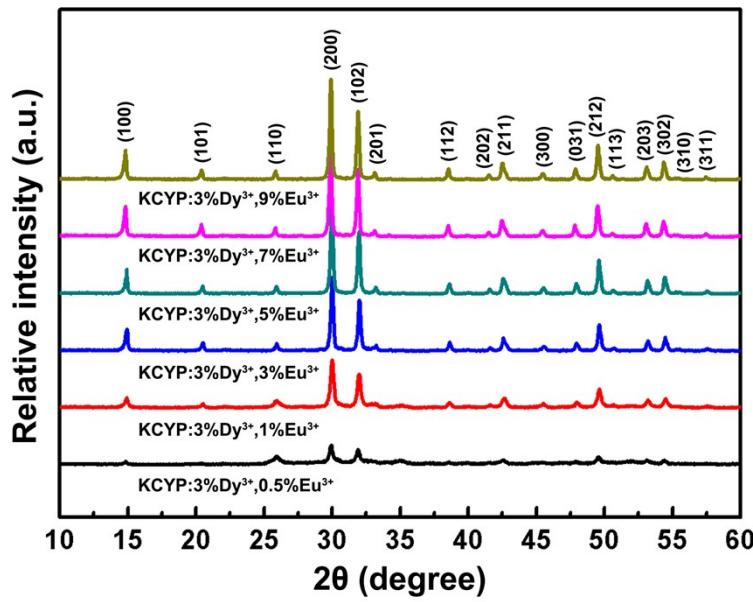


Fig. S1 Powder XRD patterns of the as-synthesized KCYP:3%Dy³⁺, $n\%$ Eu³⁺ specimens with various Eu³⁺ doping contents ($n = 0.5, 1, 3, 5, 7$, and 9).

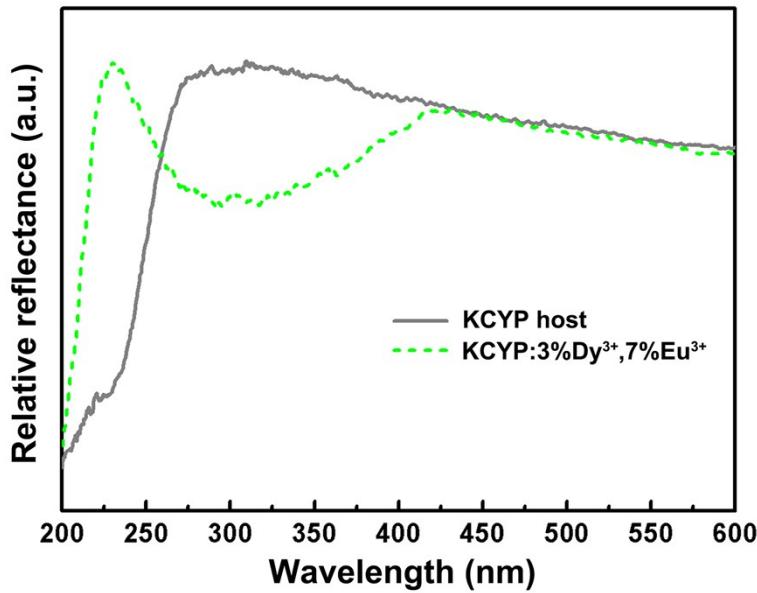


Fig. S2 Diffuse reflection spectra of the KCYP host (gray line) and the KCYP:3%Dy³⁺, 7%Eu³⁺ sample (green dash line).

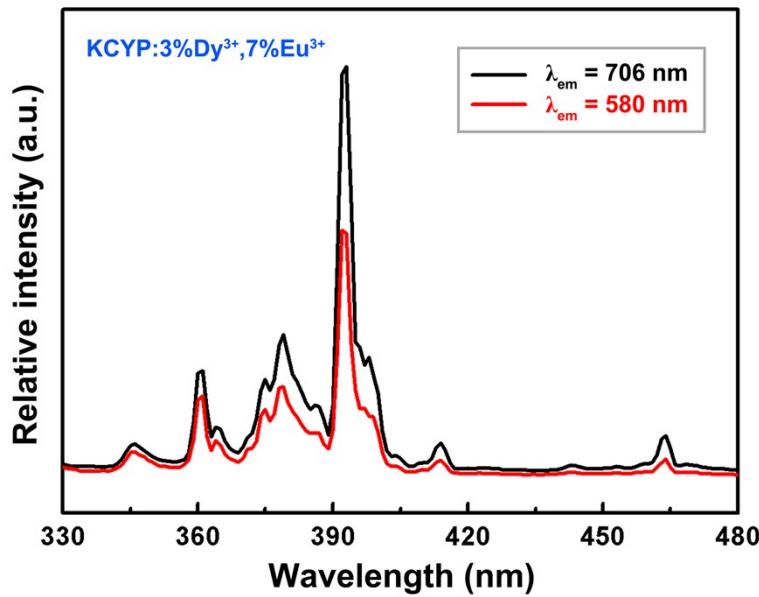


Fig. S3 Comparison of PLE spectra of the KCYP:3%Dy³⁺, 7%Eu³⁺ sample upon different monitoring wavelength.

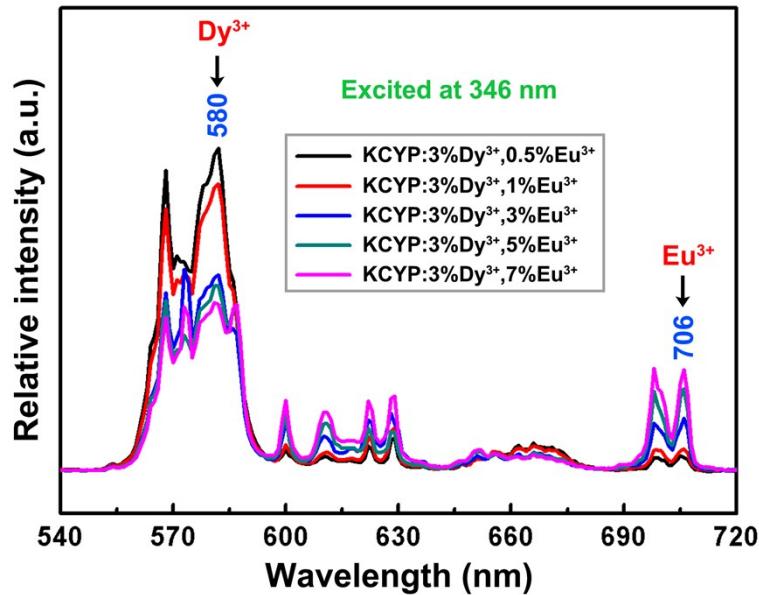


Fig. S4 PL spectra ($\lambda_{\text{ex}} = 346 \text{ nm}$) of the KCYP:3%Dy³⁺, n%Eu³⁺ nanophosphor with changed n values ($n = 0.5, 1, 3, 5$, and 7).

Table S1. Time-resolved fluorescence dynamics of the KCYP:Dy³⁺, Eu³⁺ nanosized phosphor.

Samples	τ_1 (μ s)	τ_2 (μ s)	τ_{ave} (μ s)
KCYP:3%Dy ³⁺	121 (13%)	796 (87%)	708
KCYP:3%Dy ³⁺ , 0.5%Eu ³⁺	114 (15%)	740 (85%)	646
KCYP:3%Dy ³⁺ , 1%Eu ³⁺	108 (18%)	688 (82%)	584
KCYP:3%Dy ³⁺ , 3%Eu ³⁺	100 (22%)	668 (78%)	543
KCYP:3%Dy ³⁺ , 5%Eu ³⁺	93 (25%)	650 (75%)	511
KCYP:3%Dy ³⁺ , 7%Eu ³⁺	82 (28%)	642 (72%)	485

Table S2. CIE 1931 chromaticity coordinates and correlated color temperature (CCT) of the KCYP:Dy³⁺, Eu³⁺ nanoscale phosphor.

Sample composition	CIE coordinates (x, y)	CCT (K)
KCYP:3%Dy ³⁺ , 0.5%Eu ³⁺	(0.354, 0.377)	4781
KCYP:3%Dy ³⁺ , 1%Eu ³⁺	(0.373, 0.378)	4199
KCYP:3%Dy ³⁺ , 3%Eu ³⁺	(0.403, 0.372)	3338
KCYP:3%Dy ³⁺ , 7%Eu ³⁺	(0.427, 0.369)	2766
KCYP:3%Dy ³⁺ , 9%Eu ³⁺	(0.464, 0.366)	2131

Table S3. Full set of eight CRIs and the average Ra value of the KCYP:3%Dy³⁺, 7%Eu³⁺ sample.

CRI	R1	R2	R3	R4	R5	R6	R7	R8	Ra
Sample	90	89	67	81	96	84	75	71	81.6

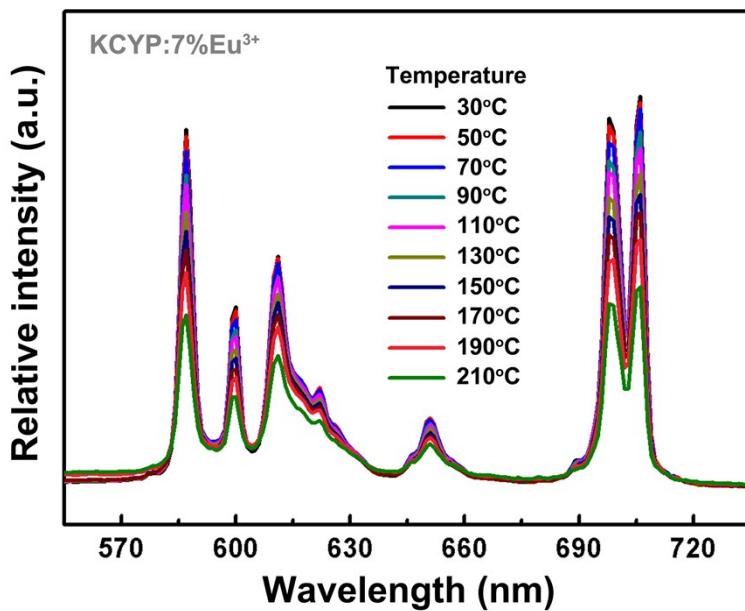


Fig. S5 Emission spectra ($\lambda_{\text{ex}} = 391$ nm) of the KCYP:7%Eu³⁺ sample at temperature between 30 and 210 °C.

Table S4. Comparison of optical properties and temperature characteristic of some reported $\text{MCaLn}(\text{PO}_4)_2$ ($\text{M}^+ = \text{Na, K}$; $\text{Ln}^{3+} = \text{Y, Gd}$) based phosphors with excellent performance.

Phosphors	Preparation method	Emission color	^a CCT (K)	^b CRI (Ra)	^c I_{150} (%)	^d E_a (eV)	References
NCYP:Eu	Solid-state reaction (1250 °C)	Blue	—	—	65	0.201	1
KCGP:Eu	Solid-state reaction (1350 °C)	Blue	—	—	68	0.038	2
KCYP:Eu,Mn	Solid-state reaction (1250 °C)	White	> 4500	< 70	80	0.025	3
KCYP:Dy,Eu	Solution-combustion (400 °C)	Warm-white	2766	81.6	72	0.174	This work

NCYP:Eu represents NaCaY(PO₄)₂:1%Eu²⁺.

KCGP:Eu represents KCaGd(PO₄)₂:1%Eu²⁺.

KCYP:Eu,Mn represents KCaY(PO₄)₂:1%Eu²⁺, 5%Mn²⁺.

KCYP:Dy,Eu represents KCaY(PO₄)₂:3%Dy³⁺, 7%Eu³⁺.

^aCCT: Correlated color temperature.

^bCRI: Color-rendering index.

^c I_{150} : The relative PL intensity when the temperature is increased up to 150 °C compared to the initial intensity at room temperature and the values may change a little because of the difference in the measurement method and the activator concentration.

^d E_a : Activation energy.

References

- 1 W. R. Liu and P. C. Lin, *Opt. Express.* 2014, **22**, A446–A451.
- 2 W. R. Liu, C. H. Huang, C. W. Yeh, Y. C. Chiu, Y. T. Yeh and R. S. Liu, *RSC Adv.* 2013, **3**, 9023–9028.
- 3 W. R. Liu, C. H. Huang, C. W. Yeh, J. C. Tsai, Y. C. Chiu, Y. T. Yeh and R. S. Liu, *Inorg. Chem.*, 2012, **51**, 9636–9641.